QUESTION-06

$$Na_3PO_4 + Ba(NO_3)_2 \rightarrow Ba_3(PO_4)_2 + NaNO_3$$

 $Na_3PO_4 : \begin{bmatrix} 3\\4\\0\\0\\0 \end{bmatrix}$, $Ba(NO_3)_2 : \begin{bmatrix} 0\\0\\6\\1\\2 \end{bmatrix}$, $Ba_3(PO_4)_2 : \begin{bmatrix} 0\\2\\8\\3\\0 \end{bmatrix}$, $NaNO_3 : \begin{bmatrix} 1\\0\\3\\3\\0 \end{bmatrix} \rightarrow Ra$

The coefficients in the eq. ase: $x_1. Na_3 PO_4 + x_2. Ba (NO_3)_2 \rightarrow x_3. Ba_3 (PO_4)_2 + x_4. (NaNO_3)$ $x_1 \begin{bmatrix} 3 \\ 4 \\ 4 \\ 0 \end{bmatrix} = \begin{bmatrix} 0 \\ 2 \\ 8 \\ 3 \end{bmatrix} + x_4 \begin{bmatrix} 1 \\ 0 \\ 3 \\ 0 \end{bmatrix}$

$$\begin{bmatrix} 3 & 0 & 0 & -1 & 0 \\ 1 & 0 & -2 & 0 & 0 \\ 4 & 6 & -8 & -3 & 0 \\ 0 & 1 & -3 & 0 & 0 \\ 0 & 2 & 0 & -1 & 0 \end{bmatrix}$$
 Replace R1 and R2
$$\begin{bmatrix} 1 & 0 & -2 & 0 & 0 \\ 3 & 0 & 0 & -1 & 0 \\ 4 & 6 & -8 & -3 & 0 \\ 0 & 1 & -3 & 0 & 0 \\ 0 & 2 & 0 & -1 & 0 \end{bmatrix}$$

Replace R2 & R4
$$\begin{bmatrix} 1 & 0 & -2 & 0 & 0 \\ 0 & 1 & -3 & 0 & 0 \\ 0 & 6 & 0 & -3 & 0 \\ 0 & 0 & 8 & -1 & 0 \\ 0 & 2 & 0 & -1 & 0 \end{bmatrix}$$

$$R_{3+}(-6)R_{2} \begin{bmatrix} 1 & 0 & -2 & 0 & 0 \\ 0 & 1 & -3 & 0 & 0 \\ 0 & 0 & 18 & -3 & 0 \\ 0 & 0 & 6 & -1 & 0 \\ 0 & 2 & 0 & -1 & 0 \end{bmatrix}$$

The general solution is:
$$X_1 = (\frac{1}{3})X_4$$
, $X_2 = (\frac{1}{2})X_4$, $X_3 = (\frac{1}{6})X_4$ with X_4 free

Taking
$$Xu = 6$$

 $X_1 = 2$
 $X_2 = 3$
 $X_3 = 1$

The balanced equation is:

2Na3 PO4 + 3Ba (NO3)2 -> Ba3 (PO4)2 + 6 NaNO3

JUESTION-07

NaHCO3:
$$\frac{1}{1}$$
 H3CcHcO7: $\frac{1}{8}$ Na3CcHsO7: $\frac{1}{5}$ H2O: $\frac{1}{2}$ CO2: $\frac{1}{2}$ $\frac{1}{3}$ $\frac{1}{4}$ $\frac{1}{3}$ $\frac{1}{4}$ $\frac{1}{3}$ $\frac{1}{4}$ $\frac{1}{$

$$\begin{array}{c|c} x_1 & \begin{bmatrix} 1 \\ 1 \\ 1 \end{bmatrix} + x_2 & \begin{bmatrix} 6 \\ 8 \\ 6 \\ 7 \end{bmatrix} = x_3 \begin{bmatrix} 3 \\ 5 \\ 4 \end{bmatrix} + x_4 \begin{bmatrix} 6 \\ 2 \\ 7 \end{bmatrix} + x_5 \begin{bmatrix} 6 \\ 0 \\ 1 \end{bmatrix} \\ \begin{bmatrix} 2 \\ 2 \end{bmatrix} \end{array}$$

