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Fabrication of Single- and Multilayer MoS₂ Film-Based Field-Effect Transistors for Sensing NO at Room Temperature

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Two-dimensional nanomaterials have received much attention in recent years owing to their unusual properties that stem from their quantum and surface effects.^[1–26] Graphene,^[27,28] a single-layer 2D carbon material, exhibits exceptional physical properties such as a high electron conductivity and excellent mechanical strength. Very recently, other 2D semiconducting nanomaterials like transition metal dichalcogenides have also attracted significant research interest and hold great potential for many applications.^[9,11,13,14,16,18,24,25] For example, MoS₂, one kind of the transition metal dichalcogenide, is currently being explored and used in lubrication, as a catalyst for hydrodesulfonation, and for photoelectrochemical hydrogen production.^[29–32] Single-layer MoS₂ prepared from mechanical exfoliation exhibits a dramatically increased luminescence quantum efficiency compared to its bulk counterpart.^[16] Mechanically exfoliated MoS₂ monolayers have also been used to fabricate field-effect transistors (FETs) with high mobilities and current On/Off ratios exceeding 1×10^8 at room temperature, which is comparable to those obtained in graphene nanoribbon-based FETs.^[9] Hence, it is of scientific importance to synthesize and characterize MoS₂ sheets ranging from a single layer to a few layers in order to better apply them in various applications.

Detection of environmental pollution, especially toxic gas, is important and critical to industry, agriculture, and public health.^[33] For example, NO_x gas is one of the most common air pollutants. It is a source of acid rain and can cause serious diseases.^[8] As such, there is an increasing demand to develop

highly sensitive, low cost, and portable gas sensors with low power consumption.^[2] Electronic gas sensors based on FETs have shown promising results.^[2,4–6,8,26,33–35] One-dimensional semiconductors, such as carbon nanotubes^[33–35] and semiconductor nanowires,^[36–38] are popular channel materials used for FET-based sensors. 2D graphene sheets are also favored in electronic sensing because of their unique electronic properties and large specific surface area.^[1–8] In addition, the 2D configuration of graphene sheets, compared to 1D semiconductors, allows for the better adsorption of gas molecules and leads to lower electrical noise and lower detection limits.^[2,4–6,26] As a result, the graphene-based gas sensor has achieved a sensitivity down to the single-molecule level.^[2] MoS₂ sheets, the semiconducting analogue of graphene, are thus expected to be a potential candidate for sensing applications.

In this communication, we report a systematic study of the fabrication of single- and multilayer MoS₂ films using the mechanical exfoliation method. The fabricated MoS₂ FET devices show n-type doping behavior. As a proof of concept, the single- and multilayer MoS₂ FETs were used to detect the adsorption of NO. It was found that the FET sensors based on bilayer (2L), trilayer (3L), and quadrilayer (4L) MoS₂ films exhibited a high sensitivity to NO with detection limit of 0.8 ppm, while the single-layer (1L) MoS₂ device showed a rapid but unstable response.

Single- and multilayer MoS₂ nanosheets were deposited onto Si substrates covered with a 300 nm-thick SiO₂ layer, referred to as Si/SiO₂, using the mechanical exfoliation technique.^[1,9,15] Similar to graphene, MoS₂ nanosheets with different layer numbers show distinguishable contrast on the Si/SiO₂ substrates, as observed by optical microscopy (Figure 1A–D). The thicknesses of MoS₂ films with different optical contrasts (Figure 1A–D) were measured by AFM (Figure 1E–H), which showed that the average height of a single layer of MoS₂ (Figure 1A) is ≈ 0.8 nm (Figure 1E), which is consistent with previous reports.^[9,15,18] The heights of 2L, 3L, and 4L MoS₂ sheets (Figure 1B–D) measured by AFM are 1.5, 2.1, and 2.9 nm, respectively (Figure 1F–H).

