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ARTICLE TYPE

Energy level tuning of TPB-based hole-transporting materials for high efficient perovskite solar cells

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Two TPB-based HTMs were synthesized and their energy levels were tuned to match with perovskite by introducing electron-donating groups asymmetrically. The TPBC based doping-free perovskite solar cell afforded an impressive PCE of 13.10% under AM 1.5G illumination, which is the first case for effective device with TPB-based doping-free HTMs.

hybrid organic-inorganic perovskites recently. (CH₃NH₃PbX₃, where X corresponds to halogenes) solar cells 15 have attracted much attention due to their simple structure, low production cost and superb photovoltaic performance. Using 2,2',7,7'-tetrakis(N,N-di-p-methoxyphenylamine)-9,9'spirobifluorene (spiro-OMeTAD) as a hole transporting material (HTM) to instead of liquid electrolytes in this hybrid solar cell, 20 power conversion efficiencies (PCE) of over 15.0% were achieved under 1 sun (100 mW/cm²). 1,2,3,4 Although spiro-OMeTAD has been considered as the best candidate HTM for perovskite solar cells, intensive effort has been devoted to develop cheaper alternatives to spiro-OMeTAD. Several polymer 25 hole conductors have been employed as HTM and the highest PCE was 12.0% based on PTAA. Compared to polymer HTMs, small-molecule HTMs benefit from the fact that they are easy to purify and easily form crystalline films for the fabrication of desired high efficiency devices. Various small-molecule HTMs, 30 such as 3,4-ethylenedioxythiophene-, 6 pyrene-, 7 linear π conjugated structure, 8 butadiene-, 9 swivel-cruciform thiophene-, ¹⁰ oligothiophene-, ¹¹ tetrathiafulvalene-, ¹² quinolizinoacridine, ¹³ and 1,3,5-triazine-14 based HTMs have been developed and gave efficiency of 8.8~13.8%. Our group have reported the synthesis 35 of several triphenylamine-based small-molecule HTMs, and impressive performances (PCE=11.6%) have been achieved with these HTMs. 8,9 From the commercialization viewpoint, the exploration of efficient and cost-effective HTM is still the key challenge for further advancement of the hybrid perovskite solar 40 cell.

N,N,N',N'-tetraphenyl-benzidine (**TPB**), as one essential HTM, has been widely used in diverse organic electronic devices such as organic light emitting diode (OLED), organic field-effect transistor (OFET), dye-sensitized solar cells (DSSC)^{15,16,17,18,19} etc. It also usually serve as building blocks for construction of largely conjugated HTM molecules due to its superb hole-transport ability.²⁰Unfortunately, to the best of our knowledge, development and application of **TPB**-based HTM in perovskite

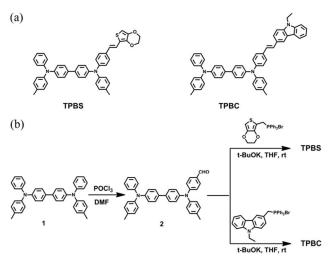


Fig. 1 (a) Molecular structures of HTMs; (b) Synthetic route for HTMs.

solar cell has been neglected. In our previous work, **TPB** was introduced into the perovskite solar cells to engineer the metal-semiconductor interface. ²¹ The PCE increased by about 28 % after the interface modification. We also proved that **TPB** could senhance the interfacial resistance of $CH_3NH_3PbI_3/Au$ interface to favor the reduction of dark current effectively and lead to improvement in open-circuit voltage (V_{oc}). The higher LUMO of **TPB** compared to perovskite could block the electron transport and hence suppress the carrier recombination. However, the **TPB** exhibits deeper HOMO than that of perovskite, which is unfavorable to the hole transportation. The antagonistic contribution of electron-blocking and hole-transport lead to a relatively low PCE (6.71 %). ²¹

Herein, we report the synthesis and characterization of two TPB-based HTMs (TPBS and TPBC) as shown in Fig.1 (a), as well as their application in perovskite solar cells. The device, fabricated with TPBC as HTM, achieves a PCE of 13.10% without doping under AM 1.5G (100 mW/cm²) illumination. This results is comparable with that obtained using the well-known p-70 type doping spiro-OMeTAD. By simply introducing 3,4-ethylene-dioxythiophene (EDOT) and N-ethylcarbazol (NEC) to TPB skeleton, the energy levels of the TPB-core HTMs are tuned to match with the perovskite, which is expected to benefit hole transportation and retard electron transportation. Additionally, the 75 nonplanar configuration of TPB core could be expected to

prevent close contact between the CH₃NH₃PbI₃ and HTMs, which favors to reduce electronic coupling and charge recombination.

