Thermodynamics and Statistical Physics

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The basic concepts in Thermodynamics

Lecture 1: Basic concepts in thermodynamics

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- 1.1.1 Macroscopic picture
- 1.1.2 Thermodynamic equilibrium
- 1.1.3 Equation of state
- 1.2 Thermodynamic processes
- 1.2.1 Reversibility versus irreversibility
- 1.2.2 Some definitions
- 1.3 The laws of thermodynamics at a glance
- 1.3.1 Zeroth law
- 1.3.2 First law
- 1.3.3 Second law
- 1.3.4 Third law



The basic laws of thermodynamics

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Universal thermodynamic identities

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Thermodynamics of the ideal classical gas

3.4.1 Calorimetric coefficients

Thermodynamic potentials and equilibrium

Lecture 4: Thermodynamic potentials

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- 4.1.1 Law of entropy increase
- 4.1.2 Equilibrium conditions of two coupled systems
- 4.1.3 About the zeroth law of thermodynamics
- 4.1.4 Stability conditions of the thermodynamic equilibrium
- 4.2 Closed systems
- 4.2.1 Preliminary: Mechanical equilibrium
- 4.2.2 Closed isochoric systems (or mechanically isolated)
- 4.2.3 Closed isobaric systems
- 4.3 Generalization and application to open systems
- 4.3.1 Generic thermodynamic system
- 4.3.2 Open systems

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