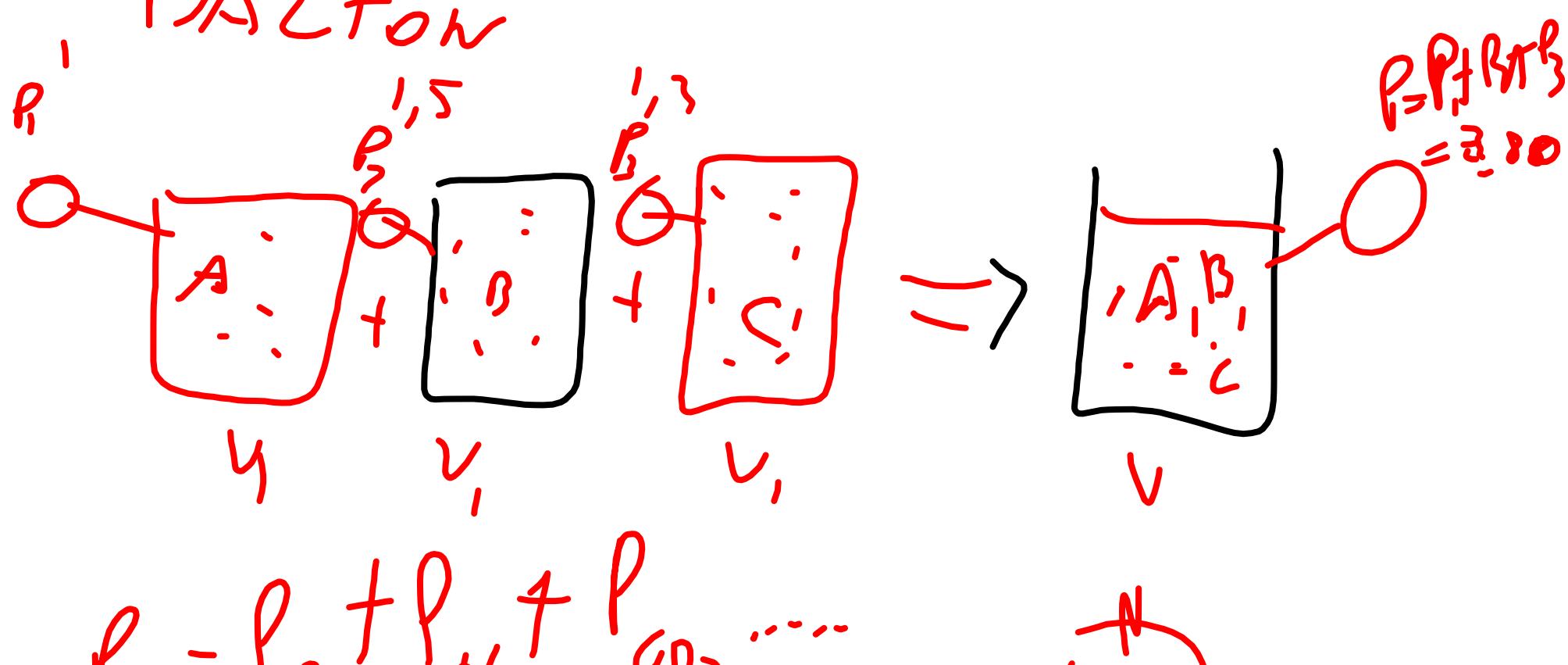
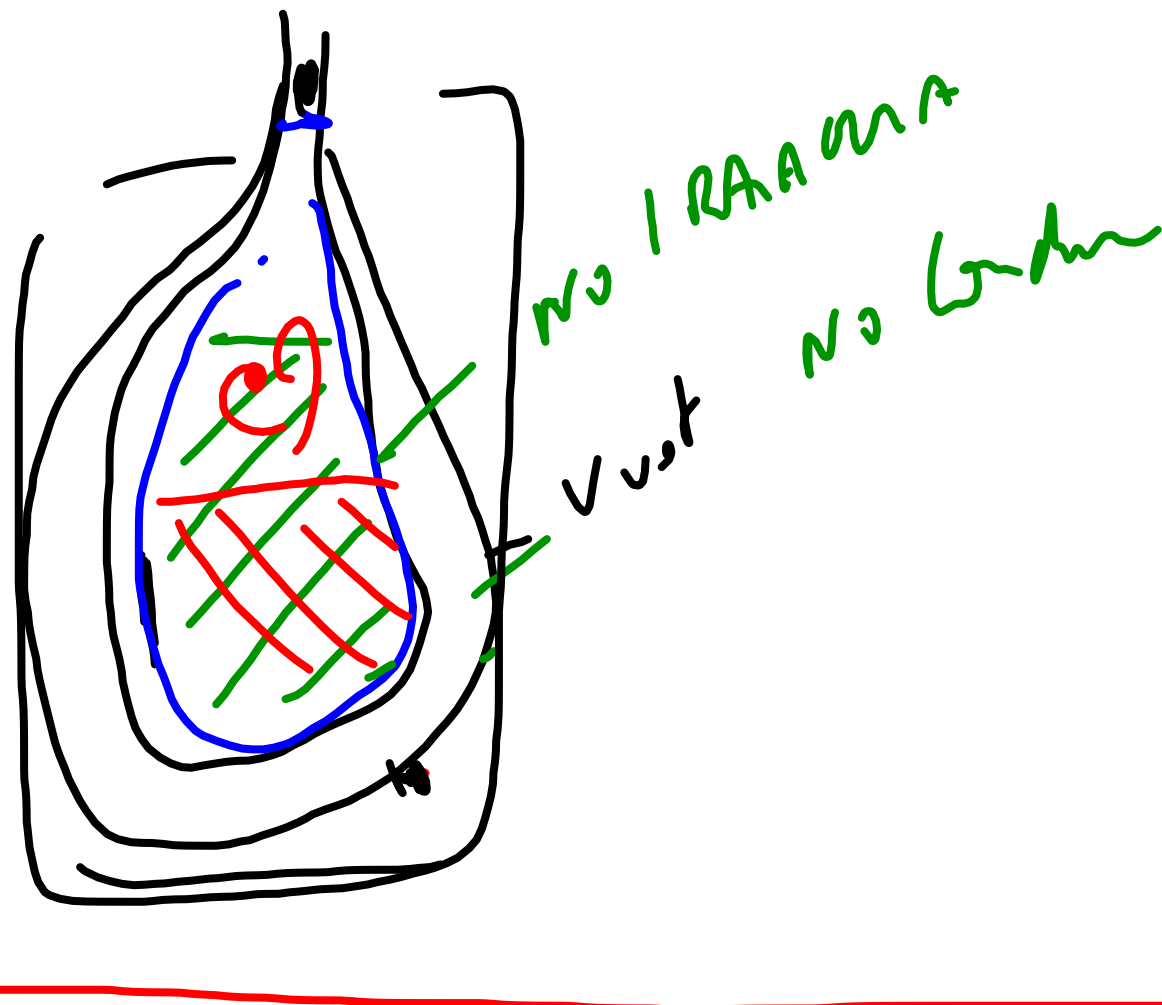
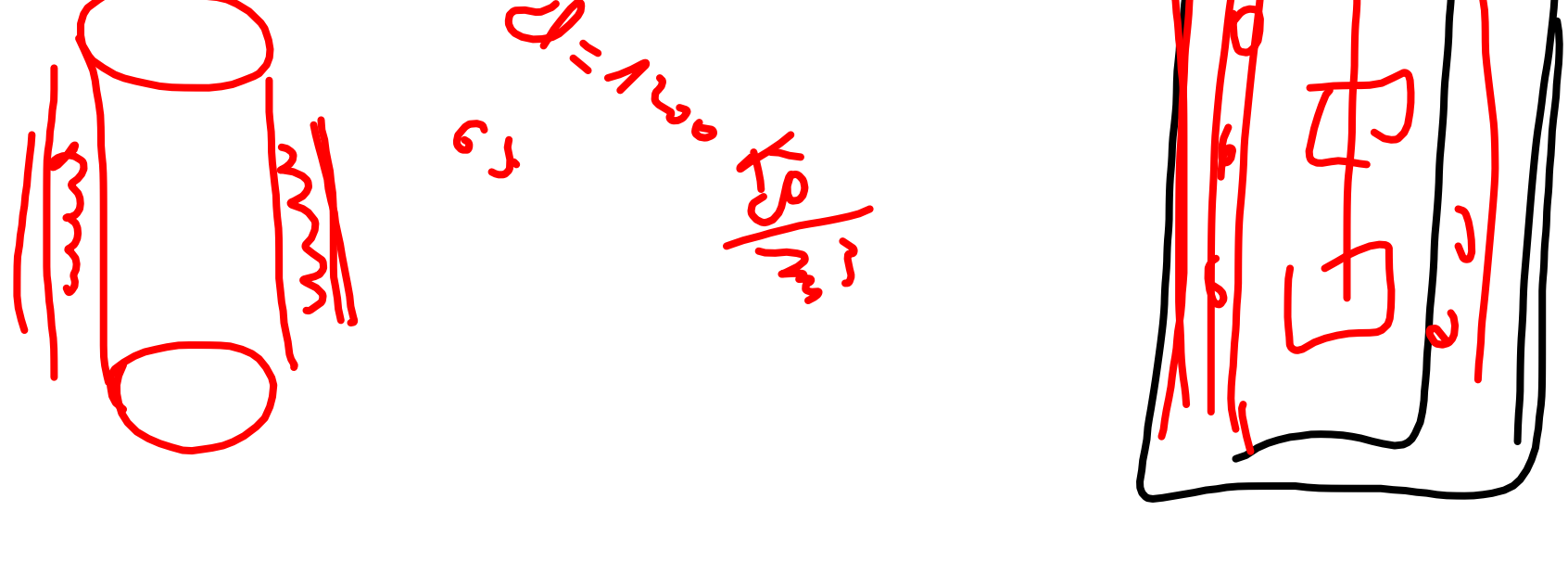


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 ISOBARA
 ISOBARA
 ISOTERMA
 ADIABATI

$V = k$
 $P = k$
 $T = k$
 N, Q

P, T
 V, T
 V, P



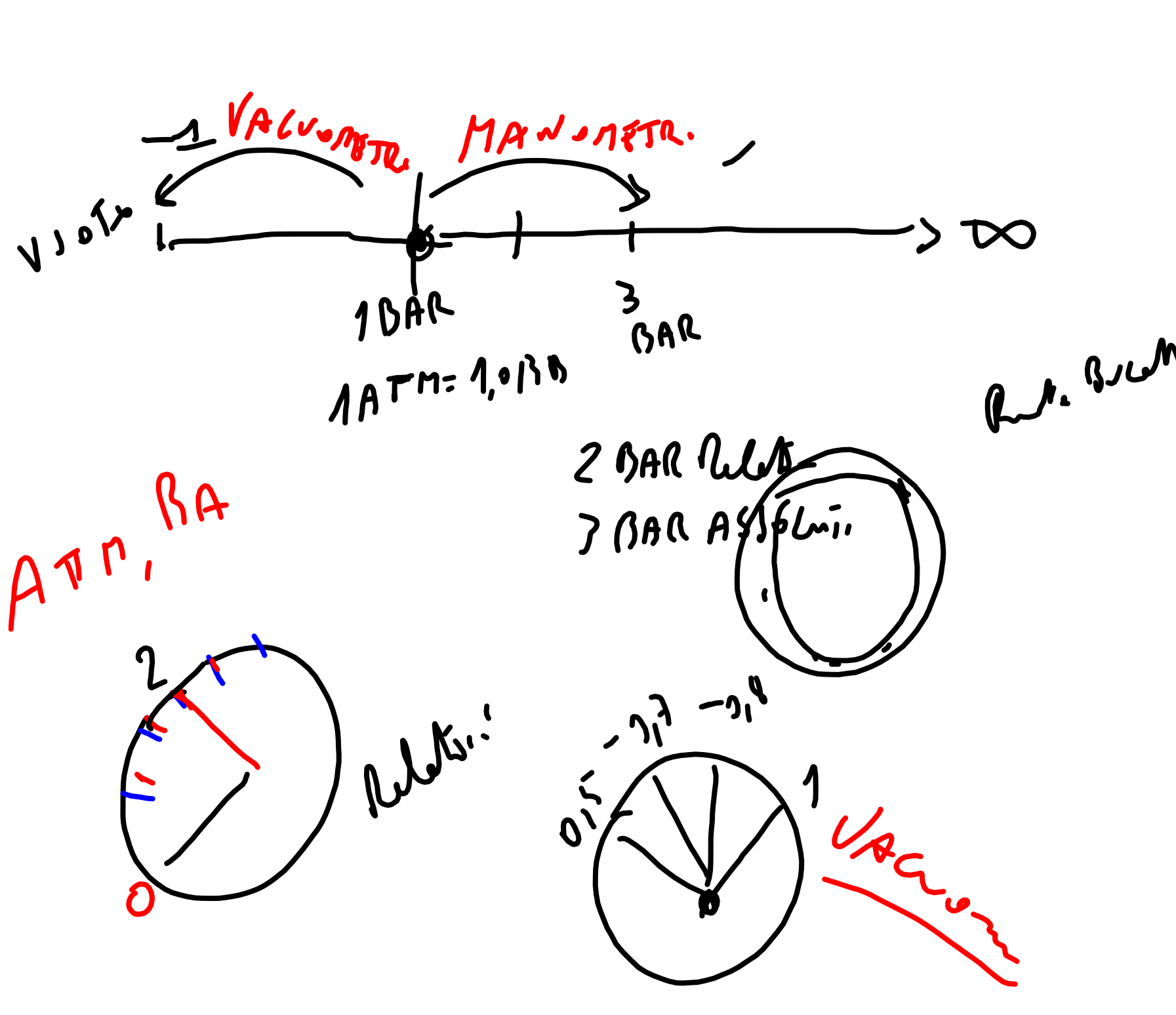
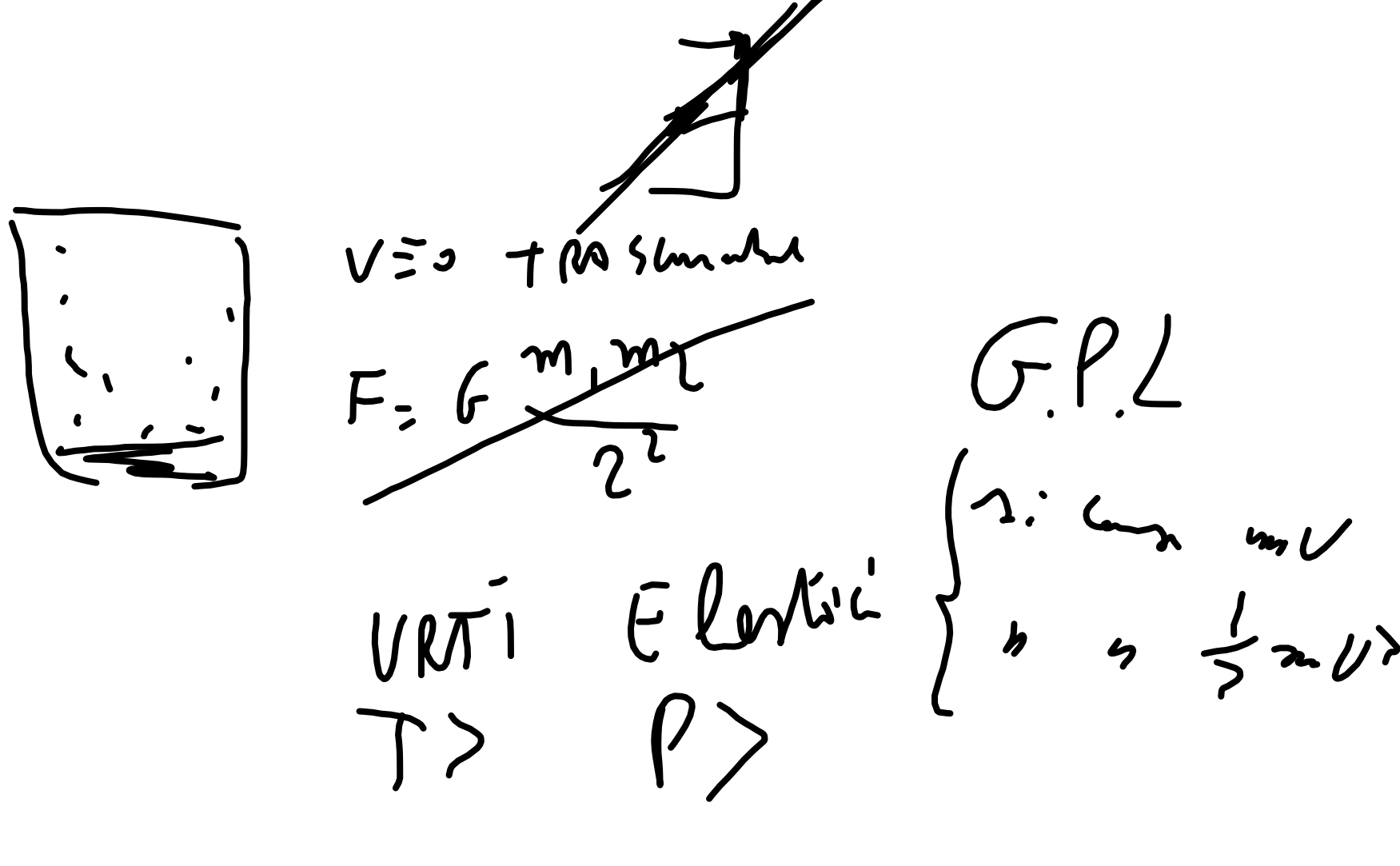
