Lista de fórmulas de interés

$$PV = nRT$$

$$P_{tot} = P_A + P_B + ... P_n$$

$$P_A = x_A P_{tot}$$

$$\frac{\mathbf{v}_1}{\mathbf{v}_2} = \sqrt{\frac{M_2}{M_1}}$$

Ec
$$_{prom} = 3/2 RT$$

$$(v_{prom}) = (3RT/M)^{1/2}$$

$$E_{ret} \propto - |\mathbf{q}_c \, \mathbf{q}_a / (r_c + r_a)|$$

$$q = m Ce \Delta T$$

$$\Delta E = q + w$$

$$w = - P \Delta V$$

$$w = -nRT \ln(Vf/Vi)$$

$$\Delta H_{reacc}$$
 = $\sum m \Delta H^{\circ} f \text{ prod - } \sum n \Delta H^{\circ} f \text{ react}$

$$\Delta S^{\circ}_{reacc}$$
 = $\sum m S^{\circ} f prod - \sum n S^{\circ} f react$

$$\Delta G^{\circ}_{reacc}$$
 = $\sum m \Delta G^{\circ} f prod - \sum n \Delta G^{\circ} f react$

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta E = \Delta H - RT \Delta n$$

$$K_{C} = \frac{[C]^{c} [D]^{d}}{[A]^{a} [B]^{b}}$$

$$K_{p} = K_{c} (RT)^{\Delta n}$$

$$pH = -log [H_3O^+]$$

$$\Delta E^{\circ}$$
celda = E° cátodo - E° ánodo

$$\Delta$$
Ecelda = Δ E°celda –(RT/nF) ln Q

$$\Delta$$
Ecelda = Δ E°celda –(0.059/n) log Q

$$I = Q / t$$