

The following properties were recorded for the 20 common amino acids, using the Molecular Operating Environment (MOE) software package (version 2012.10), developed by the Chemical Computing Group (Montreal, Canada).

a_acc Number of hydrogen bond acceptor atoms (not counting atoms but counting atoms that are both hydrogen bond don acceptors such as -OH).  a_acid Number of acidic atoms.	
acceptors such as -OH).	ors and
- /	
a_acid Number of acidic atoms.	
a_aro Number of aromatic atoms.	
a_base Number of basic atoms.	
a_count Number of atoms (including implicit hydrogens). This is cal as the sum of $(1 + h_i)$ over all non-trivial atoms $i$ .	culated
a_don Number of hydrogen bond donor atoms (not counting basic	atoms
but counting atoms that are both hydrogen bond donors a	
ceptors such as -OH).	
a_heavy Number of heavy atoms $\#\{Z_i Z_i>1\}$ .	
a_hyd Number of hydrophobic atoms.	
a_IC Atom information content (total). This is calculated to be	a_ICM
times $n$ .	
a_ICM Atom information content (mean). This is the entropy of	the ele-
ment distribution in the molecule (including implicit hydrog	
not lone pair pseudo-atoms). Let $n_i$ be the number of occur	
of atomic number i in the molecule. Let $pi = n_i/n$ where i	
sum of the $n_i$ . The value of a_ICM is the negative of the su	m over
all $i$ of $p_i \log p_i$ .	
a_nB Number of boron atoms $\#\{Z_i Z_i=5\}$ .	
a_nBr Number of bromine atoms $\#\{Z_i Z_i=35\}$ .	
a_nC Number of carbon atoms $\#\{Z_i Z_i=6\}$ .	
a_nCl Number of chlorine atoms $\#\{Z_i Z_i=17\}$ .	
a_nF Number of fluorine atoms $\#\{Z_i Z_i=9\}$ .	
a_nH Number of hydrogen atoms (including implicit hydrogens)	
is calculated as the sum of $h_i$ over all non-trivial atoms $i$ p	lus the
number of non-trivial hydrogen atoms.	
a_nI Number of iodine atoms $\#\{Z_i Z_i=53\}$ .	
a_nN Number of nitrogen atoms $\#\{Z_i Z_i=7\}$ .	
a_n0 Number of oxygen atoms $\#\{Z_i Z_i=8\}$ .	
a_nP Number of phosphorus atoms $\#\{Z_i Z_i=15\}.$	
a_nS Number of sulfur atoms $\#\{Z_i Z_i=16\}$ .	
AM1_dipole The dipole moment calculated using the AM1 Hami [MOPAC1993].	ltonian
AM1_E The total SCF energy (kcal/mol) calculated using the AM1	Hamil-
tonian [MOPAC1993].	
AM1_Eele The electronic energy (kcal/mol) calculated using the AM1	Hamil-
tonian [MOPAC1993].	

Property	Description
AM1_HF	The heat of formation (kcal/mol) calculated using the AM1 Hamil-
	tonian [MOPAC1993].
AM1_HOMO	The energy (eV) of the Highest Occupied Molecular Orbital calcu-
	lated using the AM1 Hamiltonian [MOPAC1993].
AM1_IP	The ionization potential (kcal/mol) calculated using the AM1
	Hamiltonian [MOPAC1993].
AM1_LUMO	The energy (eV) of the Lowest Unoccupied Molecular Orbital cal-
	culated using the AM1 Hamiltonian [MOPAC1993].
apol	Sum of the atomic polarizabilities (including implicit hydrogens)
_	with polarizabilities taken from [CRC1994].
ASA	Water accessible surface area calculated using a radius of 1.4 Å for
	the water molecule. A polyhedral representation is used for each
	atom in calculating the surface area.
ASA_H	Water accessible surface area of all hydrophobic ( $ q_i  < 0.2$ ) atoms.
ASA minus	Water accessible surface area of all atoms with negative partial
	charge (strictly less than 0).
ASA_P	Water accessible surface area of all polar $( q_i  \ge 0.2)$ atoms.
