Campinas Advanced School of Thermodynamics Workshop on Molecular Simulations Session 2 –MD of ethanol with MoSDeF

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Department of Chemical and Biomolecular Engineering

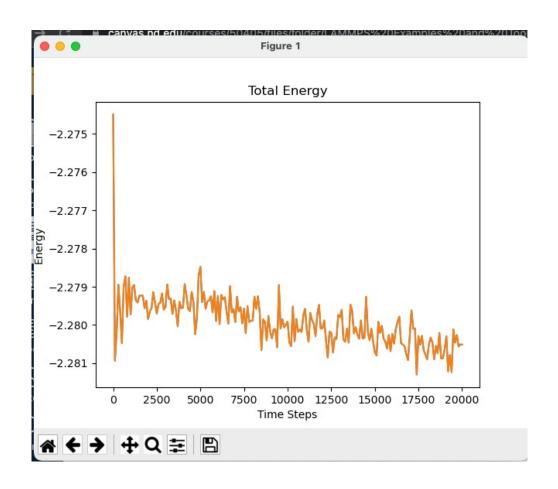
University of Notre Dame, USA

ed@nd.edu

All materials available on Maginn Group GitHub site



Computing properties

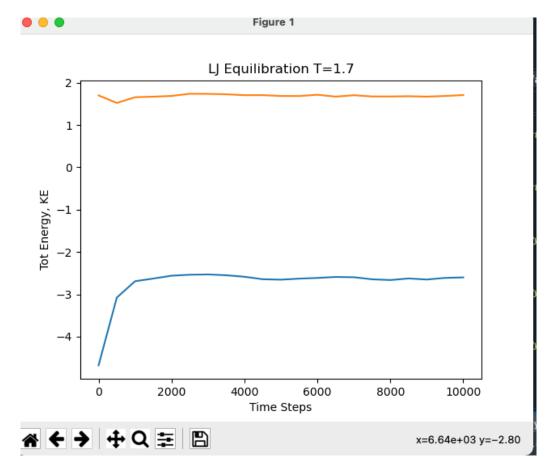


System is clearly "not equilibrated"

Procedure:

- 1. Run an "equilibration phase" where system relaxes to equilibrium. Throw this out.
- 2. Restart from the end of equilibration run, compute properties from this.

