Real Gas Effect in Reacting Flow

This code repository is used to evaluate the advantages of the proposed real gas equation of state (EoS)

1. Code Environment

```
# Cantera used in /opt/cantera_libs/cantera, one can modified code in `src` and
build & install by
conda activate cantera
python3 /usr/bin/scons build
python3 /usr/bin/scons install
```

```
# The code in this repo can be run by
source /usr/local/bin/setup_cantera
python3 src/xxx.py
```

2. Basic Theory

1.0 Generalized EoS

Helmholtz Free Energy: $F \equiv U - TS = A(T, v, n)$, where U is the internal energy of the system, T is the absolute temperature, S is the entropy of the system.

- 1st law of thermodynamics: $dU = \delta Q + \delta W$
- 2nd law of thermodynamics: $\delta Q = TdS$
- pressure work: $\delta W = -pdV$

Thus we have:

- dU = TdS pdV
- $\Rightarrow dU = d(TS) SdT pdV$
- $\Rightarrow d(U TS) = -SdT pdV$
- $\Rightarrow dF = -SdT pdV$

Therefore.

$$\begin{split} S &= -\left(\frac{\partial F}{\partial T}\right)_{V} \\ p &= -\left(\frac{\partial F}{\partial V}\right)_{T} \\ U &= F + TS = F - T\left(\frac{\partial F}{\partial T}\right)_{V} \\ H &= U + pV = F - T\left(\frac{\partial F}{\partial T}\right)_{V} - V\left(\frac{\partial F}{\partial V}\right)_{T} \\ C_{p} &= \left(\frac{\partial H}{\partial T}\right)_{p} = T\left[\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{V,T}^{2} / \left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} - \left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V}\right] \\ C_{V} &= \left(\frac{\partial U}{\partial T}\right)_{V} = -T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} \end{split}$$

1.1 Ideal Gas

$$p = \frac{nRT}{V_m}$$

where V_m is the molar volume of the gas .

1.2 Van der Waals (vdW) real gas equation 1783

$$p = \frac{RT}{V_m - b} - \frac{a}{V_m^2}$$

where b is the volume that is occupied by one mole of the molecules. vdW is a cubic type EoS.

1.4 Redlich-Kwong (RK) real gas equation 1948

$$p = rac{RT}{V_m - b} - rac{a}{\sqrt{T}V_m(V_m + b)}$$

where a is a constant that corrects for attractive potential of molecules and b is a constant that corrects for volume. The constants are different depending on which gas is being analyzed and can be calculated from the critical point data of corresponding gas:

$$a = rac{1}{9(\sqrt[3]{2} - 1)} rac{R^2 T_c^{2.5}}{P_c} \ b = rac{\sqrt[3]{2} - 1}{3} rac{RT_c}{P_c}$$

Where T_c is the temperature at the critical point and P_c is the pressure at the critical point. Noted that a and b are solved analytically from the thermodynamic criteria for the critical point:

$$\left(rac{\partial P}{\partial V}
ight)_T=0, \left(rac{\partial^2 P}{\partial V^2}
ight)_T=0.$$

For more details, please refer to original paper or wikipedia of Redlich Kwong equation of state.

1.5 Soave modified Redlich-Kwong (SRK) real gas equation 1972

In 1966, Barner noted that the acentric factor ω can be used to improve RK EoS. SRK is proposed by Soave as:

$$p = rac{RT}{V_m - b} - rac{alpha}{V_m(V_m + b)}$$

where

$$lpha = \left[1 + (0.480 + 1.574\omega - 0.176\omega^2)(1 - \sqrt{T_r})\right]^2 \ a = rac{1}{9(\sqrt[3]{2} - 1)} rac{R^2 T_c^2}{P_c} \ b = rac{\sqrt[3]{2} - 1}{3} rac{RT_c}{P_c}$$

where $T_r = T/T_c$ is the reduced temperature of the compound. Noted that $\frac{1}{\sqrt{T}}$ term in RK is vanished and its effect is put into a in SRK.

