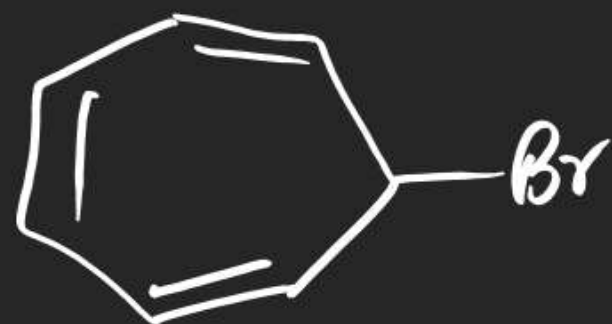
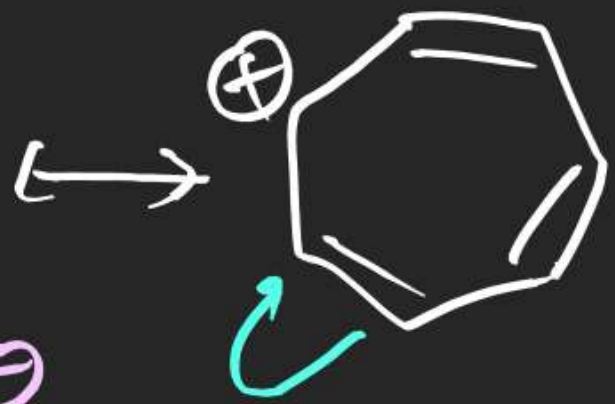
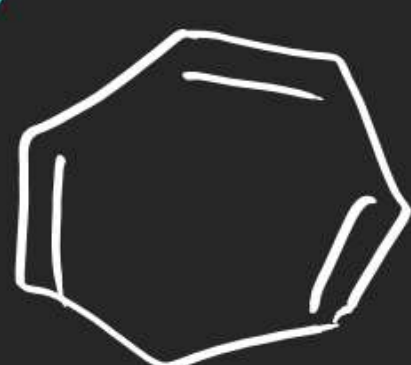
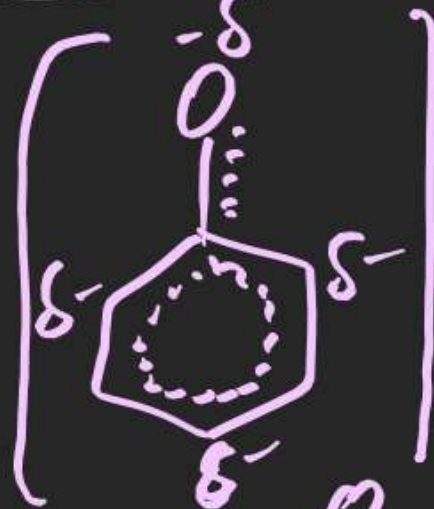


Ex-19

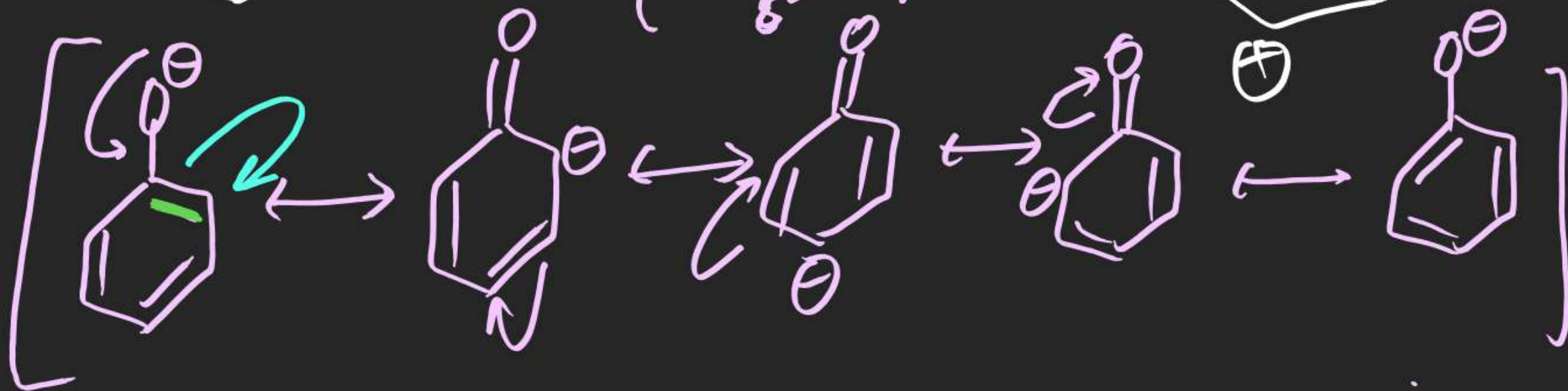


Ex-20:



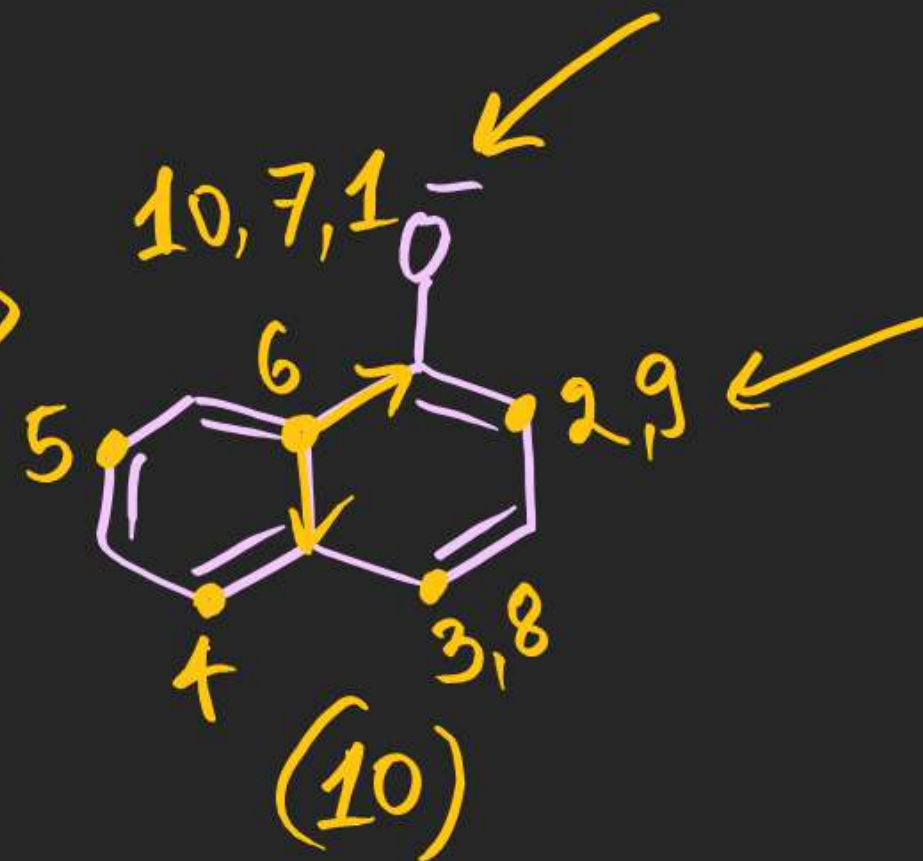
(5)

