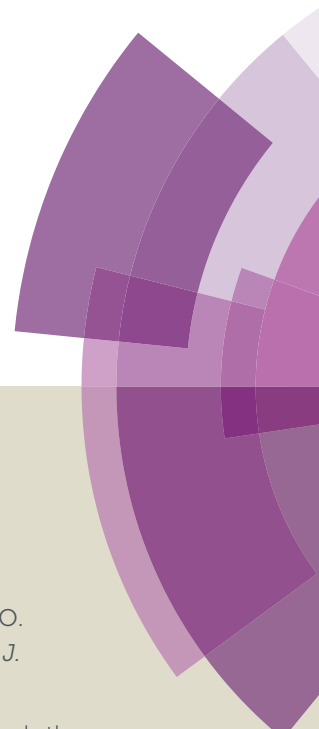


# Journal of Materials Chemistry A

Accepted Manuscript



This article can be cited before page numbers have been issued, to do this please use: S. D. Chavhan, O. Miguel, H. J. Grande, V. Gonzalez-Pedro, R. S. Sánchez, E. M. Barea, I. Mora-Seró and R. TENA-ZAERA, J. Mater. Chem. A, 2014, DOI: 10.1039/C4TA01310G.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

## ARTICLE

# Organo-metal halide perovskite-based solar cells with CuSCN as inorganic hole selective contact

Cite this: DOI: 10.1039/x0xx00000x

Sudam Chavhan,<sup>a</sup> Oscar Miguel,<sup>a</sup> Hans-Jurgen Grande,<sup>a</sup> Victoria Gonzalez-Pedro,<sup>b</sup> Rafael S. Sánchez,<sup>b</sup> Eva M. Barea,<sup>b</sup> Iván Mora-Seró<sup>b,\*</sup> and Ramón Tena-Zaera<sup>a,\*</sup>Received 00th January 2012,  
Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

CuSCN is proposed as a cost-competitive hole selective contact for the emerging organo-metal halide perovskite-based solar cells. The CuSCN films have been deposited by solution casting technique, which has proven to be compatible with the perovskite films, obtaining planar-like heterojunction-based glass/FTO/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN/Au solar cells with power conversion efficiency of 6.4 %. Among the photovoltaic parameters, the fill factor (i.e. 62%) suggests good carrier selectivity and, therefore, efficient functionality of the TiO<sub>2</sub> and CuSCN charge carrier selective contacts. However, the open-circuit voltage (V<sub>OC</sub>), which remains low in comparison to the state of the art of perovskite-based solar cells, appears to be the main limiting parameter. This is attributed to the short diffusion length as determined by the impedance spectroscopy. However, the recombination losses are not only affected by the CuSCN, but also by the perovskite film. Indeed, variations of 20 °C in the thermal annealing of the perovskite films result in changes larger than 200 mV in the V<sub>OC</sub>. Furthermore, a detailed analysis of the quantum efficiency spectra contributes significant insights into the influence of the selective contacts on the photocurrent of the planar heterojunction perovskite solar cells.

## Introduction

Previously investigated as light emitters<sup>1</sup>, the organometal methyl ammonium lead trihalide perovskites (CH<sub>3</sub>NH<sub>3</sub>PbX<sub>3</sub>, where X = I, Cl, Br) have recently emerged as promising photovoltaic materials because of their unique combination of versatile processing and superior opto-electronic properties such as large absorption coefficient and ambipolar charge transport. The extremely rapid evolution of the perovskite-based solar cells during the last 2 years<sup>2-7</sup>, reaching power conversion efficiencies up to 16.2 %<sup>8</sup> makes them a very appealing cost- and performance-competitive emerging technology for photovoltaics.

Different alternatives of methyl ammonium lead trihalide perovskites, such as CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub><sup>2,9</sup>, CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub><sup>3</sup>, CH<sub>3</sub>NH<sub>3</sub>PbBr<sub>3</sub><sup>10,11</sup> and CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Br<sub>x</sub><sup>12</sup> have been investigated for the development of photovoltaic devices. But, additionally to the solar light harvester material, appropriate electron and hole selective contacts are being used in order to minimize the charge carrier recombination<sup>13</sup> and consequent losses in the solar cell performance. Although the viability of using organic electron selective contacts (e.g. (6,6)-phenyl C61-butyric acid methyl ester (PCBM))<sup>14,15</sup> have been recently demonstrated, the best performance perovskite solar cells are based on metal oxides such as ZnO<sup>7</sup> and TiO<sub>2</sub><sup>4,5</sup> or hybrid derivative nanocomposites<sup>6</sup>. In contrast, the research on the

selective hole contacts has been mainly focused on organic compounds such as 2,2(7,7(-tetrakis-(N,N-dimethoxyphenylamine) 9, 9 (-spirobifluorene))) (spiro-OMeTAD)<sup>2,3</sup>, poly-3-hexylthiophene (P3HT)<sup>16</sup>, (poly-[[9-(1-octylonyl)-9H-carbazole-2,7-diyl]-2,5-thiophenediyl-2,1,3-benzothiadiazole-4,7-diyl-2,5-thiophenediyl ] ) (PCDTBT)<sup>17</sup>, poly-triarylamine (PTAA)<sup>12</sup>, and 4-(diethylamino)-benzaldehyde diphenylhydrazone (DEH)<sup>18</sup> and N,N'-dialkyl perylenediimide (PDI)<sup>11</sup>. Among them, the spiro-OMeTAD demonstrated the best device performance<sup>5,6</sup>. However, its relatively high cost may represent a potential bottleneck for the future commercialization of perovskite solar cells. Solution processed inorganic p-type semiconductors appear to be an appealing cost-effective alternative for hole selective contacts. Furthermore, the integration of inorganic films could pave the way to obtain devices insensitive to the ambient environment. Therefore, the integration of inorganic selective contacts may have significant beneficial impact on the production cost and robustness of the perovskite solar cell technology. Nevertheless, up to our best knowledge, there is only one report on the use of inorganic hole selective contacts i.e. CuI<sup>19</sup>. However, CuSCN appears to be a good candidate because of its good performance as hole transport material in different kinds of photovoltaic technologies such as solid-state dye-sensitized<sup>20,21</sup>, semiconductor-sensitized<sup>22,23</sup> and polymer solar cells<sup>24</sup>.

