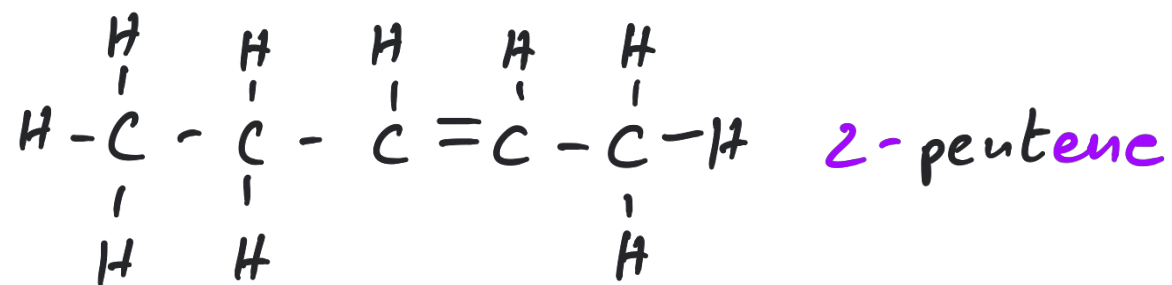
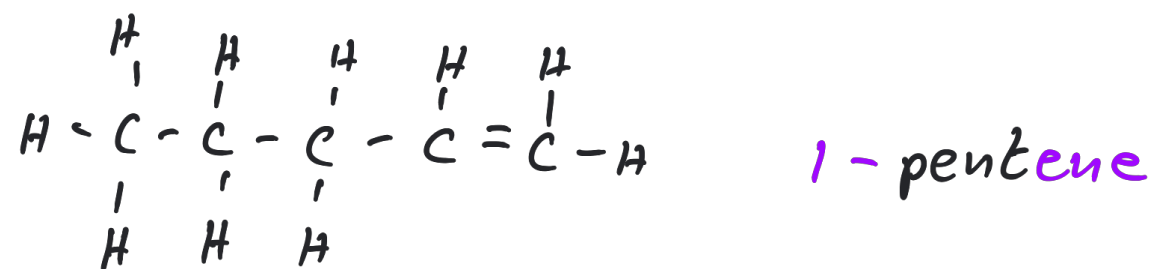


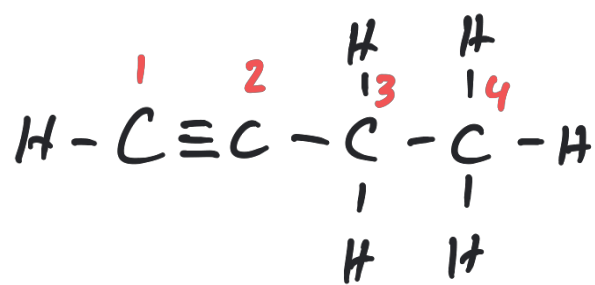
## Alkenes $C_nH_{2n}$ suffix -ene

- contain at least one double bond
- naming: root must contain both C-atoms of the double bond, even if it is not the longest chain
- chain is numbered from the end closer to the double bond; position is indicated by first C-atom in it

