

FoP 3B Part II

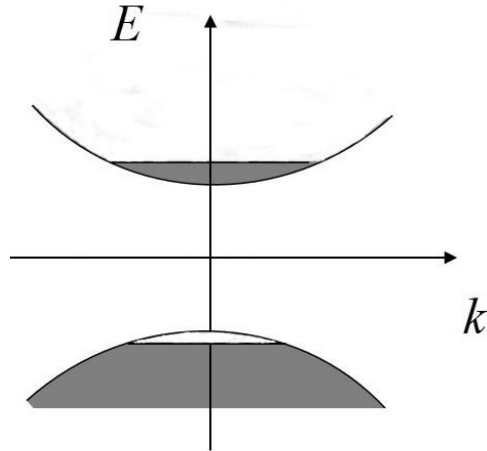
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Room 151

Lecture 4: Extrinsic semiconductors

Summary of Lecture 3

Electron/hole concentrations:



$$n = \int f(E) g_e(E) dE$$

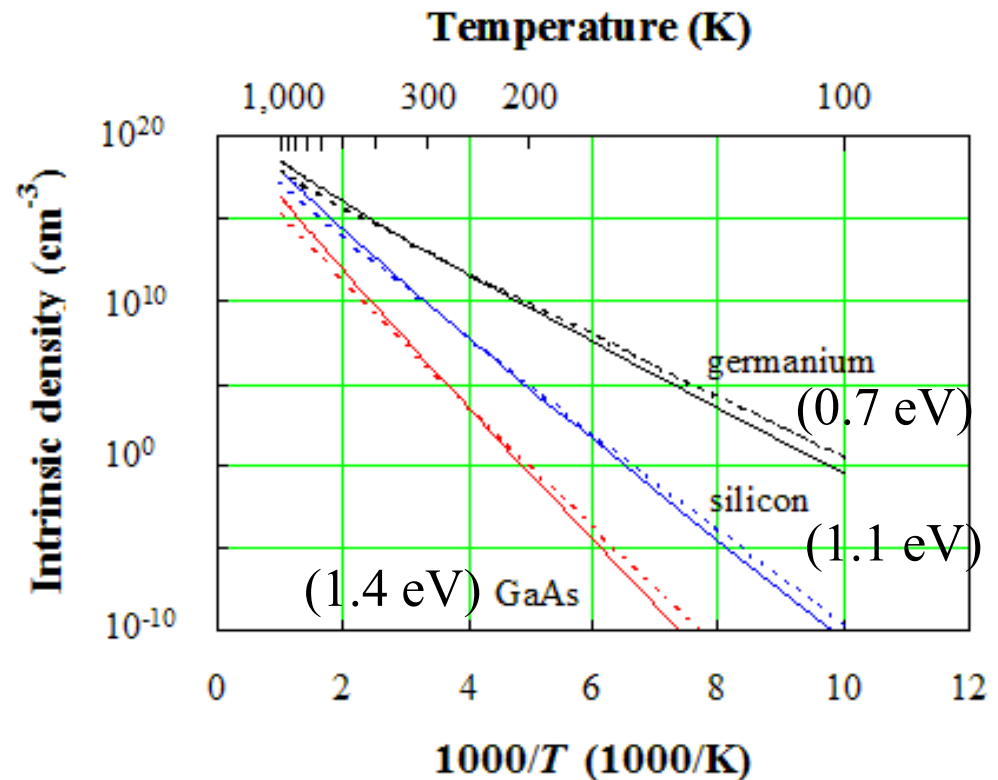
$$p = \int [1 - f(E)] g_h(E) dE$$

Law of mass action:

$$np = N_c N_v \exp\left(-\frac{E_g}{kT}\right)$$

Conductivity vs temp:

$$\mathbf{J} = (en\mu_e + ep\mu_h)\mathbf{E} = \sigma\mathbf{E}$$



Aim of today's lecture

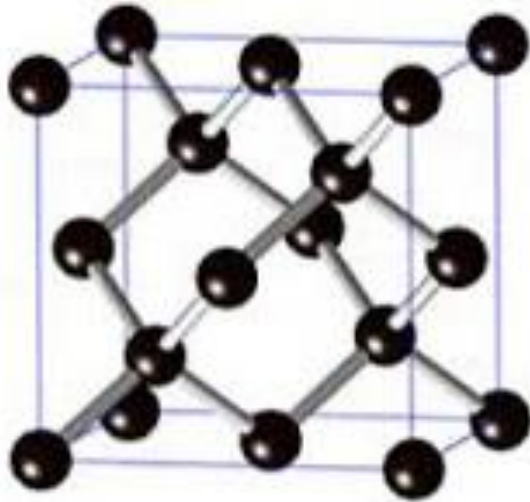
Q: How can we manipulate conductivity at room temperature?