Figure 2 shows the Raman spectra of single- and multilayer MoS₂ films. Single-layer MoS₂ exhibited strong bands at 384 and 400 cm⁻¹, which are associated with the in-plane vibrational (E₁_{2g}) and the out-of-plane vibrational (A_{1g}) modes, respectively.^[10,15] As the layer number increased from

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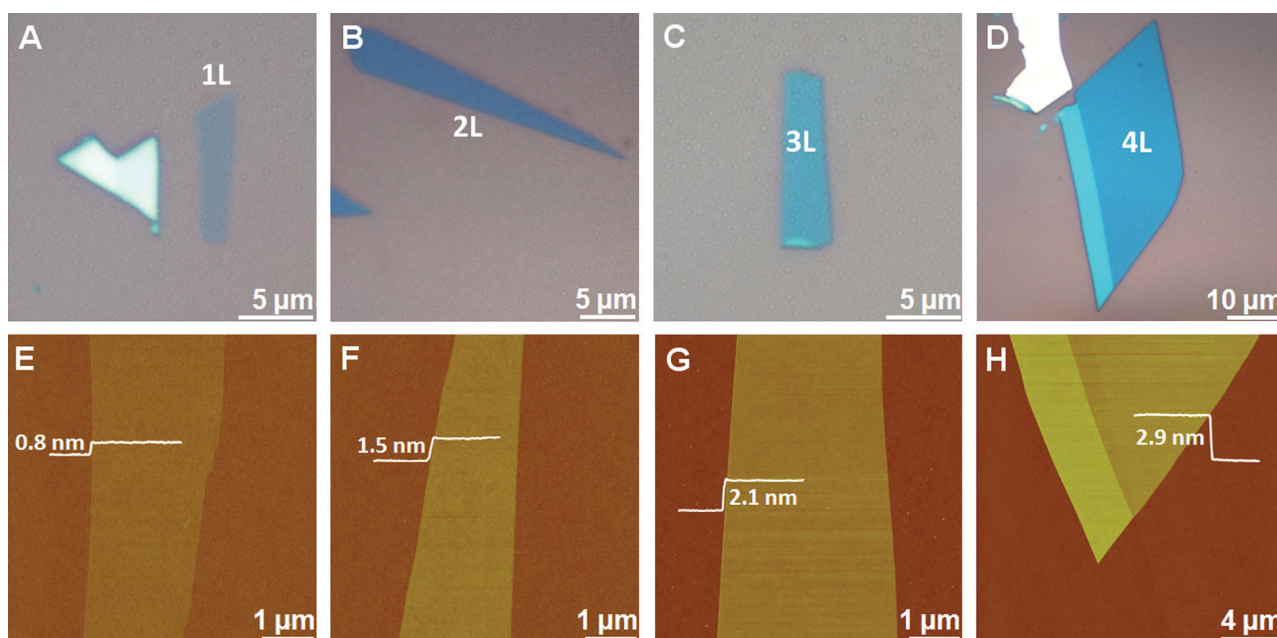


Figure 1. Mechanically exfoliated single- and multilayer MoS₂ films on Si/SiO₂. Optical microscope images of single-layer (1L), bilayer (2L), trilayer (3L), and quadrilayer (4L) MoS₂ films (A–D). Panels E–H show the corresponding AFM images of the 1L (thickness: ≈0.8 nm), 2L (thickness: ≈1.5 nm), 3L (thickness: ≈2.1 nm), and 4L (thickness: ≈2.9 nm) MoS₂ films shown in (A–D).

1L to 4L, a red-shift of the E_{1g} band and a blue-shift of the A_{1g} bands were observed (see Figure 2), which is in consistent with the previous report.^[15]

FET devices are fabricated with exfoliated 1L to 4L MoS₂ films. As an example, a 2L MoS₂ film and its corresponding FET device are shown in **Figures 3A** and **3B**, respectively. The recorded I_{ds} – V_g curve (Figure 3C) is typical of the 2L MoS₂ FET with an n-type channel, which is in agreement with the previous report.^[9] In fact, all the MoS₂ FET devices investigated here possess n-type doping character (see Figure S1, Supporting Information (SI)). The current On/Off ratio of 2L, 3L, and 4L MoS₂ devices is above 10³ (V_g from –10 to 10 V), while it is comparably smaller (≈10²) for the 1L MoS₂ FET. It is important to note that our devices show a lower On/Off

ratio compared to the previously reported single-layer MoS₂ transistor, in which a mobility booster (HfO₂) was used.^[9]

