Real Gas Effect in Reacting Flow

This code repository is used to evaluate the advantages of the proposed real gas equation of state (EoS)

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1. Code Environment

```
# Repo for cantera implementation
https://github.com/SuXY15/cantera
# Repo for this code
https://github.com/SuXY15/realGasEoS
# the file structure of this code
                  # figures
# Makefile for convenient
# mechanism files
├─ figs
├── Makefile
├─ mech
├── mech
├── README.md
├── README.pdf
                       # this document
                      # PDF version of this document
├─ ref
└─ src
                      # references, papers, etc
  └─ src
# Cantera used in /opt/cantera_libs/cantera, one can modified code in `src` and build & install by
python3 /usr/bin/scons install
# The code in this repo can be run by
source /usr/local/bin/setup_cantera
python3 src/test_PR.py
```

2. Basic Theory

1.0 Generalized EoS

Helmholtz Free Energy: $F \equiv U - TS = A(T, v, n)$, where U is the internal energy of the system, T is the absolute temperature, S is the entropy of the system.

- 1st law of thermodynamics: $dU = \delta Q + \delta W$
- ullet 2nd law of thermodynamics: $\delta Q = TdS$
- ullet pressure work: $\delta W = -pdV$

Thus we have:

•
$$dU = TdS - pdV$$

• =>
$$dU = d(TS) - SdT - pdV$$

•
$$\Rightarrow d(U - TS) = -SdT - pdV$$

•
$$\Rightarrow dF = -SdT - pdV$$

Therefore,

$$\begin{split} S &= -\left(\frac{\partial F}{\partial T}\right)_{V} \\ p &= -\left(\frac{\partial F}{\partial V}\right)_{T} \\ U &= F + TS = F - T\left(\frac{\partial F}{\partial T}\right)_{V} \\ H &= U + pV = F - T\left(\frac{\partial F}{\partial T}\right)_{V} - V\left(\frac{\partial F}{\partial V}\right)_{T} \\ C_{p} &= \left(\frac{\partial H}{\partial T}\right)_{p} = T\left[\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{V,T}^{2} / \left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} - \left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V}\right] \\ C_{V} &= \left(\frac{\partial U}{\partial T}\right)_{V} = -T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} \end{split}$$

1.1 Ideal Gas

$$p = \frac{nRT}{V_m}$$

where V_m is the molar volume of the gas .

1.2 Van der Waals (vdW) real gas equation 1783

$$p = \frac{RT}{V_m - b} - \frac{a}{V_m^2}$$

where b is the volume that is occupied by one mole of the molecules. vdW is a cubic type EoS.

1.4 Redlich-Kwong (RK) real gas equation 1948

$$p = \frac{RT}{V_m - b} - \frac{a}{\sqrt{T}V_m(V_m + b)}$$

where a is a constant that corrects for attractive potential of molecules and b is a constant that corrects for volume. The constants are different depending on which gas is being analyzed and can be calculated from the critical point data of corresponding gas:

$$a = rac{1}{9(\sqrt[3]{2}-1)}rac{R^2T_c^{2.5}}{P_c} \ b = rac{\sqrt[3]{2}-1}{3}rac{RT_c}{P_c}$$

Where T_c is the temperature at the critical point and P_c is the pressure at the critical point. Noted that a and b are solved analytically from the thermodynamic criteria for the critical point:

$$\left(rac{\partial P}{\partial V}
ight)_T=0, \left(rac{\partial^2 P}{\partial V^2}
ight)_T=0.$$

For more details, please refer to original paper or wikipedia of Redlich Kwong equation of state.

1.5 Soave modified Redlich-Kwong (SRK) real gas equation 1972

In 1966, Barner noted that the acentric factor ω can be used to improve RK EoS. SRK is proposed by Soave as:

$$p = rac{RT}{V_m - b} - rac{alpha}{V_m(V_m + b)}$$

where

$$lpha = \left[1 + (0.480 + 1.574\omega - 0.176\omega^2)(1 - \sqrt{T_r})\right]^2 \ a = rac{1}{9(\sqrt[3]{2} - 1)} rac{R^2 T_c^2}{P_c} \ b = rac{\sqrt[3]{2} - 1}{3} rac{RT_c}{P_c}$$

where $T_r = T/T_c$ is the reduced temperature of the compound. Noted that $\frac{1}{\sqrt{T}}$ term in RK is vanished and its effect is put into a in SRK.