$$\begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & -\frac{1}{4} - \frac{1}{4} & 0 & 0 \\ 0 & 0 & \frac{3}{2} & \frac{3}{2} & -1 & 0 \\ 0 & 7 & 2 & -1 & -2 & 0 \end{bmatrix}$$

$$R_{4+(-7)R_{2}} \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & -1/4 & -1/4 & 0 & 0 \\ 0 & 0 & -3/1 & 3/2 & -1 & 0 \\ 0 & 0 & 1/4 & 3/4 & -2 & 0 \end{bmatrix}$$

$$R_{3} \times (-3/3) \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & -1/4 & -1/4 & 0 & 0 \\ 0 & 0 & 1 & -1 & 2/3 & 0 \\ 0 & 0 & 15/4 & 3/4 & -2 & 0 \end{bmatrix}$$

$$R_{4+}(-\frac{1}{4})R_{3} \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & -\frac{1}{4} & -\frac{1}{4} & 0 & 0 \\ 0 & 0 & 1 & -1 & \frac{2}{3} & 0 \\ 0 & 0 & 0 & \frac{9}{2} & -\frac{9}{2} & 0 \end{bmatrix} R_{4} \times (\frac{2}{4}) \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & -\frac{1}{4} & -\frac{1}{4} & 0 & 0 \\ 0 & 0 & 1 & -1 & \frac{2}{3} & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{bmatrix}$$

$$R_{2} + \begin{pmatrix} \frac{1}{4} \end{pmatrix} R_{3} = \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 & -1/3 & 0 \\ 0 & 0 & 1 & 0 & -1/3 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{bmatrix} \xrightarrow{R_{1} + \begin{pmatrix} 3 \end{pmatrix}} R_{3} = \begin{bmatrix} 1 & 0 & 0 & 0 & -1 & 0 \\ 0 & 1 & 0 & 0 & -1/3 & 0 \\ 0 & 0 & 1 & 0 & -1/3 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{bmatrix}$$

The general solution is
$$\chi_1 = \chi_5$$
, $\chi_2 = \left(\frac{1}{3}\right)\chi_5$, $\chi_3 = \left(\frac{1}{3}\right)\chi_5$, $\chi_4 = \chi_5$ where χ_5 is free

The balanced equation is

3NaHCO3 + H3C6H5O7 -> Na3C6H5O7 + 3H2O + 3CO2

UESTION-08

$$\begin{array}{c} K \, M_{n}O_{4} + M_{n}SO_{4} + H_{2}O \rightarrow M_{n}O_{2} + K_{2}SO_{4} + H_{2}SO_{4} \\ \begin{pmatrix} 1 \\ 1 \\ 4 \\ 0 \\ 0 \end{pmatrix}, M_{n}SO_{4} : \begin{pmatrix} 0 \\ 1 \\ 4 \\ 0 \\ 0 \end{pmatrix}, H_{2}O: \begin{pmatrix} 0 \\ 1 \\ 0 \\ 1 \\ 0 \\ 0 \end{pmatrix}, M_{n}O_{2} : \begin{pmatrix} 0 \\ 1 \\ 2 \\ 0 \\ 0 \end{pmatrix}, K_{2}SO_{4} : \begin{pmatrix} 2 \\ 2 \\ 0 \\ 4 \\ 1 \\ 0 \end{pmatrix}, H_{2}SO_{4} : \begin{pmatrix} 0 \\ 4 \\ 4 \\ 1 \\ 0 \end{pmatrix} \rightarrow M_{n}O_{2} : \begin{pmatrix} 0 \\ 1 \\ 2 \\ 0 \\ 0 \end{pmatrix}, K_{2}SO_{4} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2} : \begin{pmatrix} 0 \\ 4 \\ 1 \\ 0 \\ 0 \end{pmatrix} \rightarrow K_{n}O_{2$$