Soln

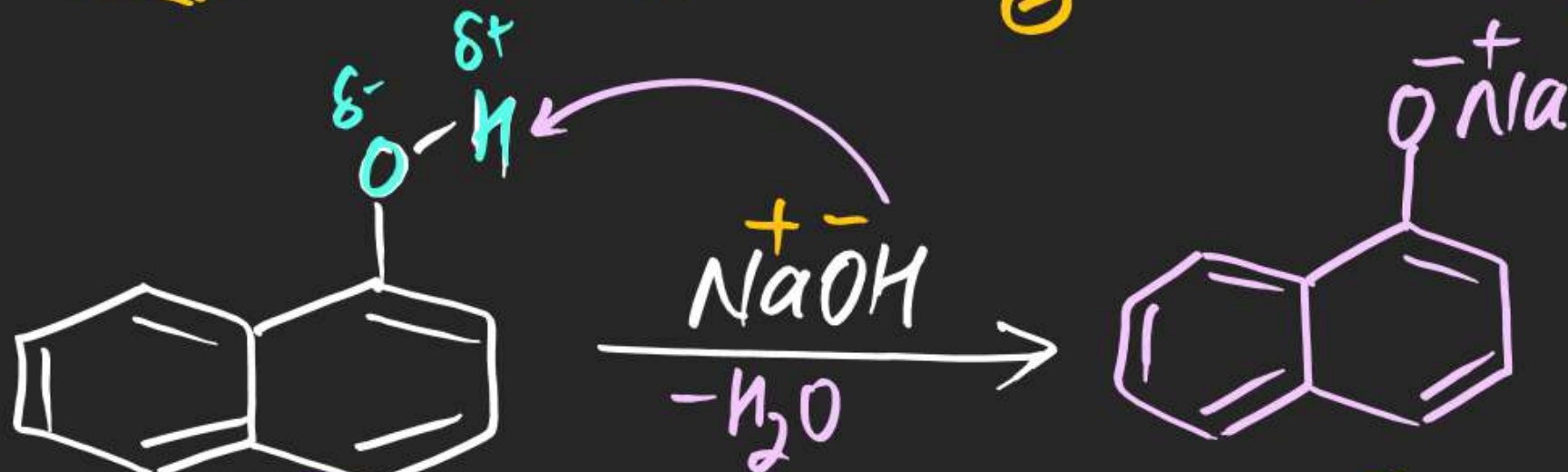




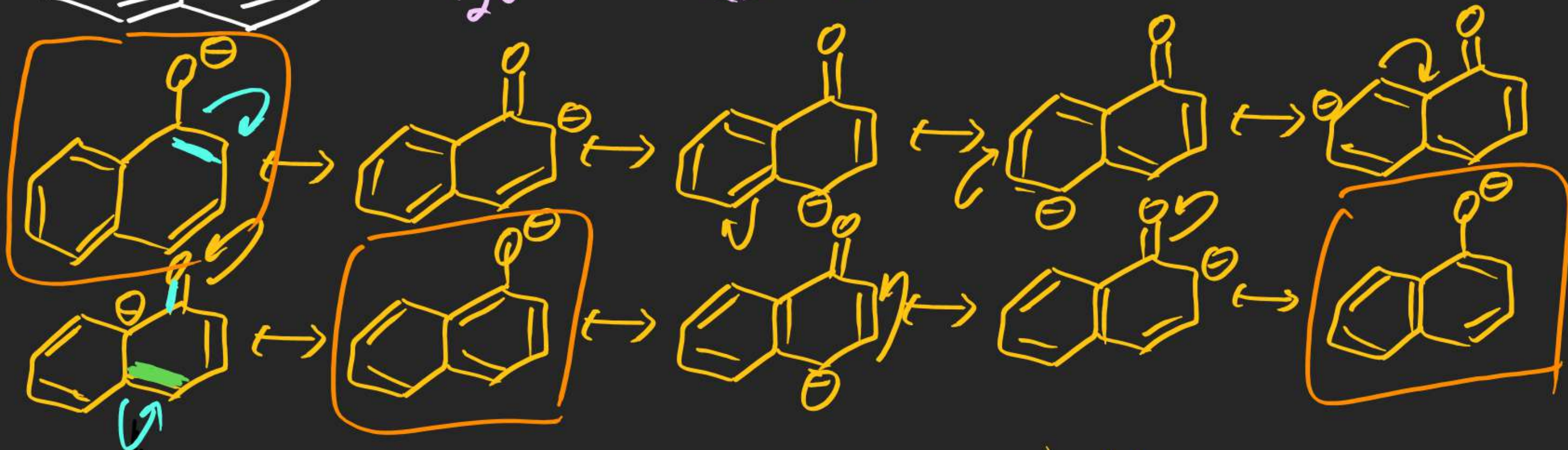
Soln!

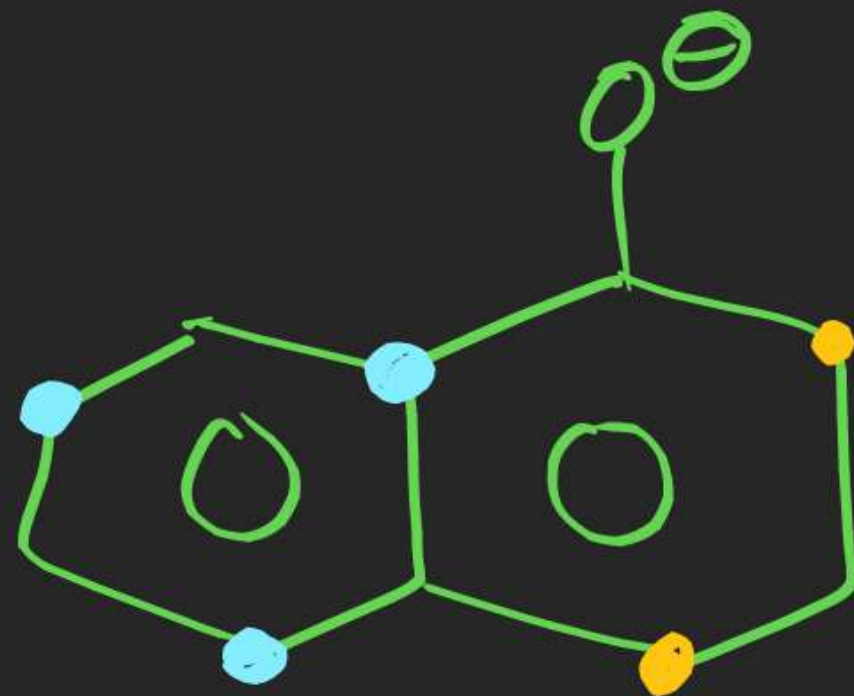
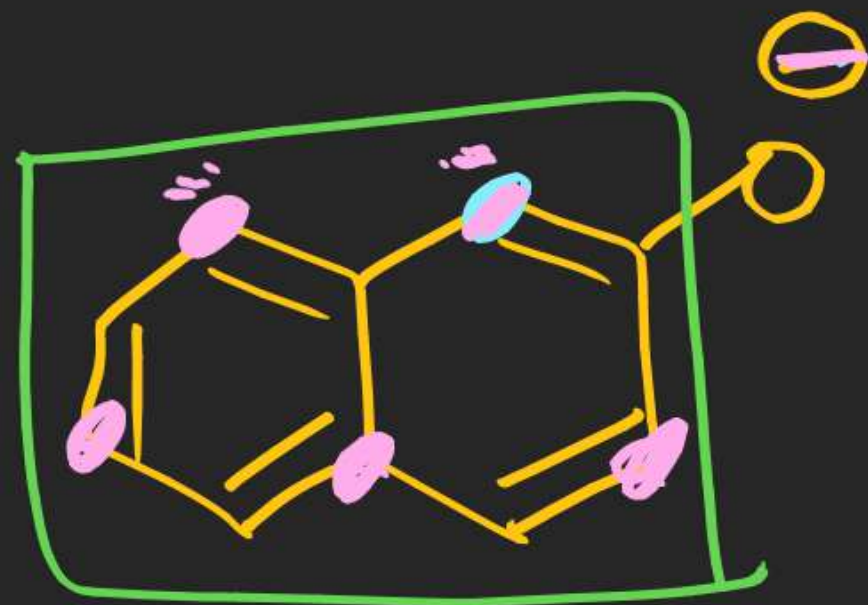


