

$$1. {}^{32}\text{P}, T = 14.3\text{d}, \lambda = \frac{\ln 2}{T} = 5.61 \times 10^{-7} \text{ s}^{-1}$$

$$A = \lambda N = \lambda \cdot \frac{1\text{mg}}{32\text{g}} \cdot N_A = 1.06 \times 10^{13} \text{ Bq}$$

$$2. {}^{131}\text{I}, T = 8.04\text{d}, A_0 = 5.6 \times 10^8 \text{ Bq}$$

$$A = A_0 e^{-\lambda t}, \lambda = \frac{\ln 2}{T} \Rightarrow A = A_0 e^{-\frac{t}{T} \ln 2} = A_0 \left(\frac{1}{2}\right)^{\frac{t}{T}}$$

$$\therefore A = 9.2 \times 10^7 \text{ Bq}$$

$$3. {}^{131}\text{I}, t=0 \text{ 时 } 0.5\text{mL 溶液}, t=11\text{d 时 要多少}, \text{已知 } T = 8.04\text{d}.$$

~~设初始放射性核素为  $N_0$ , 设  $t$  时  $N_t$~~

$$\therefore \text{设 } t \text{ 时 } {}^{131}\text{I} \text{ 浓度为 } N_0 \dots A = \lambda \cdot 0.5 N_0 = \lambda \cdot a N \Rightarrow a = \frac{N_0}{N} \cdot 0.5$$

$$N = N_0 e^{-\lambda t} = N_0 \left(\frac{1}{2}\right)^{t/T} \Rightarrow a = \frac{N_0}{N_0} 2^{t/T-1} = 1.29 \text{ mL}$$

$$4. 1.8 \times 10^{-6} \text{ g}, T = 1620 \text{ a}, {}^{226}\text{Ra}, \text{求 } 1\text{min 内的射线数}.$$

$$A = \lambda N = \frac{\ln 2}{T} \cdot \frac{1.8 \times 10^{-6} \text{ g}}{226 \text{ g}} \cdot N_A = 650.5 \text{ Bq}$$

$$\therefore 1\text{min 内有 } A \cdot 60 = 39031.5 \text{ 条射线}$$