1.6 Peng-Robinson (PR) real gas equation 1976

The Peng-Robinson EoS further modified the RK EoS by modifying the attractive term, giving

$$p = rac{RT}{V_m - b} - rac{alpha}{V_m(V_m + b) + b(V_m - b)}$$

where

$$lpha = \left[1 + (0.37464 + 1.54226\omega - 0.26992\omega^2)(1 - \sqrt{T_r})\right]^2$$
 $a = 0.457235 \frac{R^2 T_c^2}{P_c}$
 $b = 0.077796 \frac{RT_c}{P_c}$

1.7 PR-RK real gas equation 2005

The limited accuracy of the two-parameter cubic EoS is caused by the inconsistency in the density dependence of its equation rather than the empirical nature of model constants a and b. Many researchers proposed three parameters EoS. In 2005, Cismondi and Mollerup have proposed a three-parameter cubic EoS (hereafter RK-PR EoS) by introducing an additional parameter, δ_1 .

$$p = rac{RT}{V_m - b} - rac{alpha}{(V_m + \delta_1 b)(V_m + \delta_2 b)}$$

where δ_2 is not independent and $\delta_2=(1-\delta_1)(1+\delta_1)$. And it will degenerate to SRK when $\delta_1=1,\delta_2=0$; and degenerate to PR when $\delta_1=1+\sqrt{2},\delta_2=1-\sqrt{2}$.

1.8 Mixture properties

When handling mixture, to obtain the aggregate parameters a_m , b_m of the cubic EoS, mixing rules from the van der Waals one-fluid theory is applied:

$$a_m = \sum_i^{N_s} \sum_j^{N_s} X_i X_j a_{ij} \ b_m = \sum_i^{N_s} X_i b_i$$

where X_i is the mole fraction of the i^{th} components of the mixture, X_j is the mole fraction of the j^{th} components of the mixture. Besides, a_{ij} is the attractive term between a molecule of species i and species j, which is usually calculated by $a_{ij} = (a_i a_j)^{1/2}$ or $a_{ij} = (a_i a_j)^{1/2} (1 - \delta_{ij})$. δ_{ij} is an empirically determined binary interaction coefficient characterizing the binary formed by component i and component i.

The mixture properties $(a\alpha)_m$ is similarly:

$$(alpha)_m = \sum_{i=1}^N \sum_{j=1}^N X_i X_j (a_{ij}lpha_{ij}) \ (a_{ij}lpha_{ij}) = \sqrt{a_i a_j lpha_i lpha_j}$$

2. The calibration model of alpha

The lpha is previously calculated by Sovae-type functions like

$$\begin{array}{l} \bullet \ \ \ln {\rm SRK:} \ \alpha = \left[1+(0.480+1.574\omega-0.176\omega^2)(1-\sqrt{T_r})\right]^2 \\ \bullet \ \ \ln {\rm PR:} \ \alpha = \left[1+(0.37464+1.54226\omega-0.26992\omega^2)(1-\sqrt{T_r})\right]^2 \end{array}$$

But those are empirical approximations and would lead to notable effects in particular thermodynamically sensitive simulations. Here a Gaussian Process (GP) based semi-parametric method is used to utilize the information of experimental data as most as possible.

2.1 The GP model

Here we're trying to fit a function that can map the inputs including T_r to the ouput α . Using the concept of Gaussian Process (GP) model, given the observations (X,y), when applying GP to predict the target y_* at new input X_* , one have

$$\begin{split} \mu(y_*) &= \boldsymbol{K}_*^T (\boldsymbol{K} + \sigma_n^2 \boldsymbol{I})^{-1} y \\ \sigma^2(y_*) &= \boldsymbol{K}_{**} - \boldsymbol{K}_*^T (\boldsymbol{K} + \sigma_n^2 \boldsymbol{I})^{-1} \boldsymbol{K}_* \end{split}$$

where σ_n^2 is the covariance of estimated Gaussian noise from observed data (X,y), K related terms are the covariance matrices as

$$egin{aligned} oldsymbol{K} &= cov(oldsymbol{X}, oldsymbol{X}) \ oldsymbol{K}_* &= cov(oldsymbol{X}, oldsymbol{X}_*) \ oldsymbol{K}_{**} &= cov(oldsymbol{X}_*, oldsymbol{X}_*) \end{aligned}$$

2.2 Kernel functions

The covariance are usually obtained via kernel functions, which are very important for GP's data training. The most famous ones are:

• Squared exponential (or say Radial Basis Function, RBF):

$$cov(x_i, x_j) = \sigma_F^2 \exp \left(-rac{d(x_i, x_j)^2}{2\gamma^2}
ight)$$

where $d(x_i, x_j)$ is the Euclidean distance of x_i and x_j , l is the length scale of kernel size.