In this paper, the viability of using CuSCN as hole selective contact has been demonstrated in planar heterojunction-based

glass/FTO/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN/Au solar cells with power conversion efficiency of 6.4 %. As far as we know, this is the highest reported efficiency for the perovskite solar cells with inorganic hole selective contacts. Furthermore, the effect of the annealing temperature on the microstructural properties of the perovskite films and its influence on the solar cell performance has been investigated. The characterization of the devices has been carried out by measuring their current-voltage characteristics under dark and simulated sun irradiation, external quantum efficiency spectra and impedance spectroscopy (IS). The high recombination is pointed out as the main origin of the modest photovoltage (< 750 mV) in comparison with other perovskite solar cells, which is the main limiting photovoltaic parameter. The well-balanced charge carrier extraction in the selective contacts is proposed as a key to reach high photocurrents in the planar heterojunction perovskite solar cells. In particular, the influence of the hole selective contact in the solar cell performance is emphasized.

## Experimental section

### Deposition of organo-metal halide perovskite films

Organo-metal halide perovskite films were deposited from a 3 M CH<sub>3</sub>NH<sub>3</sub>I<sup>9</sup> and 1 M PbCl<sub>2</sub> (> 99.9 %, Sigma Aldrich) in anhydrous N, N – Dimethylformamide (DMF) solution<sup>25</sup>. In order to follow the same nomenclature than in previous reports<sup>26</sup>, the obtained films will be hereafter referred as CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> although unambiguous proof about the chlorine content cannot be provided. The CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> films were processed on glass/SnO<sub>2</sub>:F/TiO<sub>2</sub> substrates by spin coating at 700 rpm for 60 s then at 2000 rpm for 60 s, and annealed on a hot plate at 90, 100, 110, and 120 °C for 3 h. The deposition and annealing treatments of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> films were carried out in an Argon-filled glove-box.

### Thin film characterization techniques

The optical transmittance and reflectance of the glass/SnO<sub>2</sub>:F/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> samples were measured at room temperature with a Jascow V-570 spectrophotometer fitted with an integrating sphere, from 300 to 900 nm. The morphology and structural properties of the films were analyzed using a ULTRA plus ZEISS field emission scanning electron microscope (FE-SEM) and a Bruker AXS-D8 Advance X-ray diffractometer (XRD) using Cu K $\alpha$  radiation.

The PL spectra of the films were registered by using a spectrophotometer based on a CCD (Andor i-DUS DV420A-OE) coupled with a spectrograph as a diffraction grating (Newport 77400), Commercial blue (405 nm, 84 mW·cm<sup>-2</sup>) and red laser diodes (650 nm, 319 mW·cm<sup>-2</sup>) were used as excitation sources, whose power was adjusted by means of a mechanical pinhole.

### Solar cell preparation

Glass/SnO<sub>2</sub>:F (FTO) substrates were cleaned with soap solution and ethanol in an ultrasound bath for 30 min. In order to avoid the direct contact between FTO and hole transporting material (HTM), a ca. 60 nm thick dense TiO<sub>2</sub> layer is deposited onto the FTO by spray pyrolysis technique. Briefly, 12 g of titanium isopropoxide (IV) (Aldrich) and 12 g of acetylacetone (Aldrich) were added in to 20 ml absolute ethanol; during the deposition of TiO<sub>2</sub>, substrates temperature kept at 450 °C and subsequently calcined at 450 °C in for 30 min.

After the CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> deposition and thermal annealing, the CuSCN film was deposited by drop-casting from a saturated solution of CuSCN in propylsulfide. The solution was pumped at a constant rate of 27  $\mu$ L min<sup>-1</sup> using a syringe pump. During the CuSCN

deposition, the glass/SnO<sub>2</sub>:F/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> samples were kept at 85 °C on a hot plate. In order to control the CuSCN film thickness, 18  $\mu$ L of solution were allowed to drop per cm<sup>2</sup> of the sample. An array of gold dots, 0.35 cm in diameter, was deposited as the back contact by thermal evaporation under > 5x 10<sup>-6</sup> Torr vacuum pressure using NANO38 equipment.

### Solar Cell Characterization

The current-voltage characteristic of the cells were measured using a xenon arc lamp simulator (Sun 2000, ABET Technologies) with an AM 1.5 G spectral filter, and the intensity was adjusted to provide 1 sun (100 mW/cm<sup>2</sup>) using a calibrated silicon solar cell. The active area was defined to be 0.07 cm<sup>2</sup> by using an opaque mask. The spectral response of the device photocurrent was measured using a computerized home-built setup (xenon arc lamp, chopper, monochromator, lock-in amplifier). The external quantum efficiency (eQE) was calculated using a standard Si photodiode. The impedance spectroscopy data was recorded with a FRA-equipped PGSTAT-30 from Autolab and the measurements were carried out under illumination conditions at forward bias, applying a 20 mV AC sinusoidal signal, ranging between 400 kHz and 0.05 Hz, over the constant applied bias.

