

**Figure 3.** Effective conductivity of PVDF/xGnP nanocomposites as a function of xGnP volume fraction, measured at 1000 Hz and room temperature. The insets show the best fits of the conductivity to Equation 1.

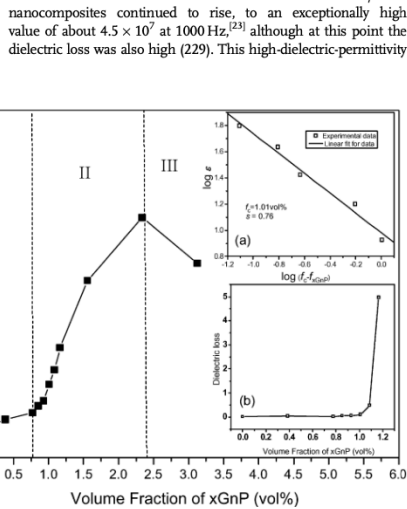
transition) at  $f_{cGnP} \approx 0.76$ – $1.55$  vol%. The conductivity of the polymer/conductive-filler composites originates from two mechanisms: 1) Ohmic conduction, through direct contact of conductive fillers, and 2) non-Ohmic conduction, through the barrier-tunneling effect between the conductive fillers separated by a polymer layer. The percolation transition can be seen as the transition from non-Ohmic conduction to Ohmic conduction, when the filler concentration approaches the percolation threshold.

The conductivity  $\sigma$  of the conductor-insulator composites near the percolation threshold can be predicted by the power laws in Equation 1a and b, as follows:

$$\sigma(f_{xGnP}) \propto (f_c - f_{xGnP})^{-t} \quad \text{for } f_{xGnP} < f_c \quad (1a)$$

$$\sigma(f_{xGnP}) \propto (f_{xGnP} - f_c)^s \quad \text{for } f_{xGnP} > f_c \quad (1b)$$

In Equation 1,  $f_c$  is the percolation threshold,  $f_{xGnP}$  is the volume fraction of xGnP, and  $s$  and  $t$  are the critical exponents in the insulating ( $f_{xGnP} < f_c$ ) and conducting ( $f_{xGnP} > f_c$ ) region, respectively. The best fits



**Figure 4.** Effective dielectric constant of the PVDF/xGnP nanocomposites as a function of the xGnP volume fraction, measured at 1000 Hz and room temperature. Inset a) shows the best fits of the conductivity to Equation 2. Inset b) shows the loss tangent of PVDF/xGnP nanocomposites as a function of xGnP volume fraction.

# Structured Extraction

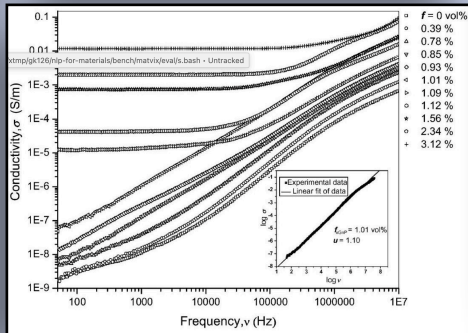
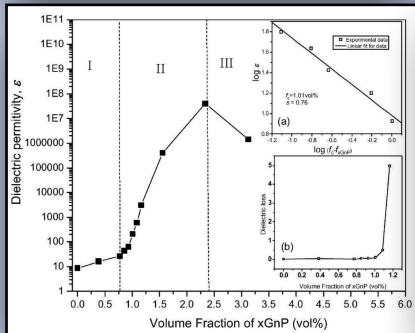
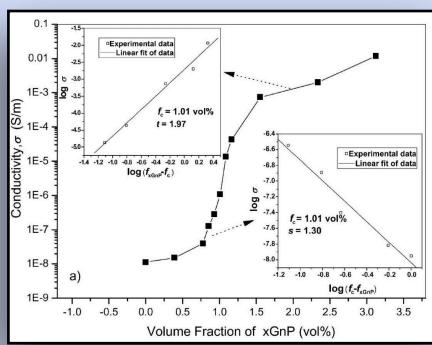
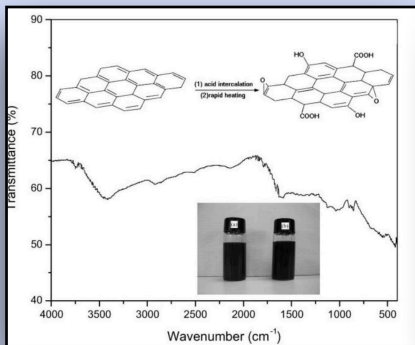
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Images



`\title{High Dielectric Permittivity and Low Percolation Threshold in Nanocomposites Based on Poly(vinylidene fluoride) and Exfoliated Graphite Nanoplates }`

`\author{By Fuan He, Sienting Lau, Helen Laiwa Chan, and Jintu Fan*}`

`\begin{document}`  
`\maketitle`  
 Ferroelectric polymers, such as poly(vinylidene fluoride) (PVDF), poly(vinylidene fluoride-co-trifluoroethylene) (P(VDF-TrFE)), and poly(vinylidene fluoride-trifluoroethylene-chlorofluoroethylene) (P(VDF-TrFE-CFE)) have great potential for applications in micro-electromechanical devices and high-charge storage capacitors `{ }^{\{1-3\}}`

`\begin{center}`  
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Figure 1c shows the transmission electron microscopy (TEM) image of a  $\text{PVDF} / \text{xGnP}$  nanocomposite obtained from solution casting. Clearly, the graphite nanoplates were well dispersed in the PVDF matrix without any aggregation, which can be ascribed to specific interactions between the functional groups of the  $\text{xGnP}$  and the  $\text{-CF}_2\text{-}$  group of the PVDF.

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