Experimental Properties of Test Compounds

Molecule ID	Structure	mp (°C) ^a	% Stability aq. buffer pH 7.4 b	Intrinsic solubility (mol/L) ^c	logP ^d	$log D_{7.4}{}^e$	PAMPA			
							Pe (cm/s) ^g	% retention ^h	$log_{Papp}{}^{i}$	pK_d^j
SM25	NH N	107.8– 108.5	100.0	9.97E–4	2.67 ±0.01	-0.09 ±0.01	1.64E- 6	-3.05E-2	-5.79 ±0.12	4.49 ±0.04
SM26	NH NH	76.2– 78.4	91.4	8.65E-2	1.04 ±0.01	-0.87 ±0.06	1.00E- 06	-5.00E-2	-6.00 ±0.12	4.91 ±0.01
SM27	O O O	ND§	100.0	8.71E–2	‡	1.56 ±0.11 [†]	6.79E– 6	-3.20E-2	-5.17 ±0.10	10.45 ±0.01
SM28	THE STATE OF THE S	135.3– 136.8	94.5	1.62E-3	‡	1.18 ±0.08 [†]	2.11E- 6	-1.20E-2	-5.68 ±0.11	>12
SM29	N. S.	71.4– 71.9	100.0	2.65E-2	1.61 ±0.03	1.61 ±0.11	8.44E- 6	0.157	-5.1 ±0.10	10.05 ±0.01
SM30	O O O O	105.8– 107.2	94.1	6.38E-4	‡	2.76 ±0.19 [†]	7.06E- 6	0.260	-5.2 ±0.10	10.29 ±0.12
SM31	M S N	ND	93.7	3.47E-2	‡	1.96 ±0.14 [†]	1.02E- 5	5.89E-2	-4.99 ±0.10	11.02 ±0.01
SM32	N.S.	115.2- 116.0	91.5	1.43E-3	‡	2.44 ±0.17 [†]	1.46E- 05	0.439	-4.8 ±0.10	10.45 ±0.02
SM33	S O O	74.2– 75.8	92.8	1.10E-3	‡	2.96 ±0.21 [†]	ND	ND	ND	>12
SM34	NH S N	56.6– 58.9	93.5	2.20E-3	‡	2.83 ±0.20 [†]	1.14E- 5	0.373	-4.9 ±0.10	11.93 ±0.05
SM35	**************************************	151.2– 154.2	100.0	2.20E-2	0.88 ±0.02	0.87 ±0.06	ND	ND	ND	9.87 ±0.01
SM36	\$,0- N,S H	135.3– 136.8	90.3	4.268E–2	‡	0.76 ±0.05 [†]	2.91E- 6	5.07E-2	-5.54 ±0.11	9.80 ±0.06
SM37	\$ 0.0 N S N	127.1– 127.8	93.4	1.06E-2	‡	1.45 ±0.10 [†]	3.41E- 6	7.13E-2	-5.47 ±0.11	10.33 ±0.02

SM38	O O O O O O O O O O O O O O O O O O O	108.3– 108.7	100.0	4.32E-2	++	1.03 ±0.07 [†]	3.66E- 06	-3.00E-3	-5.4 ±0.10	9.44 ±0.02
SM39	0, 0 S 0, 0 N H	182.5– 184.2	98.3	1.06E-3	‡	1.89 ±0.13 [†]	ND	ND	ND	10.22 ±0.15
SM40	0,0 0,0 0,0 N, N, N	100.2- 101.5	100.0	8.16E-3	1.83 ±0.05	1.82 ±0.13	4.43E– 6	0.105	-5.35 ±0.11	9.58 ±0.01
SM41	0-N S NH	158.0– 158.8	97.7	3.85E-2	0.58 ±0.02	-0.42 ±0.03	1.16E– 6	3.42E-2	-5.94 ±0.12	5.22 ±0.01
SM42	0-N S NH	164.8– 166.0	95.5	2.17E-3	1.76 ±0.03	0.99 ±0.07	5.79E- 6	-4.81E-2	-5.24 ±0.10	6.62 ±0.02
SM43	0-N S-N NH	152.5– 152.9	97.3	2.36E-2	0.85 ±0.01	0.42 ±0.03	2.87E- 6	0.131	-5.54 ±0.11	5.62 ±0.02
SM44	N=N S NH	156.0– 157.5	93.5	1.05E-2	1.16 ±0.03	0.06 ±0.00	2.42E- 6	-4.07E-2	-5.62 ±0.11	6.34 ±0.01
SM45	N=N 0,0 N=N S	163.7– 164.5	93.3	3.62E-4	2.55 ±0.04	1.06 ±0.07	4.09E– 6	2.76E-4	-5.39 ±0.11	5.93 ±0.05
SM46	N=N NH NH	158.5– 159.7	93.9	2.75E-3	1.72 ±0.01	0.69 ±0.05	3.77E- 6	-4.43E-2	-5.42 ±0.11	6.42 ±0.01

^a Melting point of crystalline material; ^b Test compound (%) remaining after 5 h of incubation at rt in aqueous buffer (pH 7.4) as determined by LC/MS analyses; ^c Intrinsic solubility determined from the general solubility equation (GSE) by using experimentally determined logP and mp values; ^d Log of the partition coefficient between *n*-octanol and water (unless otherwise noted, logP determinations were conducted via potentiometric titrations using a Sirius T3, Pion); ^c Log of the distribution coefficient between *n*-octanol and aqueous buffer at pH 7.4 (unless otherwise noted, logD_{7.4} determinations were conducted via potentiometric titrations using a Sirius T3, Pion); ^f Calculated values using ChemAxon³⁷; ^g Effective permeability (PAMPA assay run by Analiza); ^h Membrane retention; ⁱ Log of the apparent permeability coefficient (P_{app}); ^j P_{a} values determined by potentiometric titrations using a Sirius T3, Pion (values in brackets are from ³⁰); [§] Test compound was an oil; [†] logD_{7.4} value determined via shake-flask assay (experiment run by Analiza); [‡] logP value is considered equal to the logD_{7.4} as these compounds exhibit P_{a} values >9.4; ND = not determined.