

The thursal conductivity of a material is a measure of its ability to conduct heat.

It is commonly denoted by K.

where 'k' is the co-efficient of thermal conductivity is defined as the quantity of heat flowing per second through a conducted of unit area of cross section, when there is a unit Temp gradient.

1e, LK=9) if A=189 meter  $\frac{dT}{dx}=1$  Kelvin meter

## Expression for thermal conductivity ("K)".

consider a uniform sood the with the temperature of the temperature of the cold end 'C' as Is as shown in the figure

Heat is flowing from the hot end 'H' to the cold end 'C'

Let A' be the area of cross section & x' be the mean free path of the electrons between the two ends H&C. The kinetic energy of the electronsat be hot end H is greater than that of the electrons at the colder end'c'

het 'g' be the amount of heat flowing through the rod from end H to c whose length is '2 x'.

$$\varphi = KA \left(T_2 - T_1\right) + \longrightarrow 0$$

where 'k' is co-efficient of themal conductivity,

't' is the time for conduction & 21 is the length

of the rod;

.. thermal conductivity

$$K = \frac{\mathcal{G}(2\lambda)}{A(t_2-T_1)+} \rightarrow (2)$$

Let n'be the no of available conduction electrons & v'le the root mean square velocity of the electrons

Let us assume that the free electrons in the metal are having equal probability to more in all sen possible directions. Therefore an average of t electrons can travel in any one direction (Translation + Rotation)

WKT, the free electrons are assumed to be free moving gas molecules. the average K.E of the electron at cold endit of Temp T2 W 3/2 KB T2 -> B) unly, the average K. F of the electron at cold end 'C' of Temp T, US 3/2 KBT, ->(4) where KB + Boltzmann const The no of electrons crossing the area at & persecond = tnv ->(5) therefore the resultant heat energy transferred per unit area por unit time from H to C = No of electrons X Ang K.E of electrons moving from 115, 9= 1 no x 8/2 KB (T2-T1) -> (6) WKT. Tenp gradient dT = T2-T1 dx - T2-T1 Rubstituting ogn (6) &(7) in (8)  $K \mathcal{G} = \frac{n v}{4} k_B \left(\overline{r_2} - \overline{r_1}\right), K = \frac{\mathcal{G}(2\lambda)}{A \left(\overline{r_2} - \overline{r_1}\right) t}$  $K = \frac{(n \cdot k_{B} (T_{2} - T_{1}))}{(T_{2} - T_{1})} (2x)$ lif A=150m t=1sec

K= 1 nvKB> -1(8)

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From the classical free electron theory wky, the electronic specific heat capacity of the metales

where R=nKB R-> gas const

nks= 2/3 C. -> (0)

substituting ( in (8)

K= ないかをかたの

K = Y2VX(2/3C)

1. K= 10 CVA -> (1)

eq (8) & (11) depresents the enpression for thermal conductivity of a metal. depending

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the how states that the ratio of the Thermal Conduction - Wity & electrical Conductivity of a metal is idirectly proportional to the Absolute Temp of the metal,

 $ie_{i}$   $\frac{K}{\nabla} \times \top$   $\rightarrow 0)$ 

 $\frac{k}{T} = LT \rightarrow (2)$ 

where 'L' is the proportionality const-called horeity const

whose value L= 1.12×108 WILK WKT thermal conductivity, K = 1 nv kg x -3) Electrical conductivity, J= net T - 14)  $\frac{K}{\sigma} = \frac{\gamma v k_B \lambda}{2 \left(\gamma e^2 \tau\right)} = \frac{v k_B \lambda m}{2 e^2 \tau}$ K = IV(I)m (KB)  $\frac{k}{\tau} = \frac{1}{2} \operatorname{mn}^{2} \left( \frac{k_{B}}{e^{2}} \right) \rightarrow (5)$ By kinetic theory of gases K. E of electron = 3/2 KBT 1/2 mv2 = 3/2 KBT Eq7(5) => \( \frac{k}{2} \) K = 3/2 (kB) T  $\frac{K}{\sqrt{8}} = 3/2 \left(\frac{k_B}{c}\right)^2 + \cdots + (6)$ K = LT Jeon (92 (2) K = LT where h= 3/2 (kg) -) (7) substituting the value of 'L' is found to be L = 3 (1,38 x 1523)2 2 (1.6 × 10 19)2 = 1.1 × 108 W2/K2 NOTE! the expth value of 'h' is found to be 2.44×10° 10 2/1×2. This is due to the failure of the idensical theory to give the correct value of the Thermal conductivity of metals.

Ace to quantum theory K= nx2kb=TI

L=K
T+ (L=2.444×108002/12)

thus it confiams that wo-flaw is verified using the quantum theory & it supports that it is not applicable for low Temperature.

## failures of devisical free electron theory

- I It fails to enplain the electric specific heat & the specific heat capacity of metals.
- 2. It fails to explain superconducting properties of metals
- 3. It fails to enplais new phenomena like photo-electric effect, compton effect, Black body radiation etc.
- 4. classical theory states that electrons absorbs all the energy. But all the electrons will not absorb energy (E=h1)
- 5. classical theory will not emplais seni-conductor & Insulators.