(22)



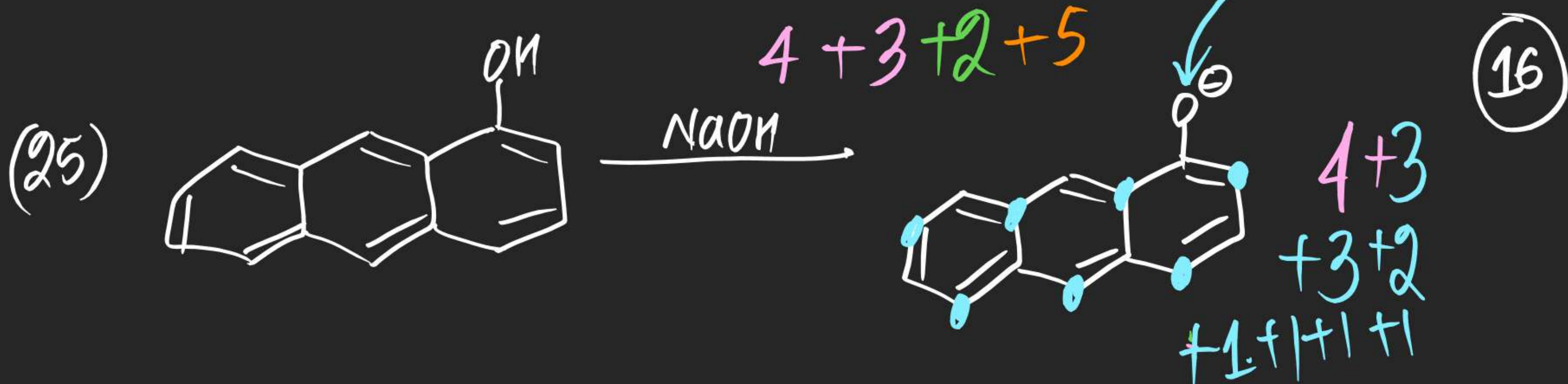
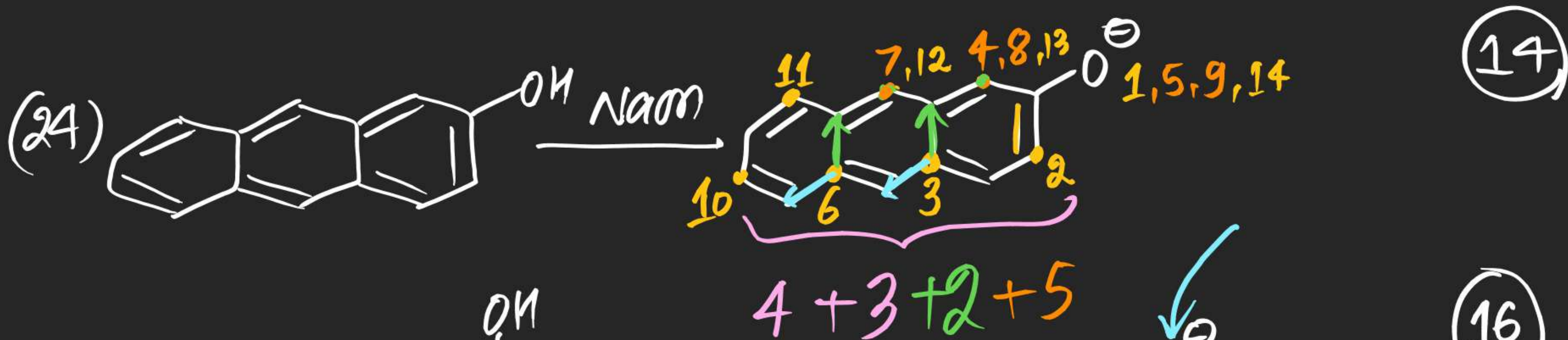
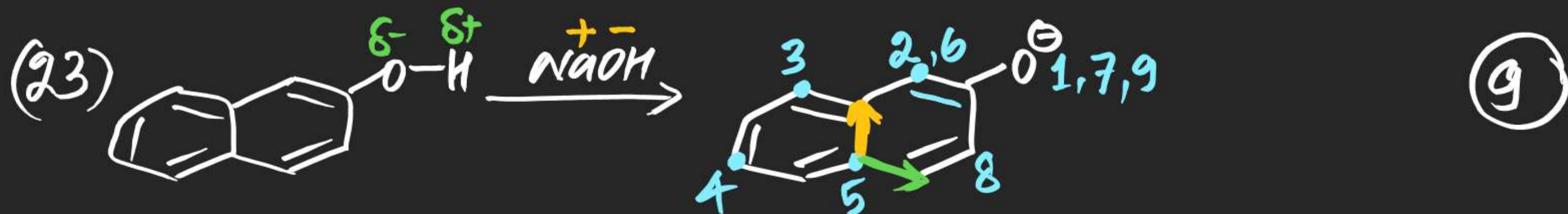
Soln!

