

Statistical Physics: Weekly Problem 6 (SP6)

- (1) In this question you will need to evaluate integrals of the form

$$I_n(\alpha) = \int_0^\infty x^n e^{-\alpha x^2} dx.$$

Simply look up the expressions for these standard integrals and use them (they are also given in the lectures).

The probability distribution of speeds of particles in a gas is given by the Maxwell-Boltzmann (MB) distribution

$$p(v)dv = \sqrt{\frac{2}{\pi}} \left( \frac{m}{k_B T} \right)^{3/2} v^2 \exp \left( -\frac{mv^2}{2k_B T} \right) dv.$$

- (a) Sketch  $p(v)$  as a function of  $v$  for (i) a low temperature and (ii) a high temperature. Indicate the  $v$  dependence for small  $v$ , and the trend at large  $v$ . [2 marks]
- (b) The expressions for the most probable speed  $v_{max}$ , the mean speed  $\bar{v}$ , and the r.m.s. speed  $v_{rms}$  were evaluated in lectures. State them here and mark them on the graph. [2 mark]
- (c) Calculate  $v_{max}$ ,  $\bar{v}$ , and  $v_{rms}$  for a system composed of gaseous neon atoms at room temperature. The mass of a neon atom is  $3.37 \times 10^{-26}$  kg. [2 marks]
- (d) In two dimensions the probability distribution of speeds of particles in a gas is

$$p(v) dv = C v \exp \left( -\frac{mv^2}{2k_B T} \right) dv$$

where  $C$  normalises the probability.

- (i) Calculate  $C$ . [2 marks]
- (ii) Find the expressions for the most probable speed  $v_{max}$ , the mean speed  $\bar{v}$ , and the r.m.s. speed  $v_{rms}$  of particles in this two dimensional gas. [2 marks]