Matern:

$$k(x_i,x_j) = rac{1}{\Gamma(
u)2^{
u-1}} \Biggl(rac{\sqrt{2
u}}{l} d(x_i,x_j)\Biggr)^
u K_
u \Biggl(rac{\sqrt{2
u}}{l} d(x_i,x_j)\Biggr),$$

where Γ is the Gamma function, $K_{
u}$ is the modified Bessel function of order u.

2.3 Basis functions

The basis function provides an important ability to model the mean of y to zero.

$$y(\boldsymbol{x}) = f(\boldsymbol{x}; \boldsymbol{\theta}) + b_{\boldsymbol{\theta}}(\boldsymbol{x}) + \epsilon$$

where ϵ is the (Gaussian) stochastic error in the database, $f(x; \theta)$ is a basis function that can model y(x) as much as possible, and $b_{\theta}(x)$ is the bias term that can compensate the error between $f(x; \theta)$ and y(x).

To solve this optimization problem, let $f(x; \theta) = \theta_3 x_2 + \theta_2 x_1 + \theta_1$ as a linear combination of the input variables x, where the input variable are $x = [x_1, x_2]$, x_1 and x_2 are the input from each dimension; and $\theta = [\theta_1, \theta_2, \theta_3]$ are corresponding parameters in the linear combination function f.

The corresponding kernel function then be formed as:

$$k(x_i, x_j) = k0(x_i, x_j) - h_{\theta}(x_i)^T \boldsymbol{H}_{\theta} h_{\theta}(x_j)$$

where $h_{\theta}(x)$ is the 1st order global sensitivity obtained from:

$$h_{ heta}(x) = \int_{\mathbf{Y}}
abla_{ heta} f(\xi) k_0(x,\xi) d\xi$$

the 2nd order global sensitivity, i.e., Hessian matrix is

$$oldsymbol{H}_{ heta} = \int_{oldsymbol{X}} \int_{oldsymbol{X}} \left(
abla_{ heta} f(\xi)
ight) \! \left(
abla_{ heta} f(\zeta)
ight)^T \! k_0(\xi,\zeta) d\xi d\zeta$$

The parameter gradient of \boldsymbol{f} shows

$$abla_{ heta}f(oldsymbol{x}) = rac{\partial}{\partialoldsymbol{ heta}}f(oldsymbol{x};oldsymbol{ heta}) = egin{bmatrix} 1 \ x_1 \ x_2 \end{bmatrix}$$

The matrix constructed by gradients' outer product of randomly samples ξ,ζ

$$(
abla_{ heta}f(oldsymbol{\xi}))(
abla_{ heta}f(oldsymbol{\zeta}))^T = egin{bmatrix} 1 \ \xi_1 \ \xi_2 \end{bmatrix} [1 \quad \zeta_1 \quad \zeta_2] = egin{bmatrix} 1 \quad \zeta_1 \quad \zeta_2 \ \xi_1 \quad \xi_1\zeta_1 \quad \xi_1\zeta_2 \ \xi_2 \quad \xi_2\zeta_1 \quad \xi_2\zeta_2 \end{bmatrix}$$

The kernel function k_0 is:

$$k_0(\xi,\zeta) = \sigma_F^2 \exp\left(-\sum_i rac{(\xi_i - \zeta_i)^2}{2\gamma_i^2}
ight) = \sigma_F^2 \exp\left(-rac{(\xi_1 - \zeta_1)^2}{2\gamma_1^2}
ight) \exp\left(-rac{(\xi_2 - \zeta_2)^2}{2\gamma_2^2}
ight)$$

where γ_1, γ_2 are the parameters of kernel function in dimension x_1 and x_2 .