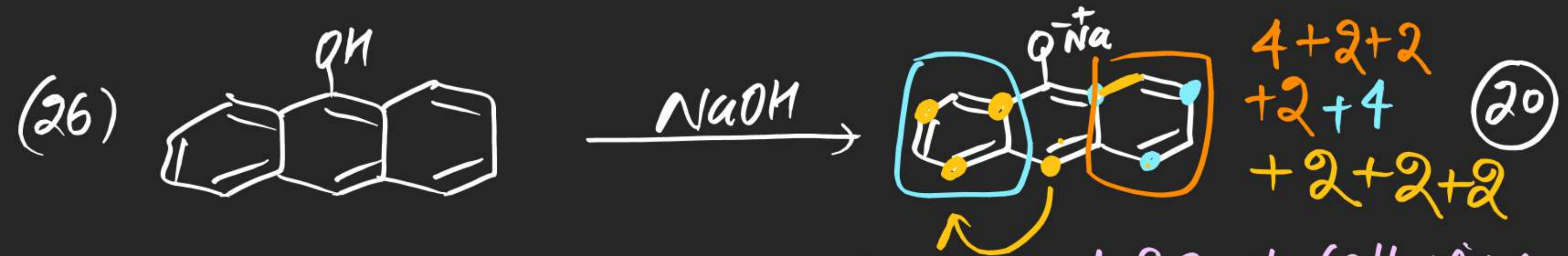




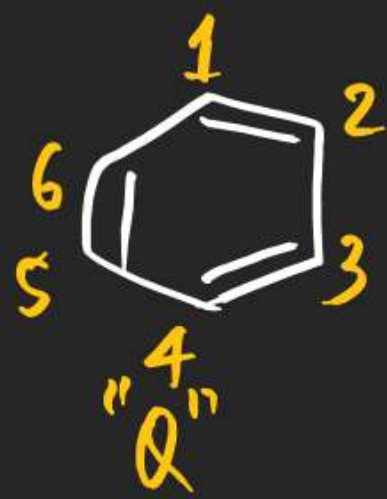
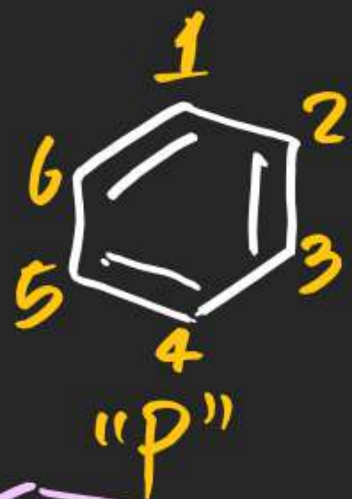
$$\underbrace{3}_{\text{wavy}} + \underbrace{2}_{\text{wavy}} + 4 = 9$$

$$\begin{array}{r} 3 + 2 + 2 \\ + 3 \\ \hline 10 \end{array}$$

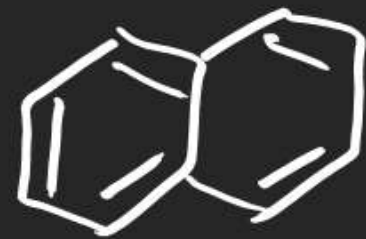




(#) Find total no. of Neutral Benzenoid R.S of following.

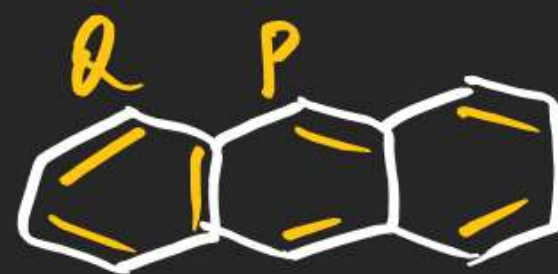
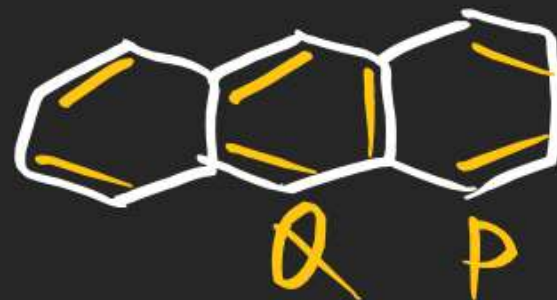
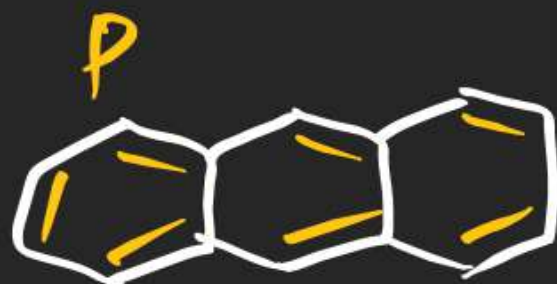
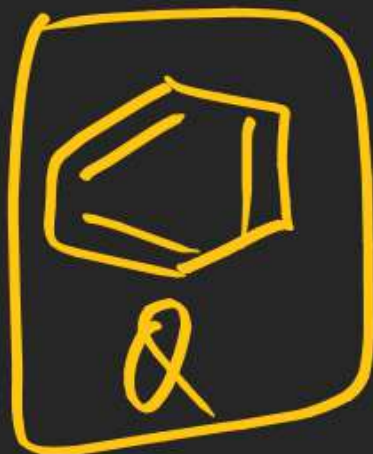


(2)



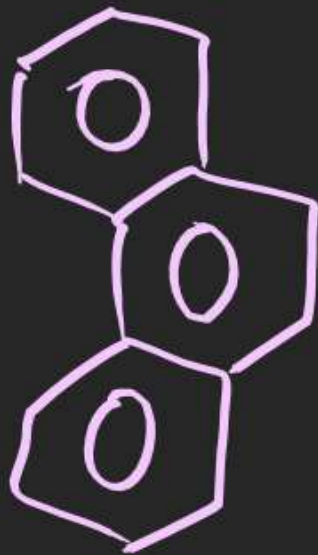
(3)

(29) Anthracene



④

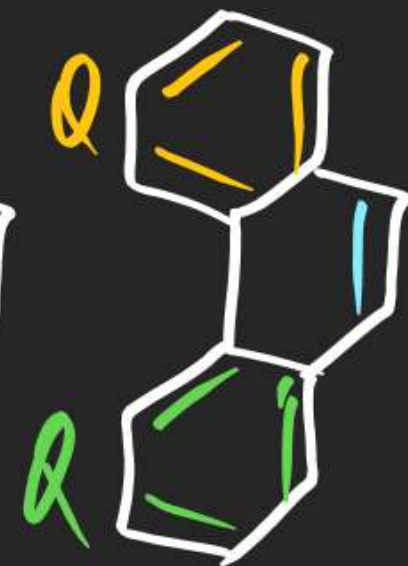
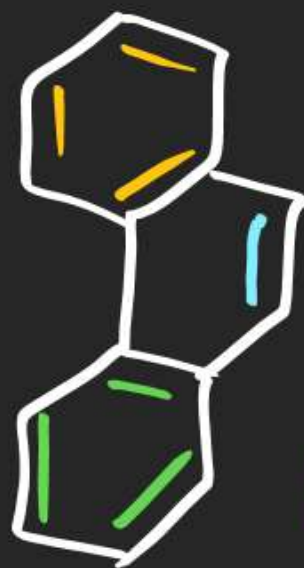
(30) Phenanthrene