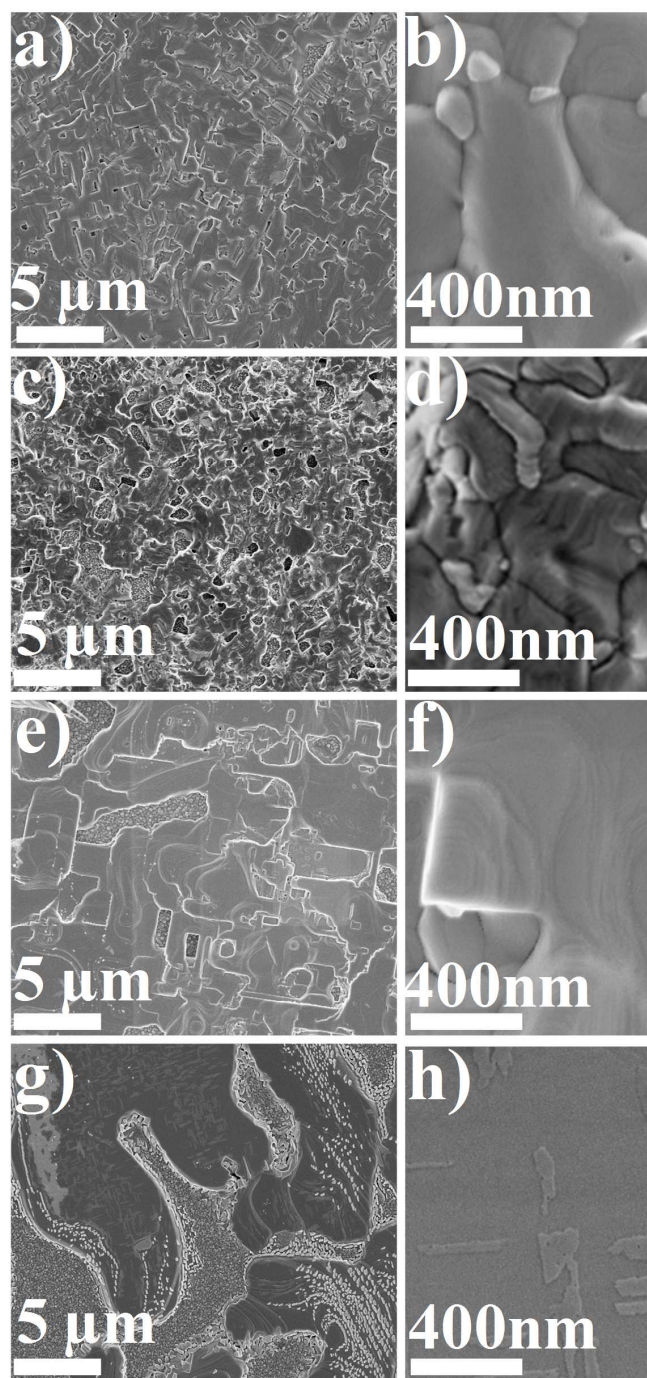
## Results and discussion

### Microstructural properties of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> films

At first, the morphology of the CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> films and its evolution with the annealing temperature was investigated by FE-SEM. Figure 1 shows representative top-view FE-SEM micrographs of the perovskite films annealed at 90, 100, 110, and 120 °C for 3 h in a glove box. Films with a relatively low density of pin-holes were obtained for the lowest annealing temperature. However, the increase of the temperature induced a transition from continuous-like films to rather discrete islands. Indeed, large areas of naked TiO<sub>2</sub> can be detected for samples annealed at 120 °C. The formation of large grains as the annealing temperature increases is emphasized in the high magnification top-view FE-SEM micrographs. It is noted that only a partial view of a representative single grain, which is larger than the micrograph area is shown for the sample annealed at 120 °C. Although the morphology variation trend versus the temperature is similar to that previously reported by Snaith's group<sup>25</sup>, the present films seem to exhibit stronger sensitivity to the temperature. Interestingly, significantly higher surface coverage is also here detected. Indeed, we obtained perovskite films with much lower coverage by using CH<sub>3</sub>NH<sub>3</sub>I synthesized by the same route than in reference<sup>25</sup> (Figure S1, Supplementary Information).

The structural properties were analyzed by XRD in Bragg-Brentano configuration. Figure 2 displays the XRD patterns of the glass/FTO/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> samples annealed at different temperatures. The tetragonal CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> phase<sup>1</sup> can be confirmed. The relatively strong intensity of the peaks versus those from the substrate, which are almost no detectable in linear scale, indicates the good crystallinity of the deposited films. Furthermore, the intensity of the peaks increases significantly with the annealing temperature. In contrast, the full width at half maximum (FWHM) of the peaks is constant regardless of the annealing temperature (Figure S2a, Supporting Information). This finding suggests the instrumental broadening as the main contribution to the peak width.

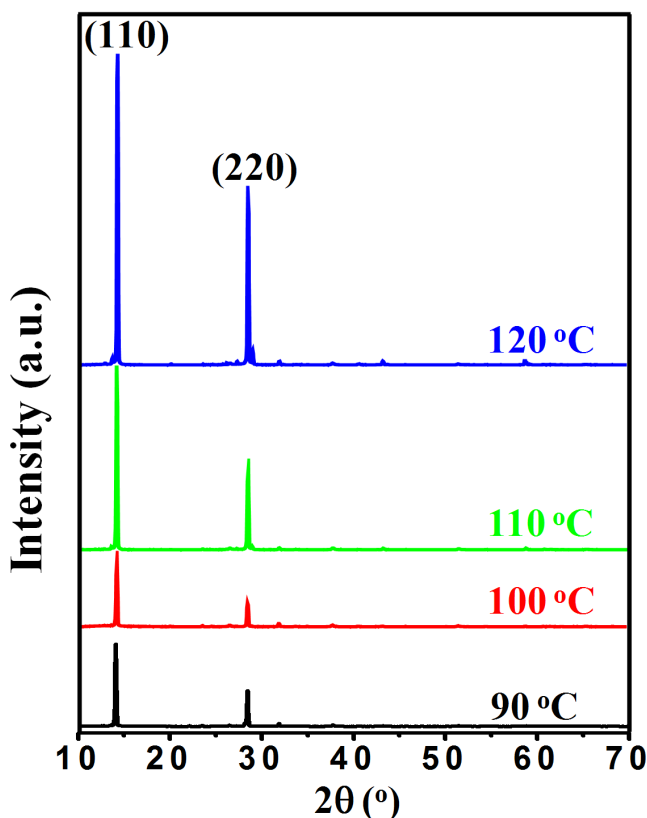




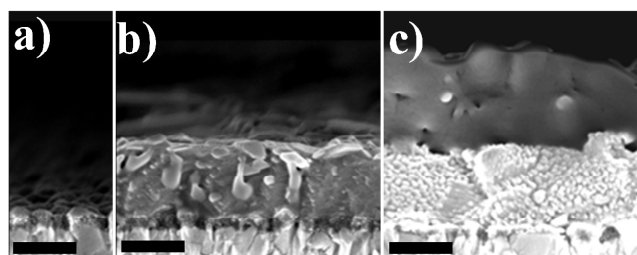
**Figure 1:** Top-view FE-SEM micrographs of the  $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  films annealed at: a & b) 90, c & d) 100, e & f) 110, and g & h) 120 °C for 3 hours in an Argon-filled glove box.