Longer run

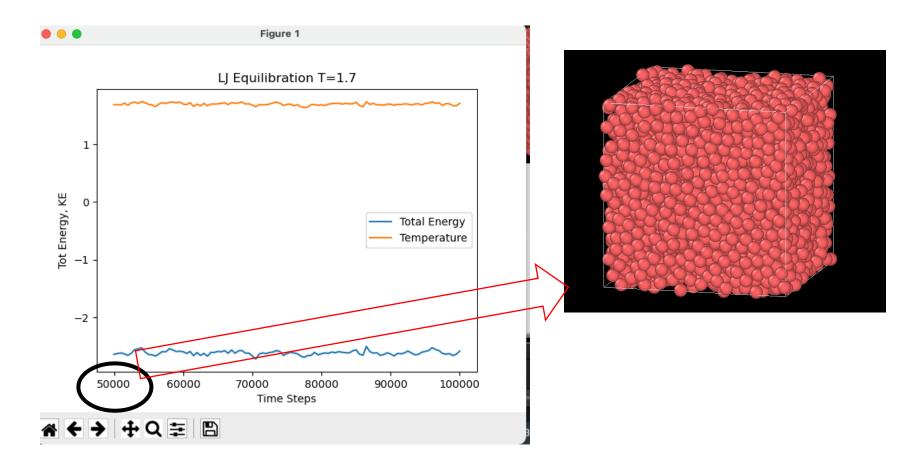


Average Energy -2.730 Average Temperature 1.686

Incorrect! Don't include part that is not equilibrated!

```
# To begin simulation from a restart file, uncomment the line below (adjusting
# the file name, if necessary) and comment the section corresponding to
# information now provided by the restart file.
#read restart ${NAME}.restart.*
# start of information provided by restart file
# (If reading data from restart file, comment this section.)
# configure and initialize system
units lj
atom style atomic
pair style lj/cut 2.5 # style args(=cutoff)
boundary p p p # x y z (p = periodic boundary conditions)
lattice fcc 0.8442 # style scale(=reduced_density)
region MYBOX block 0 10 0 10 0 10 # ID style args(=xlo,xhi,ylo,yhi,zlo,zhi)
create box 1 MYBOX # number of atom types region ID
create atoms 1 box # atom type style args
mass 1 1.0 # atom_type mass
velocity all create 3.0 87287 # group ID style args(=temp,seed)
# set forcefield parameters
pair_coeff 1 1 1.0 1.0 2.5 # atom_i atom_j args(=epsilon,sigma,cutoff)
# end of information provided by restart file
```

Restart run

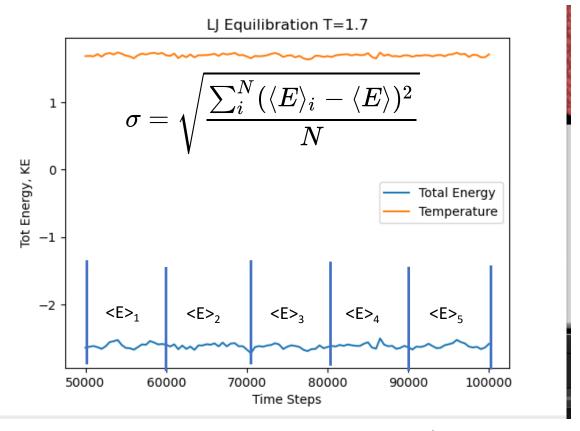


Average Energy -2.6109 Average Temperature 1.694

Proper averages and uncertainties

- Always make sure you take averages over equilibrated trajectories
 - Initial conditions are "irrelevant"
 - Like to start simulations "close" to equilibrium
 - Use restart files from other simulations close to desired statepoint
- How long should you average?
 - Longer than longest "correlation time" of system
 - Make sure running average is not changing with time
 - Rule of thumb: make sure molecules move several molecular diameters during simulation
- Uncertainties
 - Standard deviation from multiple independent simulations
 - "Block average" standard deviation

Block averages



- Blocks assumed to be independent (longer than correlation time)
- Change block size and check for convergence

Mixtures

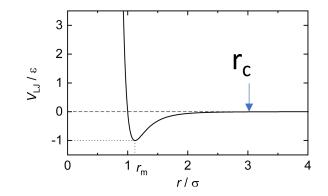
Typically relate unlike atom parameters to a combination of like parameters (combination or mixing rule):

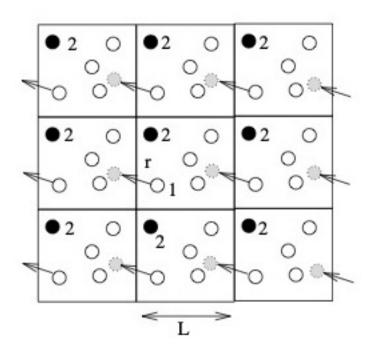
$$\epsilon_{ab}=(\epsilon_{aa}\epsilon_{bb})^{1/2}$$
 "Lorentz-Berthelot" $\sigma_{ab}=(\sigma_{aa}+\sigma_{bb})/2$

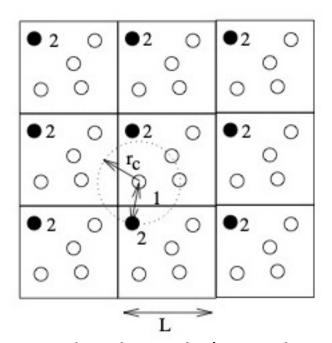
Several other possibilities (purely geometric, "size weighted", "Kong", etc.)

```
#LAMMPS mixing rules
mix geometric
mix arithmetic (this is Lorentz-Berthelot)
```

Potential Truncation







Cut off potential at no more than ½ box edge length (LJ only – charges are handled differently)