P

Q

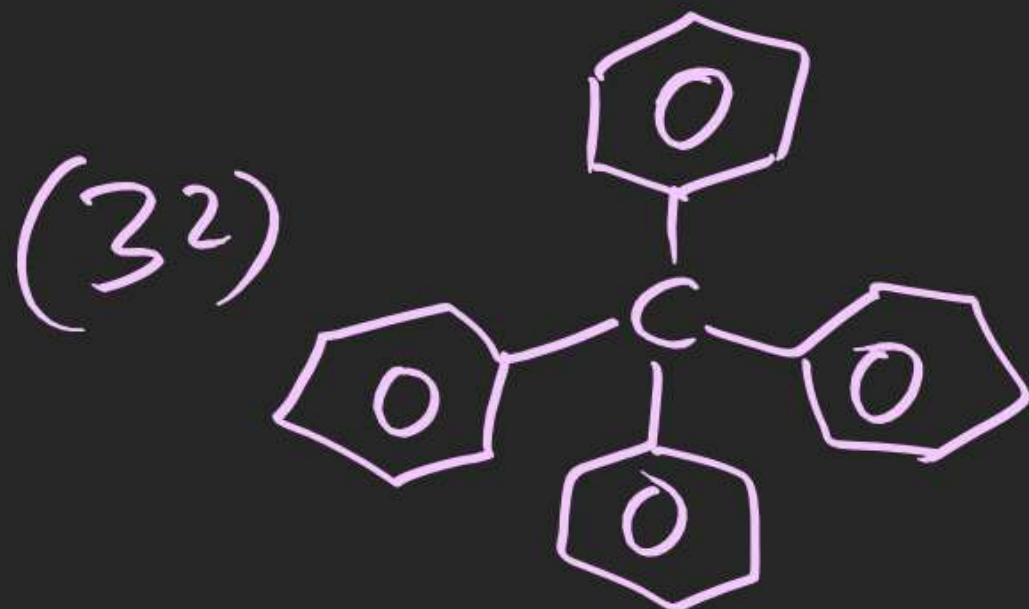
P



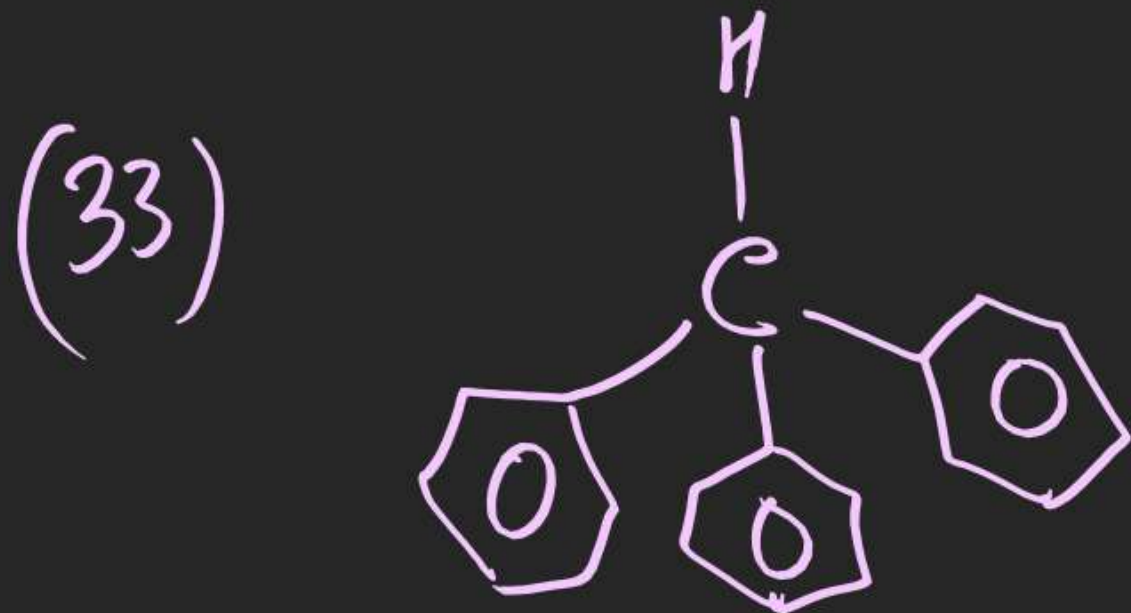
⑤



$$\begin{matrix} P & P \\ P & Q \\ Q & P \\ Q & Q \end{matrix} \quad 2^2 = 4$$



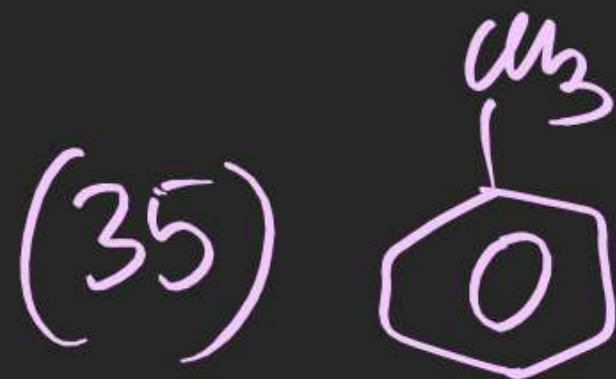
$$2^4 = 16$$



$$2^3 = 8$$

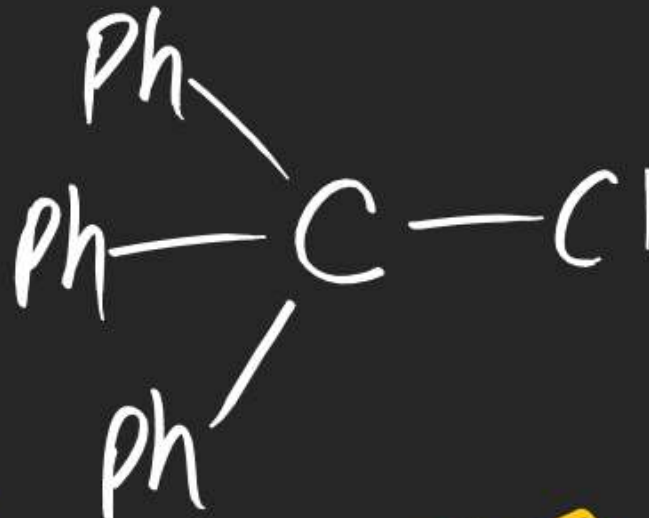


$$2^2 = 4$$

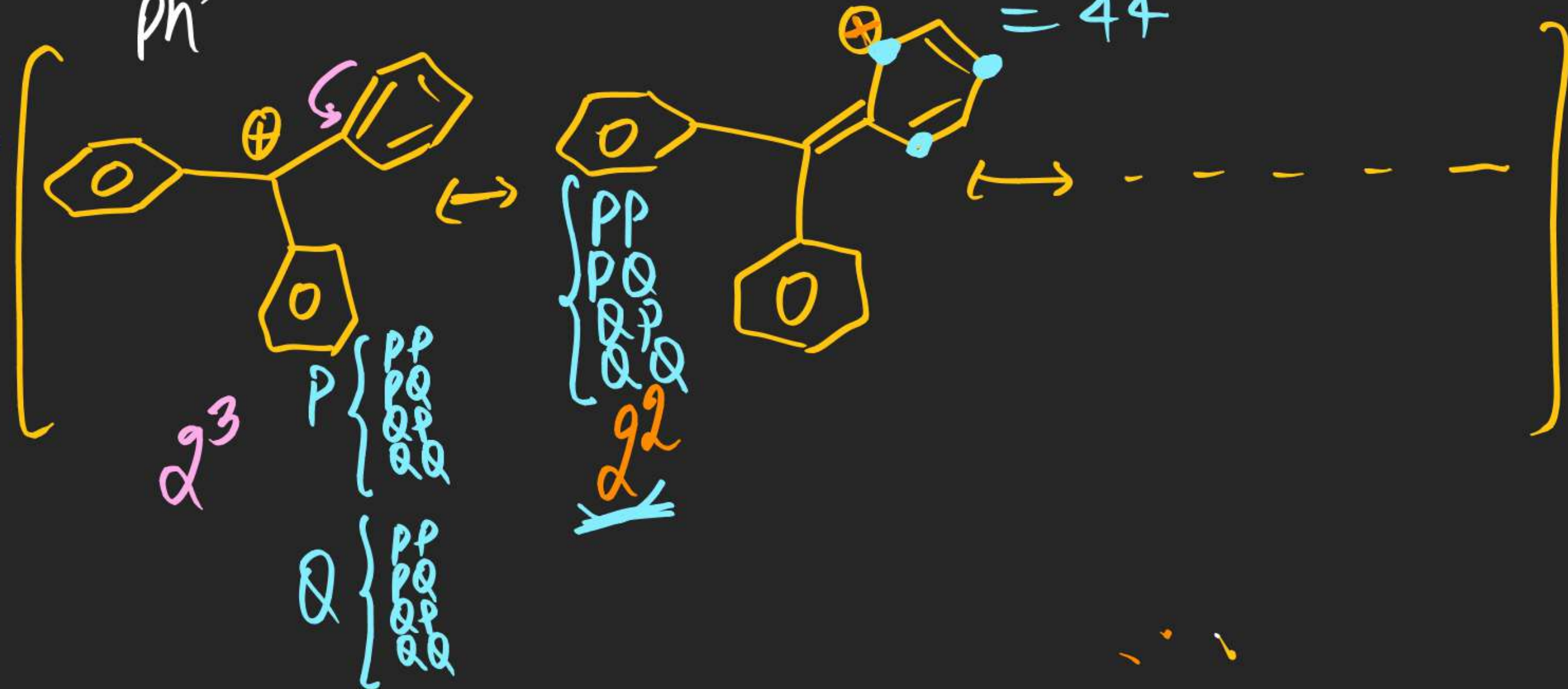


$$2^1 = 2$$

