1. Draw three constitutional isomers with the molecular formula C<sub>3</sub>H<sub>8</sub>O.

2. How many constitutional isomers can you draw for  $C_4H_{10}O$ .

- 3. Draw the cis and trans isomers for
  - a) 1-bromo-4-chlorcyclohexane b) 1-ethyl-3-methylcyclobutane.

- 4. Which of the following compounds can exist as cis-trans isomers? For those compounds that can exist as cis and trans isomers, draw and label the isomers. Also, draw the skeletal structures.
  - 1.  $CH_3CH = CHCH_2CH_2CH_3$
- 3.  $CH_3CH = CHCH_3$
- **2.** CH<sub>3</sub>CH<sub>2</sub>C=CHCH<sub>3</sub>
  CH<sub>2</sub>CH<sub>3</sub> **4.** CH<sub>3</sub>CH<sub>2</sub>CH=CH<sub>2</sub>
  CH<sub>2</sub>CH<sub>3</sub>

5. Draw four compounds with the formula C<sub>5</sub>H<sub>10</sub> that have a carbon-carbon double bond but do not have Cis-trans isomers.

- 6. Draw and label the E and Z form for each.
  - a. CH<sub>3</sub>CH<sub>2</sub>CH=CHCH<sub>3</sub>
  - **b.** CH<sub>3</sub>CH<sub>2</sub>C=CHCH<sub>2</sub>CH<sub>3</sub>

    CI

    c. CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>

    CH<sub>3</sub>CH<sub>2</sub>C=CCH<sub>2</sub>CI
  - - - CH<sub>3</sub>CHCH<sub>3</sub>

- 7. Use E and Z notation to name the following:

- 8. Which of the following has an asymmetric center?
  - a. CH<sub>3</sub>CH<sub>2</sub>CHCH<sub>3</sub>

d. CH<sub>3</sub>CH<sub>2</sub>OH

- Cl
- **b.** CH<sub>3</sub>CH<sub>2</sub>CHCH<sub>3</sub>
  - CH<sub>3</sub>

- e. CH<sub>3</sub>CH<sub>2</sub>CHCH<sub>2</sub>CH<sub>3</sub>
  - Вr

- $CH_3$
- c. CH<sub>3</sub>CH<sub>2</sub>CCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>

**f.**  $CH_2 = CHCHCH_3$  $\dot{N}H_2$ 

9. Draw the enantiomers using a) perspective diagrams and b) Fischer diagrams for:

1. CH<sub>3</sub>CHCH<sub>2</sub>OH

2. CICH<sub>2</sub>CH<sub>2</sub>CHCH<sub>2</sub>CH<sub>3</sub>

10. Assign priorities to the following sets.

—Н —СН<sub>3</sub> a.  $-CH_2OH$ —CH<sub>2</sub>CH<sub>2</sub>OH  $-CH_3$ **b.** —CH<sub>2</sub>Br —OH **c.** —CH(CH<sub>3</sub>)<sub>2</sub> —CH<sub>2</sub>CH<sub>2</sub>Br -CH<sub>2</sub>OH

-CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>Br

**d.**  $-CH=CH_2$   $-CH_2CH_3$  $-CH_3$ 

11. Does the following have R or S configuration?

12. Draw an enantiomer and a diastereomer for the following compound.

13. Draw a diastereomer for each of the following.

c. 
$$C=C$$
 $CH_3$ 
 $CH_3$ 

14. Draw the meso compound for 3,4-dimethylhexane.

15. Draw the perspective diagram for:

- a) (S)-3-chloro-1-pentanol
- b) (2R,3R) 2,3-dibromopentane

16. Name the following: