

Synopsis of the  
research conducted  
thus far



**Theoretical  
model**

**Chiller setup**

**Temperature  
acquisition**

# First trial: code everything from scratch

*Antoine equations, Raoult's law*

$$x_1 + x_2 = 1$$

$$P_{\text{bub}} = x_1 P_1^{\text{sat}} + x_2 P_2^{\text{sat}}$$

$$P_{\text{dew}} = \left( \frac{y_1}{P_1^{\text{sat}}} + \frac{y_2}{P_2^{\text{sat}}} \right)^{-1}$$

$$P_i^{\text{sat}} = 10^{\left( A_i - \frac{B_i}{T + C_i} \right)}$$

Charts for  $A_i$ ,  $B_i$ ,  $C_i$

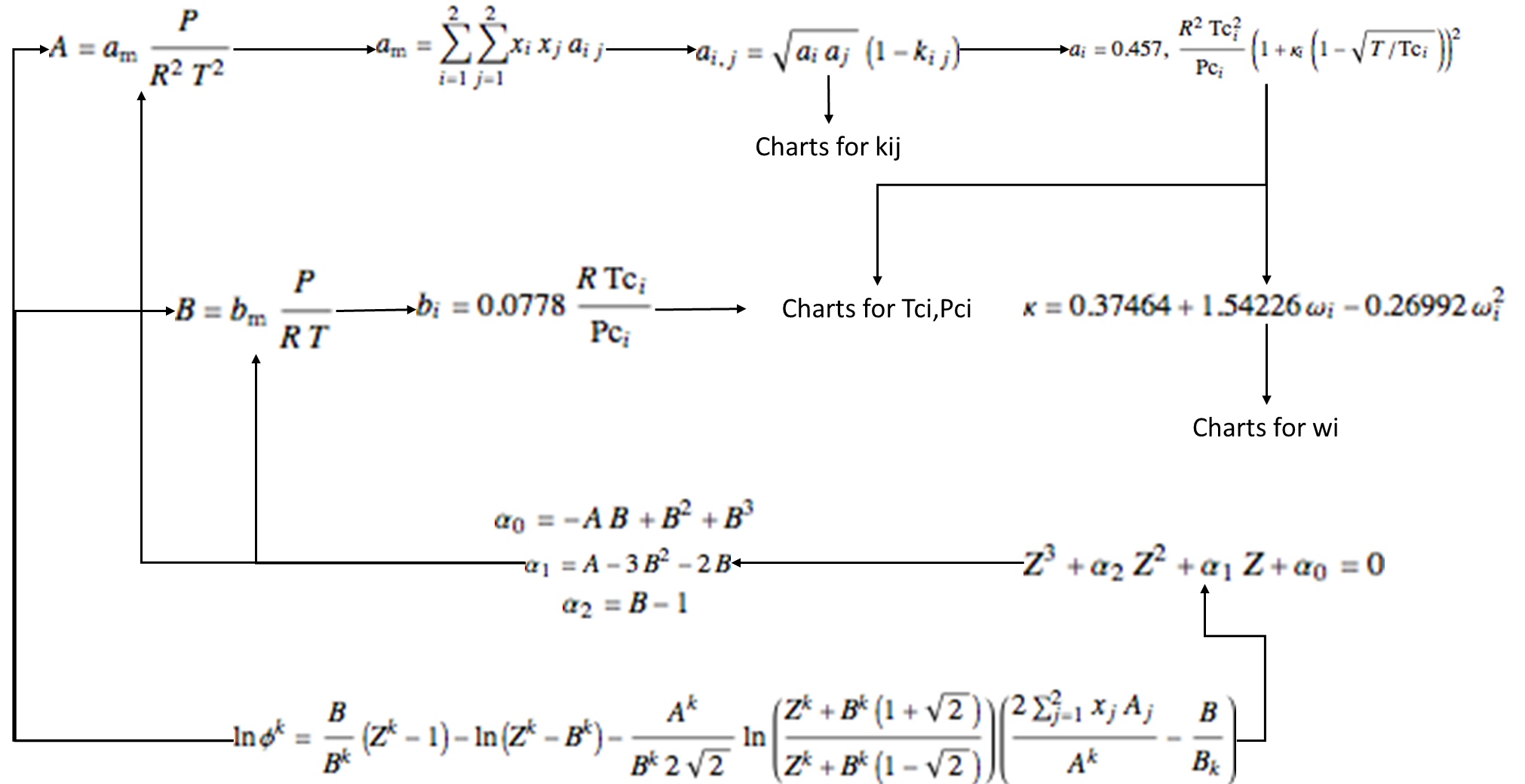
$$K_i = \frac{y_i}{x_i}$$

$$K_i = \frac{\phi_i^{\text{L}}}{\phi_i^{\text{V}}}$$

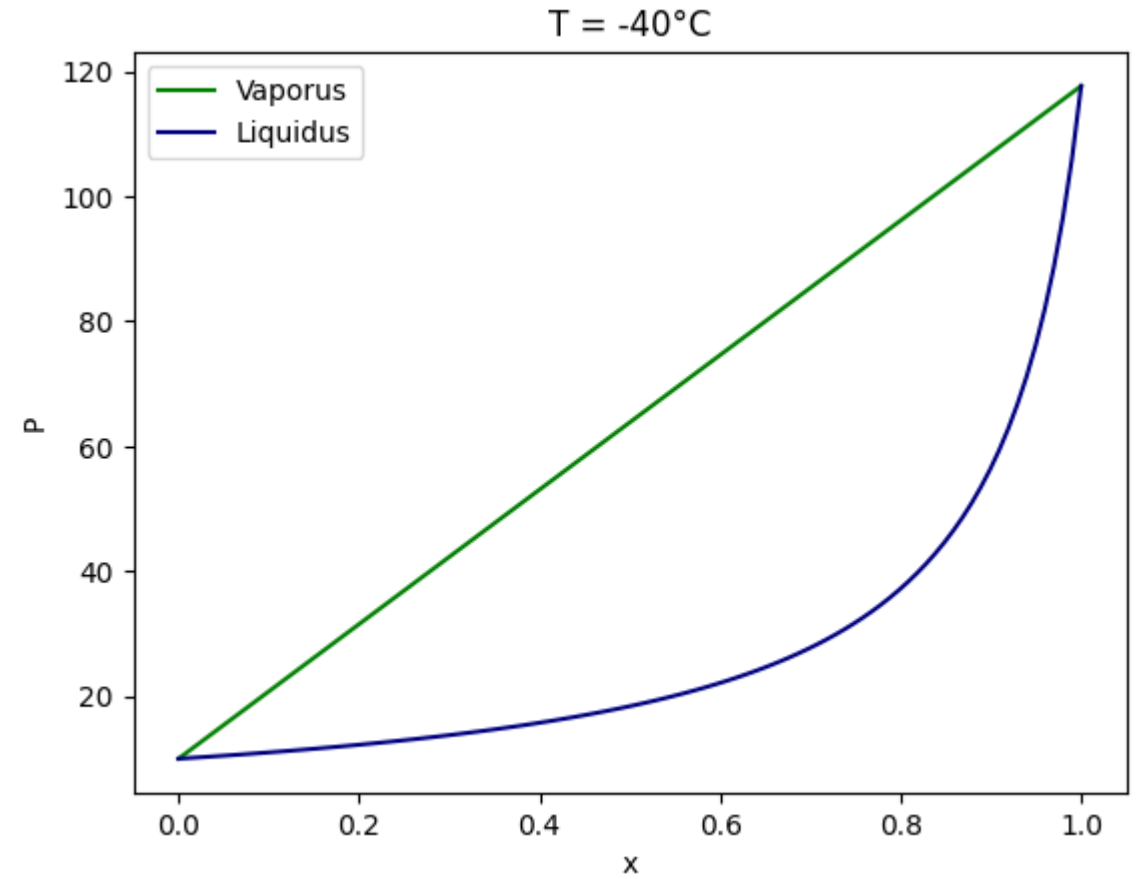
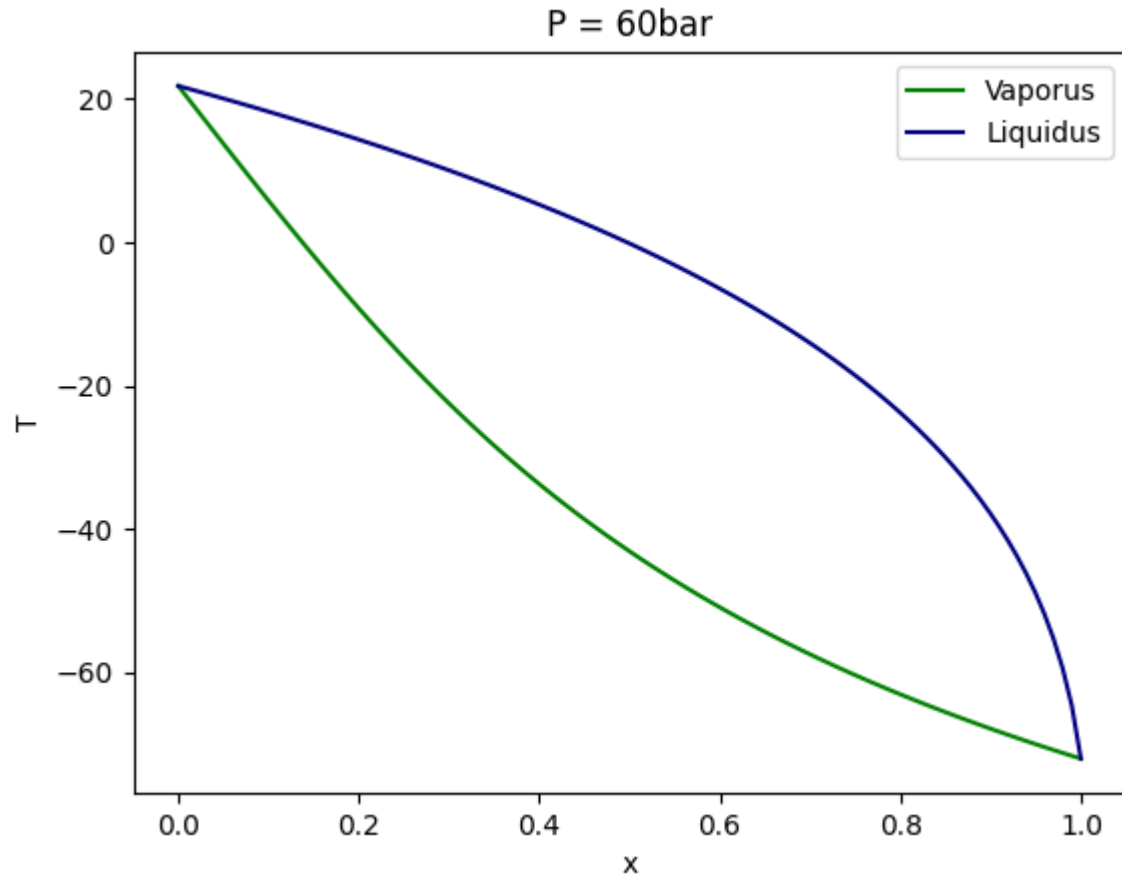
$$\ln \phi^k = \frac{B}{B^k} (Z^k - 1) - \ln(Z^k - B^k) - \frac{A^k}{B^k 2 \sqrt{2}} \ln \left( \frac{Z^k + B^k (1 + \sqrt{2})}{Z^k + B^k (1 - \sqrt{2})} \right) \left( \frac{2 \sum_{j=1}^2 x_j A_j}{A^k} - \frac{B}{B^k} \right)$$