The TPB-core HTMs were synthesized through a simple 5 process with low cost based on Wittig reaction according to our previous report.²² The synthetic route for the HTMs is depicted in Fig. 1(b) and experimental details are given in the Electronic Supporting Information (ESI†). The new TPB derivatives (TPBS and TPBC) were fully characterized by ¹H NMR spectroscopy, 10 high resolution mass spectrum, and elemental analysis (ESI†). All the analytical data are consistent with the proposed structures. The two HTMs have good solubility in commonly used solvents, such as dichloromethane, chloroform, tetrahydrofuran and toluene, etc.

The Uv-vis absorption spectra of TPBS, TPBC and TPB in dichloromethane (DCM) are depicted in Fig. 2(a). As shown in the Uv-vis spectra, the absorption band of TPB located in UV range. While for TPBS and TPBC, the extended conjugation length results in a red shift of their absorption band to visible 20 range.²³ This suggests that by introduction of electron-donating moiety (EDOT and NEC) could increase the electron density in the TPB core and hence tunes the HOMO and the LUMO energy levels effectively. Besides, TPBS and TPBC exhibit fine structure (shoulder peak) in DCM due to the substitution on TPB 25 core.

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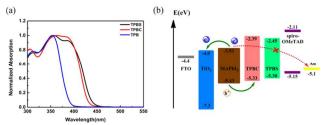


Fig. 2 (a) Normalized absorption spectra of TPBS, TPBC and TPB in DCM; (b) Energy level diagram of the corresponding materials used in perovskite solar cells.

The energy levels of these HTMs (Table 1) are determined using cyclic voltammetry (CV, FigureS1†) in dilute DCM solutions and relative data are listed in Table 1. It is found that both the HOMO levels of TPBS (-5.30 eV) and TPBC (-5.33 eV) are higher than that of TPB core (-5.52 eV), which is caused by 35 the electron-donating effect of EDOT and NEC. Fig. 2 (b) shows the energy level diagram of the corresponding materials in the devices. The HOMO levels for TPBS and TPBC could well match with the valence band of CH₃NH₃PbI₃ (-5.43 eV), which is energetically favorable to the hole transfer at the interface. 40 Furthermore, since the V_{oc} of solar cell is governed by the energy difference between the quasi-Femi levels of the electrons in the TiO₂ and the HOMO of the HTM, TPBS and TPBC could be expected to afford a higher V_{oc} than spiro-OMeTAD for perovskite solar cell because of their deeper HOMO than spiro-45 OMeTAD (-5.15 eV). Besides, the LUMO levels of TPBS (-2.45 eV) and TPBC (-2.39 eV) are higher than the conduction band edge of CH₃NH₃PbI₃ (-3.93 eV), which could block the electron transportation from CH₃NH₃PbI₃ to Au and hence suppress the

Thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC) measurements show that TPBS and TPBC

Table 1 Summary of the electrochemical, and thermal properties of TPBS, TPBC, TPB and spiro-OMeTAD

HTM	HOMO(eV)	LUMO(eV)	$E_{gap}(eV)$	T_g
TPBS	-5.30	-2.45	2.85	104
TPBC	-5.33	-2.39	2.94	105
TPB	-5.52	-2.27	3.25	77^{20}
spiro-OMeTAD	-5.15	-2.11	3.04	125^{10}

have high decomposition temperatures (T_d, 381 °C and 474 °C for 55 **TPBS** and **TPBC**, respectively) and glass transition temperatures $(T_g, 104 \, ^{\circ}\text{C} \text{ and } 105 \, ^{\circ}\text{C} \text{ for TPBS} \text{ and TPBC}, respectively)$ (Figure S2†). This result reveals that these two HTMs have excellent thermal stability to favor their practical application in device.

