

## Chapter 21

### Organic Chemistry



About half of all men's colognes contain at least some patchouli alcohol ( $C_{15}H_{26}O$ ), an organic compound (pictured here) derived from the patchouli plant. Patchouli alcohol has a pungent, musty, earthy fragrance.

"Organic chemistry just now is enough to drive one mad. It gives one the impression of a primeval, tropical forest full of the most remarkable things...."

—Friedrich Wöhler (1800–1882)

#### ✓ Learning Outcomes

- 21.1 Fragrances and Odors
- 21.2 Carbon: Why It Is Unique
- 21.3 Hydrocarbons: Compounds Containing Only Carbon and Hydrogen
- 21.4 Alkanes: Saturated Hydrocarbons
- 21.5 Alkenes and Alkynes
- 21.6 Hydrocarbon Reactions
- 21.7 Aromatic Hydrocarbons
- 21.8 Functional Groups
- 21.9 Alcohols
- 21.10 Aldehydes and Ketones
- 21.11 Carboxylic Acids and Esters
- 21.12 Ethers
- 21.13 Amines

Key Learning Outcomes

**ORGANIC CHEMISTRY IS THE STUDY** of carbon-containing compounds. Carbon is unique in the sheer number of compounds that it forms. Millions of organic compounds are known, and researchers discover new ones every day. Carbon is also unique in the diversity of compounds that it forms. In most cases, a fixed number of carbon atoms can combine with a fixed number of atoms of another element to form many different compounds. For example, 10 carbon atoms and 22 hydrogen atoms can form 75 distinctly different compounds. With carbon as the backbone, nature can take the same combination of atoms and bond them together in slightly different ways to produce a huge variety of substances. In organic chemistry, we see the theme that structure determines properties played out over and over again. It is not surprising that life is based on the chemistry of carbon because life needs diversity to exist, and organic chemistry is nothing if not diverse. In this chapter, we peer into Friedrich Wöhler's "primeval tropical forest" (see chapter-opening quotation) and discover the most remarkable things.

Not for Distribution