# First trial: code everything from scratch

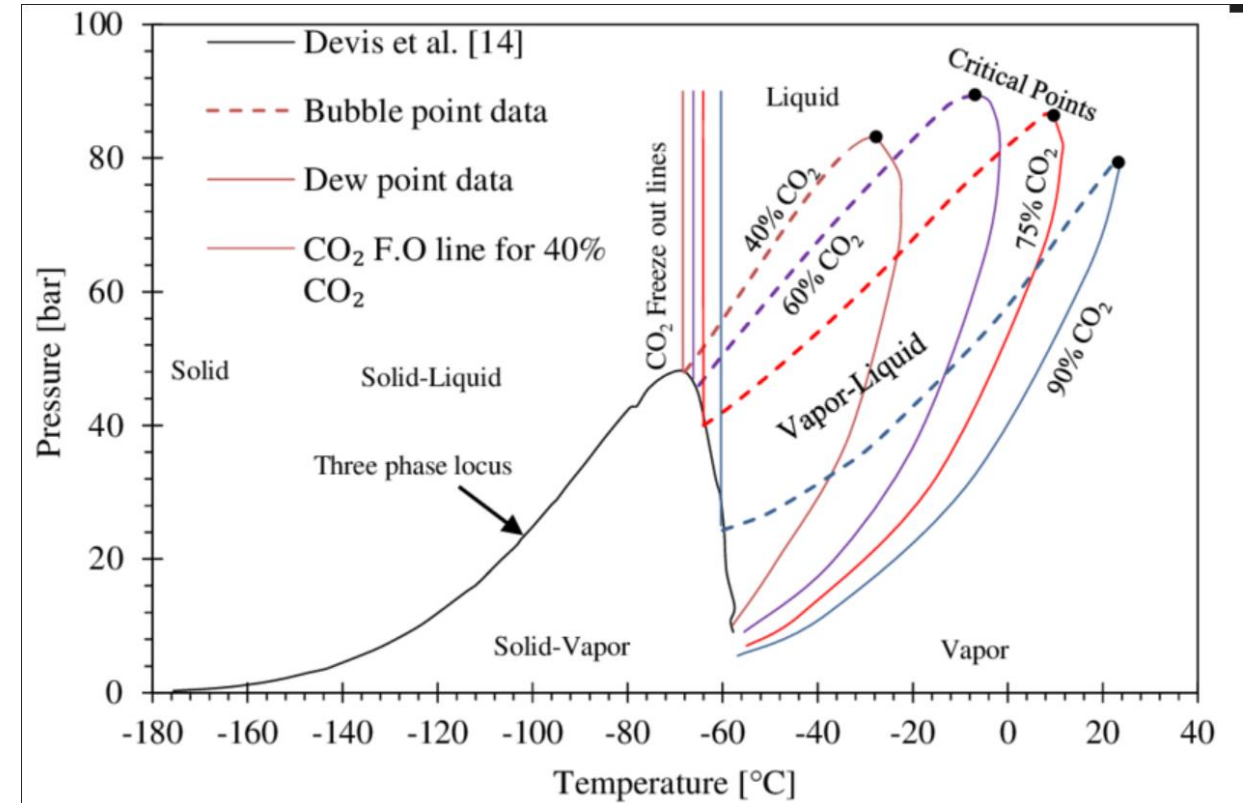
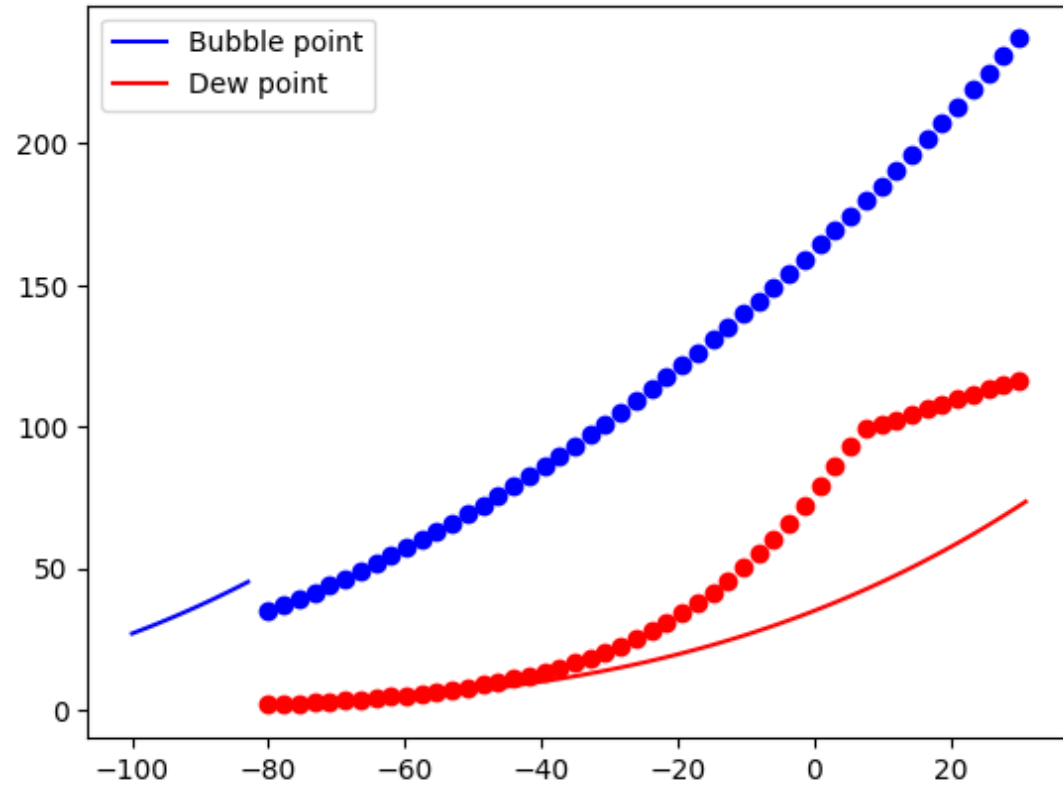
## Peng-Robinson EOS