ASA_plus	Water accessible surface area of all atoms with positive partial
	charge (strictly greater than 0).
b_1rotN	Number of rotatable single bonds. Conjugated single bonds are not
	included (e.g. ester and peptide bonds).
b_1rotR	Number of rotatable single bonds: b_1rotN divided by b_heavy.
b_ar	Number of aromatic bonds.
b_count	Number of bonds (including implicit hydrogens). This is calculated
	as the sum of $(d_i/2 + h_i)$ over all non-trivial atoms i.
b_double	Number of double bonds. Aromatic bonds are not considered to be
	double bonds.
b_heavy	Number of bonds between heavy atoms.
b_rotN	Number of rotatable bonds. A bond is rotatable if it has order 1, is
	not in a ring, and has at least two heavy neighbors.
b_rotR	Fraction of rotatable bonds: b_rotN divided by b_heavy.
b_single	Number of single bonds (including implicit hydrogens). Aromatic
	bonds are not considered to be single bonds.
b_triple	Number of triple bonds (including implicit hydrogens). Aromatic
	bonds are not considered to be triple bonds.
balabanJ	Balaban's connectivity topological index [Balaban1982].
BCUT_PEOE_0	The BCUT descriptors [Pearlman1998] are calculated from the
	eigenvalues of a modified adjacency matrix. Each ij entry of the
	adjacency matrix takes the value $1/\operatorname{sqrt}(b_{ij})$ where $b_{ij}$ is the formal
	bond order between bonded atoms $i$ and $j$ . The diagonal takes the
	value of the PEOE partial charges. The resulting eigenvalues are
	sorted and the smallest, 1/3-ile, 2/3-ile and largest eigenvalues are
DOUT DECE 4	reported.
BCUT_PEOE_1	See above.
BCUT_PEOE_2	See above.
BCUT_PEOE_3	See above.
BCUT_SLOGP_0	The BCUT descriptors using atomic contribution to logP (using the
	Wildman and Crippen SlogP method) instead of partial charge.

Property	Description
BCUT_SLOGP_1	See above.
BCUT_SLOGP_2	See above.
BCUT_SLOGP_3	See above.
BCUT_SMR_0	The BCUT descriptors using atomic contribution to molar refrac-
	tivity (using the Wildman and Crippen SlogP method) instead of
	partial charge.
BCUT_SMR_1	See above.
BCUT_SMR_2	See above.
BCUT_SMR_3	See above.
bpol	Sum of the absolute value of the difference between atomic polar-
	izabilities of all bonded atoms in the molecule (including implicit
	hydrogens) with polarizabilities taken from [CRC1994].
CASA_minus	Negative charge weighted surface area, ASA_minus times max { $q_i <$
	0 } [Stanton1990].
CASA_plus	Positive charge weighted surface area, ASA_plus times max $\{q_i > 0\}$
	} [Stanton1990].
chi0	Atomic connectivity index (order 0) from [Hall1991] and [Kier1977].
	This is calculated as the sum of $1/\operatorname{sqrt}(d_i)$ over all heavy atoms $i$
	with $d_i > 0$ .
chi0_C	Carbon connectivity index (order 0). This is calculated as the sum
	of $1/\operatorname{sqrt}(v_i)$ over all carbon atoms $i$ with $d_i > 0$ .
chi0v	Atomic valence connectivity index (order 0) from [Hall1991] and
	[Kier1977]. This is calculated as the sum of $1/\operatorname{sqrt}(v_i)$ over all heavy
	atoms $i$ with $v_i > 0$ .
chi0v_C	Carbon valence connectivity index (order 0). This is calculated as
	the sum of $1/\operatorname{sqrt}(v_i)$ over all carbon atoms $i$ with $v_i > 0$ .
chi1	Atomic connectivity index (order 1) from [Hall1991] and [Kier1977].
	This is calculated as the sum of $1/\operatorname{sqrt}(d_id_j)$ over all bonds between
	heavy atoms $i$ and $j$ where $i < j$ .
chi1_C	Carbon connectivity index (order 1). This is calculated as the sum
	of $1/\operatorname{sqrt}(d_id_j)$ over all bonds between carbon atoms $i$ and $j$ where
1 '4	i < j.
chi1v	Atomic valence connectivity index (order 1) from [Hall1991] and
	[Kier1977]. This is calculated as the sum of $1/\operatorname{sqrt}(v_i v_j)$ over all
chi1v_C	bonds between heavy atoms $i$ and $j$ where $i < j$ .
CIIIIV_C	Carbon valence connectivity index (order 1). This is calculated as the sum of $1/\operatorname{sqrt}(v_iv_j)$ over all bonds between carbon atoms $i$ and
	j where $i < j$ .
chiral	The number of chiral centers.
chiral_u	The number of chiral centers.  The number of unconstrained chiral centers.
DASA	Absolute value of the difference between ASA_plus and ASA_minus.
DCASA	Absolute value of the difference between CASA_plus and CASA_minus  Absolute value of the difference between CASA_plus and CASA_minus
Donon	[Stanton1990].
dens	Mass density: molecular weight divided by van der Waals volume
	as calculated in the vol descriptor.
dongity	Molecular mass density: Weight divided by vdw_vol (amu/ų).
density diameter	
	Largest value in the distance matrix [Petitjean1992].  Dipole moment calculated from the partial charges of the molecule.
dipole	Dipole moment calculated from the partial charges of the molecule.

Property	Description
dipoleX	The x component of the dipole moment (external coordinates).
dipoleY	The y component of the dipole moment (external coordinates).
dipoleZ	The z component of the dipole moment (external coordinates).
E	Value of the potential energy.
E_ang	Angle bend potential energy.
E_ele	Electrostatic component of the potential energy.
E_nb	Value of the potential energy with all bonded terms disabled.
E_oop	Out-of-plane potential energy.
E_sol	Solvation energy.
E_stb	Bond stretch-bend cross-term potential energy.
E_str	Bond stretch potential energy.
E_strain	Local strain energy: the current energy minus the value of the energy at a near local minimum. The current energy is calculated as for the E descriptor. The local minimum energy is the value of the E descriptor after first performing an energy minimization.
E_tor	Torsion (proper and improper) potential energy.
E_vdw	Van der Waals component of the potential energy.
FASA_H	Fractional ASA_H calculated as ASA_H/ASA.
FASA_minus	Fractional ASA_minus calculated as ASA_minus/ASA.
FASA_P	Fractional ASA_P calculated as ASA_P/ASA.
FASA_plus	Fractional ASA_plus calculated as ASA_plus/ASA.
FCASA_minus	Fractional CASA_minus calculated as CASA_minus/ASA.
FCASA_plus	Fractional CASA_plus calculated as CASA_plus/ASA.
FCharge	Total charge of the molecule (sum of formal charges).
GCUT_PEOE_0	The GCUT descriptors are calculated from the eigenvalues of a modified graph distance adjacency matrix. Each $ij$ entry of the adjacency matrix takes the value $1/\operatorname{sqrt}(d_{ij})$ where $d_{ij}$ is the (modified) graph distance between atoms $i$ and $j$ . The diagonal takes the value of the PEOE partial charges. The resulting eigenvalues are sorted and the smallest, $1/3$ -ile, $2/3$ -ile and largest eigenvalues are reported.
GCUT_PEOE_1	See above.
GCUT_PEOE_2	See above.
GCUT_PEOE_3	See above.
GCUT_SLOGP_0	The GCUT descriptors using atomic contribution to logP (using the Wildman and Crippen SlogP method) instead of partial charge.
GCUT_SLOGP_1	See above.
GCUT_SLOGP_2	See above.
GCUT_SLOGP_3	See above.
GCUT_SMR_O	The GCUT descriptors using atomic contribution to molar refractivity (using the Wildman and Crippen SMR method) instead of partial charge.
GCUT_SMR_1	See above.
GCUT_SMR_2	See above.
GCUT_SMR_3	See above.

Property	Description
glob	Globularity, or inverse condition number (smallest eigenvalue di-
	vided by the largest eigenvalue) of the covariance matrix of atomic
	coordinates. A value of 1 indicates a perfect sphere while a value of
	0 indicates a two- or one-dimensional object.
Kier1	First kappa shape index: $(n-1)^2/m^2$ [Hall1991].
Kier2	Second kappa shape index: $(n-1)^2/m^2$ [Hall1991].
Kier3	Third kappa shape index: $(n-1)(n-3)^2/p_3^2$ for odd n, and $(n-1)(n-3)^2/p_3^2$
	$(3)(n-2)^2/p_3^2$ for even $n$ [Hall1991].
KierA1	First alpha modified shape index: $s(s-1)^2/m^2$ where $s=n+a$
	[Hall1991].
KierA2	Second alpha modified shape index: $s(s-1)^2/m^2$ where $s=n+a$
	[Hall1991].
KierA3	Third alpha modified shape index: $(s-1)(s-3)^2/p_3^2$ for odd n, and
	$(s-3)(s-2)^2/p_3^2$ for even n where $s=n+a$ [Hall1991].
KierFlex	Kier molecular flexibility index: (KierA1)(KierA2)/n [Hall1991].
lip_acc	The number of O and N atoms.
lip_don	The number of OH and NH atoms.
lip_druglike	One if and only if lip_violation < 2 otherwise zero.
lip_violation	The number of violations of Lipinski's Rule of Five [Lipinski2012].
logP_par_o_div_w_par_	Log of the octanol/water partition coefficient (including implicit hy-
	drogens). This property is calculated from a linear atom type model
	[LOGP1998] with $r^2 = 0.931$ , RMSE=0.393 on 1,827 molecules.
logS	Log of the aqueous solubility (mol/L). This property is calculated
	from an atom contribution linear atom type model [Hou2004] with
	$r^2 = 0.90, \sim 1,200 \text{ molecules.}$
MNDO_dipole	The dipole moment calculated using the MNDO Hamiltonian
	[MOPAC1993].
