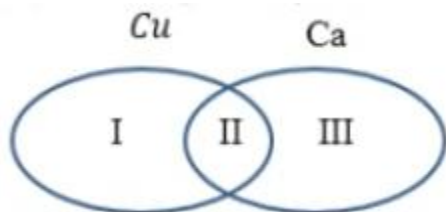


1.  $\text{KHCO}_3$  qizdirilganda 5,6 litr (n.sh) gaz ajraldi. Reaksiyaing parchalinish unumi 80% bo'lsa qoldiqdagi  $\text{KHCO}_3$  miqdorini (mol) toping.

A) 0,5    B) 0,25    C) 0,125    D) 0,625

2. Venn diagrammasida asosida metallarning o'ziga xos va umumiy xususiyatlarini moslang.



a) valentligi doim II    b)  $\text{Me}^{+2}$  ionida 18 ta electron bor    c)  $\text{MeCl}_2$  tuzi suvda eriydi  
d) oksidi suvda erimaydi

A) I-d; II-a,c; III-b  
B) I-a; II-c; III-b,d  
C) I-b; II-c; III-a,d  
D) I-d; II-c; III-a,b

3.  $\text{NH}_3$  ga  $\text{HCl}$  ta'sir ettirilganda qanday tuz hosil bo'ladi.

A) ammiy xlorid  
B) ammoniy xlorit  
C) nordon tuz  
D) asosli tuz

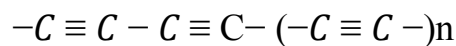
4. 0,1 mol sulfid angidrid atomlar soni x xramm arsenat kislota molekulalari soniga teng. X ni toping.

A) 24,1    B) 14,2    C) 41,2    D) 42,6

5. 2:3 mol nisbatda temir va rux aralashmasini xlorlash uchun 26,88 litr (n.sh) xlor gazi sarflandi. Hosil bo'lgan tuzlarga mo'l miqdorda natriy gidroksid eritmasi ta'sir ettirilganda hosil bo'lgan cho'kma massasi (g) topilsin.

A) 102,2    B) 59,4    C) 42,8    D) 95,4

6. Quyidagi tuzilish uglerodning qaysi allotropiyasiga mos keladi?



A) polien  
B) Grafit  
C) karbin  
D) Fluren

7. Bariy xlorat molekulasidagi (sigma) bog'lar yig'indisi, umumiy bog'lari (sigma+pi) soniga teng bo'lgan modda?

A) mochevina  $(\text{NH}_2)_2\text{CO}$   
B) bariy borat  
C) kaliy fosfat  
D) alyuminiy karbonat

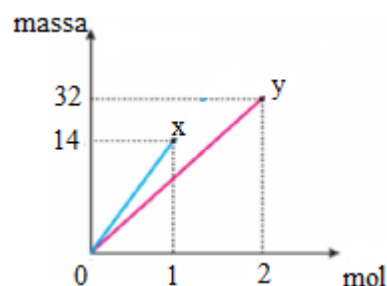
8. Teng massadagi  $\text{Mg}_3\text{N}_2$  va  $\text{CaCO}_3$  aralashmasi mo'l miqdordagi xlorid kislota eritilganda 16,8 litr (n.sh.) gaz ajraldi. Dastlabki aralashma massasini toping.

A) 200    B) 100  
C) 50    D) 25

9.  $\text{KMnO}_4 \rightarrow \text{K}_2\text{MnO}_4 + \text{MnO}_2 + \text{O}_2$  reaksiyasida ajralgan kislorod miqdori (mol) parchalanmay qolgan  $\text{KMnO}_4$  moliga teng bo'lsa,  $\text{KMnO}_4$  parchalanish unumini toping.

A) 50 %    B) 66,(6) %    C) 33,(3)    D) 40%

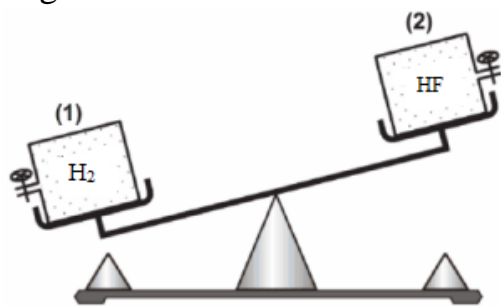
10. Quyidagi digramma orqali qaysi moddaning tarkibidagi atomlar moli va massa ifodalangan?



