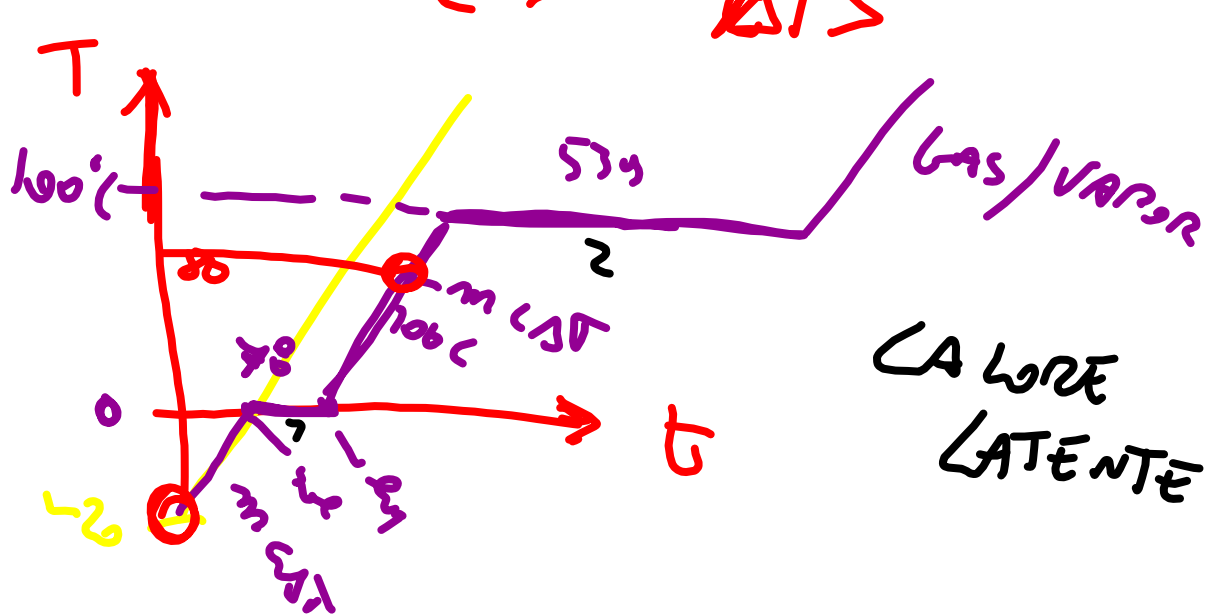


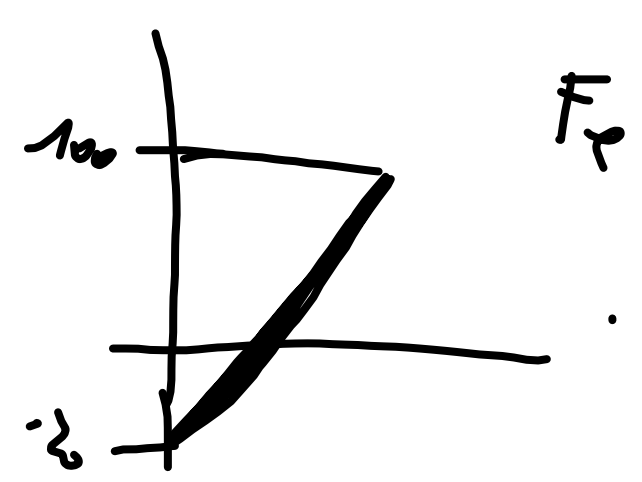
NAFTAUNA
GHIACCIO DE' CLO
CO₂
T ≈ -56,6

T - 20 H₂O
Condens
W = $\frac{L}{t}$
L = W t (J)
Energia

$Q = E = m c \Delta T = k \Delta T$
 $E > \Delta T$



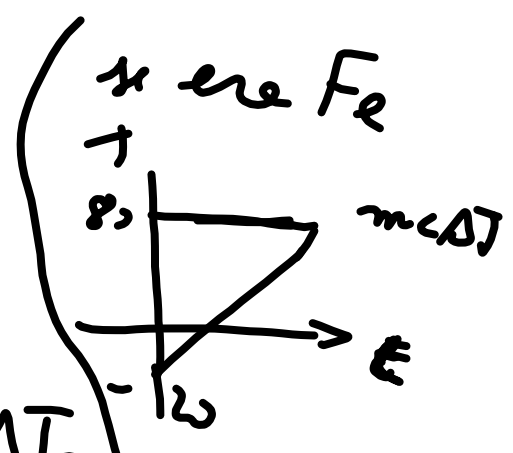
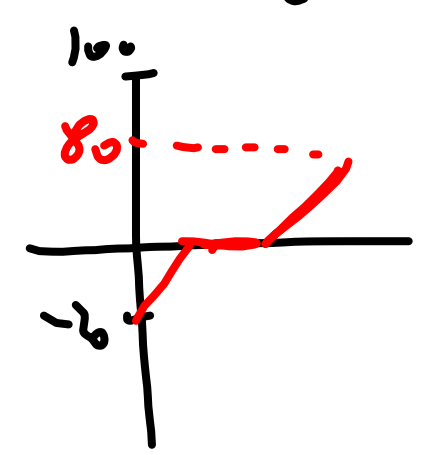
$\Delta E_1 = m \lambda_f$
 $\Delta E_2 = m \lambda_{ev}$



MATERIALI
T_f λ_f
T_e λ_e

m = 7 kg
H₂O
- 20 a 80
ΔE = ?

H₂O
T_f 0°C
T_e 100°C



$\Delta E = m c \Delta T_1 + m \lambda + m c \Delta T_2$
 $= 2000 [0 - (-20)] + 7 \cdot 334 +$

$4186 \cdot (80 - 0) =$

$\frac{J}{kg}$
 $\frac{J}{kg}$
 $\frac{J}{kg}$

= J

300 Ton FRUTTA H₂O

T = 30°C
T_f = 4°C

$Q = m c \Delta T$
 $= 300000 \cdot 4186 (4 - 30)$
 $= - 3,77 \cdot 10^{10} J$

t = 22 h

$W = \frac{L}{t} = \frac{3,77 \cdot 10^{10}}{22 \times 60 \times 60} =$ W

η = 70%

$W_{am} = \frac{W}{\eta}$