MNDO_E	The total SCF energy (kcal/mol) calculated using the MNDO
	Hamiltonian [MOPAC1993].
MNDO_Eele	The electronic energy (kcal/mol) calculated using the MNDO
	Hamiltonian [MOPAC1993].
MNDO_HF	The heat of formation (kcal/mol) calculated using the MNDO
	Hamiltonian [MOPAC1993].
MNDO_HOMO	The energy (eV) of the Highest Occupied Molecular Orbital calcu-
	lated using the MNDO Hamiltonian [MOPAC1993].
MNDO_IP	The ionization potential (kcal/mol) calculated using the MNDO
	Hamiltonian [MOPAC1993].
MNDO_LUMO	The energy (eV) of the Lowest Unoccupied Molecular Orbital cal-
	culated using the MNDO Hamiltonian [MOPAC1993].
mr	Molecular refractivity (including implicit hydrogens). This property
	is calculated from an 11 descriptor linear model [MREF1998] with
	$r^2 = 0.997$ , RMSE = 0.168 on 1,947 small molecules.
mutagenic	Indicator of the presence of potentially toxic groups. A non-zero
	value indicates that the molecule contains a mutagenic group. The
	table of mutagenic groups is based on the Kazius set [Kazius 2005].
nmol	The number of molecules (connected components).
npr1	Normalized PMI ratio pmi1/pmi3.
	Normalized PMI ratio pmi2/pmi3.

Property	Description
opr_brigid	The number of rigid bonds from [Oprea2000].
opr_leadlike	One if and only if opr_violation < 2 otherwise zero.
opr_nring	The number of ring bonds from [Oprea2000].
opr_nrot	The number of rotatable bonds from [Oprea2000].
opr_violation	The number of violations of Oprea's lead-like test [Oprea2000].
PEOE_PC_minus	Total negative partial charge: the sum of the negative $q_i$ .
PEOE_PC_plus	Total positive partial charge: the sum of the positive $q_i$ .
PEOE_RPC_minus	Relative negative partial charge: the smallest negative $q_i$ divided by
	the sum of the negative $q_i$ .
PEOE_RPC_plus	Relative positive partial charge: the largest positive $q_i$ divided by
	the sum of the positive $q_i$ .
PEOE_VSA_FHYD	Fractional hydrophobic van der Waals surface area. This is the sum
	of the $v_i$ such that $ q_i $ is less than or equal to 0.2 divided by the
	total surface area. The $v_i$ are calculated using a connection table
	approximation.
PEOE_VSA_FNEG	Fractional negative van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is negative divided by the total surface area. The
	$v_i$ are calculated using a connection table approximation.
PEOE_VSA_FPNEG	Fractional negative polar van der Waals surface area. This is the
	sum of the $v_i$ such that $q_i$ is less than -0.2 divided by the total surface
	area. The $v_i$ are calculated using a connection table approximation.
PEOE_VSA_FPOL	Fractional polar van der Waals surface area. This is the sum of the
	$v_i$ such that $ q_i $ is greater than 0.2 divided by the total surface area.
	The $v_i$ are calculated using a connection table approximation.
PEOE_VSA_FPOS	Fractional positive van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is non-negative divided by the total surface area.
	The $v_i$ are calculated using a connection table approximation.
PEOE_VSA_FPPOS	Fractional positive polar van der Waals surface area. This is the sum
	of the $v_i$ such that $q_i$ is greater than 0.2 divided by the total surface
DECE 1101 1111D	area. The $v_i$ are calculated using a connection table approximation.
PEOE_VSA_HYD	Total hydrophobic van der Waals surface area. This is the sum of
	the $v_i$ such that $ q_i $ is less than or equal to 0.2. The $v_i$ are calculated
DECE MON	using a connection table approximation.
PEOE_VSA_minus0	Sum of $v_i$ where $q_i$ is in the range $[-0.05,0.00)$ .
PEOE_VSA_minus1	Sum of $v_i$ where $q_i$ is in the range [-0.10,-0.05).
PEOE_VSA_minus2 PEOE_VSA_minus3	Sum of $v_i$ where $q_i$ is in the range [-0.15,-0.10).
PEOE_VSA_minus3	Sum of $v_i$ where $q_i$ is in the range [-0.20,-0.15).
	Sum of $v_i$ where $q_i$ is in the range [-0.25,-0.20).
