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South East
Technological
University

Physics 2 Reports

SETU - South East Technological University

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Contents

1 Lab 1 - Specific Heat Capacity by Electrical Heating, 28/01/2025	3
1.1 Theory	3
1.1.1 Aim(s)	3
1.1.2 Procedure	3
1.2 Experiment	4
1.3 Questions	5
1.4 Conclusion	5
2 Exp 3 - Calibration of a Thermocouple, 11/02/2025	6
2.1 Theory	6
2.1.1 Aim(s)	6
2.1.2 Procedure	6
2.2 Diagrams/Apparaths	6
2.3 Questions	8

1 Lab 1 - Specific Heat Capacity by Electrical Heating, 28/01/2025

1.1 Theory

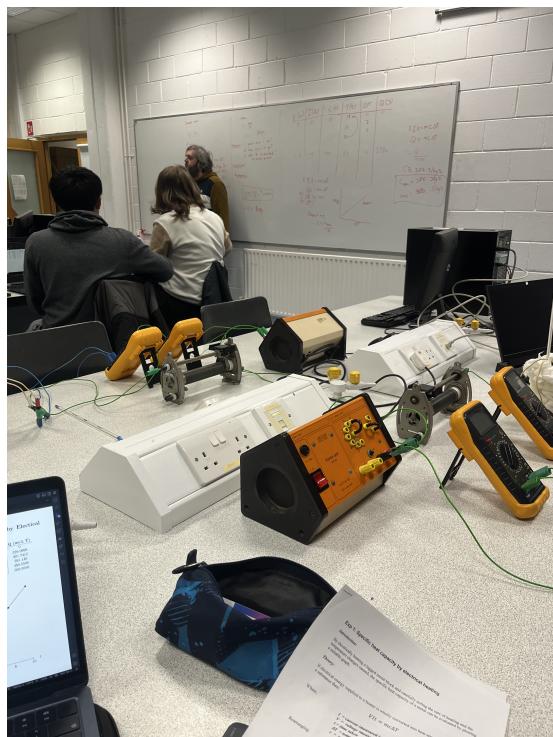
By electrically heating a lagged metal block and carefully noting the rate of heating and the temperature changes caused, the specific heat capacity of a metal can be evaluated by plotting a suitable graph.

1.1.1 Aim(s)

The aim of this experiment is to find the specific heat of the material we are heating electrically.

1.1.2 Procedure

As in Lab manual.



1.2 Experiment

V (volt)	I (Ampere)	t (s)	Temp (°C)	ΔT	Q (mcΔ T)
0	0	0	17	0	0
8.34	2.7	120	20.5	3.5	2702.16
8.34	2.7	240	27	10	5404.32
8.34	2.7	360	35	18	8106
8.4	2.7	480	43	26	10808
8.3	2.7	600	51	34	13510.8