1.6 Peng-Robinson (PR) real gas equation 1976

The Peng-Robinson EoS further modified the RK EoS by modifying the attractive term, giving

$$p = rac{RT}{V_m - b} - rac{alpha}{V_m(V_m + b) + b(V_m - b)}$$

where

$$lpha = \left[1 + (0.37464 + 1.54226\omega - 0.26992\omega^2)(1 - \sqrt{T_r})\right]^2 \ a = 0.457235 rac{R^2 T_c^2}{P_c} \ b = 0.077796 rac{RT_c}{P_c}$$

1.7 PR-RK real gas equation 2005

The limited accuracy of the two-parameter cubic EoS is caused by the inconsistency in the density dependence of its equation rather than the empirical nature of model constants a and b. Many researchers proposed three parameters EoS. In 2005, Cismondi and Mollerup have proposed a three-parameter cubic EoS (hereafter RK-PR EoS) by introducing an additional parameter, δ_1 .

$$p = rac{RT}{V_m - b} - rac{alpha}{(V_m + \delta_1 b)(V_m + \delta_2 b)}$$

where δ_2 is not independent and $\delta_2=(1-\delta_1)(1+\delta_1)$. And it will degenerate to SRK when $\delta_1=1,\delta_2=0$; and degenerate to PR when $\delta_1=1+\sqrt{2},\delta_2=1-\sqrt{2}$.

1.7 Mixture properties

When handling mixture, to obtain the aggregate parameters a_m , b_m of the cubic EoS, mixing rules from the van der Waals one-fluid theory is applied:

$$a_m = \sum_i^{N_s} \sum_j^{N_s} X_i X_j a_{ij} \ b_m = \sum_i^{N_s} X_i b_i$$

where X_i is the mole fraction of the i^{th} components of the mixture, X_j is the mole fraction of the j^{th} components of the mixture. Besides, a_{ij} is the attractive term between a molecule of species i and species j, which is usually calculated by $a_{ij}=(a_ia_j)^{1/2}$

2. The calibration model of alpha

The α is previously calculated by Sovae-type functions like

- In SRK: $lpha = \left[1 + (0.480 + 1.574\omega 0.176\omega^2)(1 \sqrt{T_r})\right]^2$
- In PR: \$\alpha = \left [1+(0.37464+1.54226\omega-0.26992\omega^2)(1-\sqrt{T r}) \right]^2 \$

But those are empirical approximations and would lead to notable effects in particular thermodynamically sensitive simulations. Here **a Gaussian Process (GP) based semi-parametric method** is used to utilize the information of experimental data as most as possible.

2.1 The GP model

Here we're trying to fit a function that can map the inputs including T_r to the ouput α . Using the concept of Gaussian Process (GP) model, given the observations (X, y), when applying GP to predict the target y_* at new input X_* , one have

$$egin{aligned} \mu(y_*) &= oldsymbol{K}_*^T (oldsymbol{K} + \sigma_n^2 oldsymbol{I})^{-1} y \ \sigma(y_*) &= oldsymbol{K}_{**} - oldsymbol{K}_*^T (oldsymbol{K} + \sigma_n^2 oldsymbol{I})^{-1} oldsymbol{K}_* \end{aligned}$$

where σ_n^2 is the covariance of estimated Gaussian noise from observed data (\boldsymbol{X},y) , \boldsymbol{K} related terms are the covariance matrices as

$$egin{aligned} oldsymbol{K} &= cov(oldsymbol{X}, oldsymbol{X}) \ oldsymbol{K}_{**} &= cov(oldsymbol{X}_{*}, oldsymbol{X}_{*}) \end{aligned}$$

2.2 Kernel functions

The covariance are usually obtained via kernel functions, which are very important for GP's data training. The most famous ones are:

• Squared exponential (or say Radial Basis Function, RBF):

$$cov(x_i,x_j) = \exp\left(-rac{d(x_i,x_j)^2}{2l^2}
ight)$$

where $d(x_i, x_j)$ is the Euclidean distance of x_i and x_j , l is the length scale of kernel size.