Therefore, it is not possible to monitor, by XRD, the evolution of the crystallite size. Additionally to the  $\text{CH}_3\text{NH}_3\text{PbI}_3$  phase-related features, a minor peak occurring at  $2\theta \sim 13.55^\circ$ , which is characteristic of films deposited from  $\text{PbCl}_2$ <sup>3,27</sup>, is also detected (Figure S2b, Supplementary Information). Furthermore, a very weak intensity feature is observed at  $2\theta \sim 12.75^\circ$ , suggesting the presence of minor traces of  $\text{PbI}_2$  phase in the films. Interestingly, the intensity of the peak increases with the

annealing temperature. Very recently, Dualeh et al.<sup>27</sup> reported similar finding and suggested the relatively fast sublimation of the organic species as a potential origin.



**Figure 2:** X-ray diffraction pattern of  $\text{FTO}/\text{TiO}_{2\text{sp}}/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  samples annealed at different temperature for 3 hours in an Argon-filled glove box.

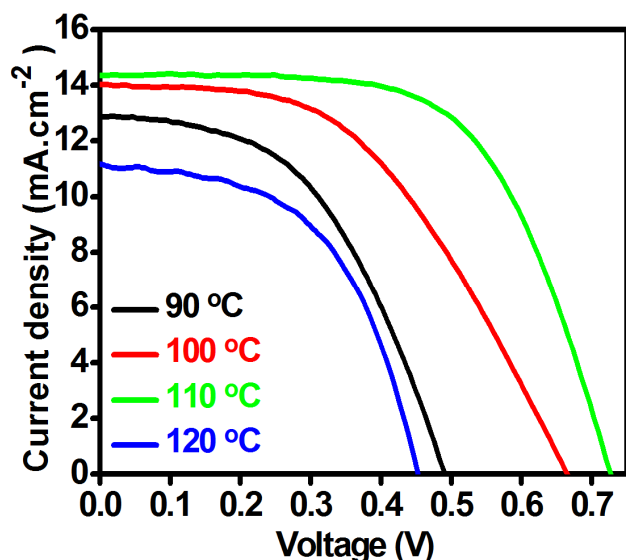


**Figure 3:** Cross section FE-SEM micrographs of a)  $\text{FTO}/\text{TiO}_{2\text{sp}}$ , b)  $\text{FTO}/\text{TiO}_{2\text{sp}}/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$ , and c)  $\text{FTO}/\text{TiO}_{2\text{sp}}/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x/\text{CuSCN}$  (400 nm scale).

### Solar cells

In order to evaluate the viability of  $\text{CuSCN}$  as a hole selective contact in perovskite solar cells,  $\text{CuSCN}$  thin films were deposited on the  $\text{glass}/\text{SnO}_2:\text{F}/\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  samples. As a result, a hybrid multilayer heterostructure (i.e.  $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x/\text{CuSCN}$ ) was obtained by a three-step deposition process. Figure 3 shows the FE-SEM micrographs of the samples after each step, monitoring the sequential deposition of compact  $\text{TiO}_2$ , perovskite and  $\text{CuSCN}$  films with mean thickness of 60, 400 and

500 nm, respectively. As previously reported<sup>5</sup>, significant deviations in the local thickness of the perovskite film were detected (Figure S3, Supporting Information). Although these local variations made difficult accurate comparative analyses for the perovskite thickness before and after the CuSCN deposition, no significant changes in the mean value were detected. Thus, it can be concluded that there is not a significant dissolution of the perovskite during the CuSCN deposition solution step.



**Figure 4:** Current density-voltage curves of solar cells, based on  $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  films annealed at different temperatures, under illumination of  $100 \text{ mW/cm}^2$  simulated sun irradiation (1.5 AM)

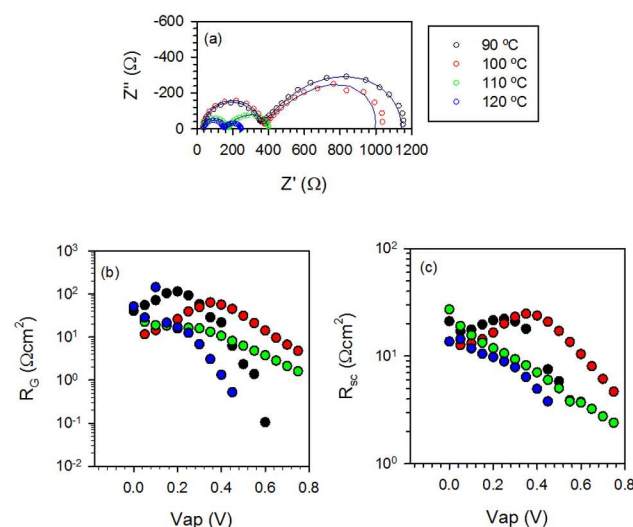
**Table 1:** Photovoltaic parameters of devices depicted in Figure 4 and integrated photocurrent from IPCE.

