

Mahindra University Hyderabad

École Centrale School of Engineering Minor-I exam

Program: B. Tech.

Branch: All Year: 1

Semester: 2

Program: b. 16

Subject: Physics 1 (PH1201)

Date: 18.04.24

Time Duration: 1.5 Hours

Start Time: 10 am

Max. Marks: 60

Instructions:

1) All the questions are compulsory

2) Calculator is allowed

3) Values of useful constants are given at the end of the question paper.

Q1

(10 + 10)

- (a) A block of mass 10 kg is attached to a spring with spring constant k = 10 N/m? What is Block's period of oscillation?
 - The block is held at rest at a position that is $X_0 = 5.00$ cm from the equilibrium position and then released. i) Find the maximum energy of the block. ii) Express the position as a function of time; assume no damping.
- (b) i) Equation of a damped harmonic oscillator is given by $\ddot{x} + \Upsilon \dot{x} + \omega_0^2 x = 0$. Here Υ and other symbols having usual meaning. If $e^{-t/2}\cos 3t$ is a solution of the above equation, what are the values of Υ and ω_0 ?
 - ii) What is the value of the quality factor (Q) for this oscillator? Assume it is a lightly damped oscillator.

Q2

(10 + 5 + 5)

- (a) In the context of photoelectric effect the work function for tungsten metal is 4.52 eV.
 - (i) What is the cutoff wavelength $\lambda_{\mathcal{C}}$ for tungsten?
 - (ii) What is the maximum kinetic energy of the ejected electrons when an incident radiation of wavelength 200 nm is used?
 - (iii) What is the stopping potential in this case?
- (b) A 500 W radio transmitter operates at a frequency of 1 kHz. How many photons per second does it emit?
- (c) The distance between two successive atomic planes of a calcite crystal is 0.3 nm. Find the smallest angle for Bragg scattering of X-rays of 0.03 nm wavelength by the Calcite crystal.

- (a) An X-ray photon of initial frequency 3.0×10^{19} Hz collides with an electron and is scattered through 90 degrees. Find the frequency of the scattered photon. How much energy is transferred to the electron in this collision?
- (b) What voltage must be applied to an X-ray tube for it to emit X-rays with a minimum wavelength of 40 pm? (1 pm = 10^{-12} m)
- (c) What is the minimum energy (in eV) that a photon must have to produce a proton-antiproton pair in the presence of matter? Note that the rest mass energy of a particle of mass m_0 is given by m_0c^2 . Also, mass of a proton = mass of an antiproton

Useful constants:

 $c = 3 \times 10^8$ m/s mass of an electron = 9×10^{-31} kg mass of a proton = 1.67×10^{-27} kg $h = 6.6 \times 10^{-34}$ J.s Or $h = 4.1 \times 10^{-15}$ eV.s