A: 'doping' with foreign atoms

Key concepts:

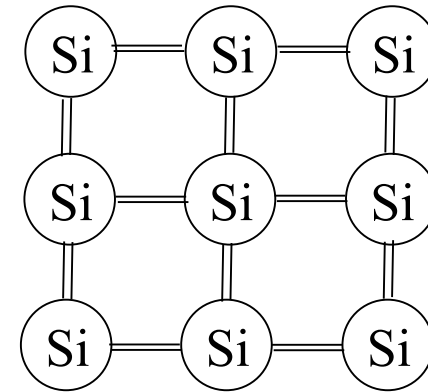
- Donor and acceptor doping
- Role of temperature on doping
- Carrier concentrations and..
- Chemical potential in extrinsic semiconductors

Donor impurities

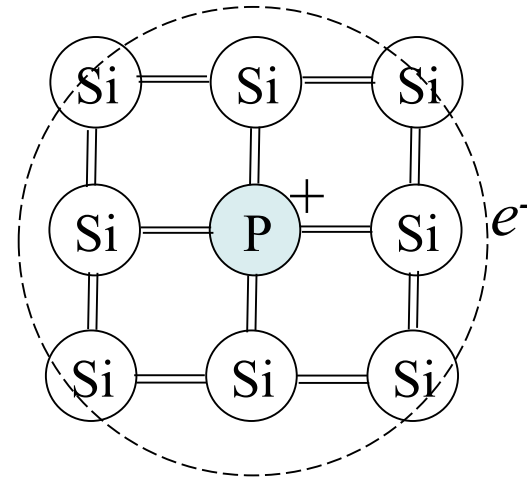


Diamond cubic crystal structure

	5 B boron 10.81 [10.806, 10.821]	6 C carbon 12.011 [12.009, 12.012]	7 N nitrogen 14.007 [14.006, 14.008]
	13 Al aluminium 26.982	14 Si silicon 28.085 [28.084, 28.086]	15 P phosphorus 30.974
12	30 Zn zinc 65.38(2)	31 Ga gallium 69.723	32 Ge germanium 72.630(8)
	33 As arsenic 74.922	34 Se selenium 78.96 [78.96, 78.97]	35 Br bromine 79.904 [79.904, 79.906]
	48 Cd cadmium 112.41	49 In indium 114.82	50 Sn tin 118.71
	51 Sb antimony 121.76	52 Te tellurium 127.6 [127.6, 127.61]	53 I iodine 126.905 [126.905, 126.909]



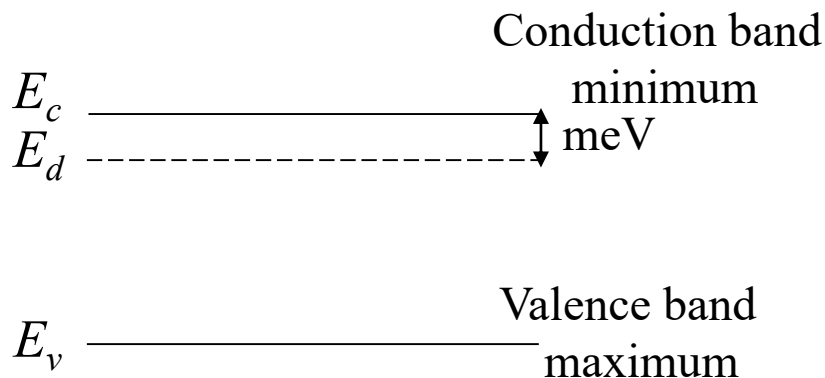
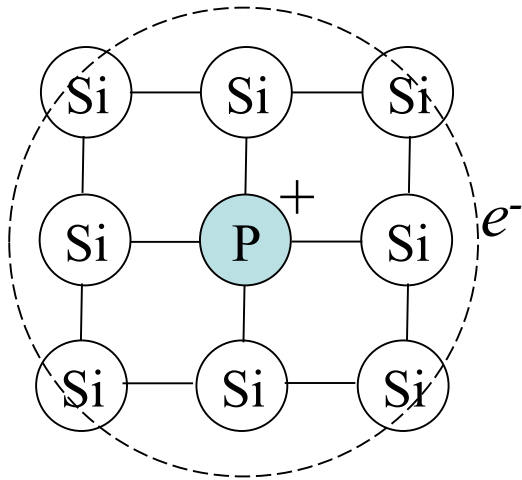
Perfect crystal