# Success: plot Pxy and Txy curves

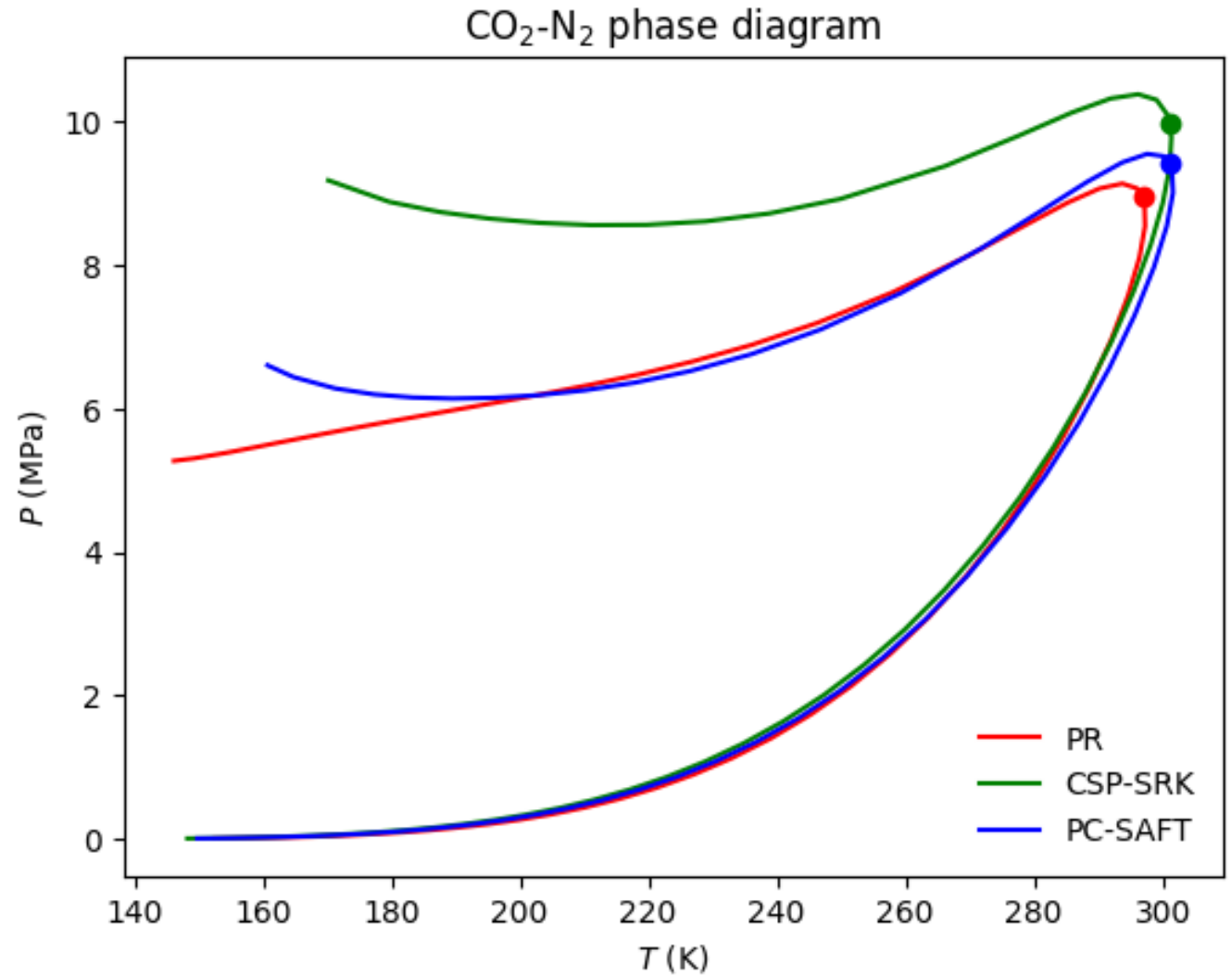


# Failure: plot PT curves



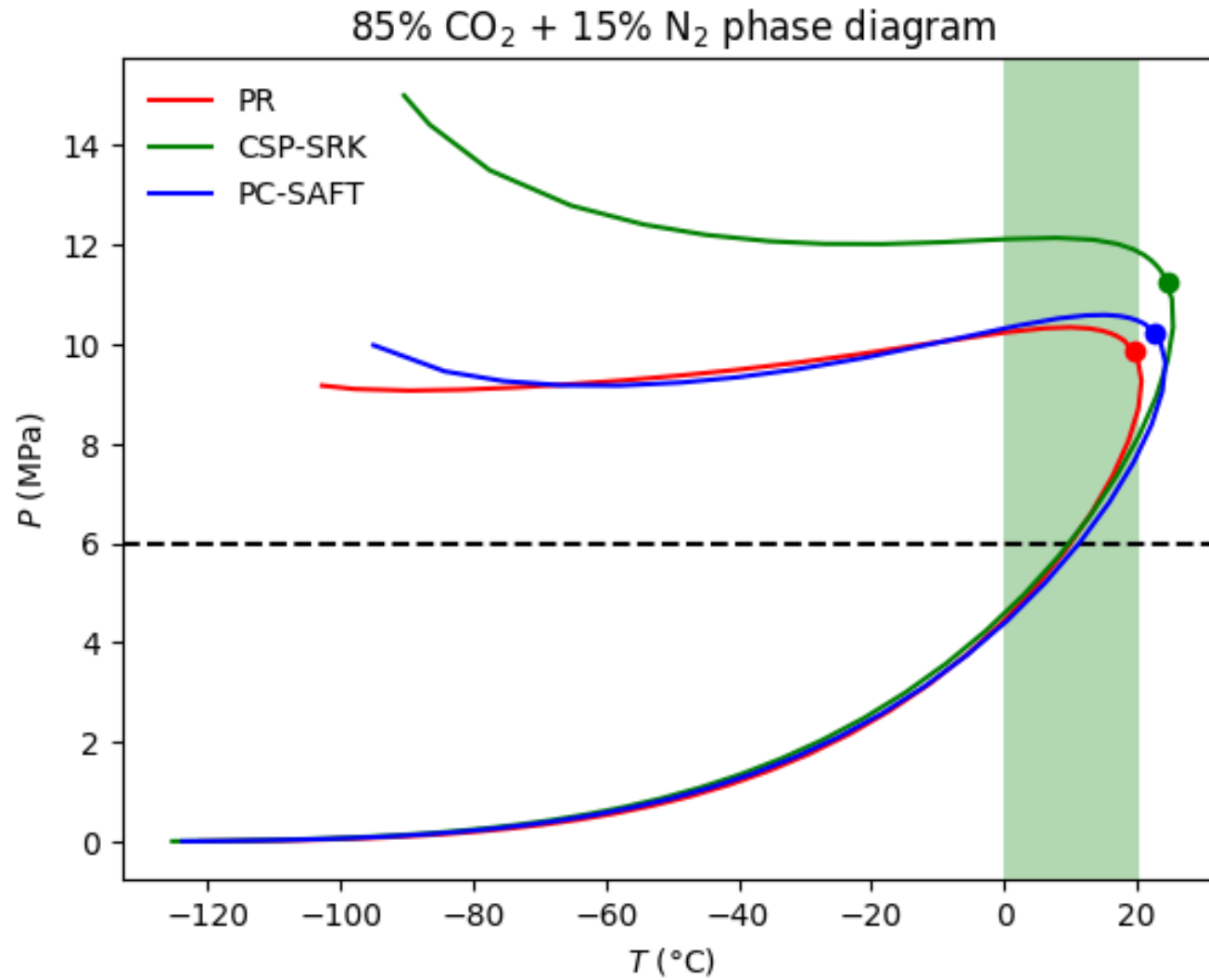
## Second tentative: use Thermopack library

Thermopack is a thermodynamics library for multi-component and multi-phase thermodynamics developed at SINTEF Energy Research and NTNU Department of Chemistry.