Account for energy beyond r_c

Long range ("tail") corrections

$$E_{tot} = E_{cut} + E_{LRC} = E_{cut} + 2\pi N
ho \int_{r_{cut}}^{\infty} r^2 \mathcal{V}(r) dr$$

For the LJ potential, this is analytic

$$E_{LRC} = 8\pi
ho N\epsilon \left[rac{\sigma^{12}}{9r_c^9} - rac{\sigma^6}{3r_c^3}
ight]$$

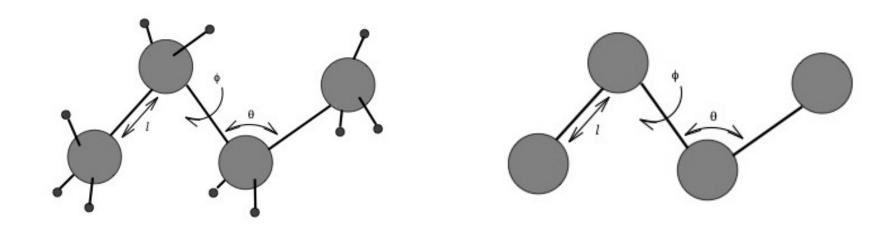
Can also correct pressure P_{LRC} is analytic

LAMMPS commands

pair_style lj/cut 2.5
pair_modify tail yes

See Leach: pp 330-334

Simulating multi-atom molecules



"all-atom" butane

"united-atom" butane

We need *intramolecular* force field terms to account for the molecule shape and motion

Intramolecular terms: Bond lengths and angles

Bond length potentials often modeled as a harmonic

$$\mathcal{V}_b(\ell) = k_b (\ell - \ell_0)^2$$

force constant nominal bond length

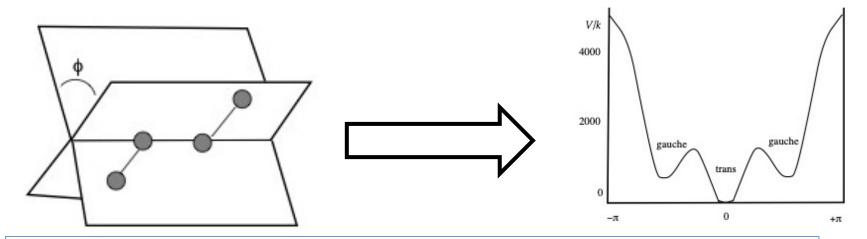
```
bond style harmonic
bond coeff 1 80.0 1.2 # type k b 1 0
```

Harmonic bond angles

$$\mathcal{V}_{ heta}(heta) = k_{ heta}(heta - heta_0)^2$$

```
angle_style harmonic
angle_coeff 1 300.0 107.0 # type k_theta theta0
```

Dihedral angles



Optimized Potentials for Liquid Simulation (OPLS) style

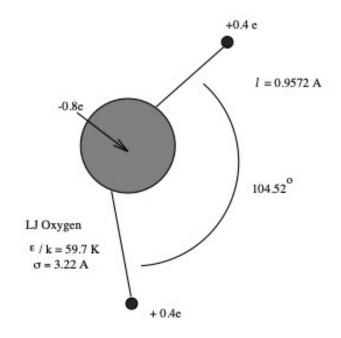
$$\mathcal{V}_{\phi} = rac{1}{2} K_1 [1 + \cos(\phi)] + rac{1}{2} K_2 [1 - \cos(2\phi)] + rac{1}{2} K_3 [1 + \cos(3\phi)] + rac{1}{2} K_4 [1 - \cos(4\phi)]$$

dihedral_style opls
dihedral coeff 1 90.0 90.0 90.0 70.0

Chemistry and HArvard Molecular Mechanics (CHARMM) style

$$\mathcal{V}_{\phi} = k_{\phi}[1+\cos(n\phi-\delta)]$$

Partial charges



Ex: Water (TIPS model)

