








## QUESTION 1

Use a weather forecast website, and utilize the psychrometric chart and the formula we went through in the class to determine the absolute humidity, the wet-bulb temperature and the mass of water vapour in the air in Classroom A (Aula A) of Piacenza campus in the moment that you are solving this exercise (provide the inputs that you utilized)

Il tempo oggi in Piacenza Martedì, 03 Dicembre 2019							
	13:00	14:00	16:00	18:00	20:00	21:00	22:00
							
	LightCloud	LightCloud	PartlyCloud	LightCloud	Sun	Sun	Sun
Temperatura effettiva	9°C	10°C	8°C	6°C	4°C	2°C	2°C
Temperatura percepita	7°C	10°C	6°C	4°C	2°C	0°C	0°C
Precipitazioni	0 mm	0 mm	0 mm	0 mm	0 mm	0 mm	0 mm
Umidità	67 %	65 %	69 %	70 %	75 %	83 %	87 %
Pressione atmosferica	1025 hPa	1025 hPa	1025 hPa	1026 hPa	1027 hPa	1027 hPa	1028 hPa
Intensità del vento	15 km/h	14 km/h	9 km/h	9 km/h	7 km/h	8 km/h	8 km/h
Direzione del vento	←	←	←	←	↗	↗	↗
	E	E	E	E	SE	SE	SE
Probabilità di nebbia	0 %	0 %	0 %	0 %	0 %	0 %	0 %
Punto di rugiada	3°C	3°C	3°C	1°C	-1°C	0°C	-1°C
Nuvole	21 %	13 %	42 %	15 %	2 %	3 %	3 %
Nuvole basse	11 %	7 %	42 %	15 %	2 %	3 %	3 %
Nuvole medie	18 %	12 %	2 %	0 %	1 %	0 %	0 %
Nuvole alte	0 %	0 %	0 %	0 %	0 %	0 %	0 %

Water tends to evaporate or vaporize by projecting molecules into the space above its surface. As the number of molecules increases until the rate at which molecules reenter the liquid is equal to the rate at which they leave, the pressure at which water vapour is in *thermodynamic equilibrium with its condensed state* at a given temperature is the **saturation pressure**.

**Online Water saturation pressure Calculator**

The calculator below can be used to calculate the water saturation pressure at given temperature. The output pressure is given as kPa, bar, atm, psi (lb/in<sup>2</sup>) and psf (lb/ft<sup>2</sup>).

Temperature must be within the ranges 0-370 °C, 32-700 °F, 273-645 K and 492-1160 °R

Temperature

Choose the actual unit of temperature:  
☒ °C ☐ °F ☐ K ☐ °R

**Calculate water saturation pressure!**

The saturation pressure of water depends on temperature as shown below:

See [Water and Heavy Water](#) for thermodynamic properties at standard conditions.  
 See also other properties of **Water** at varying temperature and pressure: Boiling points at high pressure, Boiling points at vacuum pressure, Density and specific weight, Dynamic and kinematic viscosity, Enthalpy and entropy, Heat of vaporization, Ionization Constant,  $pK_w$ , of normal and heavy water, Melting points at high pressure, Prandti number, Properties at Gas-Liquid Equilibrium Conditions, Specific gravity, Specific heat (heat capacity), Specific volume, Thermal conductivity, Thermal diffusivity and Vapour pressure at gas-liquid equilibrium.

www.engineeringtoolbox.com indique  
 Water saturation pressure at 4 degree C:

0.813 kPa  
 0.0081 bar  
 0.008 atm  
 0.118 psi  
 17 psf

We have:

- Temperature to be utilized:  $T = 4^\circ\text{C} = 277\text{K}$
- relative humidity:  $\phi = 0,75$  (75%)
- Air to total pressure:  $P = 102,7\text{ kPa}$   
 $= P_a + P_v$
- water saturation pressure at  $(4^\circ\text{C})$ :  $\frac{P_v}{P_g} = 0,813\text{ kPa}$