Matern:

$$k(x_i,x_j) = rac{1}{\Gamma(
u)2^{
u-1}} \Bigg(rac{\sqrt{2
u}}{l} d(x_i,x_j)\Bigg)^
u K_
u \Bigg(rac{\sqrt{2
u}}{l} d(x_i,x_j)\Bigg),$$

where Γ is the Gamma function, $K_{
u}$ is the modified Bessel function of order u.

2.3 Basis functions

The basis function provides an important ability to model the mean of y to zero.

$$y(\mathbf{x}) = f(\mathbf{x}; \boldsymbol{\theta}) + b_{\boldsymbol{\theta}}(\mathbf{x}) + \epsilon$$

where ϵ is the (Gaussian) stochastic error in the database, $f(x; \theta)$ is a basis function that can model y(x) as much as possible, and $b_{\theta}(x)$ is the bias term that can compensate the error between $f(x; \theta)$ and y(x).

To solve this optimization problem, let $f(\boldsymbol{x};\boldsymbol{\theta}) = \theta_3 x_2 + \theta_2 x_1 + \theta_1$ as a linear combination of the input variables \boldsymbol{x} , where the input variable are $\boldsymbol{x} = [x_1, x_2]$, x_1 and x_2 are the input from each dimension; and $\boldsymbol{\theta} = [\theta_1, \theta_2, \theta_3]$ are corresponding parameters in the linear combination function f.

The corresponding kernel function then be formed as:

$$k(x_i,x_j) = k0(x_i,x_j) - h_{ heta}(x_i)^T oldsymbol{H}_{ heta} h_{ heta}(x_j)$$

where $h_{\theta}(x)$ is the 1st order global sensitivity obtained from:

$$h_{ heta}(x) = \int_{oldsymbol{X}}
abla_{ heta} f(oldsymbol{x}) k_0(x, \xi) d \xi$$

the 2nd order global sensitivity, i.e., Hessian matrix is

$$oldsymbol{H}_{ heta} = \int_{oldsymbol{X}} \int_{oldsymbol{X}} (
abla_{ heta} f(\xi)) (
abla_{ heta} f(\zeta))^T k_0(\xi,\zeta) d\xi d\zeta$$

The parameter gradient of f shows

$$abla_{ heta}f(oldsymbol{x}) = rac{\partial}{\partialoldsymbol{ heta}}f(oldsymbol{x};oldsymbol{ heta}) = egin{bmatrix} 1 \ x_1 \ x_2 \end{bmatrix}$$

The matrix constructed by gradients' outer product of randomly samples ξ, ζ

$$(
abla_{ heta}f(oldsymbol{\xi}))(
abla_{ heta}f(oldsymbol{\zeta}))^T = egin{bmatrix} 1 \ \xi_1 \ \xi_2 \end{bmatrix} egin{bmatrix} 1 & \zeta_1 & \zeta_2 \end{bmatrix} = egin{bmatrix} 1 & \zeta_1 & \zeta_2 \ \xi_1 & \xi_1\zeta_1 & \xi_1\zeta_2 \ \xi_2 & \xi_2\zeta_1 & \xi_2\zeta_2 \end{bmatrix}$$

The kernel function k_0 is:

$$k_0(\xi,\zeta) = \exp\left(-\sum_i rac{(\xi_i-\zeta_i)^2}{\gamma_i}
ight) = \exp\left(-rac{(\xi_1-\zeta_1)^2}{\gamma_1}
ight) \exp\left(-rac{(\xi_2-\zeta_2)^2}{\gamma_2}
ight)$$

where γ_1, γ_2 are the parameters of kernel function in dimension x_1 and x_2 .