The coefficients in the eq. ace:

$$\begin{bmatrix} 1 & 0 & 0 & 0 & -2 & 0 & 0 \\ 1 & 1 & 0 & -1 & 0 & 0 & 0 \\ 4 & 4 & 1 & -2 & -4 & -4 & 0 \\ 0 & 1 & 0 & 0 & -1 & -1 & 0 \\ 0 & 0 & 2 & 0 & 0 & -2 & 0 \end{bmatrix} \qquad \begin{matrix} R_{2+}(-1)R_1 \\ R_$$

$$R_{4+}(-1)R_{2} \begin{bmatrix} 1 & 0 & 0 & 0 & -2 & 0 & 0 \\ 0 & 1 & 0 & -1 & 2 & 0 & 0 \\ 0 & 0 & 1 & 2 & -4 & -4 & 0 \\ 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 2 & 0 & 0 & -2 & 0 \end{bmatrix}$$

$$R_{4+(-1)R_{2}} \begin{bmatrix} 1 & 0 & 0 & 0 & -2 & 0 & 0 \\ 0 & 4 & 0 & -1 & 2 & 0 & 0 \\ 0 & 0 & 1 & 2 & -4 & -4 & 0 \\ 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 2 & 0 & 0 & -2 & 0 \end{bmatrix} R_{5+(-2)R_{3}} \begin{bmatrix} 1 & 0 & 0 & 0 & -2 & 0 & 0 \\ 0 & 1 & 0 & -1 & 2 & 0 & 0 \\ 0 & 0 & 1 & 2 & -4 & -4 & 0 \\ 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 0 & -4 & 8 & 6 & 0 \end{bmatrix}$$

$$R_{5+}(4)R_{4} = \begin{bmatrix} 1 & 0 & 0 & 0 & -2 & 0 & 0 \\ 0 & 1 & 0 & -1 & 2 & 0 & 0 \\ 0 & 0 & 1 & 2 & -4 & -4 & 0 \\ 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 0 & 0 & -4 & 2 & 0 \end{bmatrix} R_{5*} \begin{pmatrix} -\frac{1}{4} \\ -\frac{1}{4} \\ 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 0 & 0 & 1 & -3 & -1 & 0 \\ 0 & 0 & 0 & 0 & 1 & -\frac{1}{2} & 0 \\ 0 & 0 & 0 & 0 & 1 & -\frac{1}{2} & 0 \end{bmatrix}$$

The general solution is:- $\chi_1 = \chi_6, \quad \chi_2 = \left(\frac{3}{2}\right) \chi_6, \quad \chi_3 = \chi_6, \quad \chi_4 = \left(\frac{5}{2}\right) \chi_6, \quad \chi_5 = \left(\frac{1}{2}\right) \chi_6$ where χ_6 is given

The balanced equation is:

2KMn Dy + 3Mn SOu + 2H2O -> 5Mn O2 + K2SOu + 2H2SOq

luestion-09

$$\begin{array}{c} PbN_{6} + C_{8}Mn_{2}O_{8} \rightarrow Pb_{3}O_{4} + C_{82}O_{3} + MnO_{2} + NO \\ \hline \\ PbN_{6} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, C_{1}Mn_{2}O_{8} : \begin{bmatrix} 0 \\ 0 \\ 1 \\ 2 \\ 8 \end{bmatrix}, Pb_{3}O_{4} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, C_{1}2O_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, MnO_{2} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, NO : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow Pb \\ \rightarrow NO : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, MnO_{2} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}, NO : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{2} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix} \rightarrow MnO_{3} : \begin{bmatrix} 0 \\ 0 \\$$

The coefficients of eq. are: x_1 . PbN6 = x_2 Camp. $0_8 \rightarrow x_3$ Pb30u + x_4 . $C_{12}O_3 + x_5$ MnO2 + x_6 NO x_1 . PbN6 = x_2 Camp. $0_8 \rightarrow x_3$ Pb30u + x_4 . $0_9 \rightarrow x_6$ MnO2 + x_6 NO x_1 . x_2 x_3 x_4 x_4 x_5 x_5 x_6 x_6 x

