# Used Physics

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| Variable | Physical quantity | Physical unit |
|  | Amount of substance | mol |
|  | Molar change of the amount of substance | mol.s‑1 |
|  | Volume | m3 |
|  | Molar volume | m3.mol-1 |
|  | Mole fraction | mol.mol‑1 |
|  | Molarity (molar concentration) | mol.m‑3 |
|  | Molality | mol.kg‑1 |
|  | Mole fraction based activity | 1 |
|  | Change of heat energy/heat flow | J.s‑1 |
|  | Molar enthalpy of the formation | J.mol‑1 |
|  | Molar entropy of the formation | J.K-1.mol‑1 |
|  | Gas constant 8.3144621(75) | J.K‑1.mol‑1 |
|  | Temperature | K |
|  | Forward rate | s‑1 |
|  | Products to reactants ratio at equilibrium | 1 |
|  | Activity coefficients of substance | 1 |
|  | Stoichiometry coefficient | 1 |
|  | Partial pressure of substance A | Pa |
|  | Electrical charge of particle | C |
|  | Permeability of membrane for substance A | mol.s‑1 |
|  | Ratio of specific form of side s inside general form G | 1 |
|  | Change of volume or volumetric flow | m3.s‑1 |
|  | Pressure | Pa |
|  | Osmotic permeability of membrane | m3.Pa‑1.s‑1 |
|  | Relative heat energy/relative enthalpy | J |
|  | Mass | kg |
|  | Change of mass/mass flow | kg.s-1 |
|  | Specific heat of the mass | J.kg‑1.K‑1 |
|  | Heat conductance of heat convection | J.K‑1.s‑1 |
|  | Hydraulic compliance of space with elastic walls | m3.Pa‑1 |
|  | Gravitational acceleration 9.8 | m.s‑1 |
|  | Density of the mass | kg.m‑3 |
|  | Vertical height between two points | m |
|  | Hydraulic conductance | m3.Pa‑1.s‑1 |
|  | Pumping effect | 1 |
|  | Hydraulic inertance of hydraulic inertia | Pa.s2.m‑3 |
|  | Size of population=number of members | 1 |
|  | Change of population | s‑1 |
|  | Change of population per one member | s‑1 |

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| Chemical equations | |
|  | Amount of substance A |
|  | Mole fraction of the substance A |
|  | Mole fraction based activity of substance A |
|  | Chemical reaction  *sR1R1 + sR2R2 + ... <-> sP1P1 + sP2P2 +…* |
|  | Dissociation constant |
|  | Enthalpy change of the reaction |
|  | Heat energy flow from the reaction |
|  | Entropy change of the reaction |
|  | Ideal gas equation |
|  | Henry’s law |
|  | Electroneutrality |
|  | Gibbs-Donnan equilibrium |
|  | Equilibrium of form A from sides *s* in group G |
|  | Osmotic pressure |
|  | Kirchhoff’s junction rule for molar flows |

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| Thermal equations | |
|  | Heat accumulation |
|  | Temperature |
|  | Heat convection |
|  | Change of heat with change of mass (e.g. vaporization) |
|  | Ideal radiator (e.g. microcirculation) |
|  | Kirchhoff’s junction rule for heat flows |

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| Population equations | |
|  | Size of population |
|  | Change of population |
|  | Kirchhoff’s junction rule for population flows |

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| Hydraulic equations | |
|  | Volume accumulation |
|  | Pressure in container with elastic walls (e.g. blood or lymph vessels) |
|  | Hydraulic conductance/resistance |
|  | Pascal’s law |
|  | Idealized hydraulic valve |
|  | Hydraulic inertia |
|  | Kirchhoff’s junction rule for volumetric flows |

# Metric Prefixes

|  |  |  |  |  |
| --- | --- | --- | --- | --- |
| symbol | Name | Decimal logarithm | Decimal value | word |
| Y | yotta | 24 | 1 000 000 000 000 000 000 000 000 | septillion |
| Z | zetta | 21 | 1 000 000 000 000 000 000 000 | sextillion |
| E | exa | 18 | 1 000 000 000 000 000 000 | quintillion |
| P | peta | 15 | 1 000 000 000 000 000 | quadrillion |
| T | **tera** | **12** | **1 000 000 000 000** | **trillion** |
| G | **giga** | **9** | **1 000 000 000** | **billion** |
| M | **mega** | **6** | **1 000 000** | **million** |
| k | **kilo** | **3** | **1 000** | **thousand** |
| h | **hecto** | **2** | **100** | **hundred** |
| da | **deca** | **1** | **10** | **ten** |
|  |  | **0** | **1** | **one** |
| d | **deci** | **-1** | **0.1** | **tenth** |
| c | **centi** | **-2** | **0.01** | **hundredth** |
| m | **milli** | **-3** | **0.001** | **thousandth** |
| µ | **micro** | **-6** | **0.000 001** | **millionth** |
| n | **nano** | **-9** | **0.000 000 001** | **billionth** |
| p | **pico** | **-12** | **0.000 000 000 001** | **trillionth** |
| f | femto | -15 | 0.000 000 000 000 001 | quadrillionth |
| a | atto | -18 | 0.000 000 000 000 000 001 | quintillionth |
| z | zepto | -21 | 0.000 000 000 000 000 000 001 | sextillionth |
| y | yocto | -24 | 0.000 000 000 000 000 000 000 001 | septillionth |