$$\phi = \frac{m_v}{m_g} = \frac{P_v}{P_g}$$

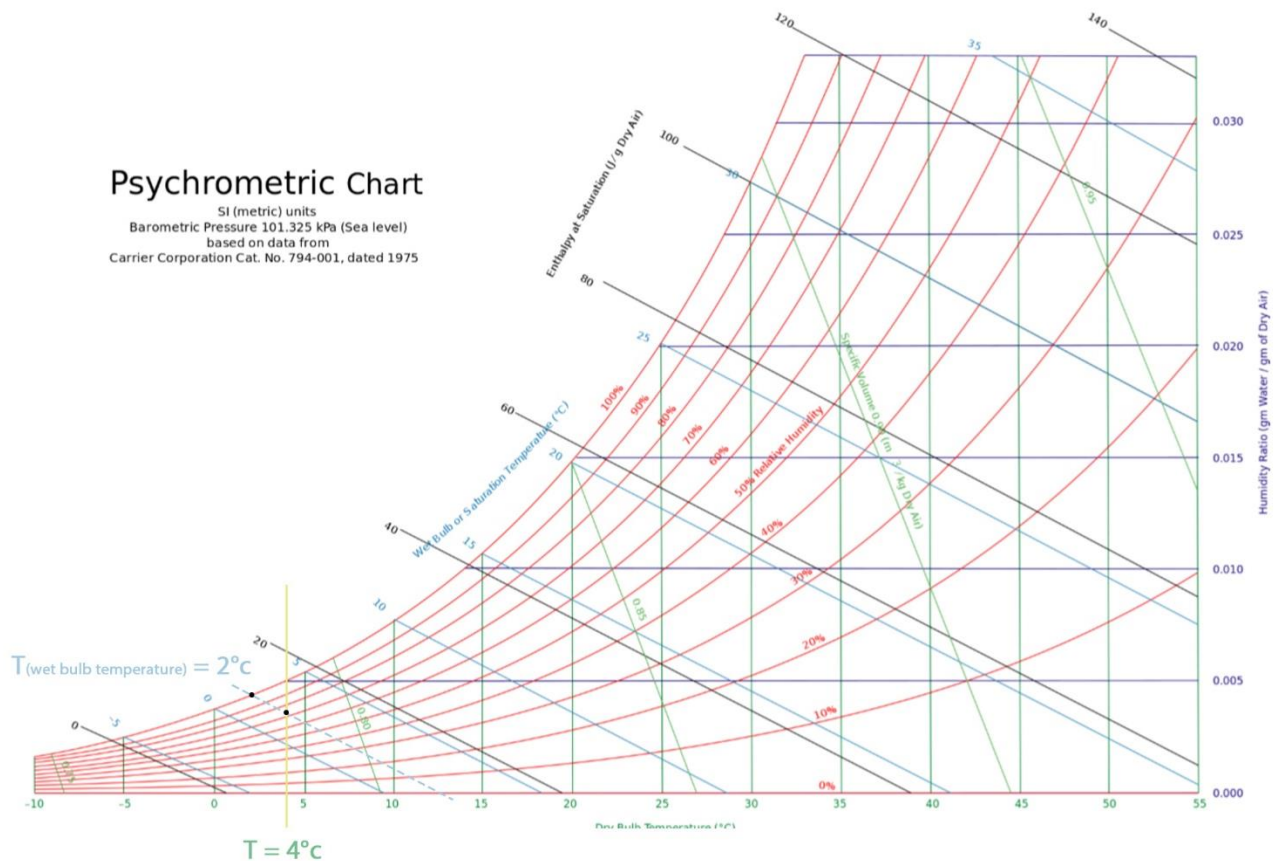
$$\Rightarrow P_v = \phi \cdot P_g = 0,75 \times 0,813 \text{ (kPa)}$$

$$\boxed{P_v = 0,60975 \text{ kPa}}$$

$$\omega = \frac{0,622 P_v}{P - P_v} = \frac{0,622 \times 0,60975}{101,325 - 0,60975}$$

$$\Rightarrow \text{Absolute Humidity} \quad \boxed{\omega = 0,0037 \frac{\text{kg vapour}}{\text{kg dry air}}}$$

- For the wet bulb temperature we use the psychrometric chart :