Figure 1: Table of measurements

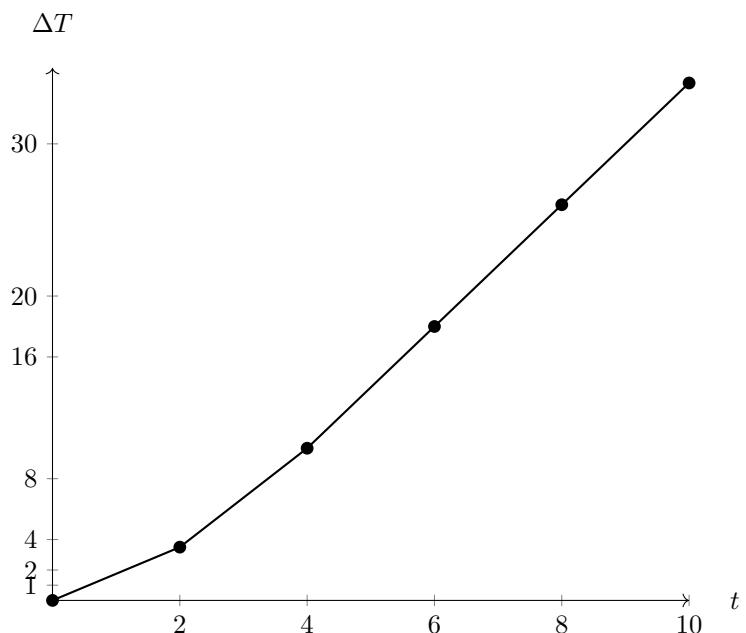


Figure 2: Graph of ΔT against t

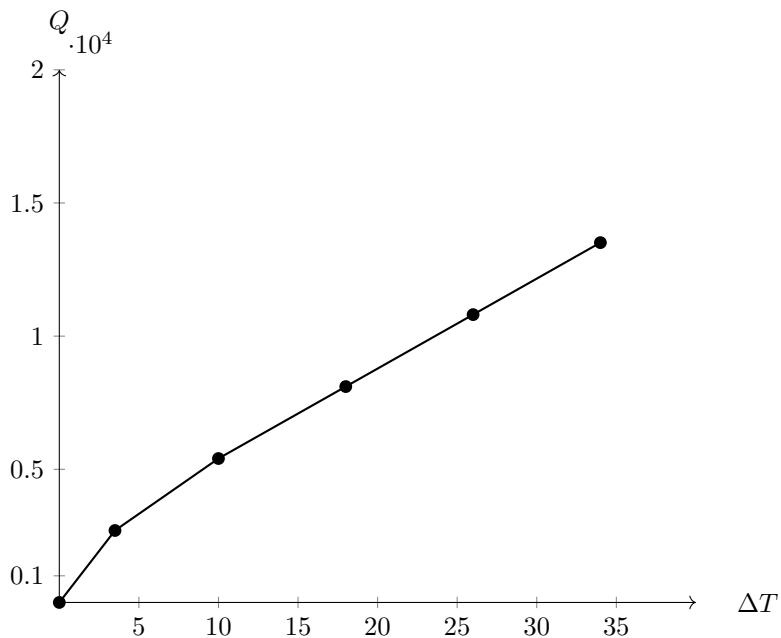


Figure 3: Graph of Q against ΔT

1.3 Questions

1. In your opinion, is it important that the block is well insulated during the experiment? Why? It is important since proper insulation ensures that most of the electrical energy supplied to the block is used to increase its temperature rather than being lost to the surroundings.
2. If you were to increase the current (A) used in this experiment, to a value higher than you used, would it change the experiment in any way? Give reasons as to your answer. Increasing the current beyond 2.7 Ampere, would definitely change the experiment. The rate of heating would increase, and the temperature of the block would increase more rapidly. This would make it more difficult to measure the temperature accurately and to record the data in a timely manner.
3. What is the function of the big variable resistor (rheostat) in this experiment? Give reasons as to your answer. The big variable resistor (rheostat) is used to control the current flowing through the circuit. By adjusting the resistance of the rheostat, the current can be adjusted to the desired value. This allows the rate of heating of the block to be controlled and the temperature changes to be measured accurately.

1.4 Conclusion

We found the slope coefficient which 379,9975. We just divide the slope by the mass of the material to get the specific heat capacity of the material. The specific heat capacity of the material is 385 J/kg°C. Which exactly matches the specific heat capacity of the material we were testing.

2 Exp 3 - Calibration of a Thermocouple, 11/02/2025

2.1 Theory

A Thermocouple is formed by joining two metals. When two different metals are joined and connected through a sensitive voltmeter, it is found that whenever a temperature difference exists between the junctions a reading is recorded on the voltmeter. The greater the temperature difference between the junctions the larger the reading. This property of the Thermocouple can be used to measure temperature. In practice the cold junction is maintained at a fix temperature i.e 0° C.

2.1.1 Aim(s)

The aim of the experiment is to calibrate a Thermocouple.

2.1.2 Procedure

As in lab manual.

2.2 Diagrams/Apparatus

Voltmeter, 2 Becker, Thermocouple, thermometer.

Temperature (°C)	Voltage (mV) Hot	Voltage (mV) Cold	Average Voltage (mV)
0	0	0	0
10	0.1	0.35	0.05
20	0.4	0.3	0.35
30	1.1	0.9	1
40	1.5	1.3	1.4
50	2.1	1.6	1.7
60	2.3	2.1	2.2
70	2.7	2.4	2.55
80	3	2.8	2.9
90	3.5	3.2	3.35
100	3.8	3.7	3.75