Annealing Temp (°C)	$J_{sc}$ (IPCE) (mA/cm <sup>2</sup> )	$J_{sc}$ (mA/cm <sup>2</sup> )	$V_{oc}$ (mV)	FF (%)	$\eta$ (%)
90	13.04	12.9	491	49.0	3.1
100	14.27	14.0	665	48.1	4.5
110	18.53	14.4	727	61.7	6.4
120	13.80	11.1	453	53.8	2.7

Figure 4 shows the current density-voltage curves recorded for  $\text{TiO}_2/\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x/\text{CuSCN}$  solar cells, based on perovskite films annealed at different temperatures, under illumination of  $100 \text{ mW/cm}^2$  simulated sun-light (1.5 AM). The solar cell performance is strongly influenced by the annealing temperature of the perovskite film, i.e. influenced by the effect of the crystallinity and continuity of the perovskite film. The solar cell parameters extracted from the J-V curves are summarized in Table 1. From Table 1, remarkable influence of the annealing temperature on the FF and  $V_{oc}$  values is noticed. The enhancement in solar cell performance is particularly dramatic for annealing temperatures of  $110^\circ\text{C}$ , leading to 6.4 % power conversion efficiency for the champion cell. To our best knowledge, this is the first efficiency reported for perovskite based solar cell employing CuSCN as hole selective contact, and the highest one obtained for inorganic semiconductor-based hole selective contact in perovskite solar cells<sup>19</sup>. A histogram of the power conversion efficiency for 168 devices prepared from  $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  films annealed at  $110^\circ\text{C}$  is shown in Figure S4

(Supporting Information). Although obtaining less performing devices, preliminary results also pointed out the viability of using CuSCN as selective hole contact in  $\text{CH}_3\text{NH}_3\text{PbI}_3$  (i.e. processed in absence from  $\text{PbI}_2$  and  $\text{CH}_3\text{NH}_3\text{I}$  solution) based solar cells (Figure S5, Supplementary Information). Therefore, the solution processed CuSCN can be suggested as a cost effective competitor to spiro-OMeTAD in perovskite solar cells. However, the reached  $V_{oc}$ , which is much lower (i.e.  $> 200 \text{ mV}$  below) than for the spiro-OMeTAD-based solar cells with similar device architecture<sup>28</sup>, appears to be the main limitation. This limitation, even higher for other inorganic hole selective contacts<sup>19</sup>, has to be overcome in order to represent a full appealing alternative in the preparation of perovskite solar cells.

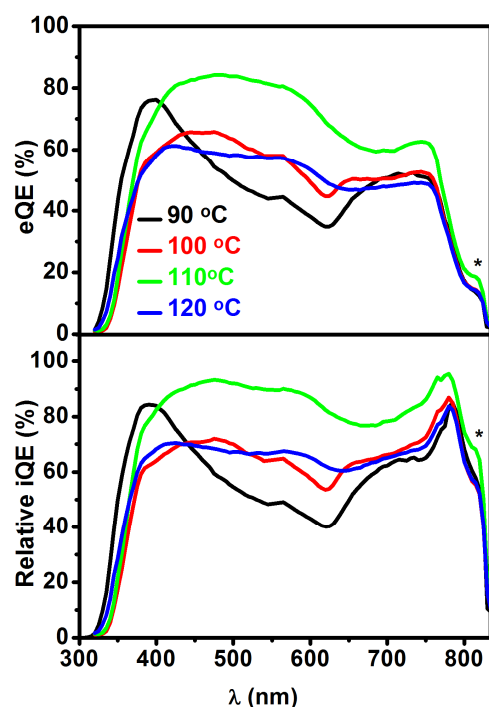
To better understand the origin of the  $V_{oc}$  limitation, impedance spectroscopy (IS) analysis was performed. This technique allows decoupling processes, which take place at different characteristic times at the cell working conditions, and allows a systematic characterization focused on the device optimization<sup>28-32</sup>. Examples of the impedance pattern obtained for tested devices at applied voltage  $V_{app} = 0.35 \text{ V}$  are plotted in Figure 5a. Nyquist plots, imaginary vs. real part of impedance  $Z''$  and  $Z'$  respectively, present two characteristic features regardless of the annealing temperatures. On one hand, the arc at high frequencies ( $hf$ ), at low  $Z'$ , is related with the selective contacts, with transport in CuSCN layer<sup>13, 33, 34</sup> and charge transfer at the interfaces between the selective contacts and perovskite<sup>13</sup>. In this sense,  $hf$  arc is also influenced by perovskite absorbing layer. On the other hand, at low frequencies ( $lf$ ) a Gerischer pattern is clearly identified. Gerischer pattern is formed by a straight line with  $45^\circ$  slope followed by an arc laying below the prolongation of the straight line<sup>30</sup>, (Figure 5a). Gerischer pattern derives from the classical transmission line pattern that becomes Gerischer when diffusion length is shorter than the sample thickness<sup>28,30</sup>. This fact indicates that the cell efficiency, and especially the  $V_{oc}$ , is limited by the low diffusion length, shorter than the perovskite thickness.



**Figure 5:** a) Nyquist plots for solar cells, annealed at different temperature, at applied voltage  $V_{app} = 0.35 \text{ V}$ . Plot of recombination resistance ( $R_G$ ) vs. applied voltage and c) plot of the resistance associated with the selective contacts ( $R_{sc}$ ) vs. applied voltage.

Despite the Gerischer pattern prevents the estimation of diffusion parameters, important qualitative information about the recombination in the solar cell is contained in the Gerischer

resistance,  $R_G$ . Indeed,  $R_G$  is the recombination resistance, inversely proportional to the recombination rate, but it is just for the part of the sample from which charge can be collected. The obtained impedance pattern can be fitted using an R-C circuit for the  $hf$  in series with a Gerischer for the  $lf$  (Figure S6, Supplementary Info). Since a capacitor and a Gerischer are ideal elements, Constant Phase Elements (CPE) and Gerischer plus R-C, were used in order to obtain a better fit, see Figure 5a. The recombination can be qualitatively analyzed from  $R_G$ , Figure 5b. Reduction of recombination resistance at high potential, for cells annealed at 90 and 120 °C, indicates a recombination increase in these samples, which is in good agreement with the reduction of  $V_{oc}$  observed. On the other hand, Figure 5c shows  $R_{sc}$ , resistance associated with the selective contacts, obtained from  $hf$  arc. This resistance is related with the charge transfer resistance between the perovskite and selective contacts<sup>13</sup> and transport resistance in the CuSCN layer<sup>13, 33, 34</sup>.  $R_{sc}$  acts as a series resistance and decreases as a function of the annealing temperature, explaining the improvement in FF as annealing temperature increased until 110 °C. Nevertheless, additional parameters seem to contribute to the decrease of the FF in devices based on the perovskite films annealed at 120 °C. The FF drop may be mainly due to the large voids in the perovskite film (Figure 2g) and the consequent contact between the electron and hole selective contacts. It is worth to mention that good reproducibility of impedance results were found, by measuring several solar cells prepared at the same conditions, see Figure S7 (Supplementary Information).