The corresponding distribution of the evaluated output y_{st} at new $\,x_{st}$ should be:

$$egin{aligned} \mu(y_*) &= f(x_*; heta) + oldsymbol{K}_*^T igg(oldsymbol{K} + rac{\sigma_n^2}{
u} oldsymbol{I} igg)^{-1} \left[y - f(oldsymbol{X}; heta)
ight] \ \sigma^2(y_*) &=
u + oldsymbol{K}_{**} -
u oldsymbol{K}_*^T igg(oldsymbol{K} + rac{\sigma_n^2}{
u} oldsymbol{I} igg)^{-1} oldsymbol{K}_* \end{aligned}$$

where u is a measurement of possible noises, its maximum likelihood estimation (MLE) $\,\hat{
u}$ is

$$\hat{\nu} = \frac{1}{N} [y - f(\boldsymbol{X}; \boldsymbol{\theta})]^T (\boldsymbol{K} + g\boldsymbol{I})^{-1} [y - f(\boldsymbol{X}; \boldsymbol{\theta})]$$

where g is the original possible noise level, which still should be estimated by optimization program.

2.4 Thermodynamic Properties

By setting at least two thermo states like p, T, ρ , one can calculate all the related thermodynamic properties of given mixture. Where pressure p and temperature T can be obtained explicitly from the equation of state (EoS). Density can be obtained by first getting V_m via solving the cubic function:

$$p = \frac{RT}{V - b} - \frac{a\alpha}{V^2 + 2Vb - b^2}$$

Other related thermodynamic properties can be calculated via:

$$egin{aligned} C_v &= \left(rac{\partial H}{\partial T}
ight)_V - V\left(rac{\partial p}{\partial T}
ight) \ C_p &= C_v - Tigg(rac{\partial p}{\partial T}igg)^2 / rac{\partial p}{\partial V} \ s(T,p) &= e(T,p) &= h(T,p) &= \end{aligned}$$

where the corresponding partial derivatives are:

$$\begin{split} \left(\frac{\partial H}{\partial T}\right)_{V} &= C_{p,0} + V \frac{\partial p}{\partial T} - R + \frac{1}{2\sqrt{2}b} \log\left(\frac{V+b}{V-b}\right) T \frac{\partial^{2}(a\alpha)_{m}}{\partial T^{2}} \\ \frac{\partial p}{\partial T} &= \frac{R}{V-b} - \frac{\partial (a\alpha)_{m}}{\partial T} \frac{1}{V^{2} + 2Vb - b^{2}} \\ \frac{\partial p}{\partial V} &= -\frac{RT}{(V-b)^{2}} + 2\frac{(a\alpha)_{m}(V+b)}{V^{2} + 2Vb - b^{2}} \end{split}$$

The alpha related properties are:

$$egin{aligned} rac{\partial (alpha)_m}{\partial T} &= \sum_{i=1}^N \sum_{j=1}^N X_i X_j rac{\partial (a_{ij}lpha_{ij})}{\partial T} \ rac{\partial^2 (alpha)_m}{\partial T^2} &= \sum_{i=1}^N \sum_{j=1}^N X_i X_j rac{\partial^2 (a_{ij}lpha_{ij})}{\partial T^2} \end{aligned}$$

where the derivatives of $(a_{ij}\alpha_{ij})$ are:

$$\begin{split} \frac{\partial(a_{ij}\alpha_{ij})}{\partial T} &= \sqrt{a_ia_j} \frac{\partial \sqrt{\alpha_i\alpha_j}}{\partial T} = \frac{1}{2} \sqrt{a_ia_j\alpha_i\alpha_j} \left(\frac{1}{\alpha_i} \frac{\partial \alpha_i}{\partial T} + \frac{1}{\alpha_j} \frac{\partial \alpha_j}{\partial T} \right) \\ \frac{\partial^2(a_{ij}\alpha_{ij})}{\partial T^2} &= \frac{1}{2} \sqrt{a_ia_j\alpha_i\alpha_j} \left[\frac{1}{\alpha_i} \frac{\partial^2\alpha_i}{\partial T} + \frac{1}{\alpha_j} \frac{\partial^2\alpha_j}{\partial T} + \frac{2}{\alpha_i\alpha_j} \frac{\partial \alpha_i}{\partial T} \frac{\partial \alpha_j}{\partial T} - \frac{1}{2} \left(\frac{1}{\alpha_i} \frac{\partial \alpha_i}{\partial T} + \frac{1}{\alpha_j} \frac{\partial \alpha_j}{\partial T} \right)^2 \right] \end{split}$$

Finally, we only need to get $\frac{\partial \alpha}{\partial T}$ and $\frac{\partial^2 \alpha}{\partial T^2}$.