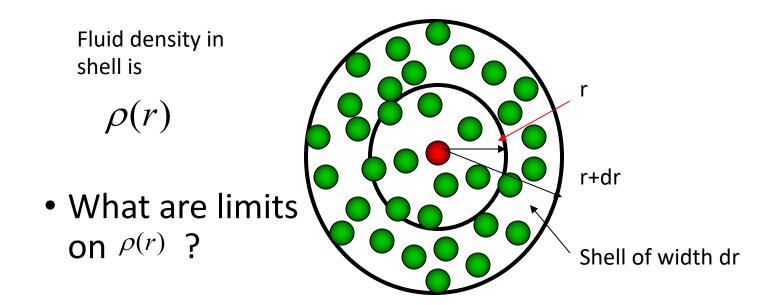
- LJ + "partial charges" on atom centers
- Charges interact via Coulomb potential

$$\mathcal{V}_{zz}(r_{ij}) = rac{z_i z_j}{4\pi\epsilon_0 r_{ij}}$$

Jorgensen, William L. (1981). "Quantum and statistical mechanical studies of liquids. 10. Transferable intermolecular potential functions for water, alcohols, and ethers. Application to liquid water". *Journal of the American Chemical Society*. American Chemical Society (ACS). **103** (2): 335–340. doi:10.1021/ja00392a016

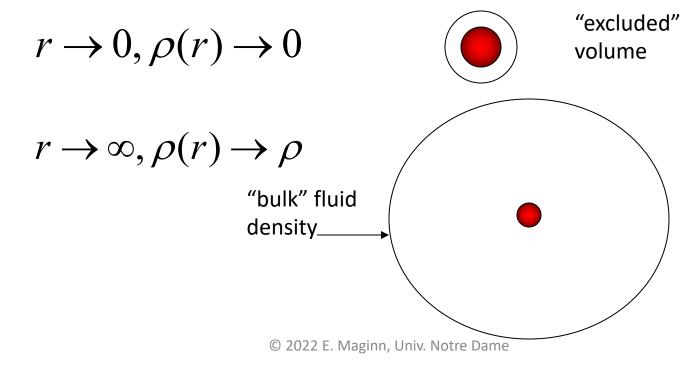
Computing Properties

- What does a fluid "look" like?
- Imagine a snapshot of an atomic liquid



Computing Properties

- What are limits on P(r)?
- As r-> 0?
- As r-> infinity?



Radial Distribution Function

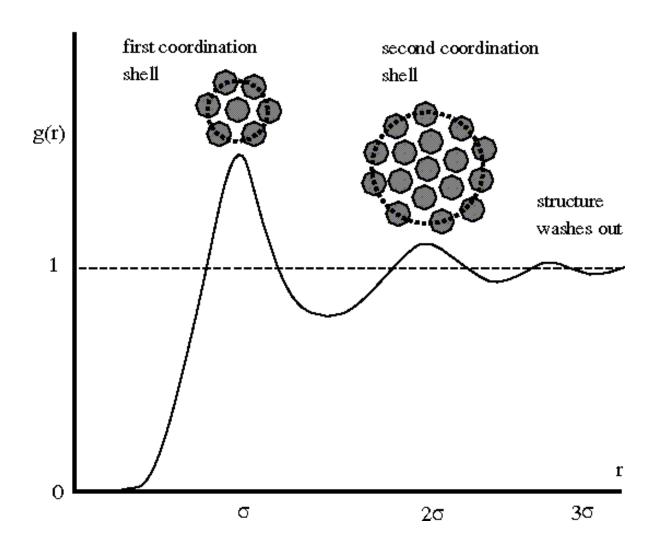
- First statement: fluids have short range "order"
- Last statement: fluids has no long range order.
- Define pair distribution function

$$g(r) \equiv \frac{\rho(r)}{\rho}$$

• Limits:

$$g(0) = 0, g(\infty) = 1$$

Liquid Structure

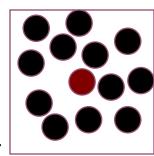


Self-diffusivity

- Right-hand side is macroscopic property
 - o applicable at macroscopic time scales

$$\left\langle r^2(t) \right\rangle = 2dDt$$
 Einstein equation

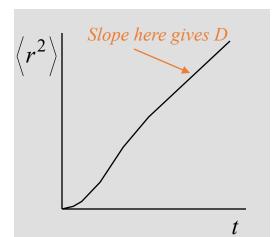
 For any given configuration, each atom represents a point of high concentration (a weak fluctuation)