The channel mobility of our devices was calculated based on the equation

$$\mu = \frac{L}{W \times (\epsilon_0 \epsilon_r / d) \times V_{ds}} \times \frac{dI_{ds}}{dV_g},$$

where L is the channel length, W is the channel width, ϵ_0 is 8.854×10^{-12} F m^{–1}, ϵ_r for SiO₂ is 3.9, and d is the thickness of SiO₂ (300 nm). We found that the mobility of our devices was between 0.03 and 0.22 cm² V^{–1} s^{–1}, which is in agreement with the previous reports.^[19,11] Importantly, the mobility of the fabricated FETs was found to increase with the layer number (1L: 0.03; 2L: 0.07; 3L: 0.17; 4L: 0.22 cm² V^{–1} s^{–1}).

As a proof of concept, the as-fabricated single- and multilayer MoS₂ FETs were used to detect NO gas at room temperature. In these experiments, Ti/Au was deposited as drain and source electrodes and the MoS₂ sheets were the active channels. We found that the current response of a single-layer MoS₂ device is not stable (see Figure S2, SI). Contrary to this, the devices fabricated with 2–4L MoS₂ films showed much better performance. **Figure 4A** shows a typical current response of a 2L MoS₂ FET device upon exposure to NO with concentrations ranging from 0.3 to 2 ppm. The decrease in current, i.e., increase in resistance, upon exposure of the device to NO is most likely due to the p-doping effect,^[2,4,6,39] which is consistent with the charge transfer mechanism taking place in the 2D graphene^[3] and 1D carbon nanotube^[34] based gas sensors.

The complete desorption of adsorbed NO molecules upon N₂ flow is very slow, which is possibly due to the strong chemisorption of NO on the MoS₂ surface. Interestingly, both the adsorption and desorption processes of NO can be divided into two steps (a rapid one and a slow one). In

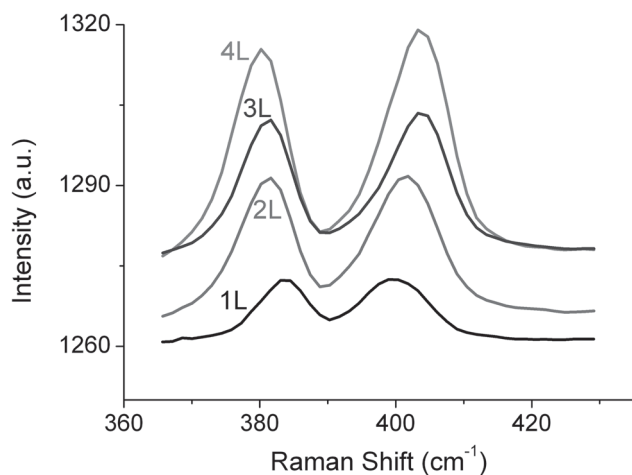


Figure 2. Raman spectra of single- and multilayer (1L to 4L) MoS₂ films.