Group V substitutional atom (e.g. N, P)

Donor electron energy level

Adapt Bohr's atomic model:



Energy level diagram

E_d : donor level

Energy level:

$$E_D = \frac{e^4 m_e^*}{2(4\pi\epsilon_r\epsilon_0\hbar)^2} = \frac{m_e^*}{m\epsilon_r^2} \underbrace{\left[\frac{e^4 m}{2(4\pi\epsilon_0\hbar)^2} \right]}_{13.6 \text{ eV}}$$

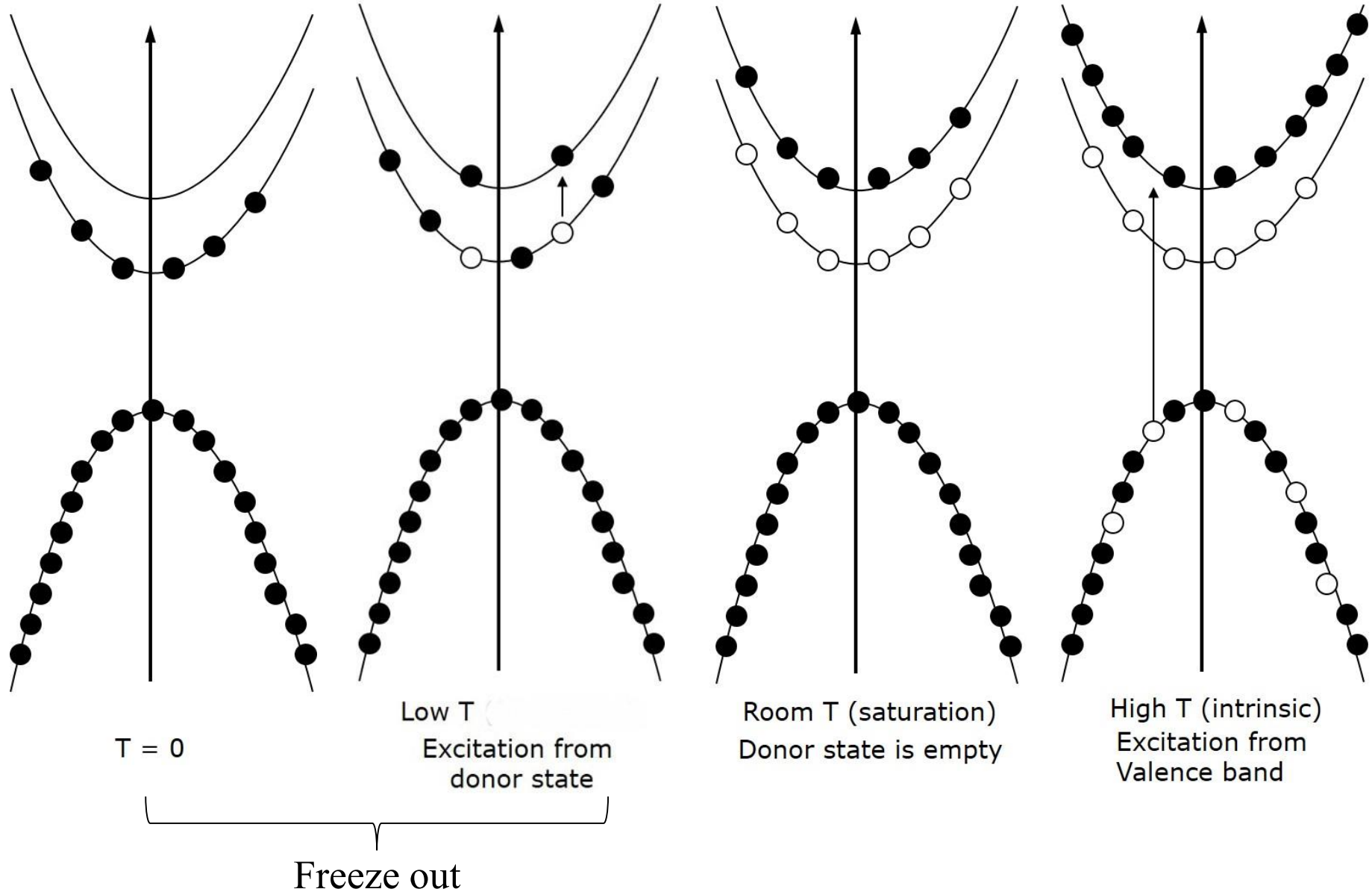
⇒ Typical energies meV

Orbit radius:

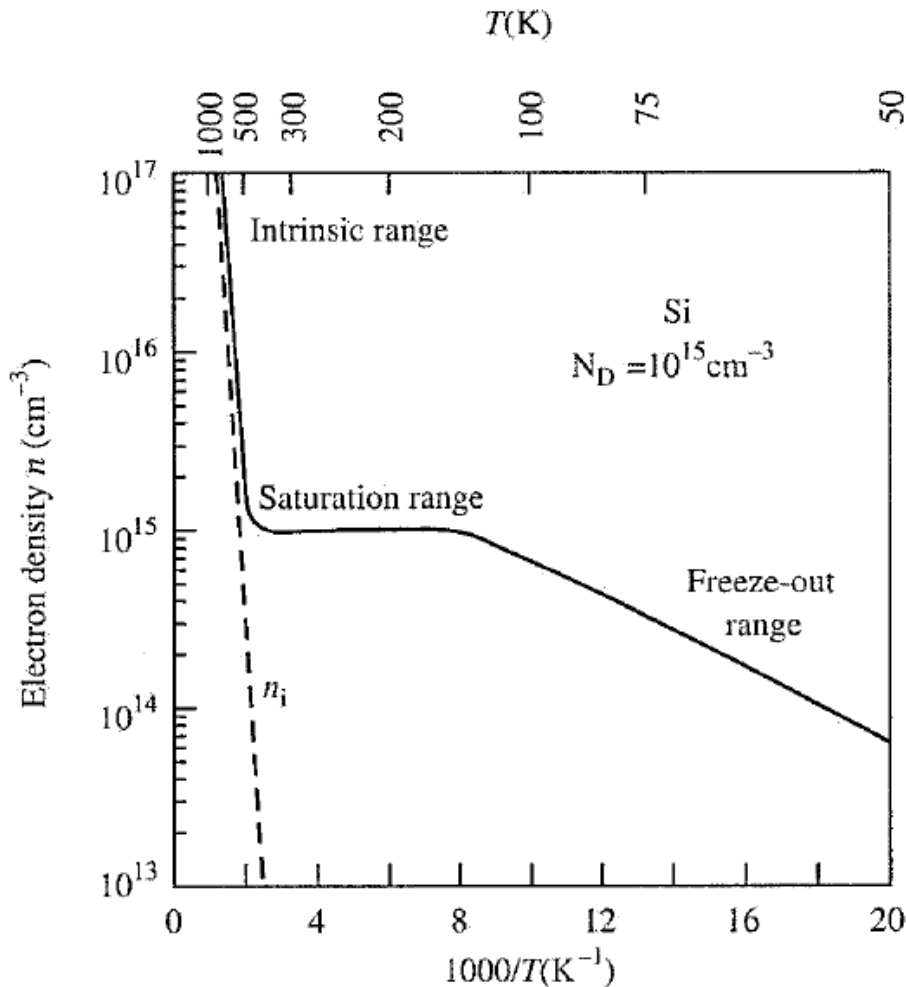
$$r = \frac{4\pi\epsilon_r\epsilon_0\hbar^2}{m_e^*e^2} = \frac{\epsilon_r m}{m_e^*} \underbrace{\left(\frac{4\pi\epsilon_0\hbar^2}{me^2} \right)}_{0.5 \text{ Å}}$$