A)  $\text{SiO}_2$     B)  $\text{Li}_2\text{O}$     C)  $\text{NO}_2$     D)  $\text{Li}_2\text{O}_2$

**11.** Ikkinchi idishda qancha massa modda bo'lsa, birinchi idishda shuncha mol modda bor. Quyidagi qaysi amallar tarozi pallalarini muvozanatlashtiradi?

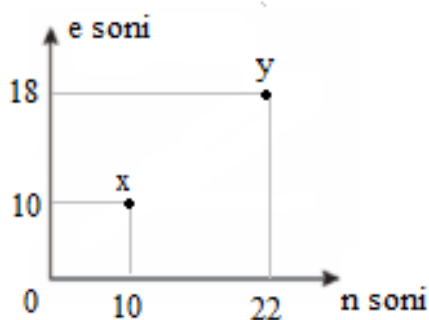
- 1)  $H_2$  ning miqdori ikki barobarga orttirilsa
- 2) HF miqdori ikki barobar orttirilsa
- 3) HF ga yana bir mol shu gazdan qo'shilsa
- 4)  $H_2$  va HF ning mol nisbati 10:1 ga tenglashtirilganda
- 5)  $H_2$  ning miqdorini 10 molga kamaytirilsa
- 6) har ikki idishdagi moddalar miqdorlarini tenglashtirish



- A) 1 3 5    B) 2 4 6    C) 1 3 4    D) 2 4

**12.** Quyida jadvalda berilgan x va y larga mos keluvchi zarrachalarni toping.

- 1)  $^{40}Ca^{2+}$     2)  $^{40}Ar$     3)  $^{20}F^-$     4)  $^{37}Cl^-$   
 5)  $^{17}O^{2-}$     6)  $^{20}Ne$     7)  $^{21}Na^+$     8)  $^{43}Sc^{3+}$



- A)  $x=6, y=2, 8$   
 B)  $x=3, 5, 6, y=1, 2, 4$   
 C)  $x=5, 6, 7, y=1, 2, 8$   
 D)  $x=3, 6, y=2, 8$

**13.** 3 mol magniy azot bilan qizdirilganda reaksiyaga kirishdi. Olingan mahsulotning massasi brom atomining massasidan qanchaga og'ir?

- A) 60    B) 100    C) 20    D) 25

**14.** Tarkibida 20% qo'shimchalari bo'lgan ohaktoshdan 33,6 l karbonat angidrid olingan bo'lsa reaksiya uchun necha gr ohaktosh olingan (reaksiya unumi-75%)

- A) 160    B) 250    C) 200    D) 240

**15.** Kumush nitrat eritmasiga kaliy xlorid qo'shilganda hosil bo'lgan cho'kma massasi dastlabki kaliy xlorid massasidan 13,8 gr ga ko'p bo'lsa, dastlabki tuzlarning massalari yig'indisini toping.

- A) 48,9    B) 34    B) 14,9    D) 28,7

**16.**  $Zn(NO_3)_2$  va  $KNO_3$  termik parchalanganda 4 mol qo'ng'ir rangli gaz va 2 mol oddiy modda hosil bo'lgan bo'lsa, dastlabki tuzlarning mollarini toping.

- A) 2;1    B) 2;2    C) 3;1    D) 1;2

**17.** Kislotali oksid, nordon tuz, ishqor berilgan javobni belgilang.

- A)  $MnO_2$ ;  $NaHS$ ;  $KOH$   
 B)  $P_2O_5$ ;  $NaHSO_4$ ;  $Be(OH)_2$   
 C)  $SO_3$ ;  $CaOHCl$ ;  $Mg(OH)_2$   
 D)  $CrO_3$ ;  $KHCO_3$ ;  $Sr(OH)_2$

**18.** 24,5 gr Bertole tuzi parchalanishidan hosil bo'lgan kislorodni qanch gr kaliypermanganat 50 % unum bilan parchalanishidan olish mumkin.