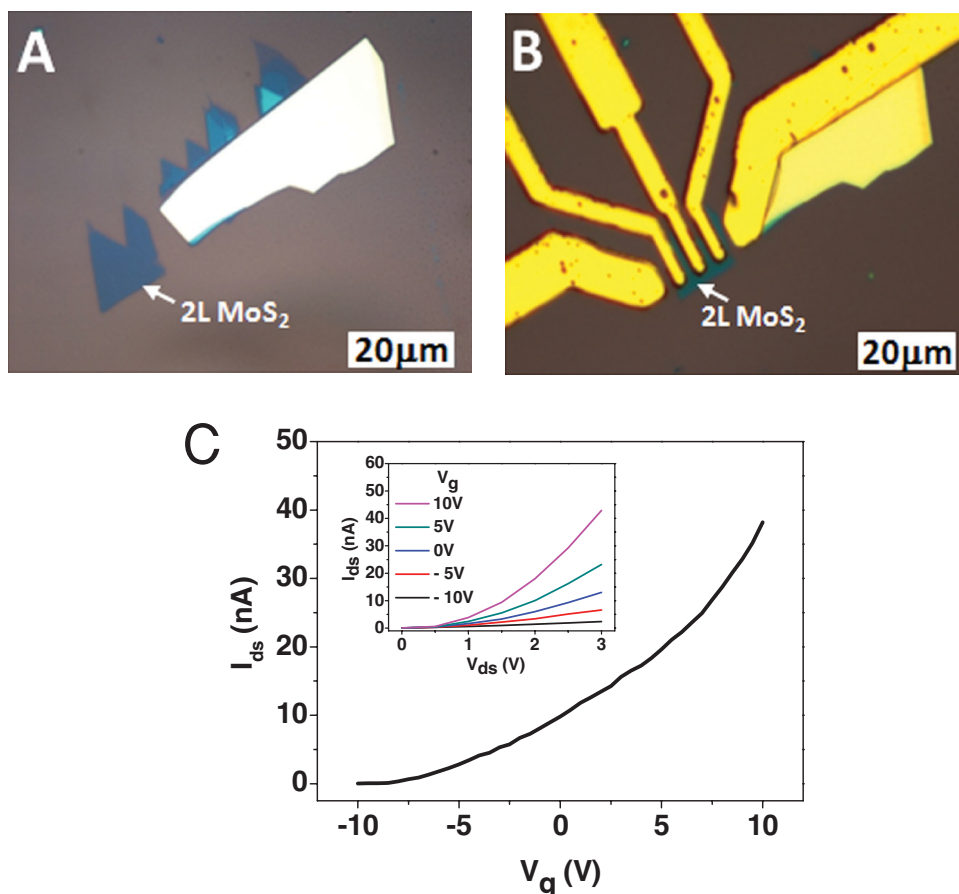


Figure 3. A) Optical microscope image of a bilayer (2L) MoS₂ film deposited onto Si/SiO₂. B) Optical microscope image of an FET device based on the 2L MoS₂ film shown in (A), where Au electrodes work as the source and drain electrodes and the substrate acts as the back gate. C) Plot of I_{ds} vs. V_g of the 2L MoS₂ FET shown in (B) at $V_{ds} = 3$ V. Inset: I_{ds} - V_{ds} curves at different V_g ranging from -10 to 10 V.

a typical adsorption process (see Figure 4A inset), the current dropped immediately after the device was exposed to NO. This lasted about 30 s (the rapid step, left solid line) and then continuously decreased for more than 2 min until the saturation of NO adsorption was reached (the slow step, left dashed line). The same two-step process was observed for

NO desorption (right solid and dashed lines in the inset in Figure 4A). A similar phenomenon was also observed in the graphene-based gas sensors, in which it was attributed to the different binding energies of gas molecules at the different defect sites on the graphene surface.^[4] However, the detailed interaction between the MoS₂ surface and the gas molecules

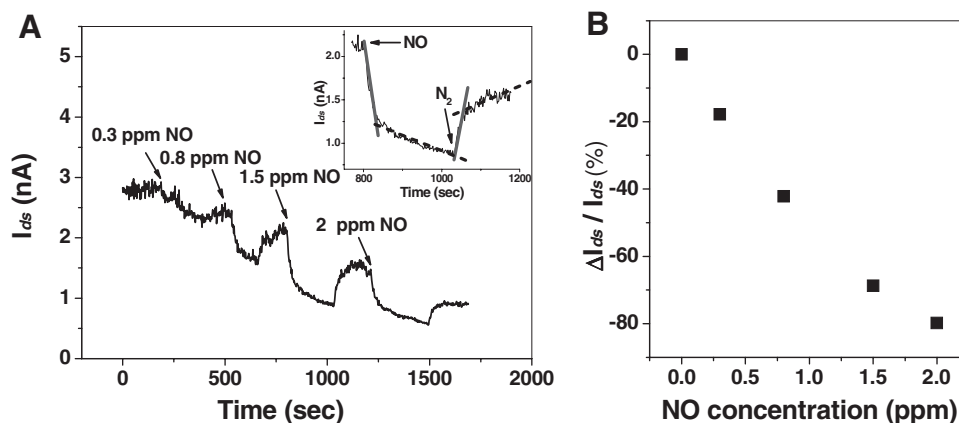


Figure 4. NO sensor based on 2L MoS₂ FET device. A) Real-time current response after exposure of the 2L MoS₂ FET to NO with increased concentration. Inset: A typical adsorption and desorption process of NO on the 2L MoS₂ FET. B) Plot of the percent change in current as a function of NO concentration.

is not clear here, which requires further investigation. The responses from 3L and 4L MoS₂ devices (see Figure S3, SI) are similar to those of the 2L MoS₂ device.