Figure 4: Table of measurements

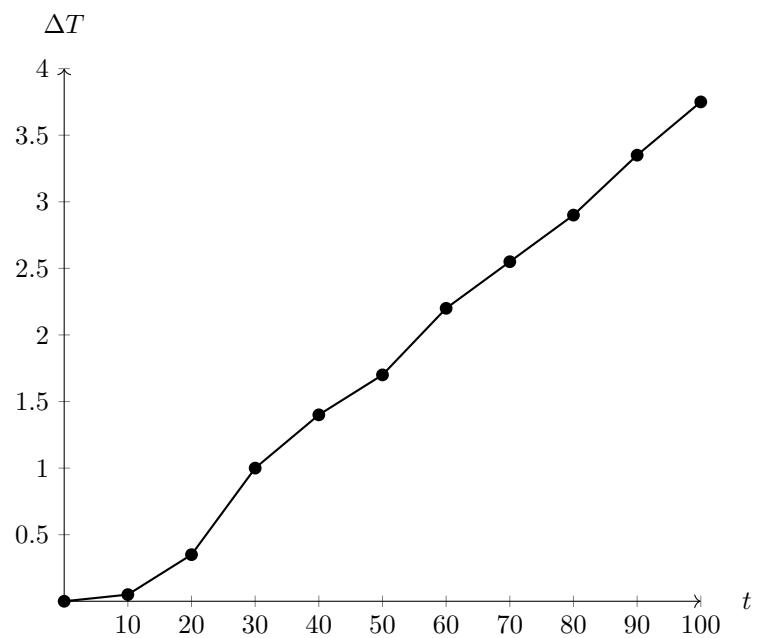
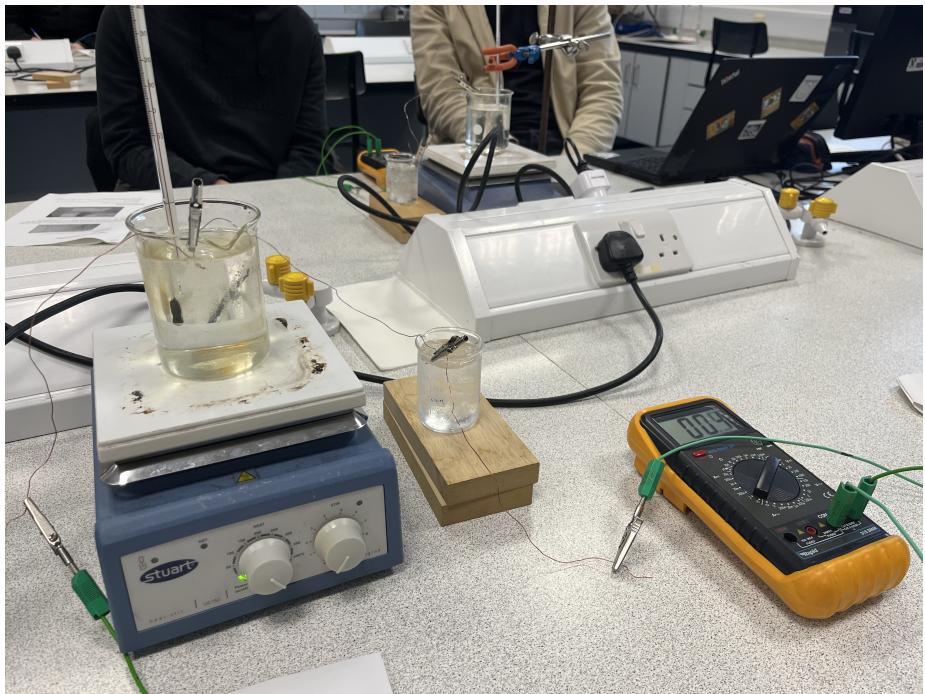


Figure 5: Graph of Average Voltage against Temperature

So we found that the equation for this graph is:

$$y = 0.04x - 0.2114$$

2.3 Questions

Use the thermocouple and the calibration graph to estimate:

- a. Room Temperature $\rightarrow 19^{\circ}\text{C}$ so using the graph equation: $y = 0.04x - 0.2114 \rightarrow 0.5486$
- b. Body Temperature $\rightarrow 37^{\circ}\text{C} \rightarrow (37 * 0.04) - 0.2114 = 1.2686$
- c. The temperature of water from the tap in the lab $\rightarrow 17.5^{\circ}\text{C}$ $(17.5 * 0.04) - 0.2114 = 0.4886$