PEOE_VSA_minus5	Sum of $v_i$ where $q_i$ is in the range [-0.30,-0.25).
PEOE_VSA_minus6	Sum of $v_i$ where $q_i$ is less than -0.3.
PEOE_VSA_NEG	Total negative van der Waals surface area. This is the sum of the
	$v_i$ such that $q_i$ is negative. The $v_i$ are calculated using a connection table approximation.
DEOF VCA pluco	Sum of $v_i$ where $q_i$ is in the range [0.00,0.05).
PEOE_VSA_plus0	Sum of $v_i$ where $q_i$ is in the range [0.05,0.10). Sum of $v_i$ where $q_i$ is in the range [0.05,0.10).
PEOE_VSA_plus1	
PEOE_VSA_plus2	Sum of $v_i$ where $q_i$ is in the range [0.10,0.15).
PEOE_VSA_plus3	Sum of $v_i$ where $q_i$ is in the range [0.15,0.20).
PEOE_VSA_plus4	Sum of $v_i$ where $q_i$ is in the range [0.20,0.25).

Property	Description
PEOE_VSA_plus5	Sum of $v_i$ where $q_i$ is in the range [0.25,0.30).
PEOE_VSA_plus6	Sum of $v_i$ where $q_i$ is greater than 0.3.
PEOE_VSA_PNEG	Total negative polar van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is less than -0.2. The $v_i$ are calculated using a
	connection table approximation.
PEOE_VSA_POL	Total polar van der Waals surface area. This is the sum of the $v_i$
	such that $ q_i $ is greater than 0.2. The $v_i$ are calculated using a
	connection table approximation.
PEOE_VSA_POS	Total positive van der Waals surface area. This is the sum of the $v_i$
	such that $q_i$ is non-negative. The $v_i$ are calculated using a connec-
	tion table approximation.
PEOE_VSA_PPOS	Total positive polar van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is greater than 0.2. The $v_i$ are calculated using
	a connection table approximation.
petitjean	Value of (diameter - radius)/diameter.
petitjeanSC	Petitjean graph Shape Coefficient as defined in [Petitjean1992]:
	(diameter - radius)/radius.
PM3_dipole	The dipole moment calculated using the PM3 Hamiltonian
-	[MOPAC1993].
PM3_E	The total SCF energy (kcal/mol) calculated using the PM3 Hamil-
	tonian [MOPAC1993].
PM3_Eele	The electronic energy (kcal/mol) calculated using the PM3 Hamil-
	tonian [MOPAC1993].
PM3_HF	The heat of formation (kcal/mol) calculated using the PM3 Hamil-
	tonian [MOPAC1993].
PM3_HOMO	The energy (eV) of the Highest Occupied Molecular Orbital calcu-
	lated using the PM3 Hamiltonian [MOPAC1993].
PM3_IP	The ionization potential (kcal/mol) calculated using the PM3
	Hamiltonian [MOPAC1993].
PM3_LUMO	The energy (eV) of the Lowest Unoccupied Molecular Orbital cal-
	culated using the PM3 Hamiltonian [MOPAC1993].
pmi	Principal moment of inertia.
pmi1	First diagonal element of diagonalized moment of inertia tensor.
pmi2	Second diagonal element of diagonalized moment of inertia tensor.
pmi3	Third diagonal element of diagonalized moment of inertia tensor.
pmiX	x component of the principal moment of inertia (external coordi-
	nates).
pmiY	y component of the principal moment of inertia (external coordi-
	nates).
pmiZ	z component of the principal moment of inertia (external coordi-
	nates).
Q_PC_minus	Total negative partial charge: the sum of the negative $q_i$ .
Q_PC_plus	Total positive partial charge: the sum of the positive $q_i$ .
Q_RPC_minus	
	the sum of the negative $q_i$ .
Q_RPC_plus	Relative positive partial charge: the largest positive $q_i$ divided by
	the sum of the positive $q_i$ .
Q_PC_minus Q_PC_plus Q_RPC_minus	nates).  Total negative partial charge: the sum of the negative $q_i$ .  Total positive partial charge: the sum of the positive $q_i$ .  Relative negative partial charge: the smallest negative $q_i$ divided by the sum of the negative $q_i$ .

Property	Description
Q_VSA_FHYD	Fractional hydrophobic van der Waals surface area. This is the sum
	of the $v_i$ such that $ q_i $ is less than or equal to 0.2 divided by the
	total surface area. The $v_i$ are calculated using a connection table
	approximation.
Q_VSA_FNEG	Fractional negative van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is negative divided by the total surface area. The
	$v_i$ are calculated using a connection table approximation.
