

* Wave aspect of particles \Rightarrow

de Broglie's Hypothesis (Matter wave) \Rightarrow

In 1923 de Broglie suggested that

the wave-particle duality is not restricted to radiation, but it must be universal.

All material particles should also display a dual wave-particle behaviour.

$$\text{for photon} \quad \left\{ \begin{array}{l} E = h\nu \\ p = \frac{h\nu}{c} = \frac{h}{\lambda} \end{array} \right.$$

Each material particle of momentum \vec{p} behaves as a group of waves (matter waves) whose wavelength λ & wavevector \vec{k}_2 will be -

$$\boxed{\begin{aligned} \lambda &= \frac{h}{p} = \frac{h}{m\omega} \\ \vec{k}_2 &= \frac{\vec{p}}{h} \end{aligned}}$$

$$\text{where, } \hbar = \frac{e_1}{2\pi}$$

This is de-Broglie relation. It connects the momentum of a particle with the wavelength & wave vector of the wave corresponding to this particle.

Homework

Q \Rightarrow Calculate the de Broglie wavelength for

(a) \Rightarrow A proton of K.E. 70 MeV. &

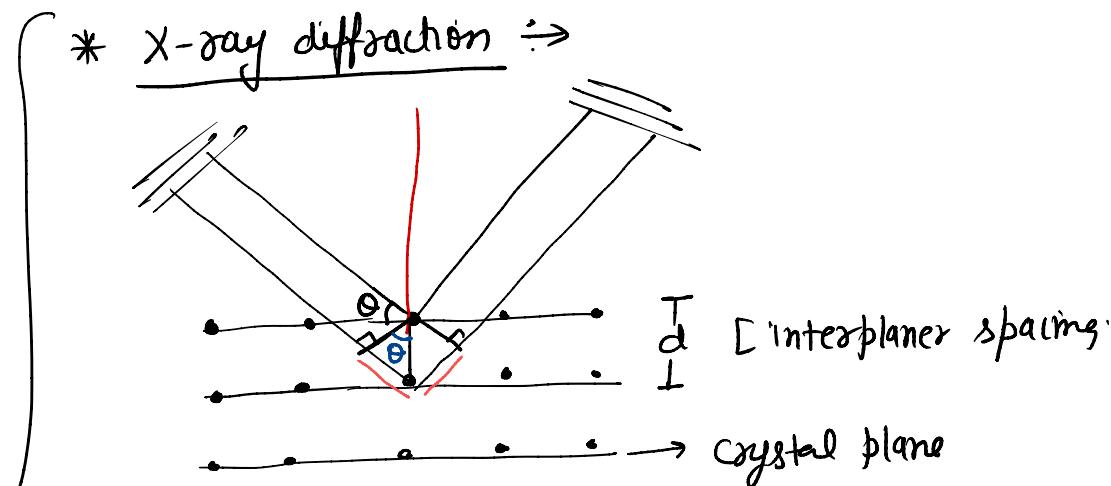
(b) \Rightarrow A 100g bullet moving at 900 m/sec.

$$\underline{\underline{A}} \Rightarrow (a) \Rightarrow 3.4 \times 10^{-15} \text{ m}$$

$$(b) \Rightarrow 7.4 \times 10^{-36} \text{ m}$$

Davisson-Germer Experiment \Rightarrow

This experiment confirmed the existence of de Broglie waves.