• For the original Peng-Robinson EoS framwork,

$$lpha = \left[1 + \kappa \left(1 - \sqrt{T/T_c}
ight)
ight]^2$$

Therefore,

$$rac{\partial lpha}{\partial T} = rac{\kappa^2}{T_c} - (\kappa^2 + \kappa) rac{1}{\sqrt{Tc}} T^{-1/2} \ rac{\partial^2 lpha}{\partial T^2} = rac{1}{2} (\kappa^2 + \kappa) rac{1}{\sqrt{T_c}} T^{-3/2}$$

• For the Gaussian Process's framework without basis function, we have

$$\alpha(x_*) = \mu(\alpha_*) = \boldsymbol{K}_*^T (\boldsymbol{K} + \sigma_n^2 \boldsymbol{I})^{-1} \alpha_{obs}$$

where α_{obs} is the given observed data of α from NIST or experimental data at specific temperatures and pressures. The right hand side (RHS) term $(\boldsymbol{K} + \sigma_n^2 \boldsymbol{I})^{-1} \alpha_{obs}$ is known during the prediction of $\alpha(x_*)$, hence is denoted as \boldsymbol{m} here. For a single new evaluation point $x_* = (T_{r,*}, P_{r,*})$, the \boldsymbol{K}_* is matrix of dimension $N \times 1$, so:

$$lpha = \sum_{i=1}^N k_{*,i} m_i$$

where

$$k_{*,i} = \sigma_F^2 \exp \left(-rac{(T_{r,*} - T_{r,i})^2}{2\gamma_T^2} - rac{(P_{r,*} - P_{r,i})^2}{2\gamma_T^2}
ight)$$

Thus,

$$egin{aligned} rac{\partial lpha}{\partial T} &= \sum_{i=1}^N m_i rac{\partial k_{*,i}}{\partial T} = \sum_{i=1}^N m_i \left(-rac{1}{T_c} rac{T_{r,*} - T_{r,i}}{\gamma_T^2}
ight) k_{*,i} \ rac{\partial^2 lpha}{\partial T^2} &= \sum_{i=1}^N m_i rac{\partial^2 k_{*,i}}{\partial T^2} = \sum_{i=1}^N m_i \left(rac{1}{T_c^2} rac{(T_{r,*} - T_{r,i})^2}{\gamma_T^4} - rac{1}{T_c^2} rac{1}{\gamma_T^2}
ight) k_{*,i} \end{aligned}$$

• For the Gaussian Process's framework with basis function, we have:

$$egin{aligned} lpha(x_*) = \mu(lpha_*) = f(x_*; heta) + oldsymbol{K}_*^T igg(oldsymbol{K} + rac{\sigma_n^2}{
u} oldsymbol{I} igg)^{-1} \left[y - f(oldsymbol{X}; heta)
ight] \end{aligned}$$

Similarly, one can define the known RHS terms as $m{m} = \left(m{K} + rac{\sigma_n^2}{
u} m{I}
ight)^{-1} [y - f(m{X}; heta)]$, and $lpha = f(x_*; heta) + \sum_{i=1}^N k_{*,i} m_i$, with

$$k_{*,i} = k0(x_*,x_i) - h_{ heta}(x_*)^T oldsymbol{H}_{ heta} h_{ heta}(x_i)$$

where

$$egin{align} h_{ heta}(x_*) &= \int_{oldsymbol{X}}
abla_{ heta}f(\xi)k_0(x_*,\xi)d\xi \ h_{ heta}(x_i) &= \int_{oldsymbol{X}}
abla_{ heta}f(\xi)k_0(x_i,\xi)d\xi \ \end{split}$$

so $m{H}_{ heta}h_{ heta}(x_i)$ is a known term and is hereby denoted as $g_{ heta}(x_i)=[g_{i,1},g_{i,2},g_{i,3}].$ Therefore,

$$k_{*,i} = k0(x_*, x_i) - h_{\theta}(x_*)^T g_{\theta}(x_i) = k0(x_*, x_i) - g_{i,1} \sum_{j=1}^N k0(x_*, x_j) - g_{i,2} \sum_{j=1}^N x_{j,1} k0(x_*, x_j) - g_{i,3} \sum_{j=1}^N x_{j,2} k0(x_*, x_j)$$