Q_VSA_FPNEG	Fractional negative polar van der Waals surface area. This is the
	sum of the $v_i$ such that $q_i$ is less than -0.2 divided by the total surface
	area. The $v_i$ are calculated using a connection table approximation.
Q_VSA_FPOL	Fractional polar van der Waals surface area. This is the sum of the
	$v_i$ such that $ q_i $ is greater than 0.2 divided by the total surface area.
	The $v_i$ are calculated using a connection table approximation.
Q_VSA_FPOS	Fractional positive van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is non-negative divided by the total surface area.
	The $v_i$ are calculated using a connection table approximation.
Q_VSA_FPPOS	Fractional positive polar van der Waals surface area. This is the sum
	of the $v_i$ such that $q_i$ is greater than 0.2 divided by the total surface
	area. The $v_i$ are calculated using a connection table approximation.
Q_VSA_HYD	Total hydrophobic van der Waals surface area. This is the sum of
	the $v_i$ such that $ q_i $ is less than or equal to 0.2. The $v_i$ are calculated
	using a connection table approximation.
Q_VSA_NEG	Total negative van der Waals surface area. This is the sum of the
	$v_i$ such that $q_i$ is negative. The $v_i$ are calculated using a connection
	table approximation.
Q_VSA_PNEG	Total negative polar van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is less than -0.2. The $v_i$ are calculated using a
	connection table approximation.
Q_VSA_POL	Total polar van der Waals surface area. This is the sum of the $v_i$
	such that $ q_i $ is greater than 0.2. The $v_i$ are calculated using a
	connection table approximation.
Q_VSA_POS	Total positive van der Waals surface area. This is the sum of the $v_i$
	such that $q_i$ is non-negative. The $v_i$ are calculated using a connec-
O HOLDOO	tion table approximation.
Q_VSA_PPOS	Total positive polar van der Waals surface area. This is the sum of
	the $v_i$ such that $q_i$ is greater than 0.2. The $v_i$ are calculated using
1.	a connection table approximation.
radius	If $r_i$ is the largest matrix entry in row $i$ of the distance matrix $D$ ,
	then the radius is defined as the smallest of the $r_i$ [Petitjean1992].
reactive	Indicator of the presence of reactive groups. A non-zero value in-
	dicates that the molecule contains a reactive group. The table of
	reactive groups is based on the [Oprea2000] set and includes metals,
	phospho-, N/O/S-N/O/S single bonds, thiols, acyl halides, Michael
raur	Acceptors, azides, esters, etc.  Radius of gyration.
rgyr	The number of rings.
rings	
RPC_minus	Relative negative partial charge: the smallest negative $q_i$ divided by the sum of the negative $q_i$ .
	one sum of the negative $q_i$ .

Property	Description
RPC_plus	Relative positive partial charge: the largest positive $q_i$ divided by
	the sum of the positive $q_i$ .
rsynth	A value in [0,1] indicating the synthetic reasonableness, or feasibil-
	ity, of the chemical structure. A value of 0 means it is unlikely that
	the molecule can be synthesized while a value of 1 means that it is
	likely that the molecule can be synthesized. The value reflects the
	fraction of heavy atoms in the molecule that can be traced back to
	starting materials fragments resulting from retrosynthetic discon-
	nection rules.
SlogP	Log of the octanol/water partition coefficient (including implicit
	hydrogens). This property is an atomic contribution model
	[Wildman1999] that calculates logP from the given structure; i.e.
	the correct protonation state (washed structures). Results may vary
	from the logP_par_o_div_w_par_ descriptor. The training set for
G1 NGAO	SlogP was ~7000 structures.
SlogP_VSA0	Sum of $v_i$ such that $L_i \leq -0.4$ .
SlogP_VSA1	Sum of $v_i$ such that $L_i$ is in $(-0.4, -0.2]$ .
SlogP_VSA2	Sum of $v_i$ such that $L_i$ is in $(-0.2,0]$ .
SlogP_VSA3	Sum of $v_i$ such that $L_i$ is in $(0,0.1]$ .
SlogP_VSA4	Sum of $v_i$ such that $L_i$ is in $(0.1,0.15]$ .
SlogP_VSA5	Sum of $v_i$ such that $L_i$ is in $(0.15,0.20]$ .
SlogP_VSA6	Sum of $v_i$ such that $L_i$ is in $(0.20,0.25]$ .
SlogP_VSA7	Sum of $v_i$ such that $L_i$ is in $(0.25,0.30]$ .
SlogP_VSA8	Sum of $v_i$ such that $L_i$ is in $(0.30,0.40]$ .
SlogP_VSA9	Sum of $v_i$ such that $L_i > 0.40$ .