(36) Total no. of RS of Product of following Reaction

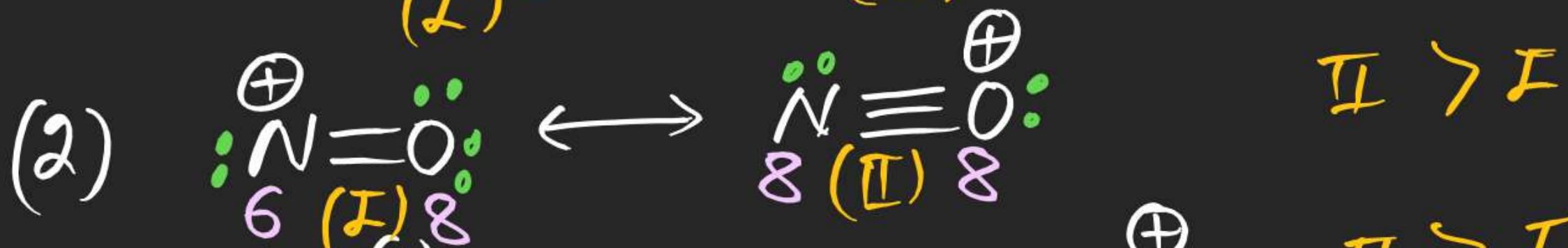
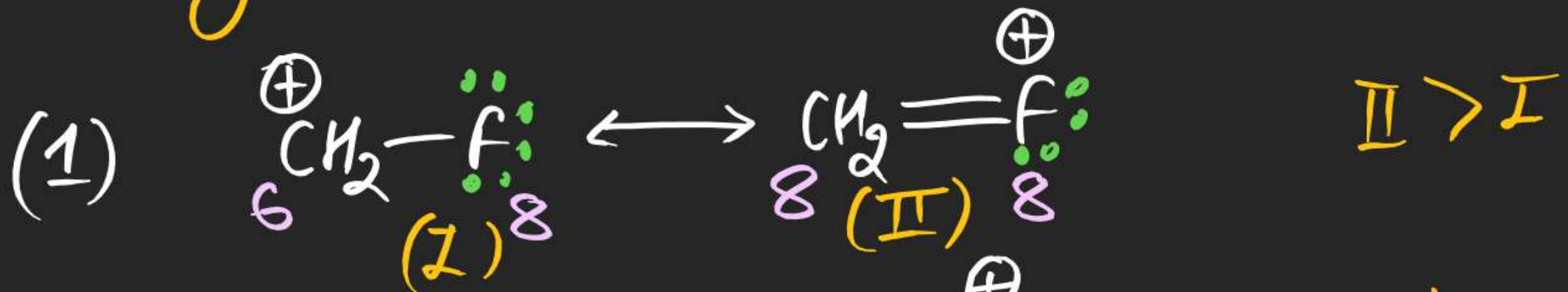


$$[2^3 + \underbrace{3 \times 2^2}_{+inph} + \underbrace{3 \times 2^2}_{+inph} + \underbrace{3 \times 2^2}_{+inph}]$$
$$= 8 + 12 + 12 + 12$$
$$= 44$$

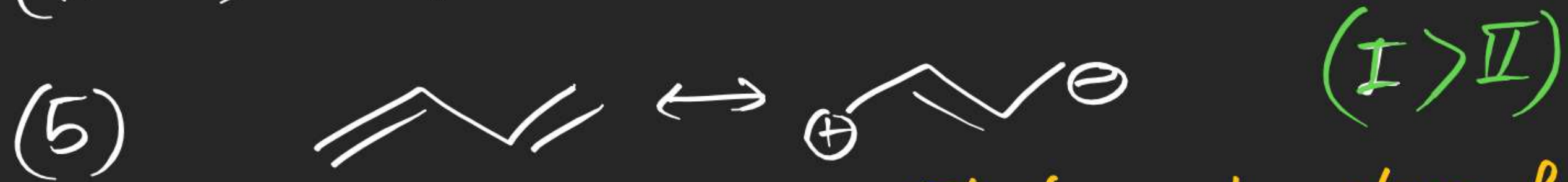


(#) Rules for Stability of RS :