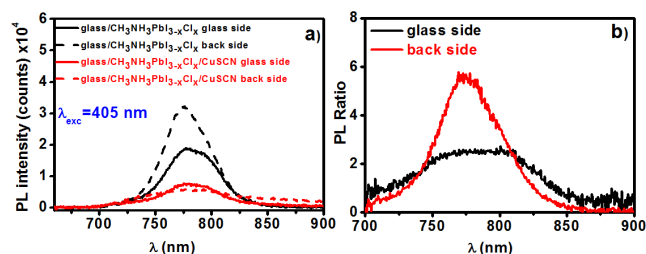
**Figure 6:** a) eQE and b) relative iQE of solar cells based on different annealing temperature of  $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$  films. \* The feature observed at  $\sim 815$  nm is due to an artifact in the eQE estimation (see Figure S8, Supporting Information). As the absorption of the glass/ $\text{SnO}_2\text{:F/TiO}_2\text{:CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x\text{:CuSCN}$  samples is negligible in this wavelength, the feature is stronger in the relative iQE values spectra.

Although less significant than  $V_{oc}$  deviations, the  $J_{sc}$  of the present perovskite solar cells is also lower than those reached in planar

heterojunction based perovskite solar cells with spiro-OMeTAD as hole selective contact<sup>25,28</sup>. To gain a further insight into the  $J_{sc}$  limiting mechanism, the photoresponse spectra were measured in the 300-850 nm range. Figure 6a shows the external quantum efficiency (eQE) spectra of representative solar cells based on perovskite films annealed at different temperatures. In general, an improvement in photoresponse for increasing temperatures, peaking at 110 °C annealing temperature, is noticed. Furthermore, the integrated currents (included in Table 1) are in reasonable good agreement with those obtained from the J-V measurements. The largest deviation is detected for solar cells with the best performance (i.e. based on perovskite films annealed at 110 °C), suggesting better performance under low irradiation. In general, the eQE for wavelengths close to the perovskite absorption edge is relatively lower than for short wavelengths. This is mainly due to the optical losses (i.e. lower absorptance for wavelengths close to the absorption edge, Figure S9, Supplementary Information). Indeed, the relative internal quantum efficiency (riQE) -i.e. eQE versus the absorptance before the Au contact deposition- spectra show similar values for the blue and red photons, or slightly higher for the latter (Figure 6a). It is noted that the peak centered at  $\lambda \sim 780$  nm may be overestimated because the reflections in the back contact could enhance the light harvesting in the solar cells in comparison to the glass/ $\text{SnO}_2\text{:F/TiO}_2\text{:CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x\text{:CuSCN}$  samples, for which we performed the optical characterization (Figure S9). Similar peak has been recently reported by Edri et al. for meso-superstructured perovskite solar cells based on  $\text{CH}_3\text{NH}_3\text{PbBr}_3$  and  $\text{CH}_3\text{NH}_3\text{PbBr}_{3-x}\text{Cl}_x$ <sup>35</sup>. A valley-like shape is clearly detected in the riQE spectra, pointing out significant losses in the collection of charges generated from green and yellow photons (i.e. especially for devices with perovskite films annealed at  $\leq 100$  °C). Taking into account the variation of the penetration depth as a function of the photon wavelength, it can be assumed that blue photons are absorbed relatively close to the  $\text{TiO}_2$ /perovskite. Furthermore, the decrease of the absorptance detected for wavelengths larger than 600 nm (Figure S9) seems to suggest that the penetration depth of the red photons is comparable to the perovskite film. Therefore, the absorption of the red photons is expected to occur along all the perovskite film. In contrast, the green and yellow photons may not reach the proximities of the perovskite/CuSCN interface. Based on this assumption, a tentative model is here proposed. The photogenerated electrons from the blue photons close to  $\text{TiO}_2$ /perovskite interface may be quickly injected into the  $\text{TiO}_2$  selective contact. The holes are then quite efficiently transported along the perovskite film until reaching the CuSCN, as the electron density is reduced (and consequently recombination probability) due to an efficient electron extraction. However, the charge collection efficiency decreases when some charge carriers are also generated in the bulk of the perovskite. Interestingly, better collection efficiency occurred when some electron-hole pairs are generated close to the perovskite/CuSCN interface. Thus, the fast hole injection to the CuSCN seems to be crucial, enhancing the collection of carriers photogenerated near to its interface, but also of those generated in the bulk of the perovskite film. Therefore, additionally to the ambipolar conduction of the perovskite film, well-balanced extraction in the selective contacts plays a vital role to reach high performance planar heterojunction perovskite solar cells. Edri et al.<sup>36,37</sup> reported recently a very similar charge collection profile, by using electron beam-induced current (EBIC) technique, in planar heterojunction  $\text{CH}_3\text{NH}_3\text{PbBr}_{3-x}\text{Cl}_x$ -based solar cells. The particular two-peak EBIC profile was presented as the first direct experimental evidence of the p-i-n mode of operation for solar cells with  $\text{CH}_3\text{NH}_3\text{PbBr}_{3-x}\text{Cl}_x$  thickness comparable to the free charge carrier diffusion length<sup>37</sup>. Potentially complementary insights are provided here from less specific experimental data such as the



quantum efficiency spectra. Therefore, the present results and discussion may be of significant interest and utility for the perovskite solar cell community



**Figure 7:** a) Photoluminescence spectra of glass/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> films and glass/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN samples, b) ratio of photoluminescence yield between glass/perovskite and glass/perovskite/CuSCN for the different excitation configurations.