# Focus on our range

Our domain: 6MPa, 0°C to room temperature (20°C), 85% CO<sub>2</sub> + 15% N<sub>2</sub>.





# Comparison with real data

85% CO<sub>2</sub> + 15% N<sub>2</sub> phase diagram

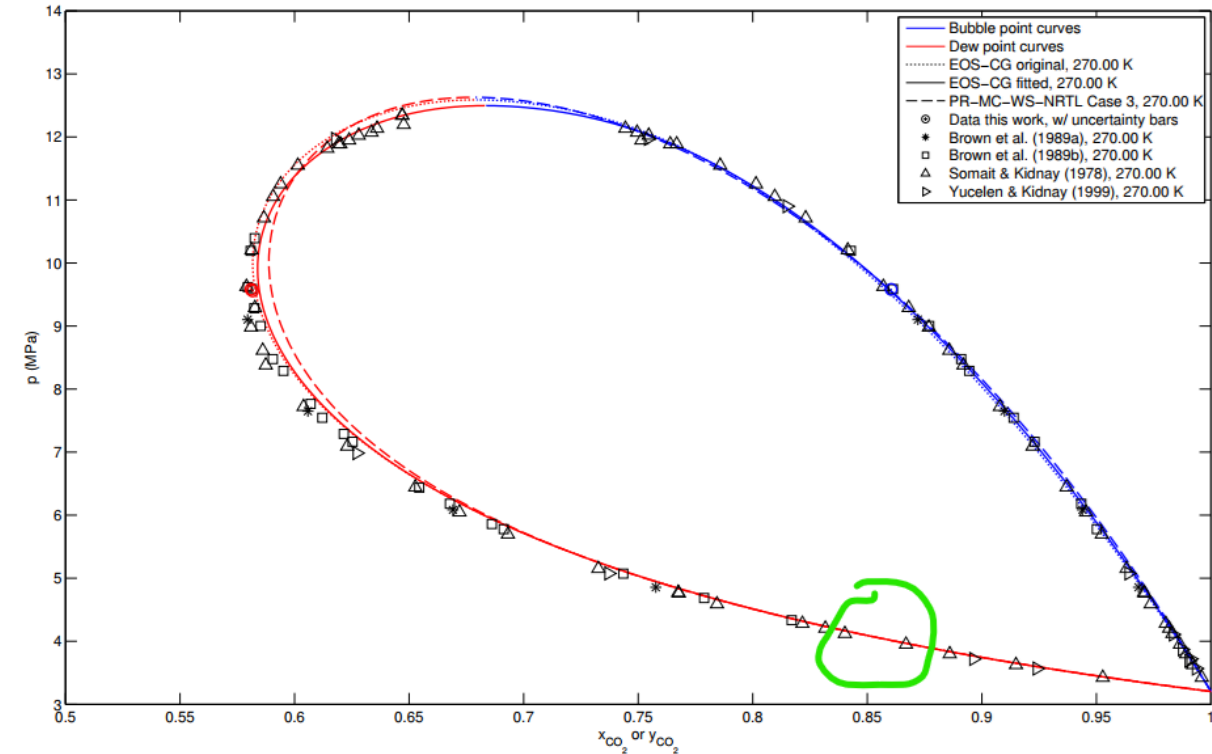
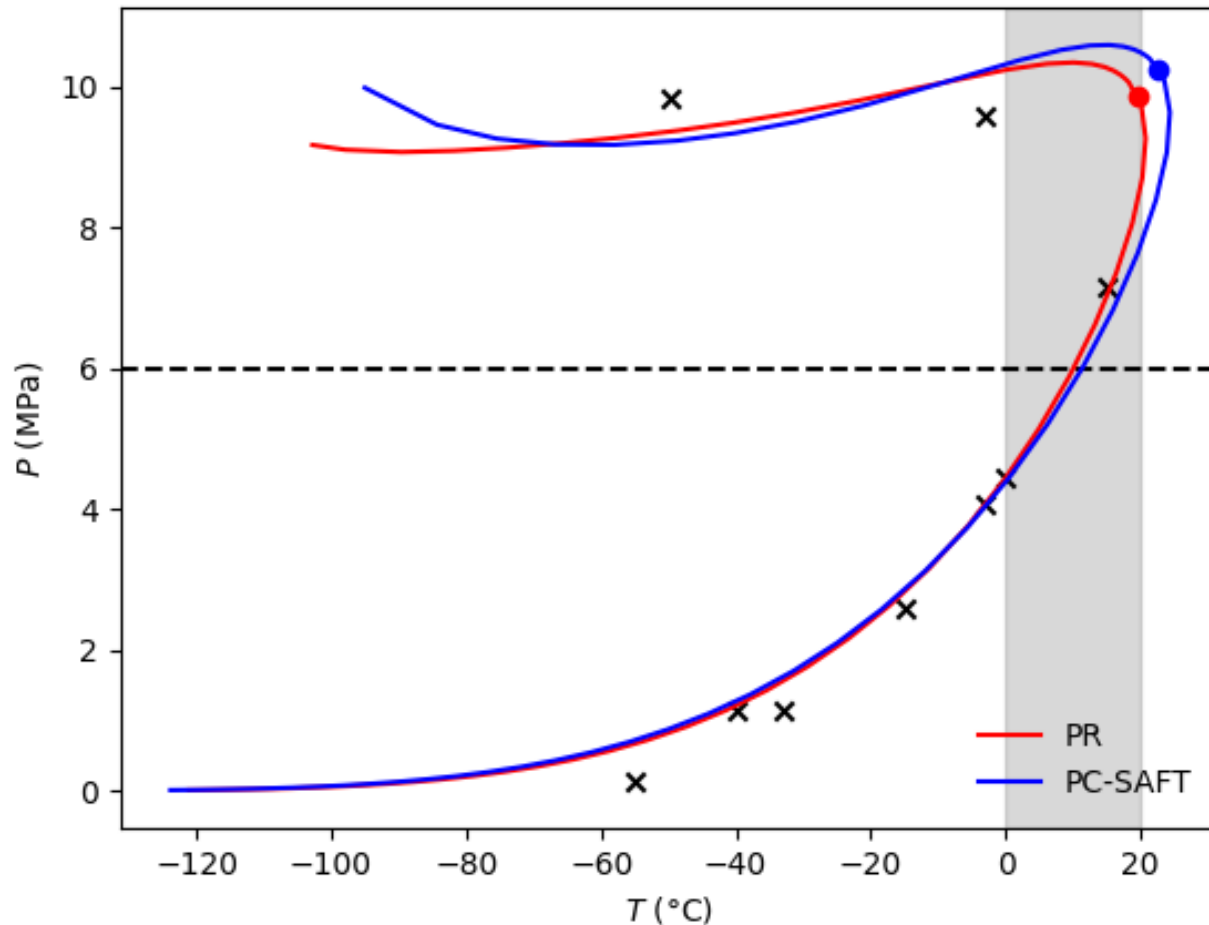
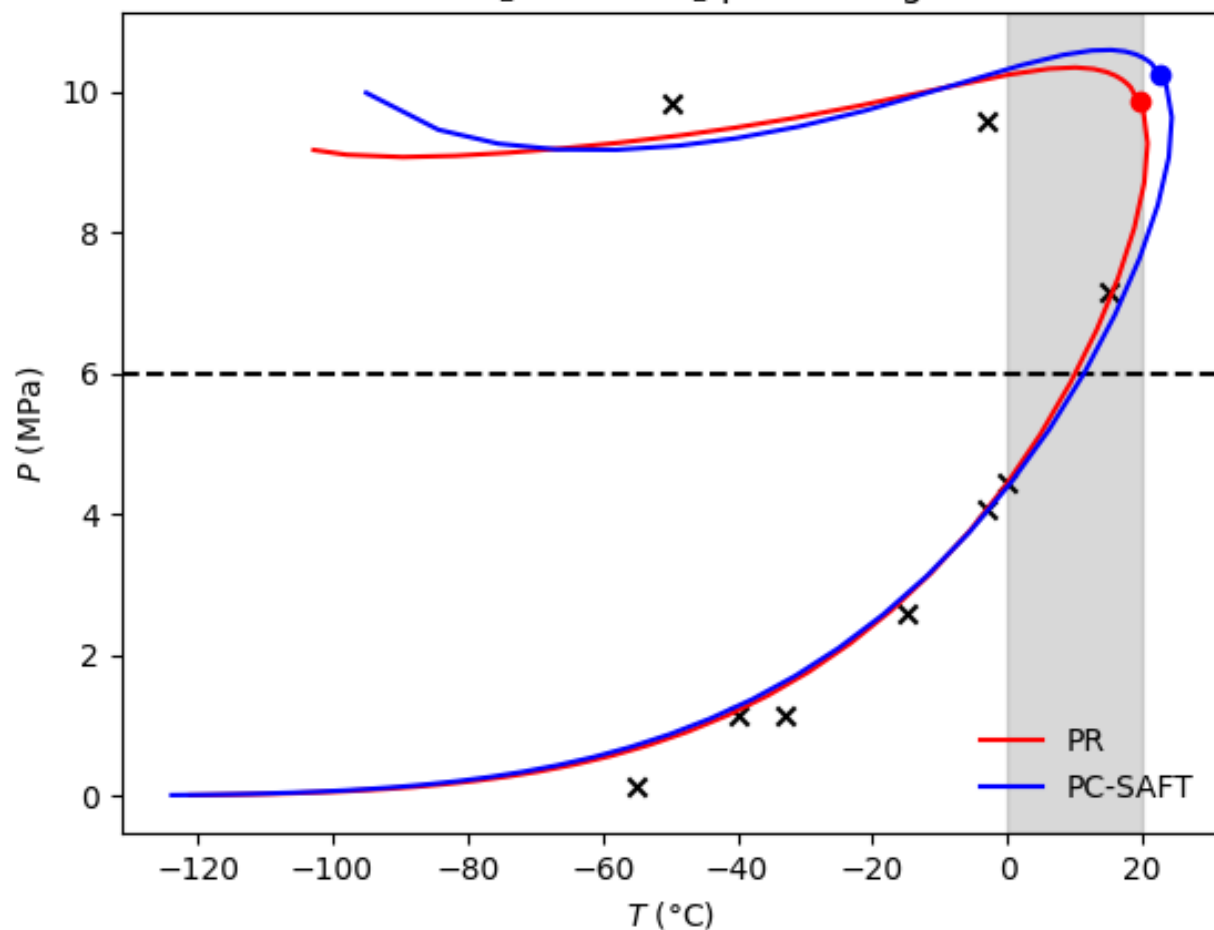