The corresponding distribution of the evaluated output y_* at new x_* should be:

$$egin{aligned} \mu(y_*) &= f(x_*; heta) + oldsymbol{K}_*^T igg(oldsymbol{K} + rac{oldsymbol{K}_{**}}{
u} oldsymbol{I} igg)^{-1} \left[y - f(oldsymbol{X}; heta)
ight] \ \sigma(y_*) &=
u + oldsymbol{K}_{**} -
u oldsymbol{K}_*^T igg(oldsymbol{K} + rac{oldsymbol{K}_{**}}{
u} oldsymbol{I} igg)^{-1} oldsymbol{K}_* \end{aligned}$$

where ν is a measurement of possible noises, its maximum likelihood estimation (MLE) $\hat{\nu}$ is

$$\hat{\nu} = \frac{1}{N}[y - f(\boldsymbol{X}; \boldsymbol{\theta})]^T (\boldsymbol{K} + g\boldsymbol{I})^{-1}[y - f(\boldsymbol{X}; \boldsymbol{\theta})]$$

where g is the original possible noise level, which still should be estimated by optimization program.

2.4 Implementation

2.5 Validation

3. Appendix

Using the chain rule from Table 5, Page 4 of <u>Partial derivatives of thermodynamic state properties</u> <u>for dynamic simulation</u>,

$$C_p = \left(rac{\partial H}{\partial T}
ight)_n = \left(rac{\partial H}{\partial T}
ight)_V - \left(rac{\partial H}{\partial V}
ight)_T \left(rac{\partial p}{\partial T}
ight)_V \left(rac{\partial p}{\partial V}
ight)_T^{-1}$$

where

$$\begin{split} \left(\frac{\partial H}{\partial T}\right)_{V} &= \left(\frac{\partial F}{\partial T}\right)_{V} - \left(\frac{\partial F}{\partial T}\right)_{V} - T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} - V\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ &= -T\left(\frac{\partial^{2} F}{\partial T^{2}}\right)_{V} - V\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ \left(\frac{\partial H}{\partial V}\right)_{T} &= \left(\frac{\partial F}{\partial V}\right)_{T} - T\left(\frac{\partial^{2} F}{\partial T \partial V}\right)_{V,T} - \left(\frac{\partial F}{\partial V}\right)_{T} - V\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \\ &= -T\left(\frac{\partial^{2} F}{\partial T \partial V}\right)_{V,T} - V\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \\ \left(\frac{\partial P}{\partial T}\right)_{V} &= -\left(\frac{\partial^{2} F}{\partial V \partial T}\right)_{T,V} \\ \left(\frac{\partial P}{\partial V}\right)_{T} &= -\left(\frac{\partial^{2} F}{\partial V^{2}}\right)_{T} \end{split}$$

therefore,

$$\begin{split} C_p &= \left(\frac{\partial H}{\partial T}\right)_V - \left(\frac{\partial H}{\partial V}\right)_T \left(\frac{\partial p}{\partial T}\right)_V \left(\frac{\partial p}{\partial V}\right)_T^{-1} \\ &= -T \left(\frac{\partial^2 F}{\partial T^2}\right)_V - V \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} + \left[T \left(\frac{\partial^2 F}{\partial T \partial V}\right)_{V,T} + V \left(\frac{\partial^2 F}{\partial V^2}\right)_T\right] \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} / \left(\frac{\partial^2 F}{\partial V^2}\right)_T \\ &= -T \left(\frac{\partial^2 F}{\partial T^2}\right)_V + T \left(\frac{\partial^2 F}{\partial T \partial V}\right)_{V,T} \left(\frac{\partial^2 F}{\partial V \partial T}\right)_{T,V} / \left(\frac{\partial^2 F}{\partial V^2}\right)_T \\ &= T \left[\left(\frac{\partial^2 F}{\partial V \partial T}\right)_{V,T}^2 / \left(\frac{\partial^2 F}{\partial V^2}\right)_T - \left(\frac{\partial^2 F}{\partial T^2}\right)_V\right] \end{split}$$