Chemica Reactions

Sept. 20

$$\underbrace{\mathcal{E}_{x}}: \quad x_{1} CH_{\mu} + x_{2}O_{a} \longrightarrow x_{3}CO_{a} + x_{4}H_{a}O$$

$$\xrightarrow{x_{1}} \quad x_{2} \quad x_{3} \quad x_{4}$$

Carbon:
$$x_1 = x_3$$

Hydrogen:
$$4x_1 = 2x_4$$

Oxygen:
$$\partial x_2 = \partial x_3$$

Carbon:
$$x_1 = x_3$$

Hydrogen: $4x_1 = 3x_4$

Oxygen: $3x_2 = 3x_3$
 $\begin{bmatrix} 1 & 0 & -1 & 0 \\ 4 & 0 & 0 & -2 \\ 0 & 3 & -3 & 0 \end{bmatrix}$

$$\begin{bmatrix} 1 & 0 & 0 & -\frac{1}{2} \\ 0 & 1 & 0 & -1 \\ 0 & 0 & 1 & -\frac{1}{2} \end{bmatrix}$$

$$\begin{array}{ccc}
x_{4} = t \\
\times_{3} = \frac{1}{2}t \\
\times_{2} = t
\end{array}$$

$$\begin{bmatrix}
x_{1} \\
x_{2} \\
x_{3} \\
x_{4}
\end{bmatrix} = t \begin{bmatrix}
1/2 \\
1/2 \\
1 \\
1/2 \\
1
\end{bmatrix}$$

Ex: x, B10 H2CNH3 +XNiCla + X3NaOH -> X,Nay (B10 H10 CNH2) ANi+ X5NaCl+xH2O

$$\begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \\ x_5 \\ x_6 \end{bmatrix} = \begin{bmatrix} t_{13} \\ t_{16} \\ t \\ t_{16} \\ t_{13} \\ t \end{bmatrix}$$

$$\longrightarrow HM_nO_4 + AsH_3 + CrS_3O_{12} + H_2O.$$
FOR YOU!

Dimensional Analysis

There density

There density temperature of the turkey must be dimensionless

no units.

Buckingham T Theorem: # variables - # units = # of ways to group variables for f.

$$[m] = kq \qquad \qquad [J] = \frac{kg \cdot m^2}{s^2}$$

$$\begin{bmatrix} K \end{bmatrix} = \frac{m \cdot K}{m \cdot K} = \left(\frac{k \cdot 3 \cdot m}{k \cdot 3 \cdot K} \right) / m \cdot K = \frac{k \cdot 3 \cdot K}{k \cdot 3 \cdot K}$$

$$[c] = \frac{\pi}{k_3 \cdot K} = \left(\frac{k_3 \cdot m^2}{s^2}\right) / k_3 \cdot K = \frac{m^2}{s^2 \cdot K} \qquad k_3, s, m, K$$

$$\left[\beta \right] = \frac{\kappa_9}{m^3} \qquad (5 - 4) = 1$$

$$\int (w_{x_1} + x_5 x_3 - x_4 x_2) = \int (x_3 + x_5) \left(\frac{x_3 - x_4}{x_2}\right) \left(\frac{x_3}{x_3}\right) \left(\frac{x_$$

$$k^{3}$$
: $x^{1} + x^{3} + x^{2} = 0$

$$s: x_2 - 3x_3 - 2x_4 = 0$$

$$m = x_2 + 2x_4 - 3x_5 = 0$$