

Polynomial Systems in Receptor Pharmacology

Gilles Gnacadja

AMGEN
Thousand Oaks, California, USA

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Industry Days: A Southern California Perspective

Polynomial Systems in Receptor Pharmacology

- 1 Binding Equilibrium Calculation in Pharmacology
 - Current Practices and How They Fall Short
 - Prior Work
 - Our Work
- 2 Polynomial System for Binding Equilibrium
 - An Example
 - The Polynomial System and its Properties
 - Solving the Polynomial System by Fixed-Point Iterations
 - Solving the Polynomial System by Cell Exclusion
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Pharmacology

Pharmacology: Study interactions between biological processes and therapeutic agents.

- **Pharmacokinetics:** *what the body does to the drug.*
- **Pharmacodynamics:** *what the drug does to the body.*
- **“Pharmacostatics”** at beginning of research pipeline:
Study binding equilibrium *in vitro* of interactions between
 - receptors (biochemical recognition units),
 - ligands (pathogenic and therapeutic molecules),
 - other molecules.

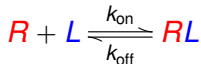
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Current Practices – Illustrated with Trivial Receptor-Ligand Interaction

Chemistry



Equilibrium state
Implicit equations
(Polynomial system)

$$[R]_{\text{Equil}} + [RL]_{\text{Equil}} = [R]_{\text{Total}}$$

$$[L]_{\text{Equil}} + [RL]_{\text{Equil}} = [L]_{\text{Total}}$$

$$[RL]_{\text{Equil}} = [R]_{\text{Equil}} [L]_{\text{Equil}} / K_d$$

Equilibrium state
Explicit formula
(Forcibly derived)

$$[R]_{\text{Total}} \ll [L]_{\text{Total}}$$

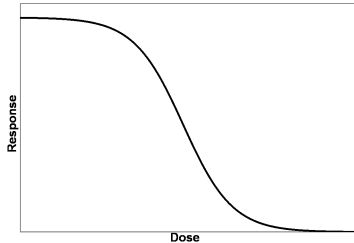
$$[L]_{\text{Equil}} = [L]_{\text{Total}} \quad (\text{"No Ligand Depletion"})$$

$$\frac{[RL]_{\text{Equil}}}{[R]_{\text{Total}}} = \frac{[L]_{\text{Total}}}{K_d + [L]_{\text{Total}}}$$

Issues with Forcibly Derived Formulas

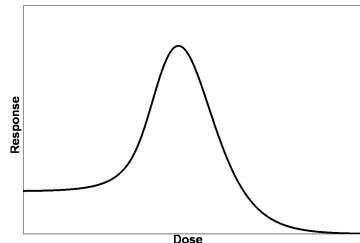
- $[R]_{\text{Total}} \ll [L]_{\text{Total}}$ is not mathematically sound.
Operational wisdom: $[R]_{\text{Total}} < [L]_{\text{Total}}/10$.
- Not always applicable experimentally,
e.g. in miniaturized assays.
- Need not be applicable *in vivo*.
- New formula derivation needed for every new interaction.
- Only possible with “*receptor-centric*” systems.
(There is one receptor and all complex molecules contain it.)

One More Issue with Forcibly Derived Formulas



This shape is expected and is what forcibly derived closed-form formulas give.

Non-monotone responses do occur and cannot be simulated with forcibly derived closed-form formulas.



Excuses for Forcibly Derived Formulas

- Somewhat admissible:

*These formulas are simple and always give an answer.
They are (often, not always) directionally correct.*

- Not admissible:

We have many sources of error, so why fix this one?

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“No Ligand Depletion” Assumption Discarded for Competitive Antagonism

Miniaturized Receptor Binding Assays: Complications Arising from Ligand Depletion

Clare M. Scaramellini Carter, Juliet R. Leighton-Davies and Steven J. Charlton
J Biomol Screen 2007; 12; 255 originally published online Jan 26, 2007



+

A

\updownarrow

RA

“No Ligand Depletion” Assumption Partially Discarded for Allosteric Modulation

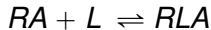
0022-3565/08/3253-927-994\$20.00

THE JOURNAL OF PHARMACOLOGY AND EXPERIMENTAL THERAPEUTICS

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 JPET 325:927-934, 2008

The Impact of Orthosteric Radioligand Depletion on the Quantification of Allosteric Modulator Interactions