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Replace

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R2 and R3

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 $\begin{array}{c} R_{3}+(-6)R_{1} \\ \sim \\ & \begin{array}{c} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 18 & 0 & 0 & 1 & 0 \\ 0 & 2 & 0 & 0 & -1 & 0 & 0 \\ 0 & 8 & -4 & -3 & -2 & -1 & 0 \end{array} \end{array}$

 $R_{4+}(-2)R_{2} \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 & -1/18 & 0 \\ 0 & 0 & 0 & 4 & -1 & 0 & 0 \\ 0 & 8 & -4 & -3 & -2 & -1 & 0 \end{bmatrix} R_{5+}(-8)R_{2} \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 & -1/18 & 0 \\ 0 & 0 & 0 & 4 & -1 & 0 & 0 \\ 0 & 0 & -4 & 13 & -2 & -1 & 0 \end{bmatrix}$

 $R_{5}+(4)R_{3} = \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 & -1/8 & 0 \\ 0 & 0 & 0 & 4 & -1 & 0 & 0 \\ 0 & 0 & 0 & 13 & -2 & -1/4 & 0 \\ \end{bmatrix}$ $R_{5}+(4)R_{3} = \begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 & -1/8 & 0 \\ 0 & 0 & 0 & 1 & -1/4 & 0 & 0 \\ 0 & 0 & 0 & 13 & -2 & -1/4 & 0 \\ 0 & 0 & 0 & 13 & -2 & -1/4 & 0 \\ \end{bmatrix}$

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$$R_{4+}(\frac{1}{4})R_{5}\begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & -2 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 & -\frac{1}{18} & 0 \\ 0 & 0 & 0 & 1 & 0 & -\frac{1}{18} & 0 \\ 0 & 0 & 0 & 1 & -\frac{1}{18} & 0 \end{bmatrix}$$

$$R_{2+}(\frac{1}{2})R_{4}\begin{bmatrix} 1 & 0 & -3 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 & 0 & -\frac{1}{243} & 0 \\ 0 & 0 & 0 & 1 & 0 & -\frac{1}{18} & 0 \\ 0 & 0 & 0 & 1 & 0 & -\frac{1}{18} & 0 \\ 0 & 0 & 0 & 0 & 1 & -\frac{1}{14} & 0 \end{bmatrix}$$

$$\begin{array}{c} R_{1+}(B)R_{3} \\ \sim \\ \end{array} \begin{bmatrix} 1 & 0 & 0 & 0 & 0 & -1/6 & 0 \\ 0 & 1 & 0 & 0 & 0 & -22/45 & 0 \\ 0 & 0 & 1 & 0 & 0 & -1/18 & 0 \\ 0 & 0 & 0 & 1 & 0 & -11/45 & 0 \\ 0 & 0 & 0 & 0 & 1 & -44/45 & 0 \end{bmatrix}$$

The general solution is:
$$X_1 = \left(\frac{1}{6}\right) \times 6, \quad X_2 = \left(\frac{12}{45}\right) \times 6, \quad X_3 = \left(\frac{1}{18}\right) \times 6, \quad X_4 = \left(\frac{11}{45}\right) \times 6, \quad X_5 = \left(\frac{14}{45}\right) \times 6$$
with X_6 free

The balanced eq. is

15 PbN6 + 44(8 Mn, 08 -> 5 Pb2O4 + 22 Co, D3 + 88 MnO2 + 90 NO

luestion - 10

Mns:
$$O_{0} = A_{2} C_{10} O_{35} + H_{2} SO_{4} \rightarrow HMnO_{4} + A_{5} H_{3} + C_{8} S_{3} O_{12} + H_{2} O_{0}$$

Mns: $O_{0} = A_{5} C_{10} O_{35} C_{10} O_{35} C_{10} O_{10} O_{10}$
 $O_{0} = A_{5} C_{10} O_{35} C_{10} O_{35} C_{10} O_{10} O_{10}$
 $O_{0} = A_{5} C_{10} O_{35} C_{10} O_{10} O_{10} O_{10} O_{10}$
 $O_{0} = A_{5} C_{10} O_{35} C_{10} O_{10} O$

The coefficients of eq. axe: XI. MnS+x2. As2Cx10Q35+X3. H2SO4 > X4HMnO4+X5A5H3+X6-C8S3O12+X7H2O