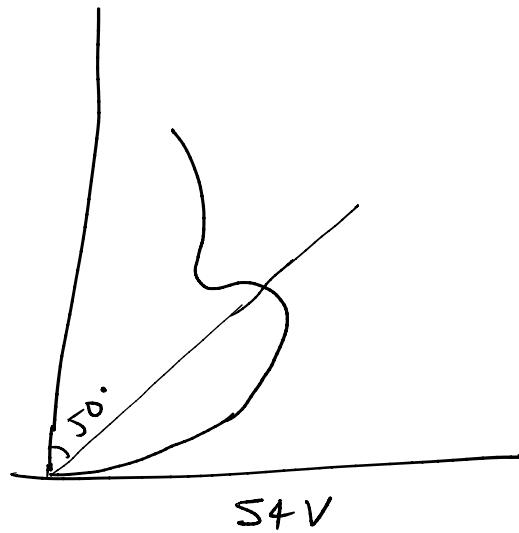


$$\text{path diff} = d \sin \theta + d \sin \theta$$

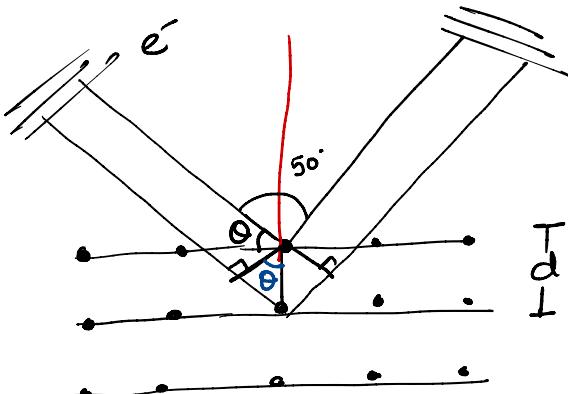
$$2d \sin \theta = n\lambda$$

n is order of diffraction.

condⁿ for constructive interference, Bragg's law.



* In the exp., energy of e^- , angle of incident & angle of detector could all be varied.



$$\theta = 90 - 25^\circ \\ = 65^\circ$$

* The spacing of the planes measured by the x-ray technique $d = 0.091 \text{ nm}$.

$$\lambda = 2 \times 91 \times \sin 65^\circ = 0.165 \text{ nm} \quad \text{--- ①}$$

* Now if we use de-Broglie formula to calculate the wavelength of e^- -

$$\lambda = \frac{h}{mv}$$

$$p = \sqrt{2mK}$$

$$= \sqrt{2 \times 9.1 \times 10^{-31} \times 54 \times 1.6 \times 10^{-19}}$$

$$= 4 \times 10^{-24} \text{ kg m/sec.}$$

$$\lambda = \frac{h}{p} = \frac{6.63 \times 10^{-34}}{4 \times 10^{-24}} = 1.66 \times 10^{-10} \text{ m} = 0.166 \text{ nm} \quad \text{--- ②}$$

$$K = \frac{1}{2}mv^2 = \frac{p^2}{2m}$$

~~Homework~~
Double-slit experiment with quantum particles like $e^- \Rightarrow$

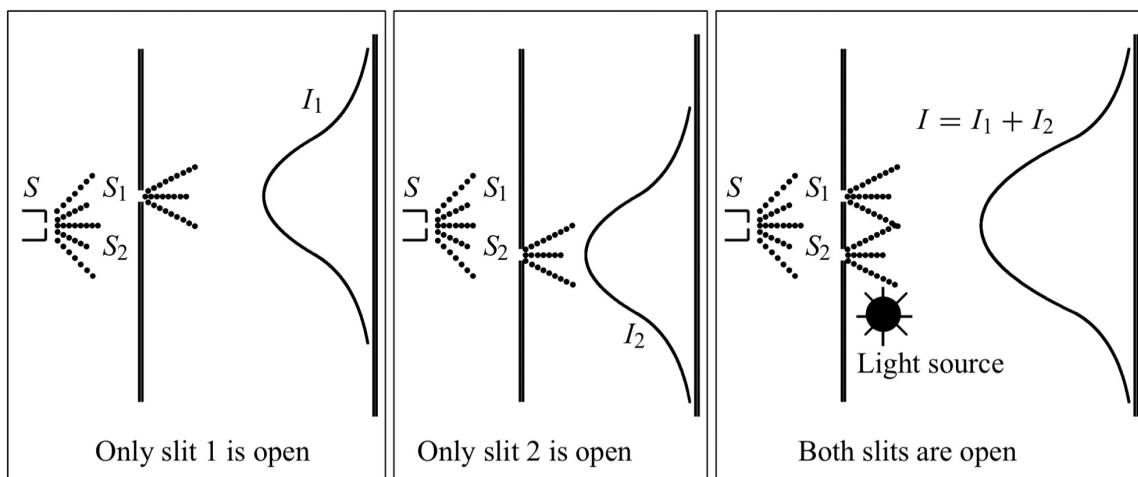


Figure 1.11 The double-slit experiment: S is a source of electrons. A light source is placed behind the wall containing S_1 and S_2 . When both slits are open, the interference pattern is destroyed and the total intensity is $I = I_1 + I_2$.