

**Level 3 Condensed Matter Physics- Part I**  
**Weekly problem 5**

(1) What is meant by  $L$ - $S$  (or Russell Saunders) coupling and under what conditions is it applicable? **[1 mark]**

(2) The magnetisation of a paramagnet consisting of isolated atoms in their ground state with total angular momentum  $J$  is given by the Brillouin function,

$$M = M_S \left[ \frac{(2J + 1)}{2J} \coth \left( \frac{(2J + 1)}{2J} y \right) - \frac{1}{2J} \coth \left( \frac{y}{2J} \right) \right]$$

where  $M_S$  is the saturated magnetisation value  $ng\mu_B J$ ,  $y = g_J\mu_B J B / (k_B T)$ ,  $B$  is the magnetic induction field and  $g_J$  is the Landé  $g$ -factor which is given by the expression,

$$g_J = 1 + \frac{J(J + 1) - L(L + 1) + S(S + 1)}{2J(J + 1)}.$$

Use the expression for  $M$  to derive Curie's law, stating any assumptions that you make.

[Hint:  $\coth x \simeq \frac{1}{x} + \frac{x}{3}$  for  $x \rightarrow 0$ ]

**[2 marks]**

(3) Use Hund's rules to determine the ground state and excited total angular momentum states of an isolated  $\text{Ti}^{2+}$  ion which has a  $3d^2$  electronic structure. Calculate the total magnetic moments of each of these states in units of  $\mu_B$ . **[3 marks]**

(4) The spin-orbit energy is given by  $E_{\text{SO}}(J) = \frac{\lambda}{2} [J(J + 1) - L(L + 1) - S(S + 1)]$ . For a  $\text{Ti}^{2+}$  ion the spin-orbit coupling constant  $\lambda = 4.5$  meV. Calculate the energies  $E_{\text{SO}}(J)$  of the total angular momentum states determined in part (3). **[2 marks]**

(5) Sketch the form of the temperature dependence of the inverse paramagnetic susceptibility of a solid of non-interacting  $\text{Ti}^{2+}$  ions, measured at low magnetic fields. Note that the probability of thermal occupation of the energy levels, determined in part (4), is proportional to  $(2J + 1)\exp(-E_{\text{SO}}(J)/k_B T)$ . **[2 marks]**