Section Summary

21.1 Planck and Quantum Nature of Light

- A blackbody will radiate energy across all wavelengths of the electromagnetic spectrum.
- Radiation of a blackbody will peak at a particular wavelength, dependent on the temperature of the blackbody.
- Analysis of blackbody radiation led to the field of quantum mechanics, which states that radiated energy can only exist in discrete quantum states.

21.2 Einstein and the Photoelectric Effect

- The photoelectric effect is the process in which EM radiation ejects electrons from a material.
- Einstein proposed photons to be quanta of EM radiation having energy $E = hf$, where f is the frequency of the radiation.
- All EM radiation is composed of photons. As Einstein explained, all characteristics of the photoelectric effect are due to the interaction of individual photons with individual electrons.
- The maximum kinetic energy KE_e of ejected electrons (photoelectrons) is given by $KE_e = hf - BE$, where hf is the photon energy and BE is the binding energy (or work function) of the electron in the particular material.

21.3 The Dual Nature of Light

- Compton scattering provided evidence that photon-electron interactions abide by the principles of conservation of momentum and conservation of energy.
- The momentum of individual photons, quantified by $\mathbf{p} = \frac{h}{\lambda}$, can be used to explain observations of comets and may lead to future space technologies.
- Electromagnetic waves and matter have both wave-like and particle-like properties. This phenomenon is defined as particle-wave duality.