- View left-hand side of formula as the movement of this atom
 - o ensemble average over all initial conditions

$$\langle r^2(t) \rangle = \int d\mathbf{p}^N \int d\mathbf{r}^N r_1^2(t) \delta(\mathbf{r}_1) \pi(\mathbf{r}^N, \mathbf{p}^N; t = 0)$$

- asymptotic linear behavior of mean-square displacement gives diffusion constant
- o independent data can be collected for each molecule $\langle r^2(t) \rangle = \frac{1}{N} \sum \langle r_i^2(t) \rangle$



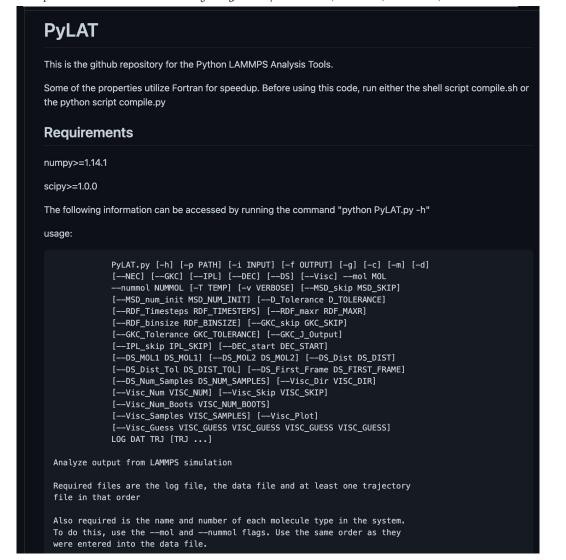
https://github.com/MaginnGroup/PyLAT



PyLAT: Python LAMMPS Analysis Tools

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Building complex force fields for LAMMPS

- Generally, need to include a "data file" in addition to input file
 - Contains basic information about the size of the simulation, the initial atomic coordinates, molecular topology, and force field parameters.
- Lots of tools to build these files: we will go into detail about using MoSDeF but there are many other ways to do this

Force field databases

http://trappe.oit.umn.edu/

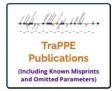
The Siepmann Group

Simulating Complex Chemical Systems

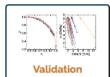
TraPPE Home Validation



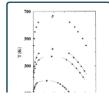
Transferable Potentials for Phase Equilibria



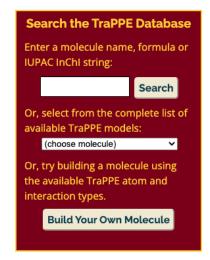
About the TraPPE Force Field



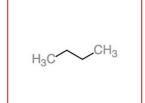
The Transferable Potentials for Phase Equilibria family of force fields is a collection of functional forms and interaction parameters useful for modeling complex chemical systems with molecular mechanics simulation techniques. TraPPE maintains a high degree of accuracy in the prediction of thermophysical properties when applied to a range of different compounds, different state points, different compositions, and different properties. This makes TraPPE one of the few force fields generally suitable for materials and industrial applications.



The development of TraPPE models prioritizes **transferability** (maximizing the ability to build new chemical compounds by minimizing the number of (pseudo-) atoms needed) and **accuracy** (quantitative prediction of phase equilibria and other thermophysical properties over a wide range of physical conditions). Starting with "bonded" 1–2 (bond length), 1–3 (bond angle) and 1–4 (dihedral angle) interactions, TraPPE relies on experimental data or electronic structure calculations to provide equilibrium bond lengths and angles, the corresponding force constants, and the dihedral potentials for



Nonbonded Interactions



n-butane (UA)

Chemical C_4H_{10} Formula:

58.12 Molecular

Weight:

Smiles String: CCCC

InChI=1S/C4H10/c1-3-4-2/h3-4H2,1-2H3

This model was assembled from previously parameterized TraPPE building blocks and was tested for accuracy as part of the transferability verification process.