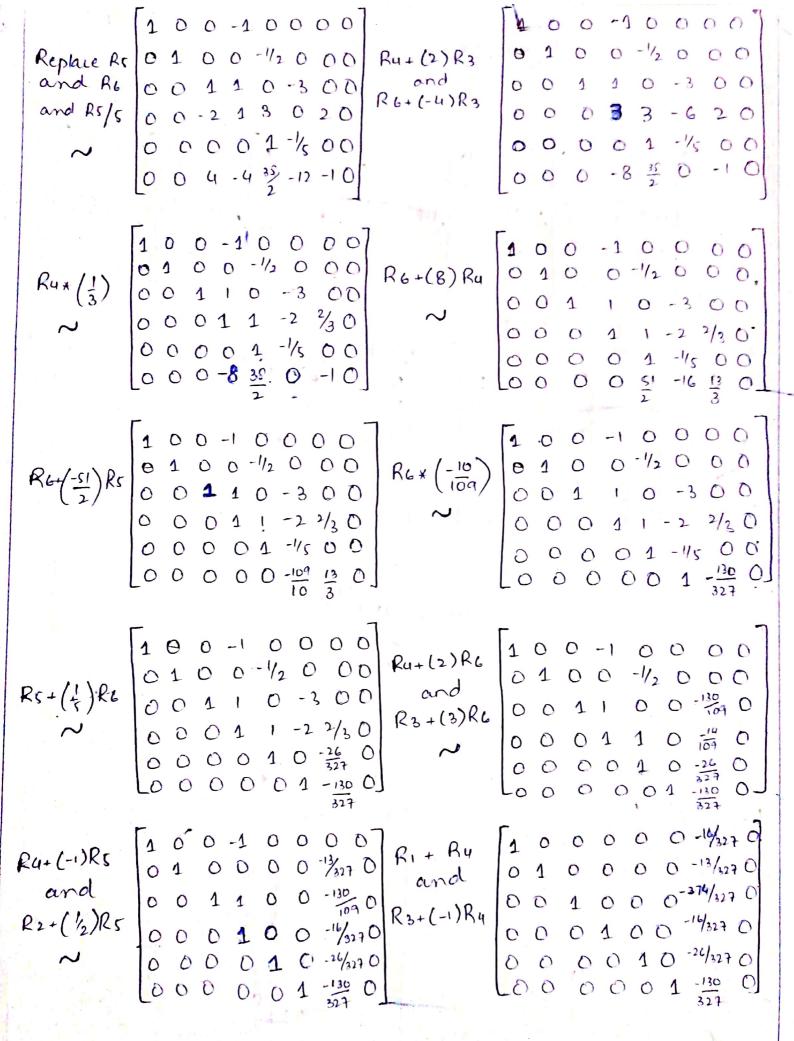
100-10000 100-100000 Replace |0100-1/2000 R4+(-10)R2 0 1 0 0 -1/2 0 Rz and Rs 010000-100 0 35 4 -4 0 -12 -1 0

100-10000 Rs+(-35)R2 0 1 0 0 -1/2 0 00 0 0 1 1 0 -3 00 00005-100 and Rux-1

Replace Ru and Re

100-100001 0 1 0 0 - 1/2 0 0 0 0 0 1 0 0 -2

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The general solution is: $\chi_1 = \left(\frac{16}{327}\right) \chi_7$

$$\chi_1 = \left(\frac{16}{327}\right) \chi_7$$

$$\chi_2 = \left(\frac{13}{317}\right)\chi_7$$

$$\chi_3 = \left(\frac{374}{327}\right)\chi_7$$

$$\chi_4 = \left(\frac{16}{327}\right)\chi_1$$

$$\chi(\zeta) = \left(\frac{26}{327}\right)\chi\gamma$$

$$16 = \left(\frac{130}{327}\right) 1$$
Where χ_1 is free

The balanced equation is:

16 MnS + 13 As2 (210 D35 + 374 H25Q4 -> 16 HMn D4 + 26 AsH3 + 130 CxS3012 + 327 H20