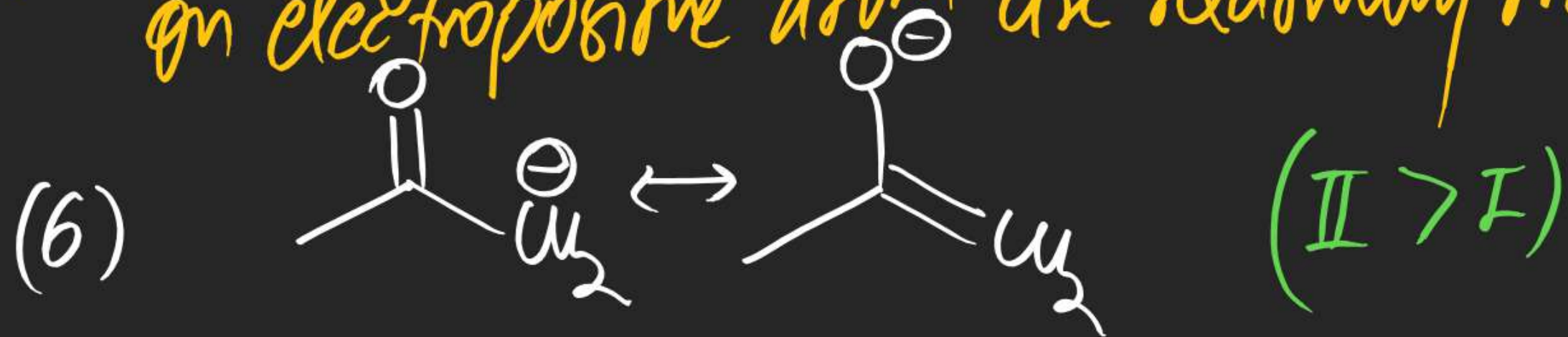
(1) RS having complete octet is more stable than having incomplete octet.



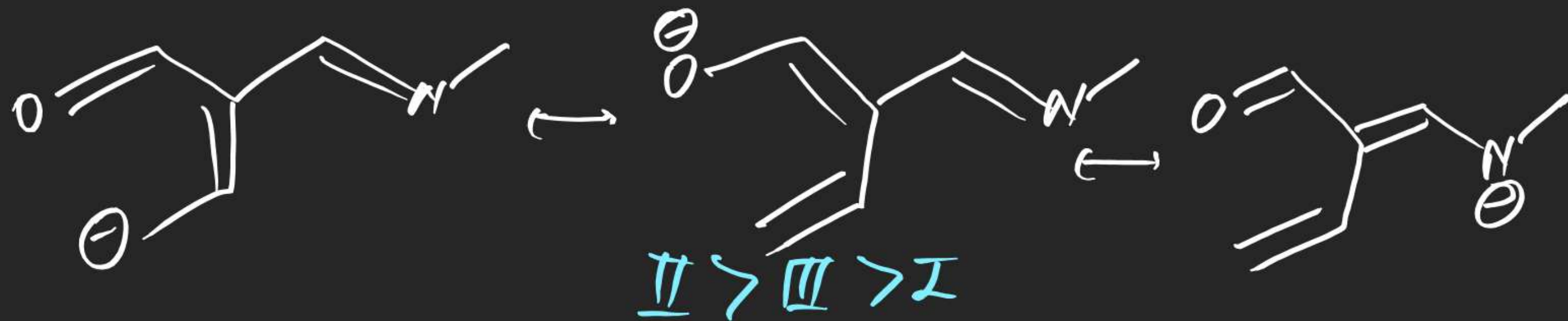
(2) RS having higher no. of Covalent Bond or less charge is more stable.



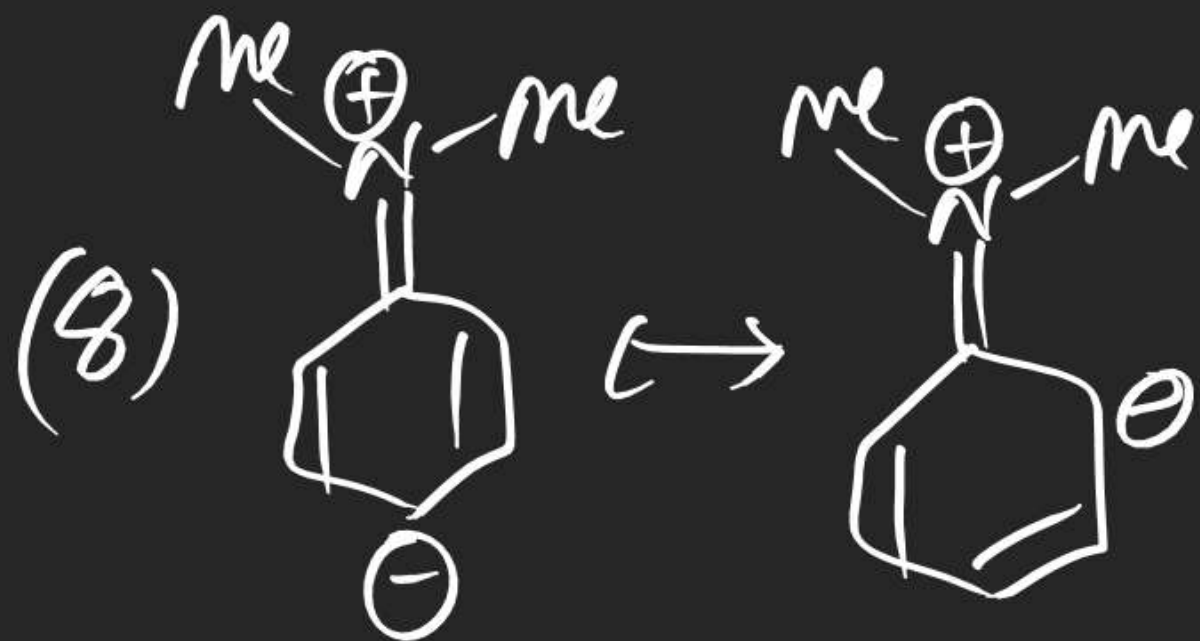
(3) RS having (-)ve charge on Electronegative atom & (+)ve charge on electropositive atom are relatively more stable.