AVOGADRO

Mole $\rightarrow N_A = 6,02 \cdot 10^{23}$

$1 O_2, 1 CO_2, 1 H_2SO_4$
 $P_{atm}, t = 0^\circ C, V = 20,4 \text{ litri}$

$M_H = 1,008 \text{ gr} = 1,008 \cdot 10^{-3} \text{ Kg}$
 $\frac{M_H}{N_A} = 1,67 \cdot 10^{-27} \text{ Kg}$

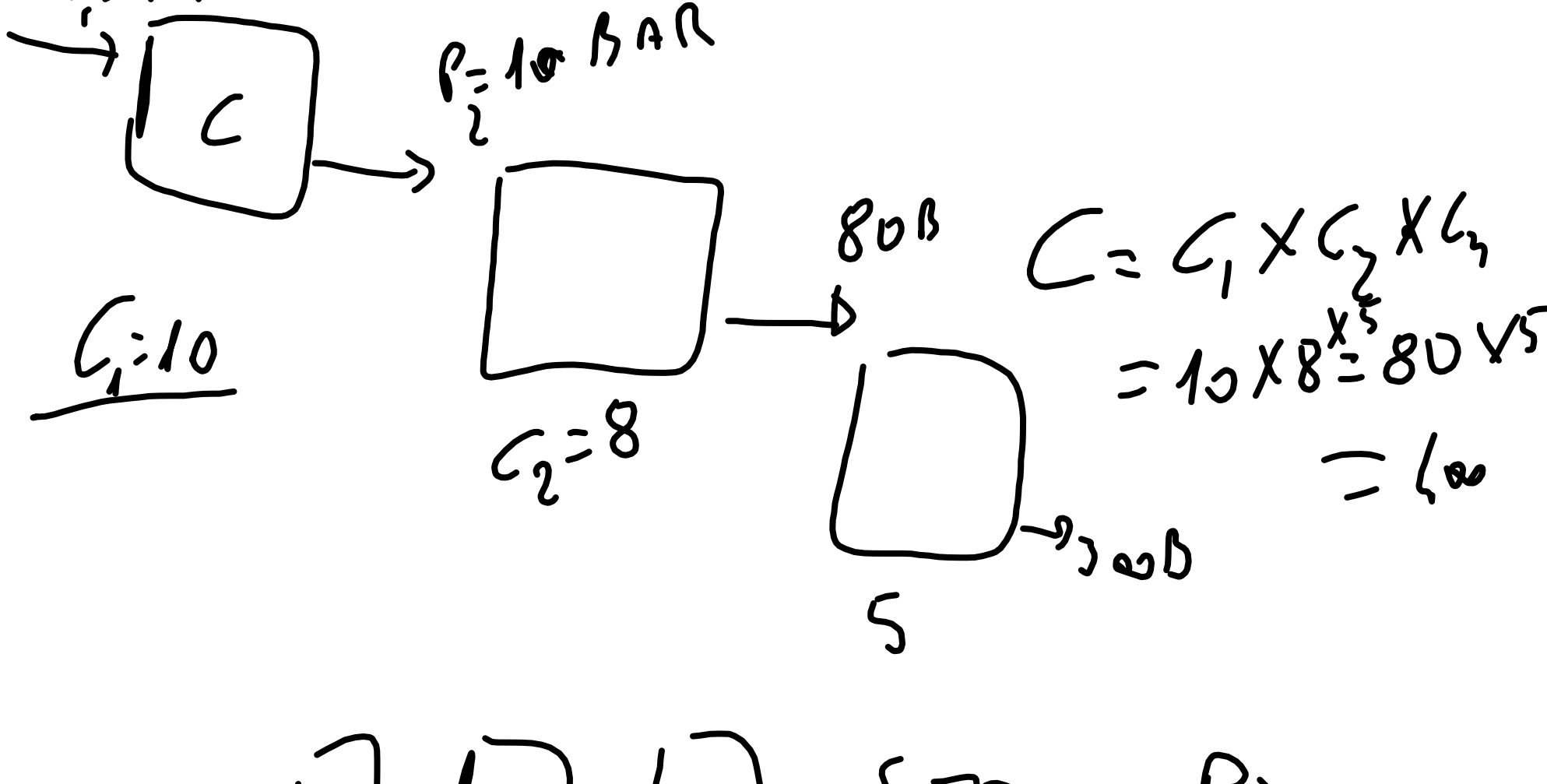
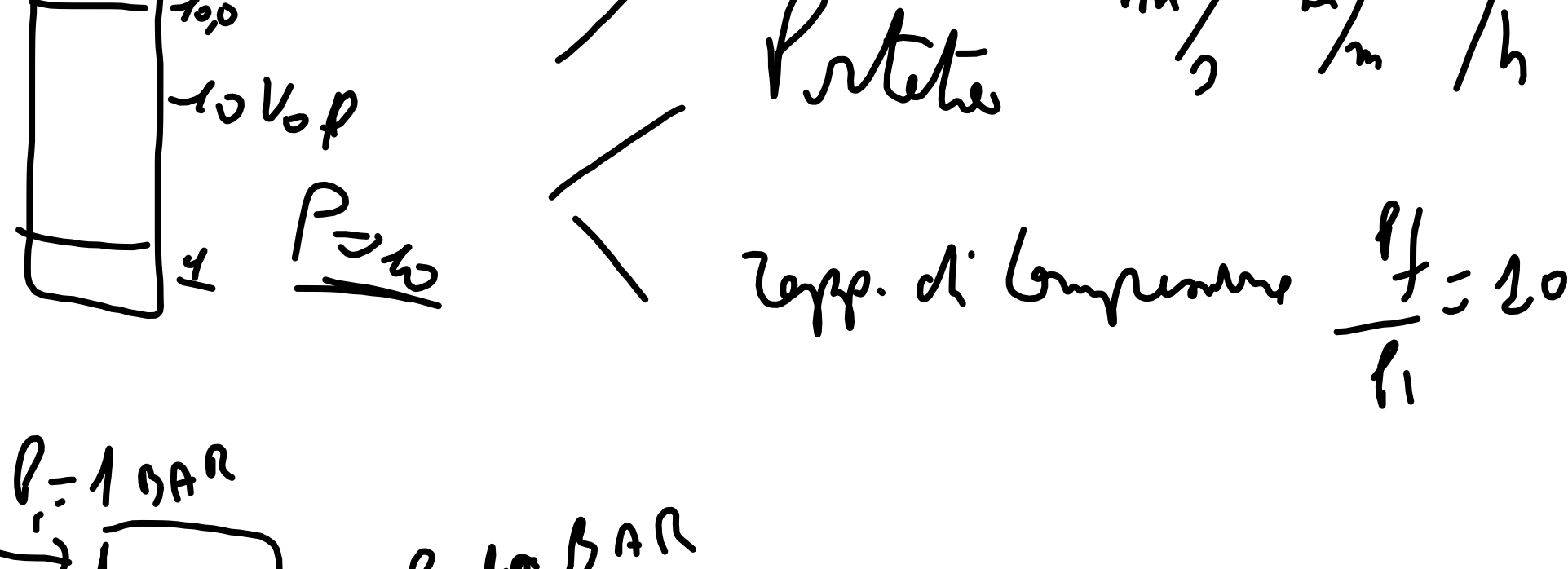
GAS PERFETTI MOLE PERFETTE