- A) 94,8    B) 47,4    C) 31,6    D) 189,6

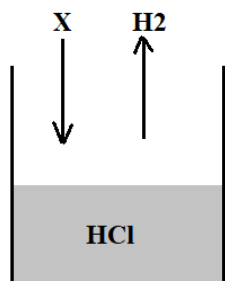
**19.**  $2X_3 + kY_m \rightarrow 3X_nY_m$  reaksiya tenglamasidagi n va k ning qiymatlarini mos ravishda aniqlang.

- A) 1; 6  
 B) 2; 3  
 C) 1; 3  
 D) 2; 1

**20.** III valentli metall nitridi tarkibida metall bilan azotning massa nisbatlari 3:2 bo'lsa modda molyar massasini(g/mol) aniqlang.

- A) 35    B) 83    C) 87    D) 49

21. X qaysi metall bo'lishi mumkin ?



- A) Cu B) Au C) Ag D) Fe

22. Laboratoriya xonasidagi reaktivlar javonining qaysi qavat(lar)ida moddalar noto'g'ri joylashtirilgan.

	Oksidlar		
I	Al <sub>2</sub> O <sub>3</sub>	ZnO	Cu <sub>2</sub> O
	Asoslar		
II	Fe(OH) <sub>3</sub>	HNO <sub>3</sub>	Ca(OH) <sub>2</sub>
	Kislotalar		
III	BaCl <sub>2</sub>	H <sub>2</sub> SO <sub>4</sub>	NaOH
	Tuzlar		
IV	K <sub>2</sub> SO <sub>4</sub>	NaNO <sub>3</sub>	BaCl <sub>2</sub>

- A) I, III B) II, IV C) I, II D) II, III

23. Ketma-ket joylashgan X, Y, Z, T, L, M elementlari tarkibidagi protonlari yigindisi 75 ga teng bolsa, M elementining eng yuqori valentligini toping.

- A) VI B) III C) IV D) V

24. Quyidagilardan faqat normal tuzlarni ko'rsating.

- 1) K<sub>2</sub>SO<sub>4</sub> 2) (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> 3) AlOHSO<sub>4</sub>  
4) Na<sub>2</sub>HPO<sub>3</sub> 5) KH<sub>2</sub>PO<sub>4</sub> 6) CaClBr  
7) CaOCl<sub>2</sub>

- A) 1,6 B) 2,4 C) 2,7 D) 3,5

25. Bir idishga 1350 gr nitrat kislota (p=1,5gr/mol) sig'ishi ma'lum bo'lsa, bu idishga necha mol atom saqlagan suv sigadi.

- A) 50 B) 150·N<sub>A</sub> C) 337,5 D) 150

26. Qaysi molekula tarkibidagi elementlar sonining atomlar soniga nisbati 1:3 ga teng?

- A) AlPO<sub>4</sub> B) Fe(NO<sub>3</sub>)<sub>2</sub>  
C) CaHPO<sub>3</sub> D) CuSO<sub>4</sub>

27. Bariy gidroksid va mo'l miqdordagi nitrat kislota ta'sirlashuvidan olingan tuz molekulasi tarkibidagi atomlar sonini aniqlang.

- A) 5 B) 7 C) 9 D) 4

28. Qaysi atomlarning massalari  $1,66 \cdot 10^{-23}$  g ga farq qiladi?  $1u = 1,66 \cdot 10^{-24}$  g.

- A) He, N  
B) Ar, K  
C) Ne, Ca  
D) Mn, Fe

29. Quyidagi ma'lumotlarning qaysilarida xlor oddiy modda tarzida ifodalangan?

- 1) Xlorning uchta elektron pog'onasi mavjud;  
2) xlor yashil tusli, bo'g'uvchi gaz;  
3) fosgenning 71,7 % ini xlor tashkil etadi;  
4) temirning xlorida yonishidan FeCl<sub>3</sub> hosil bo'ladi;  
5) xlorning geliyga nisbatan zichligi 17,75 ga teng;  
6) tabiatda xlorning ikkita (<sup>35</sup>CL va <sup>37</sup>CL) barqaror izotopi mavjud.

- A) 1, 3 B) 1, 3, 6 C) 2, 4, 5, 6 D) 2, 4,

30. 40 gr mis va kalsiy aralashmasiga xlorid kislota ta'sir etirilganda 13,44 litr (n.sh) gaz ajraldi. Dastlabki aralashmadagi B guruhcha elementining molini toping.

- A) 0,5 B) 1/3 C) 16 D) 1/4