Vimesh A. Avlani, David J. McLoughlin, Patrick M. Sexton, and Arthur Christopoulos



with

$$[A]_{\text{Equil}} = [A]_{\text{Total}}$$

More Shots at the “No Ligand Depletion” Assumption

Biochimica et Biophysica Acta, 632 (1980) 464–469
© Elsevier/North-Holland Biomedical Press

COMPETITIVE BINDING STUDIES WITH MULTIPLE SITES EFFECTS ARISING FROM DEPLETION OF THE FREE RADIOLIGAND

J.W. WELLS, N.J.M. BIRDSALL, A.S.V. BURGEN and E.C. HULME

0006-8996/80/0000-0000/\$02.00/0
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All rights of reproduction in any form reserved.
MOLECULAR PHARMACOLOGY, 21:460-469

Ligand Dissociation Constants From Competition Binding Assays: Errors Associated with Ligand Depletion

AVRAM GOLDSTEIN and RONALD W. BARRETT

Proc. Natl. Acad. Sci. USA
Vol. 84, pp. 6654–6658, October 1987
Biochemistry

An accurate method for determination of receptor–ligand and enzyme–inhibitor dissociation constants from displacement curves

AMNON HOROVITZ and ALEXANDER LEVITZKI

ARCHIVES OF BIOCHEMISTRY AND BIOPHYSICS
Vol. 284, No. 1, January, pp. 26–29, 1991

A Simple Method for Calculating the Dissociation Constant of a Receptor (or Enzyme)· Unlabeled Ligand Complex from Radioligand Displacement Measurements

Robert L. Martin, Franco Renosto, and Irwin H. Segel

0006-8996/80/0001-0007-\$750.00/0
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All rights of reproduction in any form reserved.
MOLECULAR PHARMACOLOGY, 43:107-120 (1982)

Interpretation of Binding Curves Obtained with High Receptor Concentrations: Practical Aid for Computer Analysis

STÉPHANE SWILLENS

Miniaturized Receptor Binding Assays: Complications Arising from Ligand Depletion

Clare M. Scaramellini Carter, Juliet R. Leighton-Davies and Steven J. Charlton
J Biomol Screen 2007; 12: 255 originally published online Jan 26, 2007

0007-1226/03/0007-0007-\$05.00/0
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0006-8996/03/0007-0115-12\$05.00/0
MOLECULAR PHARMACOLOGY
Copyright © 2003 The American Society for Pharmacology and Experimental Therapeutics
MO Pharmacol 73:115-126, 2004

Cell-Based and Biochemical Structure-Activity Analyses of Analogues of the Microtubule Stabilizer Dictyostatin

Brianne S. Raccor, Andreas Vogt, Rachel P. Sikorski, Charitha Madiraju,
Raghavan Balachandran, Kia Montgomery, Youseung Shin, Yoshikazu Fukui,
Won-Hyuk Jung, Dennis P. Curran, and Billy W. Day

Journal of Neuroscience Methods 188 (2008) 12–38

Concentration of receptor and ligand revisited in a modified receptor binding protocol for high-affinity radioligands: [³H]Spiperone binding to D₂ and D₃ dopamine receptors

Juan Zhen^a, Tamara Antonio^a, Aloke K. Dutta^a, Maarten E.A. Reith^a

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Goal

- Calculate binding equilibrium concentrations ...
- by *actually* solving the polynomial system of conservation and equilibrium conditions ...
- with the benefits of forcibly derived formulas, i.e. get an answer always and quickly.

