Problems

21.1 Planck and Quantum Nature of Light 22.

How many X-ray photons per second are created by an X-ray tube that produces a flux of X-rays having a power of 1.00 W? Assume the average energy per photon is 75.0 keV.

- a. 8.33×10^{15} photons
- b. 9.1×10^7 photons
- c. 9.1×10^8 photons
- d. $8.33 \times 10^{\overline{13}}$ photons

23.

What is the frequency of a photon produced in a CRT using a 25.0-kV accelerating potential? This is similar to the layout as in older color television sets.

- a. $6.04 \times 10^{-48} \text{ Hz}$
- b. $2.77 \times 10^{-48} \text{ Hz}$
- c. $3.02\times10^{18}~\mathrm{Hz}$
- d. $6.04 \times 10^{18} \text{ Hz}$

21.2 Einstein and the Photoelectric Effect 24.

What is the binding energy in eV of electrons in magnesium, if the longest-wavelength photon that can eject electrons is 337 nm?

- a. $7.44 \times 10^{-19} \text{ J}$
- b. $7.44 \times 10^{-49} \text{ J}$
- c. $5.90 \times 10^{-17} \text{ J}$
- d. $5.90 \times 10^{-19} \text{ J}$

25.

Photoelectrons from a material with a binding energy of 2.71 eV are ejected by 420-nm photons. Once ejected, how long does it take these electrons to travel 2.50 cm to a detection device?

- a. $8.5 \times 10^{-6} \text{ s}$
- b. $3.5 \times 10^{-7} \text{ s}$
- c. $43.5 \times 10^{-9} \text{ s}$
- d. $8.5 \times 10^{-8} \text{ s}$

21.3 The Dual Nature of Light 26.

What is the momentum of a 0.0100-nm-wavelength photon that could detect details of an atom?

- a. $6.626\times 10^{-27}~\mathrm{kg}~\mathrm{m/s}$
- b. $6.626 \times 10^{-32} \text{ kg} \text{ m/s}$
- c. $6.626 \times 10^{-34} \text{ kg} \text{ m/s}$

d.
$$6.626 \times 10^{-23} \text{ kg} \text{ m/s}$$

27.

The momentum of light is exactly reversed when reflected straight back from a mirror, assuming negligible recoil of the mirror. Thus the change in momentum is twice the initial photon momentum. Suppose light of intensity 1.00 kW/m2 reflectsfrom a mirror of area 2.00 m2 each second. Using the most general form of Newton's second law, what is the force on the mirror?

- a. $1.33\times10^{\text{-}5}\ \text{N}$
- b. 1.33×10^{-6} N c. 1.33×10^{-6} N d. 1.33×10^{-7} N