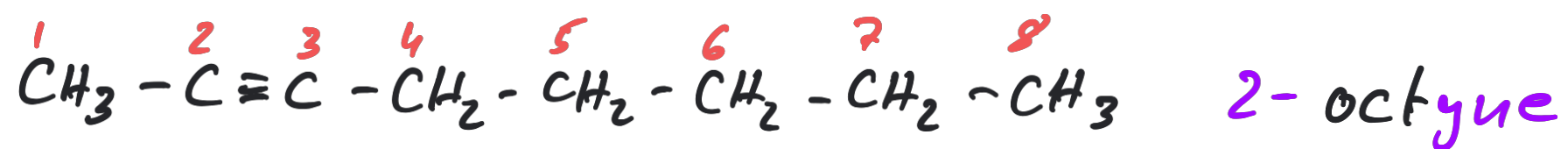


Alkynes  $C_nH_{2n-2}$  suffix -yne

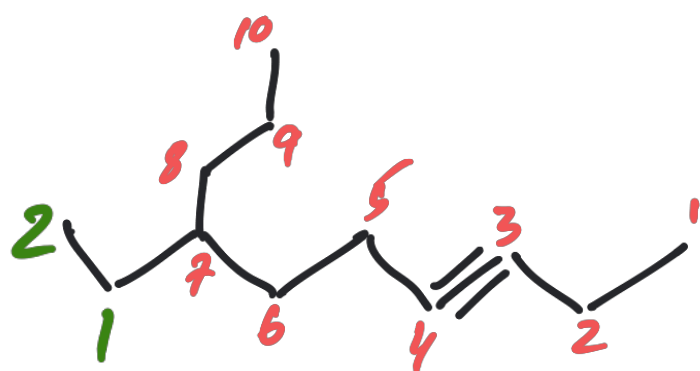
- contain at least one triple bond
- named following the same rules as for alkenes except for the suffix



1-butyne

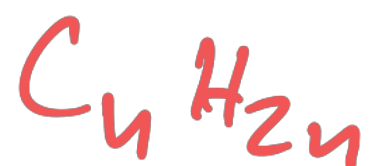


2-octyne

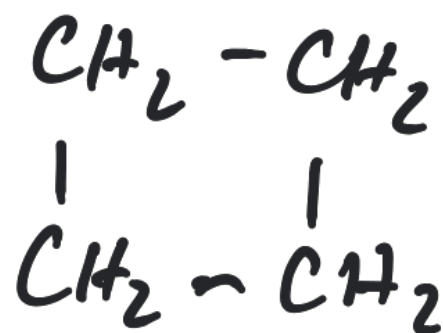


7-ethyl-3-decyne

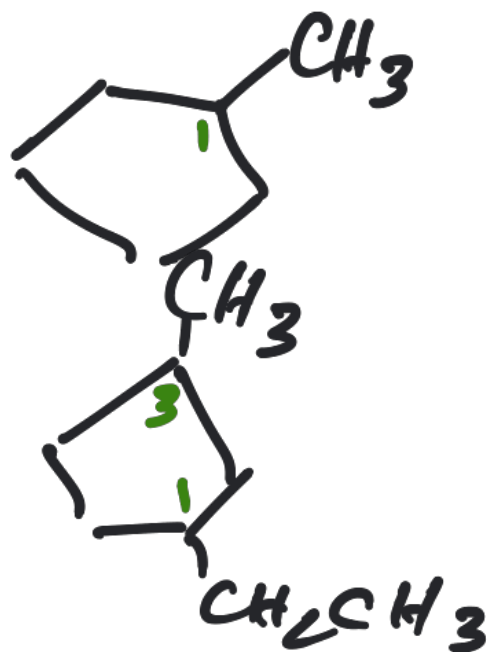
## Cycloalkanes



- formed when straight chain hydrocarb. form a ring
- prefix - cyclo      suffix - ane



cyclobutane

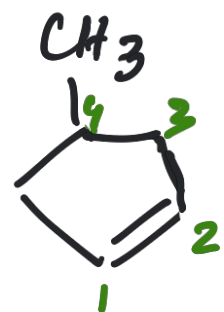


1-methyl-cyclopentane

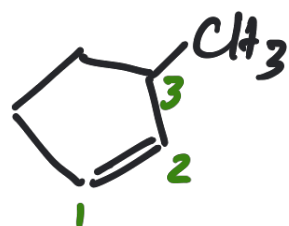
1-ethyl-3-methyl cyclopentane



cyclopentene



4-methylcyclopentene



3-methylcyclopentene

# Functional Groups

Halohydrocarbons - F, -Cl, -Br, -I

$\text{CH}_3\text{CH}_2\text{CH}_2\text{-Cl}$  propyl chloride

but:

$\text{CH}_3\text{-CH}_2\text{-}\overset{2}{\underset{\text{Br}}{\text{CH}}}\text{-}\overset{1}{\text{CH}_3}$  2-bromo butane