Particular photoluminescence measurements –by using blue excitation wavelength (405 nm) and different excitation configurations (i.e. illumination thorough the glass and through the CuSCN film, respectively)- were performed in order to gain a further insight into the relevance of the perovskite/CuSCN interface in the electron-hole dynamics. Figure 7a displays the PL spectra for glass/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> and glass/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN samples excited by the glass and CuSCN side, respectively. In general, samples with CuSCN exhibited relatively lower PL emission. However, the PL quenching (Figure 7b) is stronger when the samples are excited through the CuSCN film. Taking into account the relative short penetration depth of the blue photons, the better hole collection when the electron-hole pair is generated close to the perovskite/CuSCN interface is confirmed. As further evidences, weak PL quenching and no significant effect of the illumination side was detected for red excitation wavelength (i.e. 650 nm, Figure S10, Supplementary information.), for which electron-hole pairs are generated along whole perovskite film irrespective of the illumination side.

All in all, improving the optoelectronic properties of the perovskite films (i.e. free charge carrier transport) and the perovskite/CuSCN interface (i.e. hole transfer) are suggested to increase the power conversion efficiency –by means of an enhancement of the free charge carrier diffusion length- of TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN solar cells. As not significant sensitivity of the solar cell performance versus the CuSCN thickness was detected (Figure S11, Supporting Information), the hole transport in the CuSCN film does not seem to be a major limiting factor. However, further studies are needed.

## Conclusions

The viability of using solution processed CuSCN films as hole selective contacts in perovskite solar cells is demonstrated. Planar heterojunction-based glass/FTO/TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub>/CuSCN/Au solar cells with power conversion efficiency of 6.4 % have been obtained. The characterization of the devices by means of impedance spectroscopy pointed out the low diffusion length as the main origin of the modest photovoltage (< 750 mV), which is the main limiting photovoltaic parameter. However, the recombination losses are not only affected by the CuSCN, but also by the perovskite film. Furthermore, the detailed analysis of the quantum efficiency spectra suggest that, the well-balanced charge carrier extraction in the selective contacts is a crucial parameter to reach high photocurrents in

planar heterojunction perovskite solar cells. All in all, additionally of proposing an alternative, cost-competitive and robust hole selective contact, significant insights into the physical mechanisms involved in perovskite-based solar cells have been provided. The influence of CuSCN layer has been emphasized, paving the way for further progress in perovskite solar cells with inorganic hole selective contacts.

*Note added in proof:* During the revision of this paper, three highly relevant articles to this topic were published: S. Ito, S. Tanaka, H. Vahlman, H. Nishino, K. Manabe, P. Lund, "Carbon-Double-Bond-Free Printed Solar Cells from TiO<sub>2</sub>/CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>/CuSCN/Au: Structural Control and Photoaging Effects" *ChemPhysChem*, 2014, 15, 1194-1200. A.S. Subbiah, A. Halder, S. Ghosh, N. Mahuli, G. Hodes, S.K. Sarkar "Inorganic Hole Conducting Layers for Perovskite based Solar Cells" *J. Phys. Chem. Lett.*, 2014, DOI: 10.1021/jz500645n. (8) P. Qin, S. Tanaka, S. Ito, N. Tetreault, K. Manabe, H. Nishino, M.K. Nazeeruddin, M. Grätzel, "Inorganic hole conductor-based lead halide perovskite solar cells with 12.4% conversion efficiency" *Nat. Commun.* 2014, 5:3834 doi: 10.1038/ncomms4834.

## Acknowledgements

Financial support by the European Union (ORION CP-IP 229036-2) and Universitat Jaume I project 12I361.01/1 is gratefully acknowledged. R.T.-Z. acknowledges the support of the Program "Ramon y Cajal" of the MICINN.

## Notes and references

<sup>a</sup> Energy Division, IK4-CIDETEC, Parque Tecnológico de San Sebastián, Paseo Miramón 196, Donostia-San Sebastián 20009, Spain. rtena@cidetec.es

<sup>b</sup> Photovoltaics and Optoelectronic Devices Group, Departament de Física, Universitat Jaume I, 12071 Castello, Spain. sero@uji.es

† Footnotes should appear here. These might include comments relevant to but not central to the matter under discussion, limited experimental and spectral data, and crystallographic data.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/b000000x/

## References

- 1 K. Liang, D.B. Mitzi, M.T. Prikas, *Chem. Mater.*, 1998, **10**, 403.
- 2 H.-S. Kim, C.-R. Lee, J.-H. Im, K.-B. Lee, T. Moehl, A. Marchioro, S.-J. Moon, R.H.-Baker, J.-H. Yum, J. E. Moser, M. Grätzel, N.-G. Park, *Sci. Rep.*, 2012, **2**, 591.
- 3 M. M. Lee, J. Teuscher, T. Miyasaka, T. N. Murakami, H. J. Snaith, *Science*, 2012, **338**, 643.
- 4 J. Burschka, N. Pellet, S.-J. Moon, R.H.-Baker, P. Gao, M. K. Nazeeruddin, M. Grätzel, *Nature*, 2013, **499**, 316.
- 5 M. Liu, M.B. Johnston, H.J. Snaith, *Nature*, 2013, **501**, 395.
- 6 J. T.-W. Wang, J. M. Ball, E. M. Barea, A. Abate, J. A. Alexander-Webber, J. Huang, M. Saliba, I. Mora-Sero, J. Bisquert, H.J. Snaith, R. J. Nicholas, *Nano Lett.*, 2014, **14**, 724.
- 7 D. Liu, T. L. Kelly, *Nat. Photonics*, 2014, **8**, 133.
- 8 [http://www.nrel.gov/ncpv/images/efficiency\\_chart.jpg](http://www.nrel.gov/ncpv/images/efficiency_chart.jpg).