SMR	Molecular refractivity (including implicit hydrogens). This prop-
	erty is an atomic contribution model [Wildman1999] that assumes
	the correct protonation state (washed structures). The model was trained on ~7000 structures and results may vary from the mr de-
	scriptor.
SMR_VSAO	Sum of $v_i$ such that $R_i$ is in [0,0.11].
SMR_VSA1	Sum of $v_i$ such that $R_i$ is in $[0.11, 0.26]$ .
SMR_VSA2	Sum of $v_i$ such that $R_i$ is in $(0.26,0.35]$ .
SMR_VSA3	Sum of $v_i$ such that $R_i$ is in $(0.35,0.39]$ .
SMR_VSA4	Sum of $v_i$ such that $R_i$ is in $(0.39,0.44]$ .
SMR_VSA5	Sum of $v_i$ such that $R_i$ is in $(0.44, 0.485]$ .
SMR_VSA6	Sum of $v_i$ such that $R_i$ is in $(0.485, 0.56]$ .
SMR_VSA7	Sum of $v_i$ such that $R_i > 0.56$ .
std_dim1	Standard dimension 1: the square root of the largest eigenvalue
Bou_uimi	of the covariance matrix of the atomic coordinates. A standard
	dimension is equivalent to the standard deviation along a principal
	component axis.
std_dim2	Standard dimension 2: the square root of the second largest eigen-
~ va_aim2	value of the covariance matrix of the atomic coordinates. A stan-
	dard dimension is equivalent to the standard deviation along a prin-
	cipal component axis.
	orpur component unio.

Property	Description
std_dim3	Standard dimension 3: the square root of the third largest eigen-
	value of the covariance matrix of the atomic coordinates. A stan-
	dard dimension is equivalent to the standard deviation along a prin-
	cipal component axis.
TPSA	Polar surface area $(\text{Å}^2)$ calculated using group contributions to ap-
	proximate the polar surface area from connection table information
	only. The parameterization is that of [Ertl2000].
VAdjEq	Vertex adjacency information (equality): $-(1-f)log_2(1-f)$
	$\int f \log_2 f$ where $f = (n^2 - m)/n^2$ , n is the number of heavy atoms
	and $m$ is the number of heavy-heavy bonds. If $f$ is not in the open
	interval $(0,1)$ , then 0 is returned.
VAdjMa	Vertex adjacency information (magnitude): $1+log_2m$ where $m$ is the
	number of heavy-heavy bonds. If $m$ is zero, then zero is returned.
VDistEq	If $m$ is the sum of the distance matrix entries then $VdistEq$ is defined
	to be the sum of $log_2m - p_i log_2p_i/m$ where $p_i$ is the number of
	distance matrix entries equal to $i$ .
VDistMa	If m is the sum of the distance matrix entries then VDistMa is defined
	to be the sum of $log_2m - D_{ij}log_2D_{ij}/m$ over all $i$ and $j$ .
vdw_area	Area of van der Waals surface (Å <sup>2</sup> ) calculated using a connection
	table approximation.
vdw_vol	Van der Waals volume (Å <sup>3</sup> ) calculated using a connection table
	approximation.
vol	Van der Waals volume calculated using a grid approximation (spac-
	ing 0.75 Å).
VSA	Van der Waals surface area. A polyhedral representation is used for
	each atom in calculating the surface area.
vsa_acc	Approximation to the sum of VDW surface areas (Å <sup>2</sup> ) of pure hy-
	drogen bond acceptors (not counting acidic atoms and atoms that
	are both hydrogen bond donors and acceptors such as -OH).
vsa_acid	Approximation to the sum of VDW surface areas of acidic atoms
	$(\mathring{A}^2)$ .
vsa_base	Approximation to the sum of VDW surface areas of basic atoms
	$(\mathring{A}^2)$ .
vsa_don	Approximation to the sum of VDW surface areas of pure hydrogen
	bond donors (not counting basic atoms and atoms that are both
	hydrogen bond donors and acceptors such as -OH) $(\mathring{A}^2)$ .
vsa_hyd	Approximation to the sum of VDW surface areas of hydrophobic
	atoms ( $\mathring{A}^2$ ).
vsa_other	Approximation to the sum of VDW surface areas (Å <sup>2</sup> ) of atoms
	typed as "other".
vsa_pol	Approximation to the sum of VDW surface areas $(\mathring{A}^2)$ of polar
	atoms (atoms that are both hydrogen bond donors and acceptors),
	such as -OH.
vsurf_A	Amphiphilic moment.
vsurf_CP	Critical packing parameter.
vsurf_CW1	Capacity factor (1).
vsurf_CW2	Capacity factor (2).