$$\frac{\partial k_{*,i}}{\partial T} = \frac{\partial k0(x_*, x_i)}{\partial T} - g_{i,1} \sum_{j=1}^N \frac{\partial k0(x_*, x_j)}{\partial T} - g_{i,2} \sum_{j=1}^N \left[x_{j,1} \frac{\partial k0(x_*, x_j)}{\partial T} + \frac{\partial x_{j,1}}{\partial T} k0(x_*, x_j) \right] - g_{i,3} \sum_{j=1}^N x_{j,2} \frac{\partial k0(x_*, x_j)}{\partial T}$$

$$\frac{\partial^2 k_{*,i}}{\partial T^2} = \frac{\partial^2 k0(x_*, x_i)}{\partial T^2} - g_{i,1} \sum_{j=1}^N \frac{\partial^2 k0(x_*, x_j)}{\partial T^2} - g_{i,2} \sum_{j=1}^N \left[x_{j,1} \frac{\partial^2 k0(x_*, x_j)}{\partial T^2} + 2 \frac{\partial x_{j,1}}{\partial T} \frac{\partial k0(x_*, x_j)}{\partial T} \right] - g_{i,3} \sum_{j=1}^N x_{j,2} \frac{\partial^2 k0(x_*, x_j)}{\partial T^2}$$

And finally:

$$egin{aligned} rac{\partial lpha}{\partial T} &= rac{\partial f}{\partial T} + \sum_{i=1}^N m_i rac{\partial k_{*,i}}{\partial T} \ &rac{\partial^2 lpha}{\partial T^2} &= rac{\partial^2 f}{\partial T^2} + \sum_{i=1}^N m_i rac{\partial^2 k_{*,i}}{\partial T^2} \end{aligned}$$

2.5 Implementation

Reading Alpha related files:

```
setSpeciesCoeffs(species, a, b, w)
   -> readAlphaPara(filename, BasisTheta, KernelGamma, KernelSigma)
        -> readCSV(filename, para, nrow, ncol)
    -> readAlphaData(filename, AlphaX, Alphay)
        -> readCSV(filename, para, nrow, ncol)
```

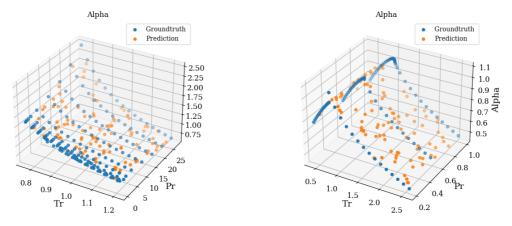
The call stack for calculating density

```
ThermoPhase::setState_TP(t, p)
   -> PengRobinsonAlphaGP::setTemperature(t)
   -> Phase::setTemperature(t)
   -> _updateReferenceStateThermo();
   -> updateMixingExpressions();
        -> mixAlpha()
        -> calculateAB(a, b, aAlpha_mix)

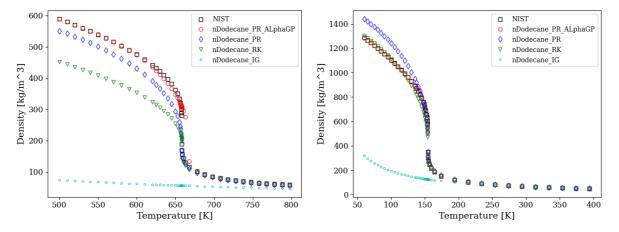
-> PengRobinsonAlphaGP::setPressure(p)
        -> densityCalc(t, p, FLUID_PHASE, rhoNow)
        -> updateAlpha(t, p)
        -> setTemperature(t)
        -> solveCubic(t, p, a, b, aAlpha_mix, Vroot)
```

2.6 Validation

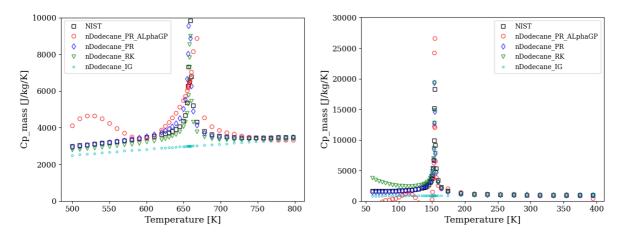
- The training data of α along with T_r, P_r is generated via quering in NIST database. For a given T, P, one can get V in NIST database and α is obtained using PR equation of state, in <code>genAlpha.py</code>.
- The training of Gaussian Process is accomplished by the Matlab code trainGP.m, using fitrgp function in Matlab.
- Validation in Python code test_GP.py of Gaussian Process, where the left panel is C12 and the right panel shows the result of oxygen.