Alcohols -OH R-OH -ol

$\text{CH}_3\text{CH}_2\text{CH}_2\text{-OH}$  propanol

 butanol

 2-butanol

## Ethers:

Functional group  $-O-$

General Formula  $R-O-R'$

Naming (R group 1) (R group 2) ether

$CH_3CH_2-O-CH_2CH_2-CH_3$  ethyl propyl ether

$CH_3-O-CH_3$  dimethyl ether

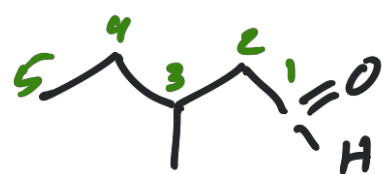
## Aldehydes:

Functional group  $\begin{array}{c} O \\ || \\ -C-H \end{array}$

General Formula  $R-\begin{array}{c} O \\ || \\ C-H \end{array}$

Naming -al

$CH_3CH_2\overset{\overset{O}{||}}{C}-H$  propanal



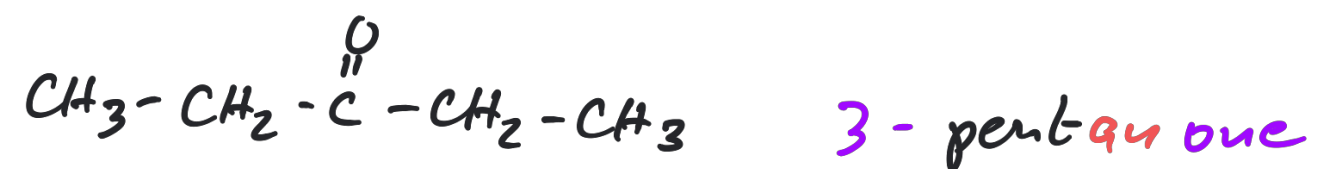
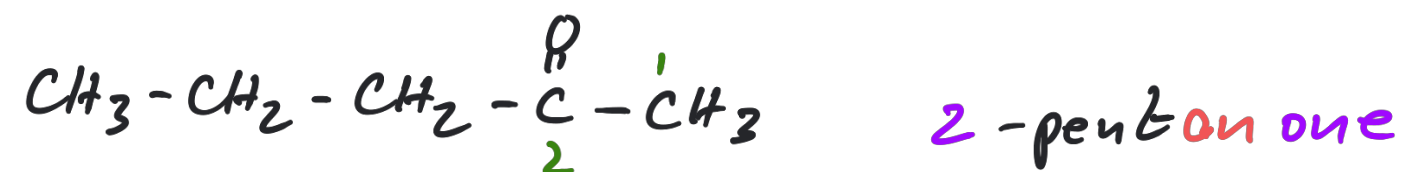
3-methylpentanal

Ketones:

Functional group  $\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}- \end{array}$

General Formula  $\text{R} - \begin{array}{c} \text{O} \\ \parallel \\ \text{C} \end{array} - \text{R}'$

