Multiple Choice

21.1 Planck and Quantum Nature of Light 29.

A perfect blackbody is a perfect absorber of energy transferred by what method?

- a. conduction
- b. convection
- c. induction
- d. radiation

30.

Which of the following is a physical entity that is quantized?

- a. electric charge of an ion
- b. frequency of a sound
- c. speed of a car

31.

Find the energy in joules of photons of radio waves that leave an FM station that has a 90.0-MHz broadcast frequency.

- a. $1.8 \times 10^{-25} \; \mathrm{J}$
- b. $1.11 \times 10^{-25} \text{ J}$
- c. $7.1 \times 10^{-43} \text{ J}$
- d. $5.96 \times 10^{-26} \text{ J}$

32.

Which region of the electromagnetic spectrum will provide photons of the least energy?

- a. infrared light
- b. radio waves
- c. ultraviolet light
- d. X-rays

33.

A hot, black coffee mug is sitting on a kitchen table in a dark room. Because it cannot be seen, one assumes that it is not emitting energy in the form of light. Explain the fallacy in this logic.

- a. Not all heat is in the form of light energy.
- b. Not all light energy falls in the visible portion of the electromagnetic spectrum.
- c. All heat is in the form of light energy.
- d. All light energy falls in the visible portion of the electromagnetic spectrum.

34.

Given two stars of equivalent size, which will have a greater temperature: a red dwarf or a yellow dwarf? Explain. Note—Our sun is considered a yellow dwarf.

- a. a yellow dwarf, because yellow light has lower frequency
- b. a red dwarf, because red light has lower frequency
- c. a red dwarf, because red light has higher frequency
- d. a yellow dwarf, because yellow light has higher frequency

21.2 Einstein and the Photoelectric Effect 35.

What is a quantum of light called?

- a. electron
- b. neutron
- c. photon
- d. proton

36.

Classical physics explains many aspects of electromagnetic radiation. Which of the following observations is consistent with the classical model?

- a. Immediate ejection of electrons upon exposure of a material to light.
- b. Exposure to light causes an electrical response in many materials.
- c. The speed of ejected electrons is independent of light intensity.
- d. Electrical activity from a given material requires a specific frequency of light.

37.

If $5\$ of energy is supplied to an electron with a binding energy of $2.3\$, with what kinetic energy will the electron be launched?

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a. 2.3\,\text{eV}
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- b. $7.3\, \text{text}\{\text{eV}\}\$
- c. 11.5\\\text{eV}
- d. $\text{text}\{2.7\}\,\text{text}\{\text{eV}\}$

38.

Which of the following terms translates to light-producing voltage?

- a. photoelectric
- b. quantum mechanics
- c. photoconductive
- d. photovoltaic

39.

Why is high frequency EM radiation considered more dangerous than long wavelength EM radiation?

- a. Long wavelength EM radiation photons carry less energy and therefore have greater ability to disrupt materials through the photoelectric effect.
- b. Long wavelength EM radiation photons carry more energy and therefore have greater ability to disrupt materials through the photoelectric effect.
- c. High frequency EM radiation photons carry less energy and therefore have lower ability to disrupt materials through the photoelectric effect.
- d. High frequency EM radiation photons carry more energy and therefore have greater ability to disrupt materials through the photoelectric effect.

40.

Why are UV, X-rays, and gamma rays considered ionizing radiation?

- a. UV, X-rays, and gamma rays are capable of ejecting photons from a surface.
- b. UV, X-rays, and gamma rays are capable of ejecting neutrons from a surface.
- c. UV, X-rays, and gamma rays are capable of ejecting protons from a surface.
- d. UV, X-rays, and gamma rays are capable of ejecting electrons from a surface.

21.3 The Dual Nature of Light 41.

What two particles interact in Compton scattering?

- a. photon and electron
- b. proton and electron
- c. neutron and electron
- d. proton and neutron

42.

What is the momentum of a 500-nm photon?

- a. $8.35 \times 10^{-26} \text{ kg} \text{ m/s}$
- b. $3.31 \times 10^{-40} \text{ kg} \text{ m/s}$
- c. $7.55 \times 10^{26} \text{ kg} \text{ m/s}$
- d. $1.33 \times 10^{-27} \text{ kg} \text{ m/s}$

43.

The conservation of what fundamental physics principle is behind the technology of solar sails?

- a. charge
- b. mass
- c. momentum
- d. angular momentum

44.

Terms like frequency, amplitude, and period are tied to what component of wave-particle duality?

- a. neither the particle nor the wave model of light
- b. both the particle and wave models of light
- c. the particle model of light
- d. the wave model of light

45.

Why was it beneficial for Compton to scatter electrons using X-rays and not another region of light like microwaves?

- a. because X-rays are more penetrating than microwaves
- b. because X-rays have lower frequency than microwaves
- c. because microwaves have shorter wavelengths than X-rays
- d. because X-rays have shorter wavelength than microwaves