

Mahindra University Hyderabad
École Centrale School of Engineering
End-semester Regular Examination, May 2024

Program: B. Tech. Branch: All Year: I Semester: II
Subject: Physics (PH1201)

Date: 28-05-2024
Time Duration: 3 Hours

Start Time: 10.00 am
Max. Marks: 100

Instructions:

- 1) All the questions are compulsory
- 2) Only non-programmable scientific calculator is allowed
- 3) Values of useful constants are given at the end of the question paper.

Q1.

Marks (10 + 10)

- (a) What are the requirements for a well-behaved wavefunction?

Indicate whether each of the functions given below is a solution of Schrodinger's equation over the given range. If it is not, write down the reason thereof in brief.

- (i) $\Phi = A e^{-x^2}$ $-\infty < x < \infty$
- (ii) $\Phi = A \sec x$ $-\infty < x < \infty$
- (iii) $\Phi = A \sin^{-1} x$ $-1 \leq x \leq 1$
- (iv) $\Phi = A i(e^{ix} + e^{-ix})$ $-\pi \leq x \leq \pi$

Note A is a non-zero constant.

- (b) A particle limited to the x -axis has the wave function

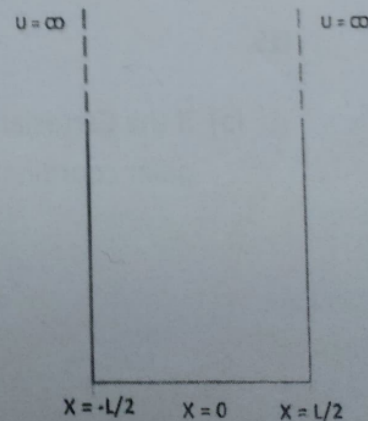
$$\begin{aligned} \Phi(x) &= Ax; & 0 < x < 1; \\ &= 0; & \text{elsewhere} \end{aligned}$$

What is the probability of finding the particle in the region $0.4 < x < 0.6$?

Q2.

Marks (10 + 10)

- (a) A particle with mass m is in an infinite square well potential with walls at $x = -L/2$ and $x = L/2$. Write the energy eigenvalues and eigenfunctions for the states $n = 1, n = 2$.



(b) The tunneling probability for a free particle of mass m and energy E , through a potential barrier of height V_0 and width L is 0.05. How will this change **qualitatively** if only

- i. the barrier height is reduced
- ii. the barrier width is reduced
- iii. the energy of the particle is reduced.
- iv. Find the new tunneling probability if $(V_0 - E)$ is changed to 0.49 times the original value, and the new width is 0.5 of its original value.

Q3.

Marks (10 + 10)

- (a) Write down the Heisenberg position-momentum uncertainty relation. Use Heisenberg's uncertainty principle to obtain an order of magnitude estimate for the minimum kinetic energy of an electron in a hydrogen atom.
- (b) Find the kinetic energy of a proton whose de Broglie wavelength is 1 pm.
(1 pm = 10^{-12} m)

Q4.

Marks (10 + 6 + 4)

- (a) Lithium, beryllium, and mercury have work functions of 2.30 eV, 3.90 eV, and 4.50 eV, respectively. Light with a wavelength of 270 nm is incident on each of these metals.
Determine which of these metals exhibit the photoelectric effect for this incident light. Explain your reasoning.
- (b) Calculate the wavelength (in nm) of light associated with energy 7.83×10^{-19} J per photon.
- (c) Find the phase and group velocity of the de Broglie waves of an electron whose speed is $0.85 c$. Here c is the velocity of light.

Q5.

Marks (4 + 6 + 10)

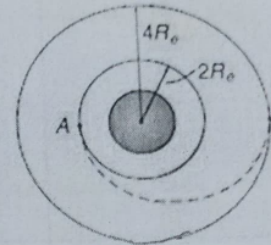
- (a) If the Cartesian coordinates (x, y) of a point are (1,1), then what are the corresponding polar coordinates (r, θ) ?

- (b) The effective potential corresponding to a pair of particles interacting through a central force is given by the expression

$$U_{\text{eff}}(r) = U(r) + \frac{L^2}{2\mu r^2} = Cr^3 + \frac{L^2}{2\mu r^2}$$

where L is the angular momentum, μ is the reduced mass and C is a constant. What is the radius r_0 of the circular orbit allowed in this potential? Express your answer in terms of L , C , and μ .

- (c) An efficient way to accomplish the transfer of satellite from one orbit to another orbit is to use a semi-elliptical orbit (known as a *Hohmann transfer orbit*), shown in the figure. What velocity changes are required at the point of intersection, point A?



-----Useful constants and unit conversions -----

$$c = 3 \times 10^8 \text{ m/s}$$

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$$

$$h = 6.6 \times 10^{-34} \text{ J.s}$$

Or

$$h = 4.1 \times 10^{-15} \text{ eV.s}$$

$$\hbar = 1.1 \times 10^{-34} \text{ J.s}$$

$$\text{Mass of an electron} = 9 \times 10^{-31} \text{ kg}$$

$$\text{Mass of a proton} = 1.67 \times 10^{-27} \text{ kg}$$

$$\text{Charge of an electron} = 1.6 \times 10^{-19} \text{ C}$$

$$\text{Mass of one hydrogen atom} \sim \text{mass of a proton} = 1.67 \times 10^{-27} \text{ kg}$$