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The Receptor-Ligand-Antagonist-Trap System

Polynomial System for Equilibrium State



$$[R] + K_{RL}[R][L] + K_{RA}[R][A] = [R]_{\text{Total}}$$

$$[L] + K_{RL}[R][L] + K_{LT}[L][T] = [L]_{\text{Total}}$$

$$[T] + K_{AT}[A][T] + K_{LT}[L][T] = [T]_{\text{Total}}$$

$$[A] + K_{AT}[A][T] + K_{RA}[R][A] = [A]_{\text{Total}}$$

The Receptor-Ligand-Antagonist-Trap System

Polynomial System for Equilibrium State, Reformulated as a Fixed-Point Equation



$$[R] = \frac{[R]_{\text{Total}}}{1 + K_{RL}[L] + K_{RA}[A]}$$

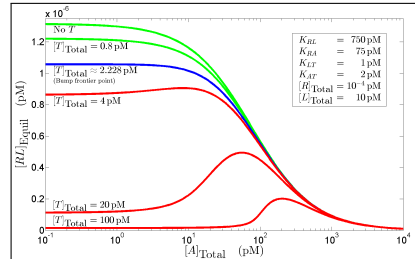
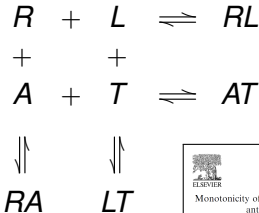
$$[L] = \frac{[L]_{\text{Total}}}{1 + K_{RL}[R] + K_{LT}[T]}$$

$$[T] = \frac{[T]_{\text{Total}}}{1 + K_{AT}[A] + K_{LT}[L]}$$

$$[A] = \frac{[A]_{\text{Total}}}{1 + K_{AT}[T] + K_{RA}[R]}$$

The Receptor-Ligand-Antagonist-Trap System

Polynomial System for Equilibrium State, Solved by Fixed-Point Iterations



- Fixed-point iterations converge, always and quickly.
 (Contraction w.r.t. Thompson metric; not L^1 , L^2 , L^∞ , etc.)
- Experimentally observed monotonicity features simulated and explained.
- Forcibly derived explicit formulas only produce always-increasing and always-decreasing curves.

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The Polynomial System

$$x_i + \sum_{\alpha \in I} \alpha_i a_{\alpha} x^{\alpha} = b_i$$

$$i = 1, \dots, n$$

Unknown $x = (x_1, \dots, x_n) \in \mathbb{R}_{\geq 0}^n$

Given

$$b_i \geq 0, \forall i = 1, \dots, n$$

$$I \text{ finite} \subset \mathbb{Z}_{\geq 0}^n \setminus \{0_n, e_{n,1}, \dots, e_{n,n}\}$$

$$a_{\alpha} \geq 0, \forall \alpha = (\alpha_1, \dots, \alpha_n) \in I$$

$$x^{\alpha} := x_1^{\alpha_1} \cdots x_n^{\alpha_n}$$

MATHEMATICAL METHODS IN THE APPLIED SCIENCES

Math. Meth. Appl. Sci. 2007; 30:201-211

Published online 23 October 2006 in Wiley InterScience

(www.interscience.wiley.com) DOI: 10.1002/mma.782

MR06 subject classification: 47H10; 74G25; 74G30; 74G15; 92C40



Fixed points of order-reversing maps in $\mathbb{R}_{>0}^n$
and chemical equilibrium

Gilles Gnacadjia

Advances in Applied Mathematics 42 (2009) 356-414

Contents lists available at ScienceDirect



Advances in Applied Mathematics

www.elsevier.com/locate/yaama



Univalent positive polynomial maps and the equilibrium
state of chemical networks of reversible binding reactions

Gilles Gnacadjia

Mathematical Biosciences 210 (2011) 115-140

Contents lists available at ScienceDirect



Mathematical Biosciences

journal homepage: www.elsevier.com/locate/mbs



A method to calculate binding equilibrium concentrations in the allosteric
ternary complex model that supports ligand depletion

Gilles Gnacadjia

Linear Algebra and its Applications 437 (2012) 612-622

Contents lists available at SciVerse ScienceDirect



Linear Algebra and its Applications

journal homepage: www.elsevier.com/locate/laa



A Jacobian criterion for the simultaneous injectivity on
positive variables of linearly parameterized polynomial maps

Gilles Gnacadjia

Equivalent Fixed-Point Formulation

$$x_i = \frac{b_i}{1 + \sum_{\alpha \in I, \alpha_i > 0} \alpha_i a_\alpha x^{\alpha - e_{n,i}}} \quad i = 1, \dots, n$$

$$x = f(a, b, x)$$

$$f(a, b, \cdot) : \mathbb{R}_{\geq 0}^n \rightarrow \mathbb{R}_{\geq 0}^n \text{ is smooth and order-reversing}$$

