

Initial Composition of feed:

Weight of methanol = 1.5 kg

Weight of ethanol = 0.75 kg

Weight of butanol = 0.75 kg

$$\text{moles of methanol} = \frac{1.5}{32.04} = 46.816 \text{ mol}$$

$$\text{moles of ethanol} = \frac{0.75}{46.07} = 16.273 \text{ mol}$$

$$\text{moles of Butanol} = \frac{0.75}{74.121} = 10.118 \text{ moles}$$

$$\text{Total moles} = 73.213 \text{ mole}$$

$$\text{mole fraction of methanol} = 0.639$$

$$\text{mole fraction of ethanol} = 0.222$$

$$\text{mole fraction of Butanol} = 0.139$$

mole fractions:

1) Total Reflux

1a) methanol →

Distillate

$$\text{Area ratio} = \frac{\text{Area of methanol}}{\text{Area of cyclohexane}}$$

$$(AR) = \frac{10.54}{75.95}$$

$$= 0.14$$

Calibration curve $\Rightarrow y = 1.293x + 0.1829$

Weight ratio = $1.293 \times 0.14 + 0.1829$
 $= 0.36$

Weight of methanol = Weight ratio \times Weight of cyclohexane
 $= 0.36 \times 5.008$
 $= 1.81$

1.2) ethanol

	Area ratio ethanol	weight ratio ethanol	weight of ethanol	mole fraction ethanol
Distillate	0.176	0.316	0.372 1.58	0.0372 0.6372
Bottom	0.183	0.327	0.376 1.63	0.0316 0.0316

1.3) Butanol

	Area ratio Butanol	weight ratio Butanol	weight of Butanol	mole fraction Butanol
Distillate	0.00197	0.008	0.09	0.013
Bottom	0.266	0.285	1.42	0.17

~~Sample~~ Sample Calculation for Distillate \rightarrow

mole fraction of methanol = $\frac{1.81}{32.04} = 0.64$

$\frac{1.81}{32.04} + \frac{1.58}{46.07} + \frac{0.09}{79.121}$

~~22 Ref~~

Total Reflux

 $R=1.9$ $R=1$

Pre-methanol-top	0.9509 atm	1.0089 atm	1.0452 atm
Post-methanol-bottom	1.5766 atm	1.7443 atm	1.9332 atm
Pre-ethanol-top	0.5342 atm	0.5703 atm	0.5930 atm
Post-ethanol-bottom	0.9336 atm	1.0437 atm	1.1689 atm
Pre-butanol-top	0.0948 atm	0.1025 atm	0.1074 atm
Post-butanol-bottom	0.1841 atm	0.2100 atm	0.2401 atm

Relative Volatility:-

$\alpha_{\text{methanol-ethanol-top}}$	1.7802	1.7691	1.7625
$\alpha_{\text{methanol-ethanol-bottom}}$	1.6887	1.6712	1.6538
$\alpha_{\text{methanol-butanol-top}}$	10.0329	9.8447	9.7348
$\alpha_{\text{methanol-butanol-bottom}}$	8.5644	8.3053	8.0522
$\alpha_{\text{ethanol-butanol-top}}$	5.6359	5.5649	5.5233
$\alpha_{\text{ethanol-butanol-bottom}}$	5.0717	4.9695	4.8689

Minimum Number of Trays:

Fenske Eqⁿ :

$$N = \frac{\log \left[\left(\frac{x_d}{1-x_d} \right) \left(\frac{1-x_b}{x_b} \right) \right]}{\log(\alpha_{avg})}$$

x_d = mole fraction of light key in distillate

x_b = mole fraction of heavy key component at bottom

α_{avg} = mean of relative volatilities at top and bottom of the column

$$\alpha_{avg} = \sqrt{\alpha_t \alpha_b}$$

$$= 9.27$$

$$P = \sqrt{10.0329 \times 8.5644}$$

$$\alpha_t = \alpha_{MB} = 10.0329$$

$$\alpha_b = \alpha_{MB} = 8.5644$$

$$N = \frac{\log \left[\frac{0.61}{1-0.61} \times \frac{1-0.81}{0.81} \right]}{\log(9.27)} = 0.912$$

We need Minimum 8.91 tray.

$$51.0 = \text{min}$$

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Minimum Reflux Ratio :

by Underwood eqⁿ :

$$\sum \left(\frac{d_i Z_i}{d_i - \theta} \right) = 1 - q$$

$$\sum \left(\frac{d_i Z_i}{d_i - \theta} \right) = 1 - q \quad R_{min} \text{ at } q = 0$$

Saturated liq. $\Rightarrow q = 0$

\therefore Substituting values,

$$\frac{9.8447 \times 0.639}{9.8447 - \theta} + \frac{5.649 \times 0.222}{5.649 - \theta} + \frac{1 \times 0.139}{1 - \theta} = 1$$

This gives $\theta = 1.14$ or 6.36

$\theta = 1.14$ gives R_{min} as negative

$\theta = 6.36$ gives

$$1 + R_{min} = \frac{9.8447 \times 0.639}{9.8447 - 6.36} + \frac{5.649 \times 0.222}{5.649 - 6.36} + \frac{0.005 \times 1}{1 - 6.36}$$

$$R_{min} = 0.12$$

$\theta = 6.36$ gives

$$R_{min} = -ve$$

$$\therefore R_{min} = 0.12$$