⇒ Typical radius nm

Temperature effects on donor ionisation



Electron, hole concentrations in saturation regime



Law of mass action:

$$np = N_c N_v \exp\left(-\frac{E_g}{kT}\right)$$

$$np = n_i(T)^2 \quad \dots (1)$$

$n_i(T)$ = electron/hole concentration for intrinsic semiconductor at temp T

Charge conservation:

$$n = p + N_D \quad \dots (2)$$

N_D = donor concentration

Combining (1) and (2):

$$n = \frac{N_D}{2} + \sqrt{\left(\frac{N_D}{2}\right)^2 + n_i^2}$$

Electron, hole concentrations in saturation regime

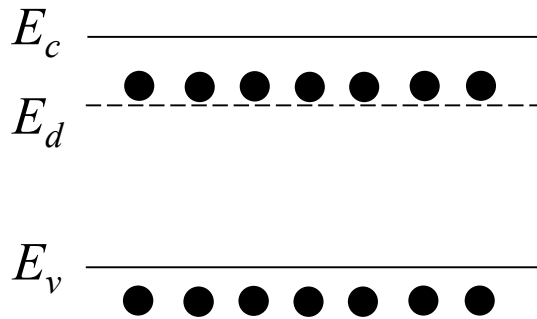
$$n = \frac{N_D}{2} + \sqrt{\left(\frac{N_D}{2}\right)^2 + n_i^2}$$

Assuming $N_D \gg n_i(T)$ then:

$$n \sim N_D$$
$$p \sim n_i^2/N_D \text{ (law of mass action)}$$

In other words $n > n_i$ and $p = n_i(n_i/N_D) < n_i$. Electrons are therefore the *majority* carriers and holes the *minority* carriers. The semiconductor is said to be doped *n*-type.

Chemical potential in *n*-type extrinsic semiconductor



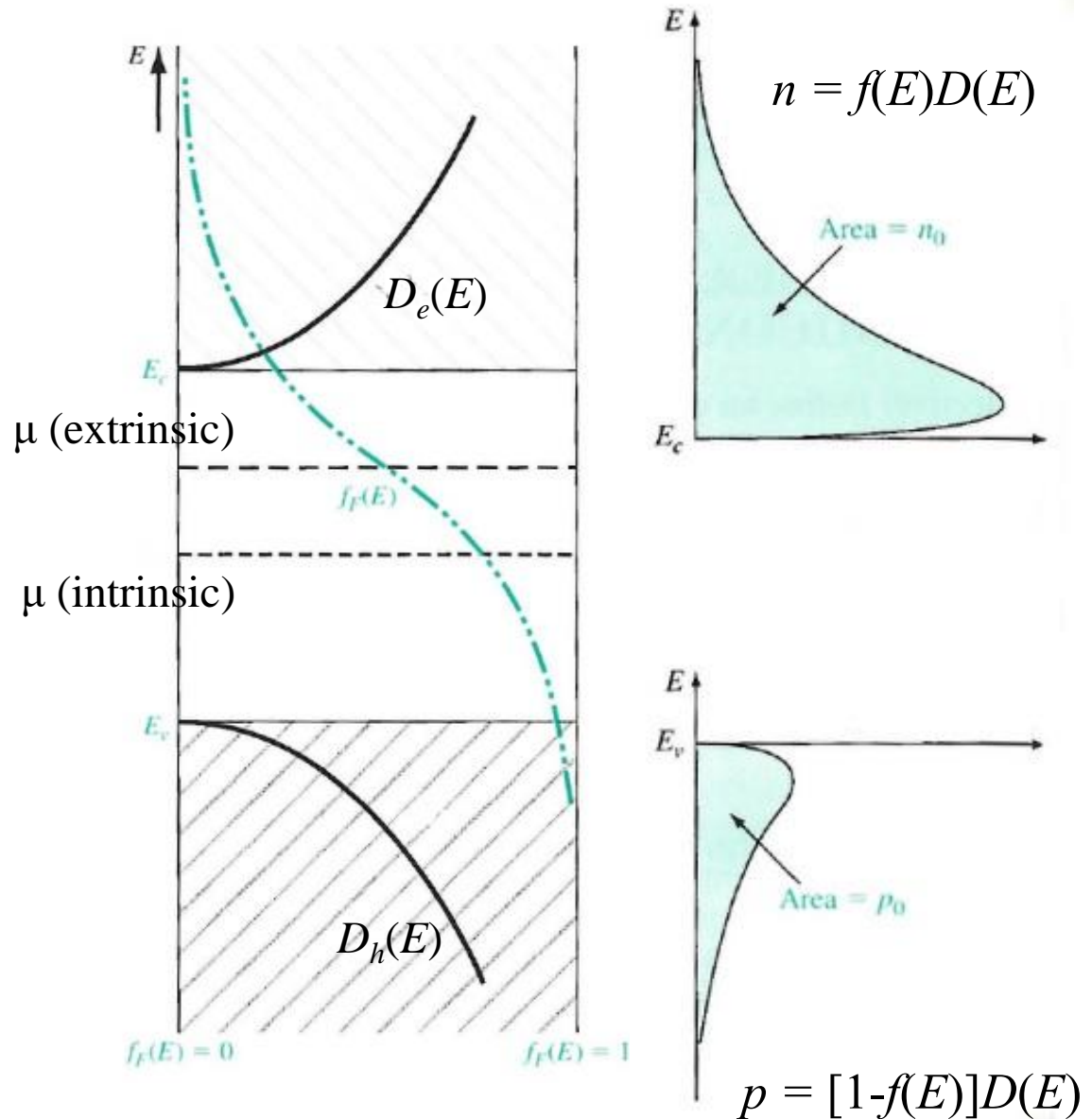
n-type semiconductor at 0K
(Fermi level above E_d)