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Solution Exists, is Unique, and is Infinitely Smooth

$$x_i + \sum_{\alpha \in I} \alpha_j a_{\alpha} x^{\alpha} = b_j \quad , \quad i = 1, \dots, n$$

Existence

Brouwer Fixed Point Theorem

Uniqueness

Gale-Nikaidô Global Injectivity Theorem

GALE, D. and H. NIKAIÐÔ
Math. Annalen 159, 81—93 (1965)

The Jacobian Matrix and Global Univalence of Mappings

By

DAVID GALE and HUKUKANE NIKAIÐÔ in Providence and Osaka

Smoothness

Inverse Function Theorem

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Solving by Fixed-Point Iterations



- 1 Fixed point iterations always converge if all complex molecules are (homo- or hetero-) dimers.
 - The map iterated is a contraction w.r.t. the Thompson metric (not L^1 , L^2 , L^∞ , etc) and is order-reversing.
 - Iterates provide a descending sequence of boxes that converges to the solution point.
- 2 With k -mers with $k \geq 3$, the descending sequence of boxes converges, either to the solution point, or to a non-point box that contains it. ("Coupled fixed points")

First part partially rediscovered: M. G. A. van Dorp, F. Berger, E. Carlon
Computing Equilibrium Concentrations for Large Heterodimerization Networks
Physical Review E, Volume 84, Issue 3, September 2011

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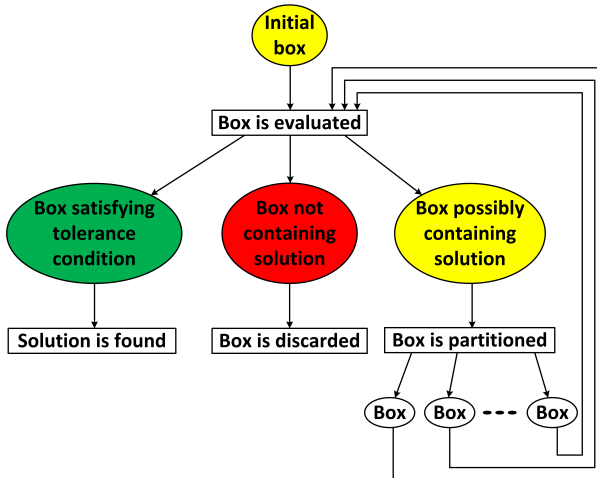
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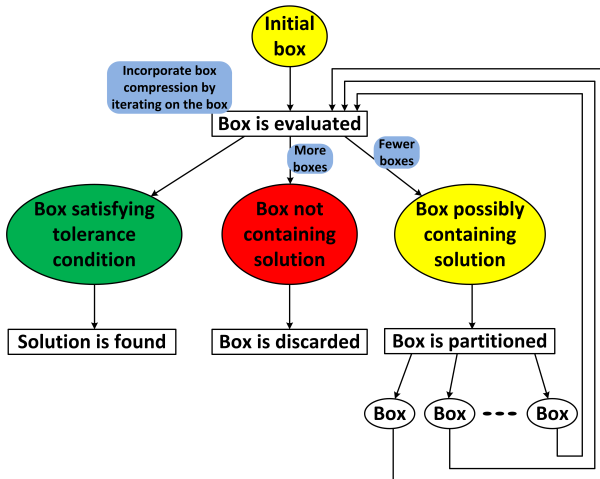
Solving by Cell Exclusion

*If the descending sequence of boxes converges to a non-point box, divide the box and try the same on each subbox.
Do this again, and again...*

Solving by Cell Exclusion



Strengthening Cell Discarding with Cell Compression



Summary

- Calculating binding equilibrium in “pharmacostatics” is about solving certain nice polynomial systems.
- Customary practice is to derive simple formulas at all cost.
 - Reduced accuracy
 - Only monotone response curves
- Algorithms exist to actually solve the polynomial systems with *a priori* assurance of success.
 - Exploit fixed-point formulation
 - Fixed-point iterations
 - Cell exclusion algorithm, strengthened by cell compression
 - Work in progress: stronger cell discarding conditions
- Enable scientists to study pharmacostatics of biochemical mechanisms that would otherwise be fudged or discarded.

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