# Physical Constants

|  |  |  |  |
| --- | --- | --- | --- |
|  | Name | Value | unit |
| *h* | *Planck constant* | *6.62606957(29)×10−34* | *J.s-1* |
| NA | **Avogadro constant** | **6.02214129(27)×1023** | **mol−1** |
| R | **Gas constant** | **8.3144621(75)** | **J.mol-1.K-1** |
| *e* | ***elementary charge*** | ***1.602176565(35)×10−19*** | ***C*** |
| *G* | *gravitation constant* | *6.67384(80)×10−11* | *m3.kg−1.s−2* |
| g | **Earth gravity acceleration** | **≈ 9.8067** | **m.s-2** |
| gMn | Moon gravity acceleration | ≈ 1.625 | m.s-2 |
| gMrs | Mars gravity acceleration | ≈ 3.728 | m.s-2 |
| *c* | *speed of light in vacuum* | *299 792 458* | *m.s-1* |
| *me* | *rest mass of electron* | *9.10938215(45)×10−31* | *kg* |
| mp | rest mass of proton | 1.672621777(74)×10−27 | kg |
| mu | atomic mass unit | 1.660538921(73)×10−27 | kg |
| kB | Boltzmann constant=R/NA | 1.3806488(13)×10−23 | J.K-1 |
| F | Faraday constant = e NA | 9.64853399(24)×104 | C.mol-1 |
| Mu | Molar mass constant = kg/g | 10-3 | kg.g-1 |
| Ɛ0 | Vacuum permittivity | 8.854 187 817… x 10−12 | F.m-1 |
| *µ0* | *Vacuum permeability* | *4π × 10−7* | *N.A-2* |
| µB | Bohr magneton | 9.27400968(20)×10−24 | J.T-1 |
| µN | nuclear magneton | 5.05078353(11)×10−27 | J.T-1 |
| VSTP | molar volume of gas at STP | **2.271 0953(21) × 10−2** | **m3.mol−1** |
| VNIST | molar volume of gas at NIST | 2.437 3845(64) × 10−2 | m3.mol−1 |
| V37 | molar volume of gas at 37°C | 2.578 7304(20) × 10−2 | m3.mol−1 |

# Selected Chemical Substances

|  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- |
| Substance | | Molar Mass |  | Substance | | Molar Mass |
| H­2O | water | 0.018 |  | Ins | insulin | 5.808 |
| H+ | proton | 0.001 |  | Lep | leptin | 16.03 |
| H­3O+ | hydronium | 0.019 |  | Glc | glucagon | 3.485 |
| [H5O2]+ | Zundel cation | 0.037 |  | PRA | renin | 48.000 |
| [H9O4]+ | Eigen cation | 0.073 |  | AGT | angiotensinogen | 56.800 |
| O­2 | oxygen | 0.032 |  | AI | angiotensin I | 1.297 |
| CO­2 | carbon dioxide | 0.044 |  | AII | angiotensin II | 1.046 |
| H2CO3 | carbonic acid | 0.062 |  | Aldo | aldosterone | 0.360 |
| HCO3- | bicarbonate | 0.061 |  | ANP | atriopeptin | 3.060 |
| Na+ | sodium | 0.023 |  | ADH | vasopressin | 1.084 |
| K+ | potassium | 0.039 |  | TSH | thyrotropin | 28.000 |
| ­Cl- | chloride | 0.035 |  | T3-4 | thyroxine | 0.777 |
| H2PO­4-, HPO­42- | phosphates | 0.095 |  | EPO | erythropoietin | 34.000 |
| SO­42- | sulfates | 0.096 |  | Epi | epinephrine | 0.183 |
| NH­4+ | amonium | 0.018 |  | NE | norepinephrine | 0.169 |
| Mg2+ | magnesium | 0.024 |  | D3 | calcitriol | 0.417 |
| Ca2+ | calcium | 0.040 |  | PTH | parathormone | 9.400 |
| Fe2+,  Fe3+ | iron | 0.056 |  |  | ethanol | 0.046 |
| Glu | glucose | 0.181 |  |  | nicotine | 0.162 |
| FA | fatty acids | 0.255 |  |  | aspirin | 0.180 |
| KA | keto acids | 0.102 |  |  | paracetamol | 0.151 |
| AA | amino acids | 0.100 |  |  | nitredipine | 0.360 |
| Lac- | lactate | 0.090 |  |  | peridopril | 0.368 |
| Triglyc | triglyceride | 0.806 |  |  | metoprolol | 0.267 |
| urea | urea | 0.060 |  |  | desglymidodrine | 0.197 |
| Hb | hemoglobin | 64.500 |  |  | digoxin | 0.781 |
| Alb | albumin | 66.440 |  |  | chlorothiazide | 0.296 |
| Glb | globulins | 34.500 |  |  | furosemide | 0.331 |