- The wet-balls temperature is  $T = 2^\circ\text{C}$
- $m_v = \frac{P_v \cdot V}{R_{sp} \cdot T}$  (we consider air an ideal gas)

we have :

$$\begin{cases} P_v = 0,60975 \text{ kPa} \\ V(\text{Aula A}) = 10 \times 5 \times 5 \text{ m}^3 \\ R_{sp} = 0,416 \\ T = 277 \text{ K} = 4^\circ\text{C} \end{cases}$$

$$m_v = \frac{0,60975 \times (10 \times 5 \times 5)}{0,416 \times 277}$$

$$m_v = 1,32 \text{ kg}$$

## QUESTION 2

Utilize the same methodology we went through in the class and determine the sensible and latent load corresponding to internal gains, the ventilation, and the infiltration in a house with a *good* construction quality and with the same geometry as that of the example which is located in Brindisi, Italy

- sensible load :

we have  $\begin{cases} A_{cf} = 200 \text{ m}^2 \\ N_{cf} = 2 \end{cases}$

$$Q_{ips} = 136 + 2,2 \times A_{cf} + 22 \times N_{cf} = 136 + 2,2 \times 200 + 22 \times 2$$

$$Q_{igl} = 20 + 0,22 \times A_{cf} + 12 \times N_{cf} = 20 + 0,22 \times 200 + 12 \times 2$$

$$\begin{cases} Q_{ips} = 620 \text{ W} \\ Q_{igl} = 88 \text{ W} \end{cases}$$

- infiltration :
- we have  $\begin{cases} A_{unit} = 1,4 \text{ m}^2/\text{m}^2 \text{ (good quality construction)} \\ A_{es} = 200 + 1,4 \times 4 = 344 \text{ (S}_{floor} + \text{S}_{walls}) \\ A_L = A_{es} \times A_{unit} = 481,6 \text{ m}^2 \\ IDF_{heating} = 0,073 \text{ L/s m}^3 \\ IDF_{cooling} = 0,0375 \text{ L/s m}^3 \end{cases}$

$$\begin{aligned} \Rightarrow \checkmark \text{ infiltration heating} &= A_L \times IDF_{heating} = 481,6 \times 0,073 \\ \checkmark \text{ infiltration cooling} &= A_L \times IDF_{cooling} = 481,6 \times 0,0375 \end{aligned}$$

⇒

$$\begin{aligned} V_{\text{filtration heating}} &= 35,156 \text{ L/S} \\ V_{\text{filtration cooling}} &= 18,06 \text{ L/S} \end{aligned}$$

• Ventilation

we have  $\begin{cases} A_{cp} = 200 \\ N_{br} = 1 \end{cases}$

$$\begin{aligned} \text{Ventilation} &= 0,05 \times A_{cp} + 3,5 \times (N_{br} + 1) \\ &= 0,05 \times 200 + 3,5 \times (1 + 1) \end{aligned}$$

$$\boxed{V_{\text{ventilation}} = 17 \text{ L/S}}$$

we have  $\begin{cases} C_{\text{sensible}} = 1,23 \\ C_{\text{latent}} = 3010 \\ \Delta_{\text{cooling}} = 0,0132 - 0,0093 = 0,0039 \\ \Delta_{\text{heating}} = 0,0190 - 0,0140 = 0,005 \end{cases}$

⇒ •  $Q_{\text{if. vent. cooling sensible}} = C_{\text{sensible}} \times V_{\text{if cooling}} \times \Delta T_{\text{cooling}}$   
 $= 1,23 \times 35,06 \times 7,9$   
 $= 340,67 \text{ W}$

•  $Q_{\text{if. vent. cooling latent}} = C_{\text{latent}} \times V_{\text{if cooling}} \times \Delta_{\text{cooling}}$   
 $= 3010 \times 35,06 \times 0,0039$   
 $= 411,57 \text{ W}$

•  $Q_{\text{if. vent. heating sensible}} = C_{\text{sensible}} \times V_{\text{if heating}} \times \Delta T_{\text{heating}}$   
 $= 1,23 \times 52,156 \times 24,8$   
 $= 1590,97 \text{ W}$

•  $Q_{\text{if. vent. heating latent}} = C_{\text{latent}} \times V_{\text{if heating}} \times \Delta_{\text{heating}}$   
 $= 3010 \times 52,156 \times 0,005$   
 $= 784,95 \text{ W}$