## Level 3 Condensed Matter Physics- Part I Weekly problem 5 solutions

- (1) **L-S coupling**; here the coupling of the **l** vectors of the individual electrons combine to form a total angular momentum  $\mathbf{L}$ , while the spin momentum vectors combine independently to form a total spin momentum vector  $\mathbf{S}$ . This means that the calculated  $\mathbf{S}$  is not affected by the value of  $\mathbf{L}$  i.e. the spin-spin and orbit-orbit couplings are much stronger than the spin-orbit coupling for each electron. This leads to a total angular momentum  $\mathbf{J}$  of an atom is the vector sum of the two non-interacting momenta  $\mathbf{L}$  and  $\mathbf{S}$ ,  $\mathbf{J} = \mathbf{L} + \mathbf{S}$ . [1 mark]
- (2) We are given that,

$$M = M_S \left[ \frac{(2J+1)}{2J} \coth\left(\frac{(2J+1)}{2J}y\right) - \frac{1}{2J} \coth\left(\frac{y}{2J}\right) \right]$$

where  $y = \frac{g_J \mu_B J B}{(k_B T)}$  and  $M_S = ng \mu_B J$ . At small magnetic fields and/or high temperatures y is very small. This means we can use the result,

$$\coth x \approx \frac{1}{x} + \frac{x}{3} \text{ for } x \to 0$$

To obtain,

$$\begin{split} M \approx n g_J \mu_{\rm B} J \left[ \frac{2J+1}{2J} \left( \frac{2J}{(2J+1)y} + \frac{(2J+1)}{2J} \times \frac{y}{3} \right) - \frac{1}{2J} \left( \frac{2J}{y} + \frac{1}{2J} \times \frac{y}{3} \right) \right] \\ = n g_J \mu_{\rm B} J \left( \frac{(J+1)}{J} \times \frac{y}{3} \right). \end{split}$$

Substituting in for y gives,

$$M \approx \frac{ng_J^2 \mu_B^2 J(J+1)B}{3k_B T} \ .$$

[1 mark]

Substituting  $B = \mu_0 H$  into this expression and re-arranging gives the paramagnetic susceptibility,

$$\chi_P = \frac{M}{H} = \frac{ng_J^2 \mu_0 \mu_B^2 J(J+1)}{3k_B T},$$

Which has the form of Curie's law,  $\chi = {}^{C}/_{T}$ 

[1 mark]

(3)  $Ti^{2+} 3d^2$  (note that the azimuthal quantum number l = 2 for the d-shell):

$m_l$	-2	-1	0	1	2
S				+1/2	+1/2

Ground state:

$$S = \sum s = 2 \times \frac{1}{2} = 1$$
,  $L = \sum m_l = 2 + 1 = 3$ ,  $J = |L - S| = |3 - 1| = 2$ .

Excited states: J = |L - S| + 1 = 3, J = L + S = 4 [2 marks]

Total magnetic moment of an atom,  $|m_I| = g_I \mu_B \sqrt{J(J+1)}$  and

$$g_J = 1 + \frac{J(J+1) - L(L+1) + S(S+1)}{2J(J+1)}.$$

$$J=2\ , g_J=1+\frac{2(2+1)-3(3+1)+1(1+1)}{4(2+1)}=0.67\ , |m_2|=0.67\times \sqrt{2(2+1)}\ \mu_{\rm B}$$
 
$$=1.64\ \mu_{\rm B}.$$

$$J=3 \ , g_J=1+\frac{3(3+1)-3(3+1)+1(1+1)}{6(3+1)}=1.08 \ , |m_2|=1.08\times\sqrt{3(3+1)}\ \mu_{\rm B}$$
 
$$=3.74\ \mu_{\rm B}.$$

$$J=4~, g_J=1+\frac{4(4+1)-3(3+1)+1(1+1)}{8(4+1)}=1.25~, |m_2|=1.25\times\sqrt{4(4+1)}~\mu_{\rm B}\\ =5.59~\mu_{\rm B}.$$

[1 mark]

(4) Energy states: 
$$E_{SO}(J) = \frac{\lambda}{2} [J(J+1) - L(L+1) - S(S+1)]$$

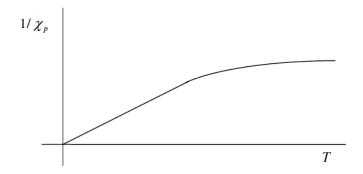
$$J = 2, E_{SO}(2) = \frac{4.5}{2} [2(2+1) - 3(3+1) - 1(1+1)] = -18 \text{ meV}$$

$$J = 3, E_{SO}(3) = \frac{4.5}{2} [3(3+1) - 3(3+1) - 1(1+1)] = -4.5 \text{ meV}$$

$$J = 4, E_{SO}(4) = \frac{4.5}{2} [4(4+1) - 3(3+1) - 1(1+1)] = 13.5 \text{ meV}$$

[2 marks]

(5) Given the form of the probability,  $P(J) \propto (2J+1) \exp(-E_{SO}(J)/k_BT)$ , it would be expected that, at very low temperatures, only the ground J=2 state, would be substantially populated and the ground state magnetic moment would dominate the paramagnetic susceptibility with  $\frac{1}{\chi_P} = \frac{3k_BT}{\mu_0Nm_2^2}$ . At higher temperatures we would expect the contribution from the J=3 state, and given the greater atomic moment, this would cause the measured paramagnetic susceptibility to be larger than that expected for the J=2 state. Hence the inverse susceptibility would have the form:



[2 marks]