#	(pseudo)atom	type	<i>€/k</i> _B [K]	σ[A]	q [e]
1	СНЗ	[CH3]-CHx	98	3.75	0.0
2	CH2	CHx-[CH2]-CHx	46	3.95	0.0
3	CH2	CHx-[CH2]-CHx	46	3.95	0.0
4	CH3	[CH3]-CHx	98	3.75	0.0

 $u_{\rm NB}(r_{ij}) = 4\varepsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ii}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ii}} \right)^{6} \right] + \frac{q_i q_j}{4\pi\varepsilon_0 r_{ij}}$

 $\sigma_{ij} = \frac{1}{2}(\sigma_{ii} + \sigma_{jj}) \epsilon_{ij} = (\epsilon_{ii} \epsilon_{jj})^{1/2}$

Parameters Simulation Data Downloads

1-2 Bonded Interactions

(TraPPE	uses	fixed	bond	<u>lengths</u>

#	Stretch	Туре	Length [Å]
1	1 – 2	CHx-CHy	1.54
2	2 - 3	CHx-CHy	1.54
3	3 – 4	CHx-CHy	1.54

1-3 Bonded Interactions

$$u_{\text{bend}}(\theta) = \frac{k_{\theta}}{2} (\theta - \theta_{\text{eq}})^{-2}$$

#	Bend	Туре	<i>θ</i> [°]	kθ/k _B [K/rad ²]
1	1 - 2 - 3	CHx-(CH2)-CHy	114	62500
2	2 - 3 - 4	CHx-(CH2)-CHy	114	62500

1-4 Bonded Interactions

(Read more about TraPPE dihedral conventions)

$$u_{\text{torsion}}(\phi) = c_0 + c_1 \left[1 + \cos(\phi) \right] + c_2 \left[1 - \cos(2\phi) \right] + c_3 \left[1 + \cos(3\phi) \right]$$

#	Torsion	Туре	c ₀ /k _B [K]	c ₁ /k _B [K]	c ₂ /k _B [K]	c ₃ /k _B [K]
1	1 - 2 - 3 - 4	CHx-(CH2)-(CH2)-CHy	0.00	355.03	-68.19	791.32

```
# Example of beginning input file using data file
           NAME index methane
variable
         ${NAME}.log
log
variable
           infile index methane.data
variable
           mytemp index 328.15
# set up simulation
units
          real
             full
atom_style
boundary
             ppp
pair_style | lj/cut/coul/long 12
kspace_style ewald 0.0001
pair modify tail yes
pair_modify mix arithmetic
special bonds amber
bond_style
             harmonic
angle_style
             harmonic
#dihedral_style opls
#improper style cvff
             ${infile}
read_data
```

LAMMPS Description (1st line of file)

```
(this must be the 3rd line, 1st 2 lines are ignored)
100 atoms
95 bonds
                    (# of bonds to be simulated)
                              (include these lines even if number = 0)
50 angles
30 dihedrals
20 impropers
                              (# of nonbond atom types)
5 atom types
10 bond types
                    (# of bond types = sets of bond coefficients)
18 angle types
20 dihedral types
                    (do not include a bond, angle, dihedral, improper type
2 improper types
                   line if number of bonds, angles, etc is 0)
-0.5 0.5 xlo xhi
                   (for periodic systems this is box size, -
                    for non-periodic it is min/max extent of atoms)
0.5 0.5
        ylo yhi
-0.5 0.5
         zlo zhi
                    (do not include this line for 2-d simulations)
Masses
1 mass ... N mass
                                                                      (N = # of
atom types)
Nonbond Coeffs
1 coeff1 coeff2 ... ... N coeff1 coeff2 ...
                                                  (N = # of atom types)
```

2500 atoms

2000 bonds

3000 angles

0 dihedrals

0 impropers

2 atom types

1 bond types

1 angle types

0.000000 40.000000 xlo xhi

0.000000 40.000000 ylo yhi

0.000000 40.000000 zlo zhi

Masses

1 12.010780 # opls_138 2 1.007947 # opls 140

Pair Coeffs # Ij

epsilon (kcal/mol) sigma (Angstrom)