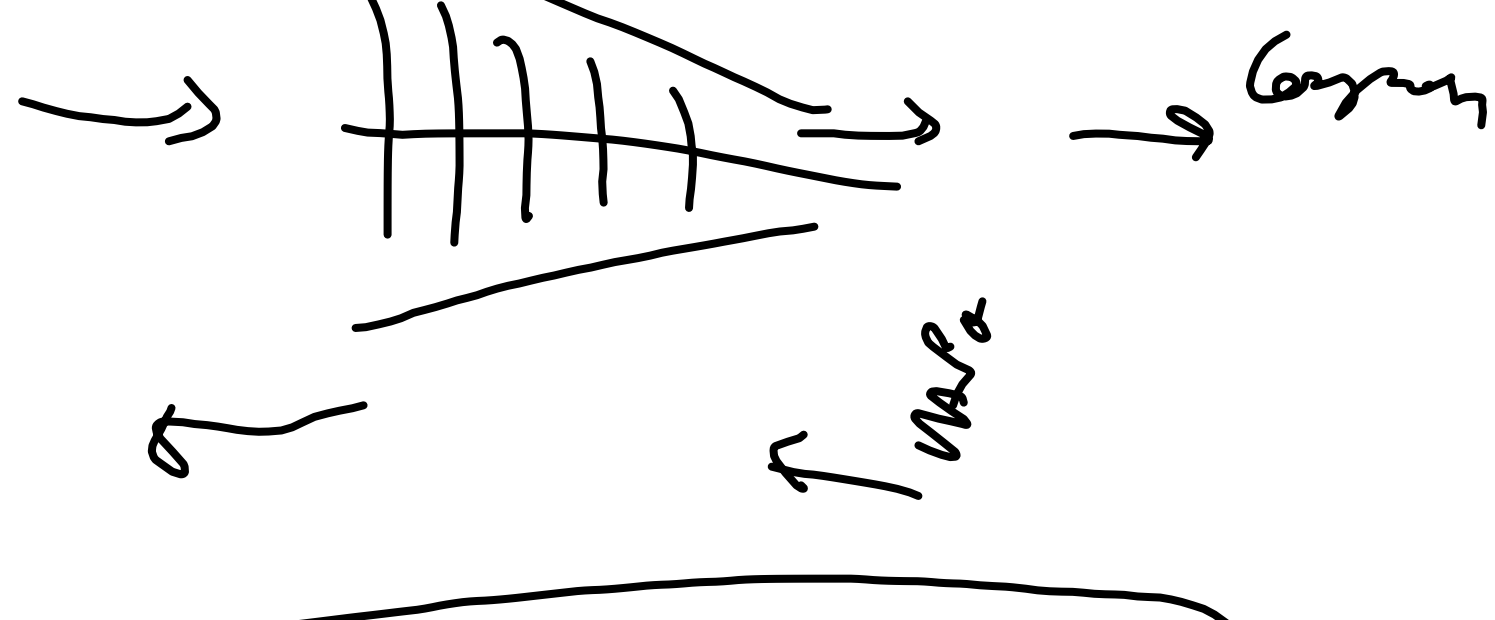
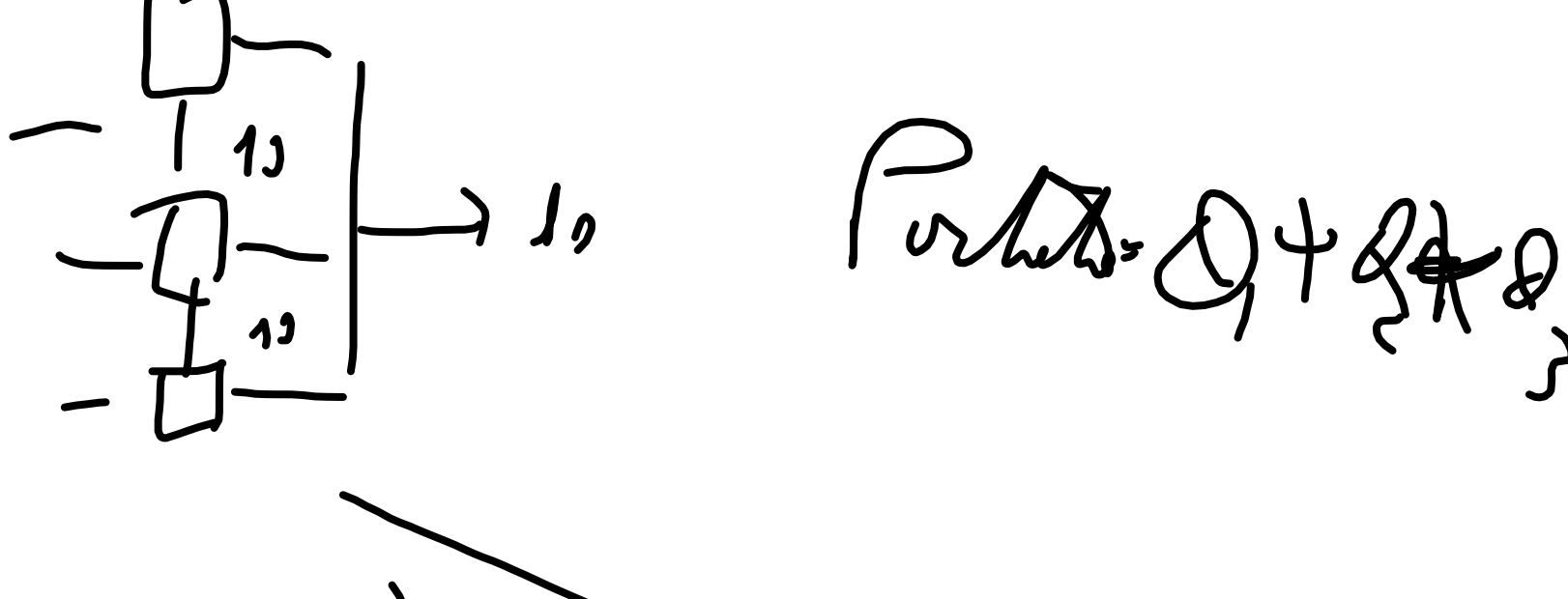
$P_A = P_2 + 1$
 $P_2 = P_A - 1$

Compressore

Potenza \rightarrow



SERIE P >



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