Figure 4B shows a plot of percent change in current in the 2L MoS₂ FET as a function of NO concentration (calculated from Figure 4A). Although the current response is very high (80% decreased at 2 ppm), the detection limit is compromised by the low signal-to-noise ratio (at 0.8 ppm the signal-to-noise ratio is approximately 3). Such a low signal-to-noise ratio might arise from our homemade gas-sensing system, since the sensing chamber is relatively big (>1000 L). On the contrary, the single-layer MoS₂ based device exhibited a much faster (decrease of 50% in current within 5 s in 0.3 ppm NO) but a rather unstable current response (see Figure S2, SI). The reason for this is not clear and further study is required.

In summary, single- and multilayer (1–4L) MoS₂ films were deposited onto Si/SiO₂ substrates using the scotch tape-based mechanical exfoliation technique. The layer numbers of MoS₂ films are confirmed by Raman spectroscopy and AFM. FETs based on single- and multilayer MoS₂ sheets, exhibiting the n-type semiconducting properties, have been successfully used for sensing NO gas. Although the single-layer MoS₂ FET showed a rapid and dramatic response upon exposure to NO, its current was found to be unstable. Contrary to this, the 2L, 3L, and 4L MoS₂ FET devices exhibit both stable and sensitive responses down to a detection limit of 0.8 ppm NO. Our experimental results presented here expand the potential application of 2D MoS₂ FETs in gas sensing.

Experimental Section

Mechanical Exfoliation of MoS₂: Single- and multilayer MoS₂ films were isolated from bulk MoS₂ (429MM-AB, SPI molybdenum disulfide, single crystals from Canada, SPI Supplies Inc., USA) and then deposited onto the freshly cleaned Si substrates covered by a 300 nm thick SiO₂ layer using the scotch tape-based mechanical exfoliation method, which is widely employed for preparation of single-layer graphene sheets.^[1] Optical microscope (Eclipse LV100D, Nikon) was used to locate the single- and multilayer MoS₂ films. As shown in Figure 1A–D, single and multilayer MoS₂ films showed clear optical contrast against the substrate. Bright flakes were observed when the layer number of MoS₂ exceeded 10 (Figure 1A and 1D). AFM (Dimension 3100 with Nanoscope IIIa controller, Veeco, CA, USA) was used to confirm the layer number by measuring the film thicknesses in tapping mode in air.

Raman Spectroscopy: Analysis of the single- and multilayer MoS₂ films by Raman spectroscopy was carried out on a WITec CRM200 confocal Raman microscopy system with the excitation line of 488 nm and an air-cooling charge-coupled device (CCD) as the detector (WITec Instruments Corp, Germany). The Raman band of Si at 520 cm⁻¹ was used as a reference to calibrate the spectrometer.

Fabrication, Characterization and Sensing Applications of MoS₂ FETs: The source and drain electrodes of MoS₂ FET devices were fabricated by traditional photolithography. The channel length for all FET devices was kept at ≈3 μm. The 5 nm Ti/50 nm Au, used as

source and drain electrodes, was deposited by the electron beam evaporator. After removal of the photoresist, the electrical properties of the fabricated MoS₂ FETs were tested by the Keithley 4200 semiconductor characterization system in air at room temperature.

The as-prepared MoS₂ FET devices were located in the chamber of the homemade sensing equipment. Keithley 4200 semiconductor characterization system was used to monitor the current change in real-time using the two-point measurement in a glove box. NO (2 ppm in N₂) and N₂ (99.999%) cylinders (National Oxygen Pte Ltd, Singapore) with attached flow meters were used to adjust the concentration of the NO gas.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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