1 0.06600 3.50000 # opls_138 2 0.03000 2.50000 # opls 140

Bond Coeffs # harmonic

k(kcal/mol/angstrom^2) req(angstrom)

1 340.0 1.09 # opls 138 opls 140

Angle Coeffs # harmonic

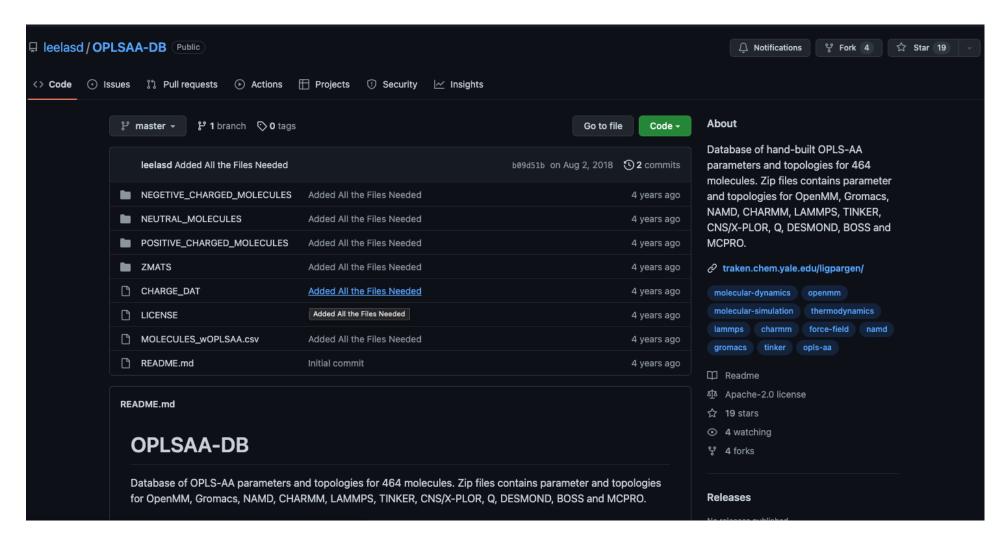
k(kcal/mol/rad^2) theteq(deg)

1 33.0 107.80000 # opls_140 opls_138 opls_140

Atoms # full

4	4	4	0.040000	26 460207	22 722742	40.050606
1	1	1	-0.240000	36.169297	32.738712	18.053636
2	1	2	0.060000	35.618914	31.986198	17.452327
3	1	2	0.060000	35.460444	33.509299	18.420370
4	1	2	0.060000	36.945195	33.219973	17.423444
5	1	2	0.060000	36.652636	32.239379	18.918402
6	1	1	-0.240000	23.235838	22.713311	15.456040
7	1	2	0.060000	22.294018	22.608173	14.879246
8	1	2	0.060000	23.222708	23.672111	16.013982
9	1	2	0.060000	24.099230	22.702595	14.759470
10	1	2	0.060000	23.327397	21.870364	16.171462
11	1	1	-0.240000	9.485563	13.833860	7.944599
12	1	2	0.060000	8.658834	13.469003	8.588158
13	1	2	0.060000	9.944858	12.976459	7.411044
14	1	2	0.060000	10.251984	14.331392	8.573749
15	1	2	0.060000	9.086575	14.558589	7.205445
16	1	1	-0.240000	31.850319	18.437411	36.362290
17	1	2	0.060000	31.507804	19.006143	37.251108
18	1	2	0.060000	32.687095	18.979813	35.876105
19	1	2	0.060000	31.011002	18.332433	35.644448
20	1	2	0.060000	32.195375	17.431255	36.677498

OPLS – "pre-built"