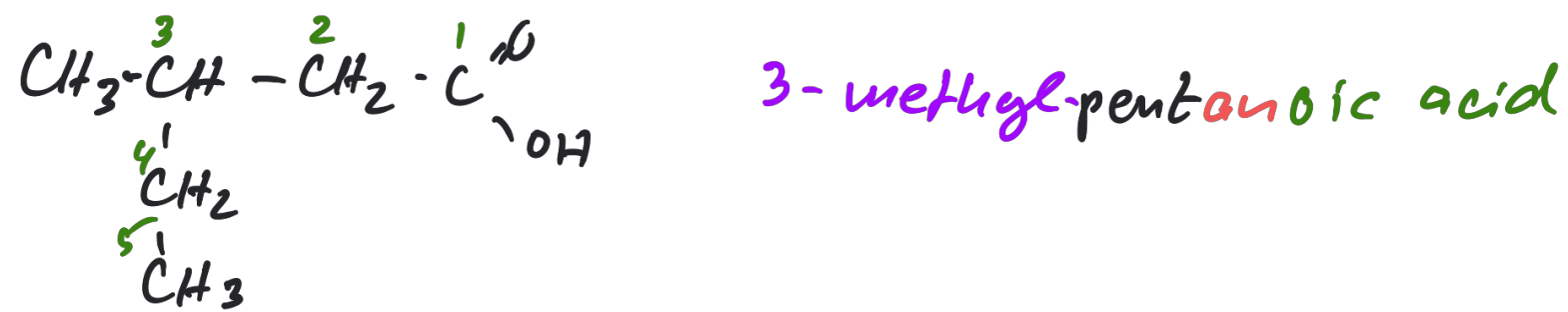
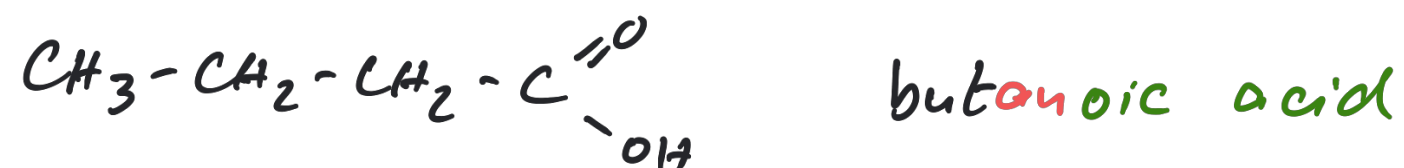
Naming -one



Carboxylic Acid

Functional group:  $\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{OH} \end{array}$

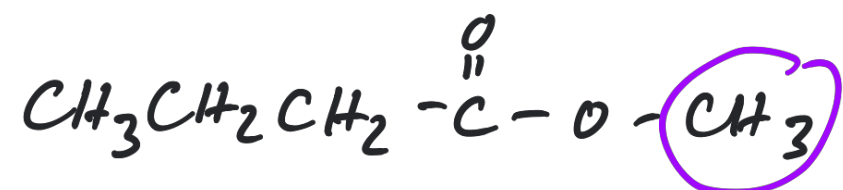
naming: -oic acid



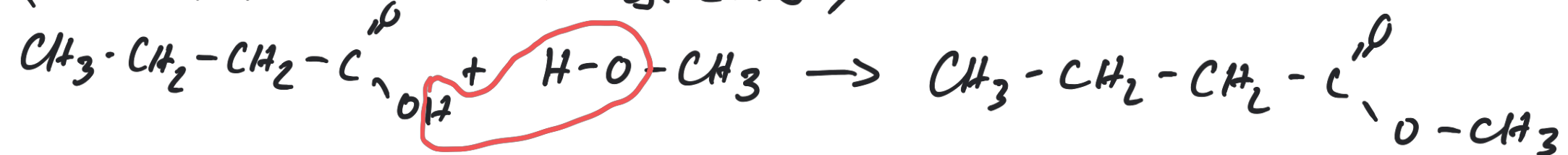
Ester



Naming: oate



metyl butanoate  
(butanoic acid methyl ester)

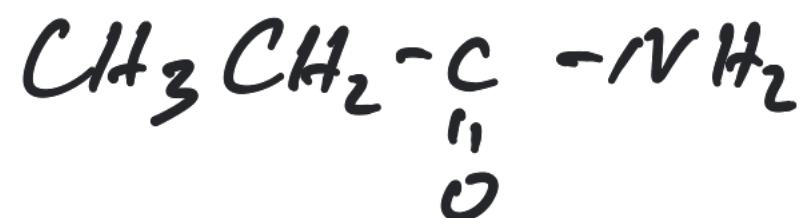
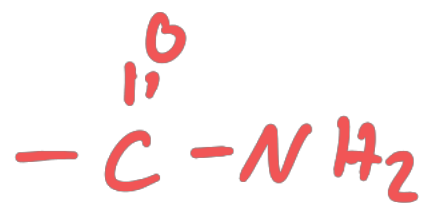


Amines:  $-\text{NH}_2$





## Amide



propanamide

Nitrile:  $-\text{C}\equiv\text{N}$



butanenitrile

## Amino Acids