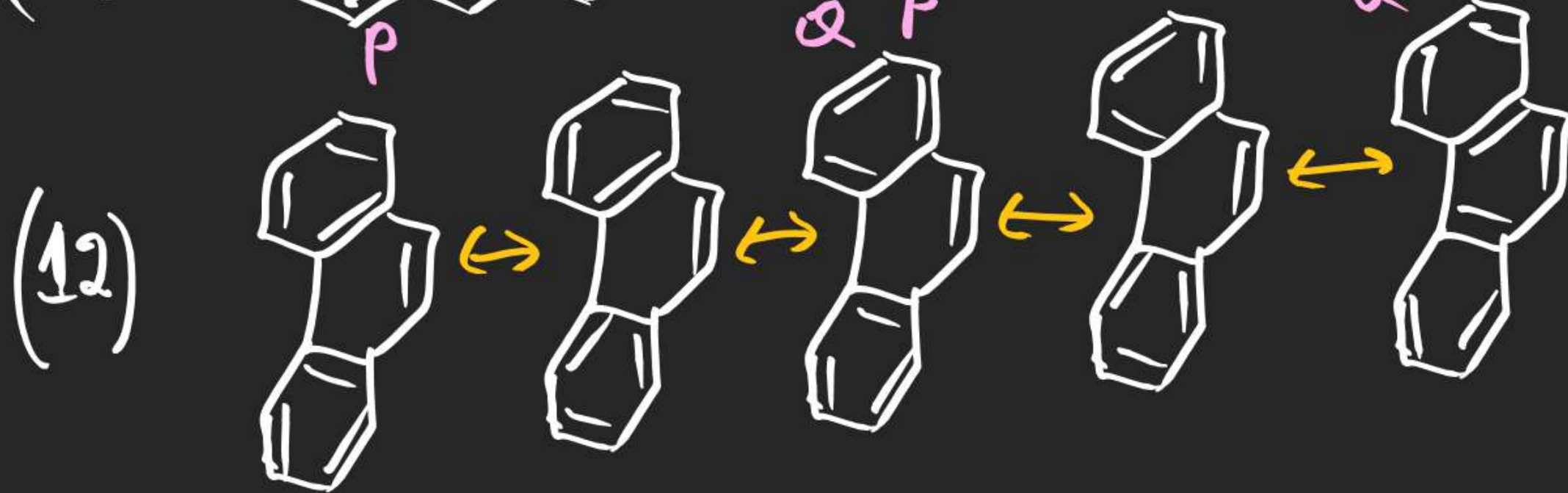
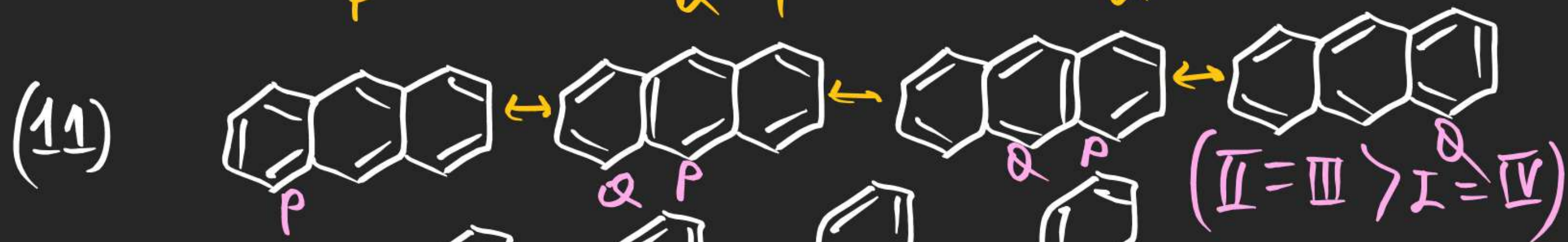
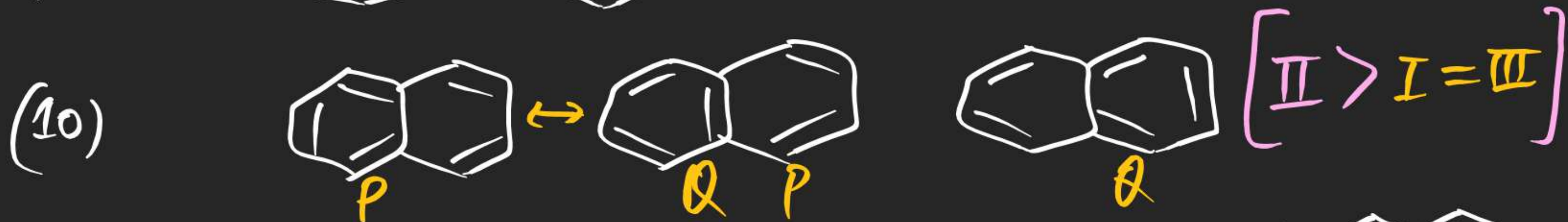
(7)



(4) RS having opp. charges closer & like charges away are more stable.



⑤ RS having higher no. of Benzenoid Segment is more stable.

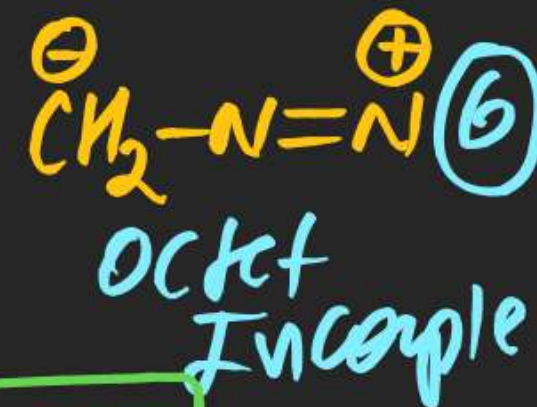
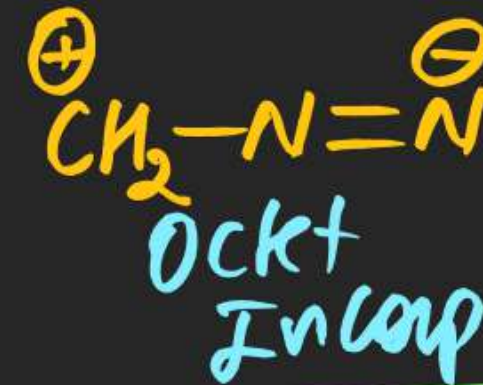
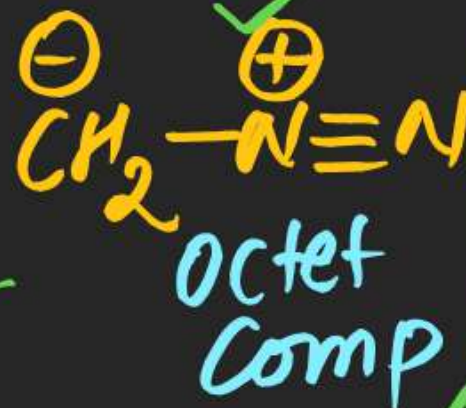
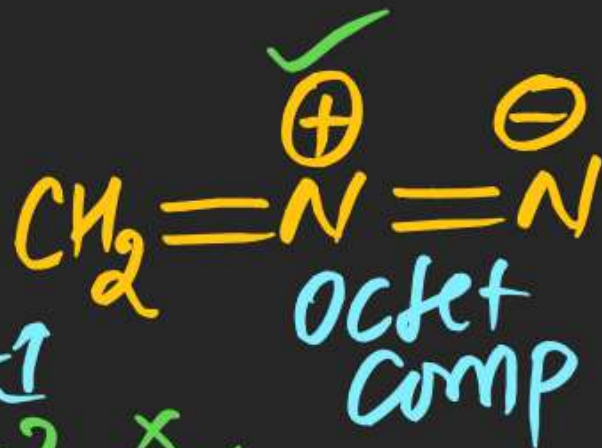


IIT Ad
(13)

Rule-1

Rule-2 x

Rule-3 ✓



Rule-2 x
Rule-3 ✓

$\text{I} > \text{II} > \text{III} > \text{IV}$

(14)



(15)



(16)



(17)



Resonance / Mesomeric Effect:-