

Chapter Thermodynamic

Key

- 1.B
- 2.B
- 3.E
- 4.C
- 5.D
- 6.B
- 7.B
- 8.A
- 9.C
- 10.C
- 11.A
- 12.E
- 13.B
- 14.D
- 15.A
- 16.E
- 17.D
- 18.E
- 19.B
- 20.E
- 21.D
- 22.E
- 23.E
- 24.C
- 25.B
- 26.B

- 27.A
28.D
29.C
30.D
31.A
32.A
33.B
34.C
35.E
36.C
37.A
38.B
39.E
40.D
41.A
42.C
43.D
44.B
45.E
46.A
47. No
48. 3.11×10^{-4}
49. -1.02 kJ/mol
50. 1.17
51. No
52. 1.58×10^{-6}
53. -46.66 kJ/mol
54. 1.11×10^3
55. 710°C
56. $71.9 \text{ J/K}\cdot\text{mol}$
57. $K_p = 7.1 \times 10^{-9}$
58. Positive
59. Positive

- 60. Negative
- 61. 639 kJ/mol
- 62. No
- 63. 2,100 K
- 64. 110. J/K·mol
- 65. -349 kJ/mol
- 66. 2160 K
- 67. 3.4×10^{-3}
- 68. 29.4 kJ/mol
- 69. right to left
- 70. -35.2 kJ/mol
- 71. left to right
- 72. -29.0 kJ/mol
- 73. 1.21×10^5
- 74. 131 J/K·mol
- 75. -123 kJ/mol
- 76. No
- 77. increases
- 78. increases
- 79. increases
- 80. decreases
- 81. $\text{O}_2(\text{g})$ at 0.5 atm
- 82. $\text{Br}_2(\text{g})$
- 83. 1 mole of $\text{N}_2(\text{g})$ in a 22.4 L container
- 84. $\text{CO}_2(\text{g})$
- 85. -110 J/K·mol
- 86. 115°C
- 87. -9.48 kJ/mol
- 88. FALSE
- 89. FALSE
- 90. FALSE