The hole transporting mobilities of TPBS and TPBC are evaluated by time-of-flight (TOF) measurements (Figure S3†). The hole mobility (μ) is calculated according to the equation $\mu = d^2/Vt_{\tau}$, where d, V and t_{τ} is the film thickness, applied bias voltage and transit time, respectively. It is found that the hole-₆₅ drift mobilities of **TPBS** and **TPBC** are 3.87×10^{-4} and 4.04×10^{-4} cm²·V⁻¹·s⁻¹ respectively at an electric field of 2.3×10⁵ V·cm⁻¹ which are little higher than that of **TPB** $(2.81 \times 10^{-4} \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1})^{18}$ and the well-known small molecular HTM spiro-OMeTAD (2×10⁻⁴cm·V⁻¹·s⁻¹ determined using the same technique²⁴). The 70 high hole mobility indicated these HTMs could have great potential for application in perovskite solar cell.

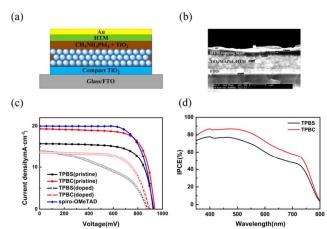


Fig. 3 (a) Diagrammatic representation of photovoltaic device structure; (b) Cross-sectional SEM image of the representative device; (c) J-V 75 curves for the perovskite solar cells fabricated with TPBS, TPBC and spiro-OMeTAD; (d) IPCE spectrum of the cell with TPBS and TPBC.

The photovoltaic performance of perovskite solar cells based on TPBS and TPBC as HTM without doping are evaluated. Device based on p-type doping spiro-OMeTAD as HTM is also 80 fabricated for comparative study. Fig 3 (a) illustrates the structure of the perovskite solar cell. Clear interfaces in the device structure are observed in the cross-sectional scanning electron microscopy (SEM) image (Fig.3 (b)). The photocurrent densityphotovoltage (J-V) curves for FTO/TiO₂/CH₃NH₃PbI₃/HTM/Au 85 solar cells are presented in Fig.3 (c) and related parameters are summarized in Table 2. As expected, the TPBS and TPBC give a higher V_{oc} than spiro-OMeTAD for device, which is commensurate with their deeper HOMO level. The best cell affords an open-circuit voltage (Voc) of 942 mV, a short-circuit 90 current density (J_{sc}) of 19.32 mA/cm² and a fill factor (FF) of

carrier recombination.²¹

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0.72, leading to a PCE of 13.10% under AM 1.5G (100mW·cm⁻²) illumination. This result is comparable to that of spiro-OMeTAD with lithium bis(trifluoromethylsulfonyl)imide (LiTFSI) and 4-tert-butylpyridine (TBP) as dopant. However, devices based on 5 p-doping **TPBS** and **TPBC** exhibit much poorer performance, especially in terms of *FF* and *J*_{sc}. Further investigation about the effect of dopant and optimization of doping process is in process. The IPCE spectrum of the cell with the two HTMs is presented in Fig. 3 (d). An integral photocurrent from the overlap of the IPCE spectrum is 15.2 mA·cm⁻² and 17.9 mA·cm⁻² for **TPBS** and **TPBC**, respectively, basically in agreement with the experimentally obtained *J*_{sc}.

Table 2 *J-V* characteristics of photovoltaic measurements ^a

HTM	$J_{sc}/\mathrm{mA\cdot cm^{-2}}$	V_{oc}/mV	FF	η/ %
TPBS	15.75	932	0.70	10.29
TPBC	19.32	942	0.72	13.10
TPBS(doped)	13.83	879	0.45	5.48
TPBC(doped)	13.72	890	0.66	8.06
spiro-OMeTAD ^b	20.03	921	0.72	13.28

^a Illumination: 100 mW⋅cm⁻² simulated AM 1.5 G solar light; the effective 15 areas of devices are 0.08 cm², ^b doping with LiTFSI and TBP.

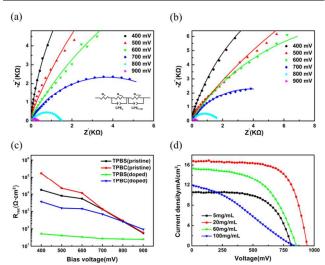


Fig.4(a) Nyquist plots of the device with thin layer **TPBS** in the dark over different forward biases, scattered point: experimental data, solid line: fitted curves; (b) Nyquist plots of the device with thin layer **TPBC** in the 20 dark over different forward biases; (c) Plots of recombination resistance (R_{rec}) vs bias voltages for the devices with **TPBS** and **TPBC**; (d) J-V curves for the perovskite solar cells with different **TPBC** concentration.

The electrochemical impedance spectroscopy (EIS) analysis are carried out, in which the potential bias is applied from 400 to $_{25}$ 900 mV in the dark in the frequency range from 10^5 to 0.1 Hz, as shown in Fig. 4 (a) and Fig. 4 (b). According to simplified transmission line model, the main arc mainly attribute to the combination of the recombination resistance (R_{rec}) and the chemical capacitance of the film (CPE_{μ}) (inset of Fig. 4(a)). 25,26,27 By fitting the Nyquist plots, the bias voltage dependence of R_{rec} for both **TPBS** and **TPBC** is obtained (Fig.4(c)). For both HTMs, the declined R_{rec} are observed as the forward bias voltage increased, which could be ascribed the elevated Fermi level in the mesoscopic TiO₂ under the forward bias (>500mV). The larger 35 R_{rec} of device with **TPBC** than **TPBS** at the same forward bias voltage suggests that **TPBC** as the HTM is superior to **TPBS** in

preventing charge recombination. The larger R_{rec} is supposed to favor improvement of photovoltaic performance, which is in agreement to the results listed in Table 2. As shown in Fig. 4(c), 40 the decreased R_{rec} of device with p-doping **TPBS** and **TPBC** indicates more charge recombination and worse photovoltaic performance, which is in consistent with the results listed in Table 2. The Nyquist plots of the device based on p-type doped **TPBS** and **TPBC** in the dark over different forward biases are 45 shown in Figure S4.

The TPBC is taken as a model HTM to further investigate the influence of HTM deposition amount on the cell performance. As shown in Fig 4 (d) and Table S1[†], the cell fabricated at very low HTM concentration (5 mg/mL) exhibits poor photovoltaic 50 performance. This could be explained by the reason that effective ohmic contact is difficult to be constructed between the metalsemiconductor interface due to the very thin HTM layer.²¹ The best device performance is achieved as the HTM concentration is 20 mg/mL. The worse cell performance at high HTM 55 concentration (>20 mg/mL) could be ascribed to the elevated series resistance caused by the increased HTM layer thickness at higher concentration. At the range of 20-100 mg/mL, the higher V_{oc} at lower HTM concentration is mainly due to the declined dark current as the HTM film thickness decreased.²⁸ This results 60 prove that suitable HTM layer thickness will facilitate the cell performance.

Time-resolved photoluminescence (PL) decay measurement has been conducted to understand the charge transfer (Figure S5) and the PL decay time (τ) are estimated by fitting the data with 65 bi-exponential decay function. 29 Without HTM layer, the TiO₂/CH₃NH₃PbI₃ interface exhibits a long decay time of 102.18 ns. While a significantly reduced τ is observed for device with **TPBS** (34.2ns) and **TPBC** (11.4 ns), which indicates fast charge transfer at the CH₃NH₃PbI₃/HTM interface. 29 Besides, the charge 70 transfer from perovskite to **TPBC** is faster than that of **TPBS**, which suggests that **TPBC** has better hole pumping capacity. 9

In addition, the reproducibility of the cells' performance based on the new HTMs (**TPBS** and **TPBC**) are evaluated by testing photovoltaic parameters of ten cells. Related data are presented in Electronic Supporting Information. (Table S2†) The standard deviations of PCE for the cells with **TPBS** and **TPBC** is 0.36% and 0.26%, respectively. These results are indicating good reproducibility in our study.

Conclusions

In summary, two **TPB**-based HTMs (**TPBS** and **TPBC**) are successfully synthesized by a simple process with low cost. The HOMO and LUMO energy levels of these **TPB** derivatives are effectively tuned to match with perovskite by introducing electron-donating groups asymmetrically to the **TPB** core, which is demonstrated by optical and electrochemical studies. The perovskite solar cell based on **TPBC** as HTM without doping affords an impressive PCE of 13.10%, which is comparable to that obtained employing the well-known p-doping spiro-OMeTAD. This results provide compelling evidence for the first time that small-molecule HTMs with **TPB** core have great potential for the application in high efficient doping-free perovskite solar cell.

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Notes and references

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- 5 † Electronic Supplementary Information (ESI) available: [Experimental; Synthesis of HTMs; ¹H NMR, HRMS, elemental analysis. Additional figures of DSC, TG curve and CV, TOF measurement]. See DOI: 10.1039/b0000000x/
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Graphical Abstract

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