Property	Description
vsurf_CW3	Capacity factor (3).
vsurf_CW4	Capacity factor (4).
vsurf_CW5	Capacity factor (5).
vsurf_CW6	Capacity factor (6).
vsurf_CW7	Capacity factor (7).
vsurf_CW8	Capacity factor (8).
vsurf_D1	Hydrophobic volume (1).
vsurf_D2	Hydrophobic volume (2).
vsurf_D3	Hydrophobic volume (3).
vsurf_D4	Hydrophobic volume (4).
vsurf_D5	Hydrophobic volume (5).
vsurf_D6	Hydrophobic volume (6).
vsurf_D7	Hydrophobic volume (7).
vsurf_D8	Hydrophobic volume (8).
vsurf_DD12	Contact distances of vsurf_DDmin (12).
vsurf_DD13	Contact distances of vsurf_DDmin (13).
vsurf_DD23	Contact distances of vsurf_DDmin (23).
vsurf_DW12	Contact distances of vsurf_EWmin (12).
vsurf_DW13	Contact distances of vsurf_EWmin (13).
vsurf_DW23	Contact distances of vsurf_EWmin (23).
vsurf_EDmin1	Lowest hydrophobic energy (1).
vsurf_EDmin2	Lowest hydrophobic energy (2).
vsurf_EDmin3	Lowest hydrophobic energy (3).
vsurf_EWmin1	Lowest hydrophilic energy (1).
vsurf_EWmin2	Lowest hydrophilic energy (2).
vsurf_EWmin3	Lowest hydrophilic energy (3).
vsurf_G	Surface globularity.
vsurf_HB1	H-bond donor capacity (1).
vsurf_HB2	H-bond donor capacity (2).
vsurf_HB3	H-bond donor capacity (3).
vsurf_HB4	H-bond donor capacity (4).
vsurf_HB5	H-bond donor capacity (5).
vsurf_HB6	H-bond donor capacity (6).
vsurf_HB7	H-bond donor capacity (7).
vsurf_HB8	H-bond donor capacity (8).
vsurf_HL1	Hydrophilic-Lipophilic (1).
vsurf_HL2	Hydrophilic-Lipophilic (2).
vsurf_ID1	Hydrophobic integy moment (1).
vsurf_ID2	Hydrophobic integy moment (2).
vsurf_ID3	Hydrophobic integy moment (3).
vsurf_ID4	Hydrophobic integy moment (4).
vsurf_ID5	Hydrophobic integy moment (5).
vsurf_ID6	Hydrophobic integy moment (6).
vsurf_ID7	Hydrophobic integy moment (7).
vsurf_ID8	Hydrophobic integy moment (8).
vsurf_IW1	Hydrophilic integy moment (1).
vsurf_IW2	Hydrophilic integy moment (2).
vsurf_IW3	Hydrophilic integy moment (3).
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Property	Description
vsurf_IW4	Hydrophilic integy moment (4).
vsurf_IW5	Hydrophilic integy moment (5).
vsurf_IW6	Hydrophilic integy moment (6).
vsurf_IW7	Hydrophilic integy moment (7).
vsurf_IW8	Hydrophilic integy moment (8).
vsurf_R	Surface rugosity.
vsurf_S	Interaction field surface area.
vsurf_V	Interaction field volume.
vsurf_W1	Hydrophilic volume (1).
vsurf_W2	Hydrophilic volume (2).
vsurf_W3	Hydrophilic volume (3).
vsurf_W4	Hydrophilic volume (4).
vsurf_W5	Hydrophilic volume (5).
vsurf_W6	Hydrophilic volume (6).
vsurf_W7	Hydrophilic volume (7).
vsurf_W8	Hydrophilic volume (8).
vsurf_Wp1	Polar volume (1).
vsurf_Wp2	Polar volume (2).
vsurf_Wp3	Polar volume (3).
vsurf_Wp4	Polar volume (4).
vsurf_Wp5	Polar volume (5).
vsurf_Wp6	Polar volume (6).
vsurf_Wp7	Polar volume (7).
vsurf_Wp8	Polar volume (8).
Weight	Molecular weight (including implicit hydrogens) in atomic mass
	units with atomic weights taken from [CRC1994].
wienerPath	Wiener path number: half the sum of all the distance matrix entries
	as defined in [Balaban1979] and [Wiener1947].
wienerPol	Wiener polarity number: half the sum of all the distance matrix
	entries with a value of 3 as defined in [Balaban1979].
zagreb	Zagreb index: the sum of $d_i^2$ over all heavy atoms $i$ .

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