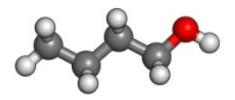


Build your own OPLS

An easy to use tool http://zarbi.chem.yale.edu/ligpargen/

1. Enter SMILES code: example

OCCCC = 1-butanol



2. Create PDB file (can do this with many other tools such as Avogadro)

```
REMARK LIGPARGEN GENERATED PDB FILE
      1 000 UNK 1
                     1.000 1.000 0.000
      2 C01 UNK 1
                    -0.421 1.000 0.000
      3 C02 UNK 1
                    -0.911 1.000 1.441
ATOM 4 C03 UNK 1 -2.436 0.998 1.516
ATOM 5 C04 UNK 1 -2.922 1.000 2.956
       6 H05 UNK 1
                     1.284 0.999 -0.930
ATOM 7 H06 UNK 1
                     -0.768 0.108 -0.531
      8 H07 UNK 1
ATOM
                     -0.768 1.892 -0.531
ATOM 9 H08 UNK 1
                     -0.506 1.877 1.962
ATOM 10 H09 UNK 1
                     -0.504 0.125 1.962
ATOM 11 HOA UNK 1
                     -2.832 0.114 1.004
ATOM 12 HOB UNK 1
                     -2.834 1.880 1.001
ATOM 13 HOC UNK 1
                     -2.566 0.114 3.490
ATOM 14 HOD UNK 1
                     -4.017 0.997 2.985
ATOM 15 HOE UNK 1
                     -2.568 1.889 3.487
TER
CONECT 1 2
CONECT 2 3
CONECT 3 4
CONECT 4 5
CONECT 1 6
CONECT 2 7
CONECT 2 8
CONECT 3 9
CONECT 3 10
CONECT 4 11
CONECT 4 12
CONECT 5 13
CONECT 5 14
CONECT 5 15
END
```

Build your own OPLS

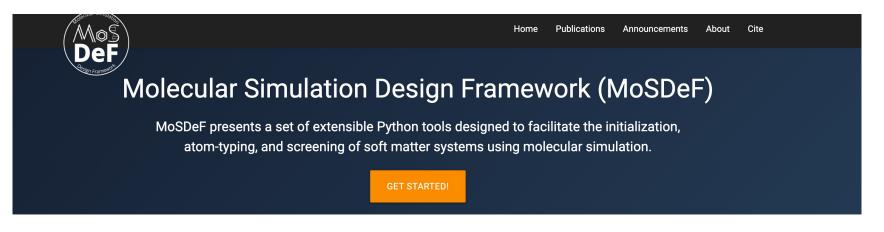
- LigParGen will spit out a LAMMPS datafile for a single molecule in addition to the pdb (and other "data files" for different packages)
- Partial charges computed using one of two methods
- You should always verify the parameters you get!
- A huge source of error and inconsistency in the literature is incorrect force field parameters.
- Lots of other resources

https://github.com/paduagroup/fftool

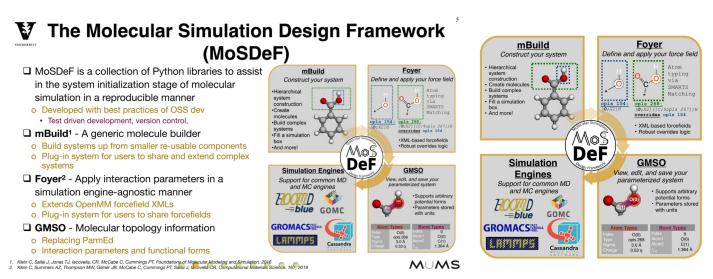
(inconsistent with LigParGen but conversion tools available)

What about other force fields?

- Lots of tools and converters to go between formats
- Eventually, you will likely need to write or modify scripts
- Great care is required because it is easy to make mistakes!
- For all but the simplest molecules, write scripts
- We will show you how to use a general tool under development called MoSDeF



The MoSDeF tools have been designed to facilitate the hierarchical construction of both atomistic and coarse-grained system configurations, the application of classical force fields (i.e., atom-typing), encapsulation of chemical topology information, and the generation of input files for a variety of molecular dynamics and Monte Carlo simulation engines, including LAMMPS, GROMACS, HOOMD-Blue, Cassandra, and GOMC.



https://github.com/mosdef-hub

mosdef.org