From

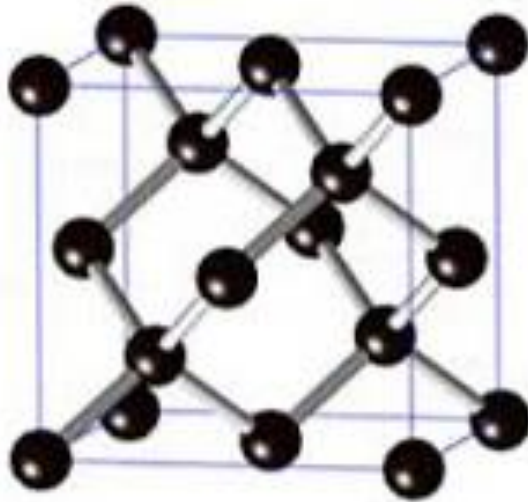
$$n = N_c \exp \left[-\frac{(E_c - \mu)}{kT} \right]$$

and substituting $n \sim N_D$

$$\mu = E_c - kT \ln \left(\frac{N_c}{N_D} \right)$$

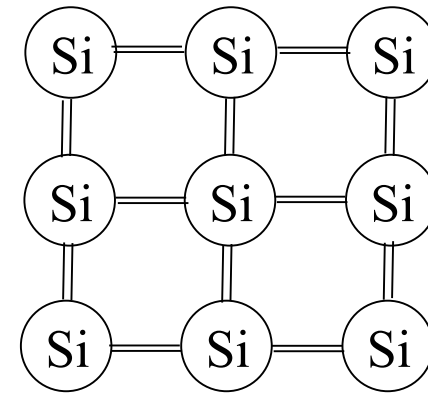


Acceptor impurities

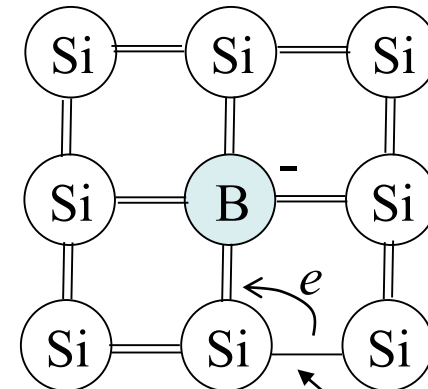


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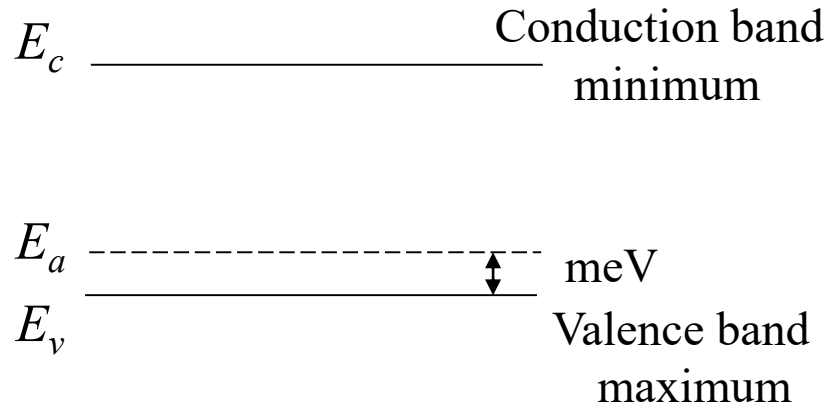
Perfect crystal



vacant site

Group III substitutional atom (e.g. B, Al)

Acceptor energy level



Energy level diagram

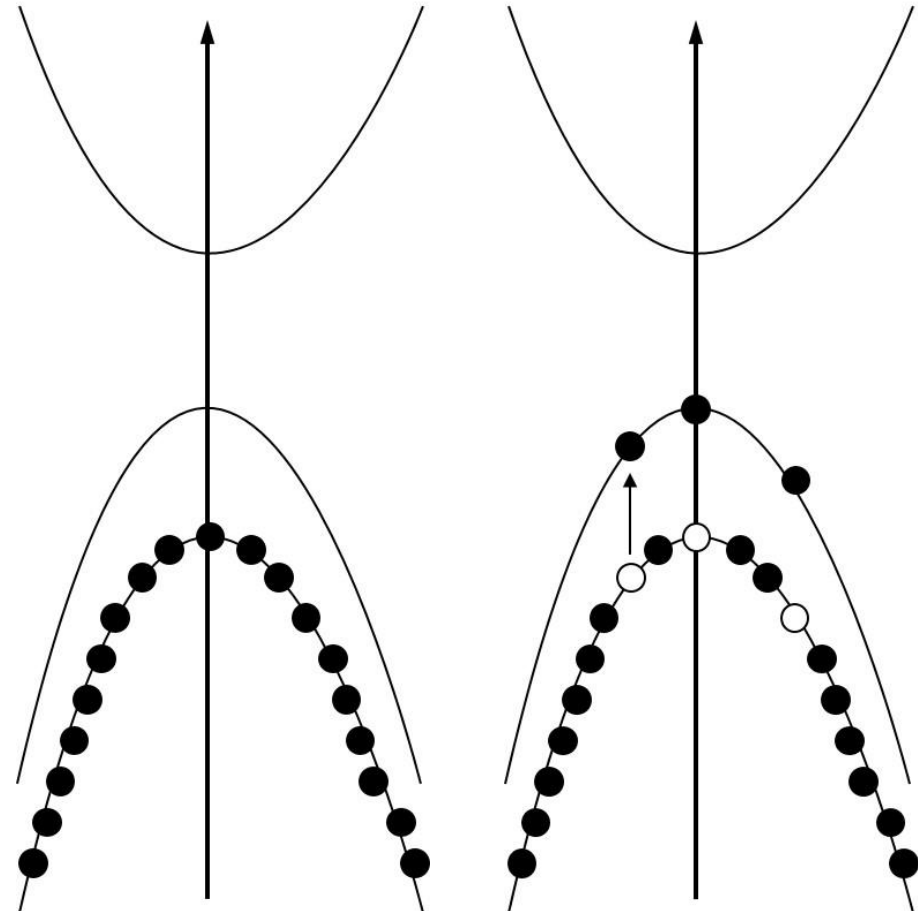
E_a : acceptor level

Saturation regime:

$p \sim N_A$ (acceptor concentration)

$n \sim n_i^2/N_A$

Holes are *majority* carriers,
electrons are *minority* carriers.
Semiconductor is *p*-type.



$T = 0$

Low T
(freeze-out)

Chemical potential in *p*-type extrinsic semiconductor

E_c —————

E_a - - - - -

E_v ● ● ● ● ● ● ●

p-type semiconductor at 0K
(Fermi level below E_a)

From

$$p = N_v \exp \left[-\frac{(\mu - E_v)}{kT} \right]$$

and substituting $p \sim N_A$

$$\mu = E_v + kT \ln \left(\frac{N_v}{N_A} \right)$$