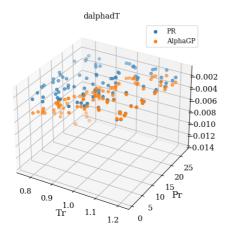
• Validation in Cantera implementation. Below is the comparison of **density** ρ , where the left panel is C12 and the right panel shows the result of oxygen.

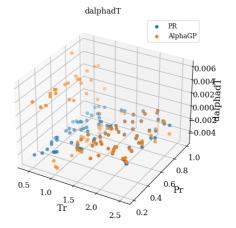


• Validation in Cantera implementation. Below is the comparison of **heat capacity** C_p , where the left panel is C12 and the right panel shows the result of oxygen.

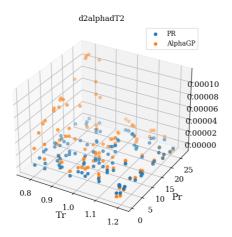


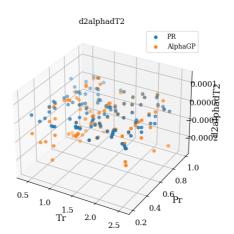
• While C_p is influenced by the 1st and 2nd order gradients of α along with temperature T, the gradients are plotted here compared with PR's gradients. The first order gradients of C12 (left) and oxygen (right):





The second order gradients of C12 (left) and oxygen (right):





3. Appendix

3.1 The equation of heat capacity C_p

Using the chain rule from Table 5, Page 4 of Partial derivatives of thermodynamic state properties for dynamic simulation,

$$C_p = \left(rac{\partial H}{\partial T}
ight)_p = \left(rac{\partial H}{\partial T}
ight)_V - \left(rac{\partial H}{\partial V}
ight)_T \left(rac{\partial p}{\partial T}
ight)_V \left(rac{\partial p}{\partial V}
ight)_T^{-1}$$

where

$$\begin{split} \left(\frac{\partial H}{\partial T}\right)_{V} &= \left(\frac{\partial F}{\partial T}\right)_{V} - \left(\frac{\partial F}{\partial T}\right)_{V} - T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} - V\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ &= -T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} - V\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ \left(\frac{\partial H}{\partial V}\right)_{T} &= \left(\frac{\partial F}{\partial V}\right)_{T} - T\left(\frac{\partial^{2} F}{\partial T \partial V}\right)_{V,T} - \left(\frac{\partial F}{\partial V}\right)_{T} - V\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \\ &= -T\left(\frac{\partial^{2} F}{\partial T \partial V}\right)_{V,T} - V\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \\ \left(\frac{\partial p}{\partial T}\right)_{V} &= -\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ \left(\frac{\partial p}{\partial V}\right)_{T} &= -\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \end{split}$$

therefore,

$$\begin{split} C_p &= \left(\frac{\partial H}{\partial T}\right)_V - \left(\frac{\partial H}{\partial V}\right)_T \left(\frac{\partial p}{\partial T}\right)_V \left(\frac{\partial p}{\partial V}\right)_T^{-1} \\ &= -T \left(\frac{\partial^2 F}{\partial T^2}\right)_V - V \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} + \left[T \left(\frac{\partial^2 F}{\partial T \partial V}\right)_{V,T} + V \left(\frac{\partial^2 F}{\partial V^2}\right)_T\right] \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} / \left(\frac{\partial^2 F}{\partial V^2}\right)_T \\ &= -T \left(\frac{\partial^2 F}{\partial T^2}\right)_V + T \left(\frac{\partial^2 F}{\partial T \partial V}\right)_{V,T} \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} / \left(\frac{\partial^2 F}{\partial V^2}\right)_T \\ &= T \left[\left(\frac{\partial^2 F}{\partial V \partial T}\right)_{V,T}^2 / \left(\frac{\partial^2 F}{\partial V^2}\right)_T - \left(\frac{\partial^2 F}{\partial T^2}\right)_V\right] \end{split}$$