- 9 J.-H. Im, C.-R. Lee, J.-W. Lee, S.-W. Park, N.-G. Park, *Nanoscale*, 2011, **3**, 4088.
- 10 A.Kojima, K. Teshima, Y. Shirai, T. Miyasaka, *J. Am. Chem. Soc.*, 2009, **131**, 6050.
- 11 E. Edri, S. Kirmayer, D. Cahen, G. Hodes, *J. Phy. Chem. Lett.* 2013, **4**, 897.
- 12 J.H. Noh, S. H. Im, J. H. Heo, T. N. Mandal, S. I .Seok, *Nano. Lett.*, 2013, **13**, 1764.
- 13 E. J. Juarez-Perez, M. Wußler, F. Fabregat-Santiago, K. Lakus-Wollny, E. Mankel, T. Mayer, W. Jaegermann, I. Mora-Sero, *J. Phy.Chem. Lett.*, 2014, **5**, 680.
- 14 O. Malinkiewicz, A. Yella, Y. H. Lee, G. M. Espallargas, M. Grätzel, M.K. Nazeeruddin, H. J. Bolink, *Nat. Photonics*, 2014, **8**, 128.
- 15 S. Sun, T. C. Sum, Y. M. Lam, *Energy. Environ. Sci.*, 2014, **7**, 399.
- 16 B. Conings, L. Baeten, C. D. Dobbelaere, J. D'Haen, J. Manca, H.-G. Boyen, *Adv. Mat.* 2013, doi: 10.1002/adma.201304803.
- 17 J. H. Heo, S. H. Im, J. H. Noh, T. N. Mandal, C.-S. Lim, J. A. Chang, Y. H. Lee, H.-J. Kim, A. Sarkar, M. K. Nazeeruddin, M. Grätzel, S. I . Seok, *Nat. Photonics*, 2013, **7**, 486.
- 18 D. Bi, L. Yang, G. Boschloo, A. Hagfeldt, E.M.J. Johansson, *J. Phys. Chem. Lett.*, 2013, **4**, 1532.
- 19 J. A. Christians, R.C.M. Fung, P.V. Kamat, *J. Am. Chem. Soc.*, 2014, **136**, 758.
- 20 E.V.A. Premalal, N. Dematage, G. R. R.A. Kumara, R.M.G. Rajapakse, M. Shimomura, K. Murakami, A. Kohn, *J. Power Sources*, 2012, **203**, 288.
- 21 B. O'Regan, D. T. Schwartz, *Chem. Mater.*, 1995, **7**, 1349.
- 22 C.L. Clement, R. Tena-Zaera, M. A. Ryan, G. Hodes, *Adv. Mater.*, 2005, **17**, 1512.
- 23 Y. Itzhak, O. Niitsoo, M. Page, G. Hodes, *J. Phy. Chem. C.*, 2009, **113**, 4254.
- 24 C. Chappaz-Gillot, S. Berson, R. Salazar, B. Lechene, D. Aldakov, V. Delaye, S. Guillerez, V. Ivanova, *Solar Ener. Mat. Sol. Cells*, 2014, **120**, 163.
- 25 G. E. Eperon, V. M. Burlakov, P. Docampo, A. Goriely, H. J. Snaith, *Adv.Funct. Mater.* 2014, **24**, 151.
- 26 J. M. Ball, M. M. Lee, A. Hey, H.J. Snaith, *Energy Environ Sci.*, 2013, **6**, 1739.
- 27 A. Dualah, N. Tetreault, T. Moehl, P. Gao, M. K. Nazeeruddin, M. Grätzel, *Adv. Funct. Mater.*, 2014, doi: 10.1002/adfm.201304022.
- 28 V. Gonzalez-Pedro, E.J. Juarez -Perez, W.-S. Arsyad, E. M. Barea, F. Fabregat-Santiago, I. Mora-Sero, J. Bisquert, *Nano. Lett.*, 2014, **14**, 888.
- 29 H.-S.Kim, I. Mora-Sero, V. Gonzalez-Pedro, F. Fabregat-Santiago, E. J. Juarez-Perez, N.-G. Park, J. Bisquert, *Nat. Commun.*, 2013, **4**, 2242.
- 30 J. Bisquert, I. Mora-Sero, F. Fabregat-Santiago, *Chem. Electro.Chem.*, 2014, **1**, 289.
- 31 H.-S. Kim, J.-W. Lee, N. Yantara, P. P. Boix, S. A. Kulkarni, S. Mhaisalkar, M. Grätzel, N.-G. Park, *Nano Lett.*, 2013, **16**, 2412.
- 32 A. Dualah, T. Moehl, N. Tetreault, J. Teuscher, P. Gao, M. K. Nazeeruddin, M. Grätzel, *ACS nano*, 2013, **8**, 362.
- 33 P. P. Boix, G. Larramona, A. Jacob, B. Delatouche, I. Mora-Seró, J. Bisquert, *J. Phys. Chem. C*, 2012, **116**, 1579.
- 34 I. Mora-Seró, S. Giménez, F. Fabregat-Santiago, E. Azaceta, R. Tena-Zaera, J. Bisquert, *Phys. Chem. Chem. Phys.*, 2011, **13**, 7162.
- 35 E. Edri, S. Kirmayer, M. Kulbak, G. Hodes, D. Cahen, *J. Phys. Chem. Lett.*, 2014, **5**, 429.
- 36 E. Edri, S. Kirmayer, A. Henning, S. Mukhopadhyay, K. Gartsman, Y. Rosenwaks, G. Hodes, D. Cahen, *Nano Lett.*, 2014, **14**, 1000.
- 37 E. Edri, S. Kirmayer, S. Mukhopadhyay, K. Gartsman, G. Hodes, D. Cahen, *Nat. Commun.* 2014, **5**, 3461 doi: 10.1038/ncomms4461



## TOC:

The viability of using solution-processed CuSCN films as inorganic hole selective contacts in perovskite solar cells is demonstrated, by reaching power conversion efficiency of 6.4 % in planar heterojunction-based devices.

