Project Introduction

$FIREWOOD^{TM}$

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1 Prologue

1.1 Customer Traits

It is reasonable to assume that, due to the inherent nature of the products offered by FirewoodTM, its customers must exhibit one common trait, namely **the need for a mobile, localised source of heat energy**. This definition, despite not immediately coming across as inherently flawed, ignroes a very important parameter of the energy source, this being the (average) thermal power output.

1.2 Power Calculations

Consider a typical BIC lighter equipped with 4.5g of butane fuel. Said fuel amount is capable of maintaining a (more or less) constant power output throughout the duration of 30min. Upon combustion, one mole of butane fuel releases a fixed amount of energy into the environment. This energy is defined by the *enthalpy of combustion*...

$$\Delta H_{c(mol)} = -2.88 \cdot 10^3 \ kJ \cdot mol^{-1}$$

Given the molar mass of butane ...

$$M = 58.124 \, g \cdot mol^{-1}$$

A full tank of butane contains:

$$n = \frac{4.5 g}{58.124 g \cdot mol^{-1}}$$
$$= 7.74 \cdot 10^{-2} mol$$

And is thus capable of delivering a fixed amount of total energy:

$$\Delta H_c = \Delta H_{c(mol)} \cdot n$$
$$= -2.23 \cdot 10^2 \, kJ$$

Spreading that energy output out over a 30-minute burn time, the average power output equals . . .

$$\bar{P} = \frac{E}{t} = \frac{2.23 \cdot 10^2 \ kJ \cdot 10^3}{30min \cdot 60s}$$
$$= \frac{2.23 \cdot 10^5}{1.7 \cdot 10^3}$$
$$= 123.8 \ W$$

Assuming that the fuel consuption is not constant, the power formula may be expressed in a continuous form:

$$P(t) = \frac{dE}{dt}$$

Thus, the average power output becomes ...

$$\bar{P} = \frac{1}{T} \cdot \int_{0}^{T} P(t) dt$$