Fig. 8. Isothermal VLE data from literature [49, 28, 50, 48], EOS calculations at mean temperature  $T = 270.00$  K, and measurements with estimated uncertainties from present work:  $\tilde{x}_{\text{CO}_2}$ ,  $\tilde{y}_{\text{CO}_2}$ ,  $\tilde{p}_i$ ,  $u_c(\tilde{x}_{\text{CO}_2})$ ,  $u_c(\tilde{y}_{\text{CO}_2})$  and  $u_c(\tilde{p}_i)$  from Tables 4 and 5.

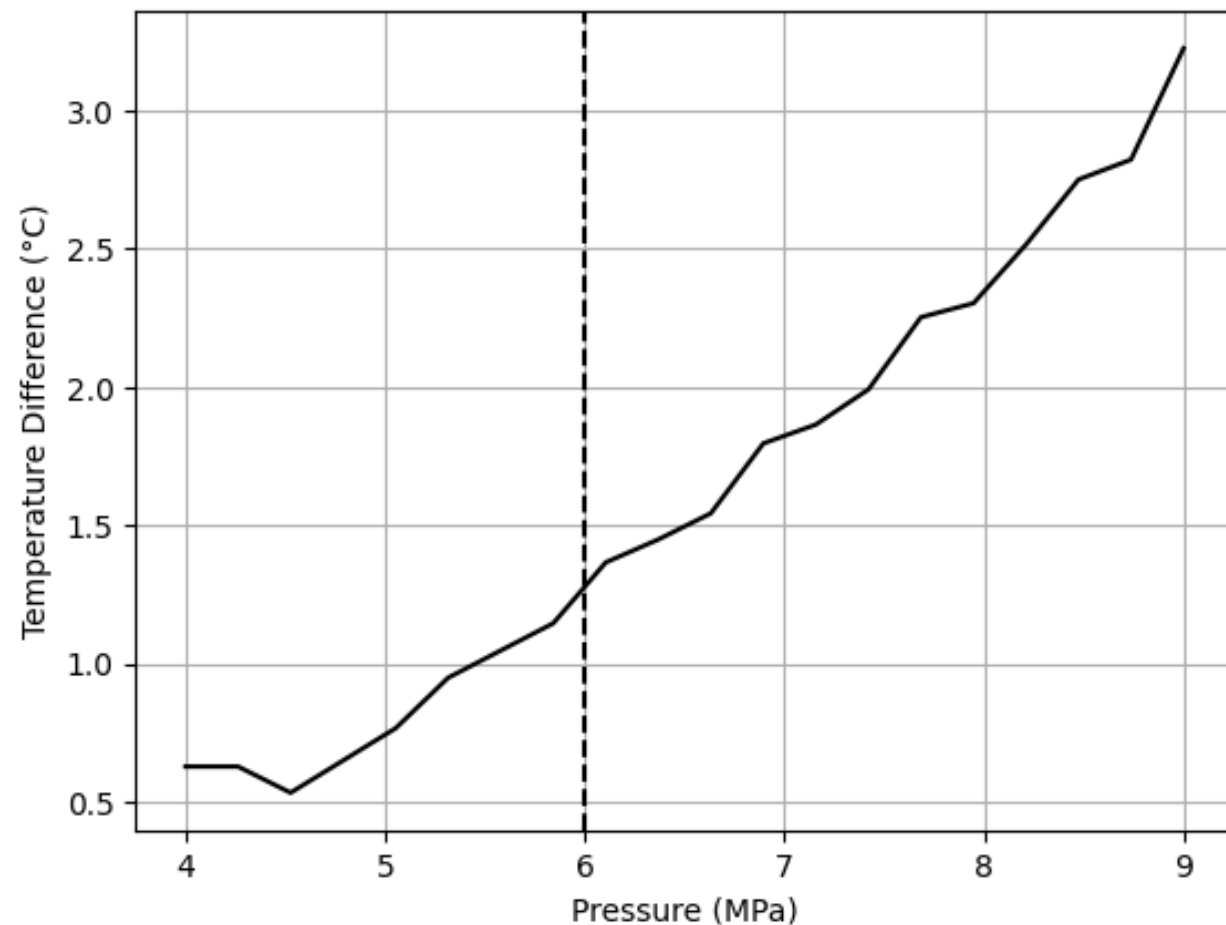
Experimental data from: S. F. Westman, H. G. J. Stang, S. W. Løvseth, A. Austegard, I. Snustad, S. Ø. Størset, and I. S. Ertesvåg, Vapor–Liquid Equilibrium Data for the Carbon Dioxide and Nitrogen (CO<sub>2</sub> + N<sub>2</sub>) System at the Temperatures 223, 270, 298 and 303 K and Pressures up to 18 MPa, Fluid Phase Equilibria 409, 207 (2016).

# Comparison between models

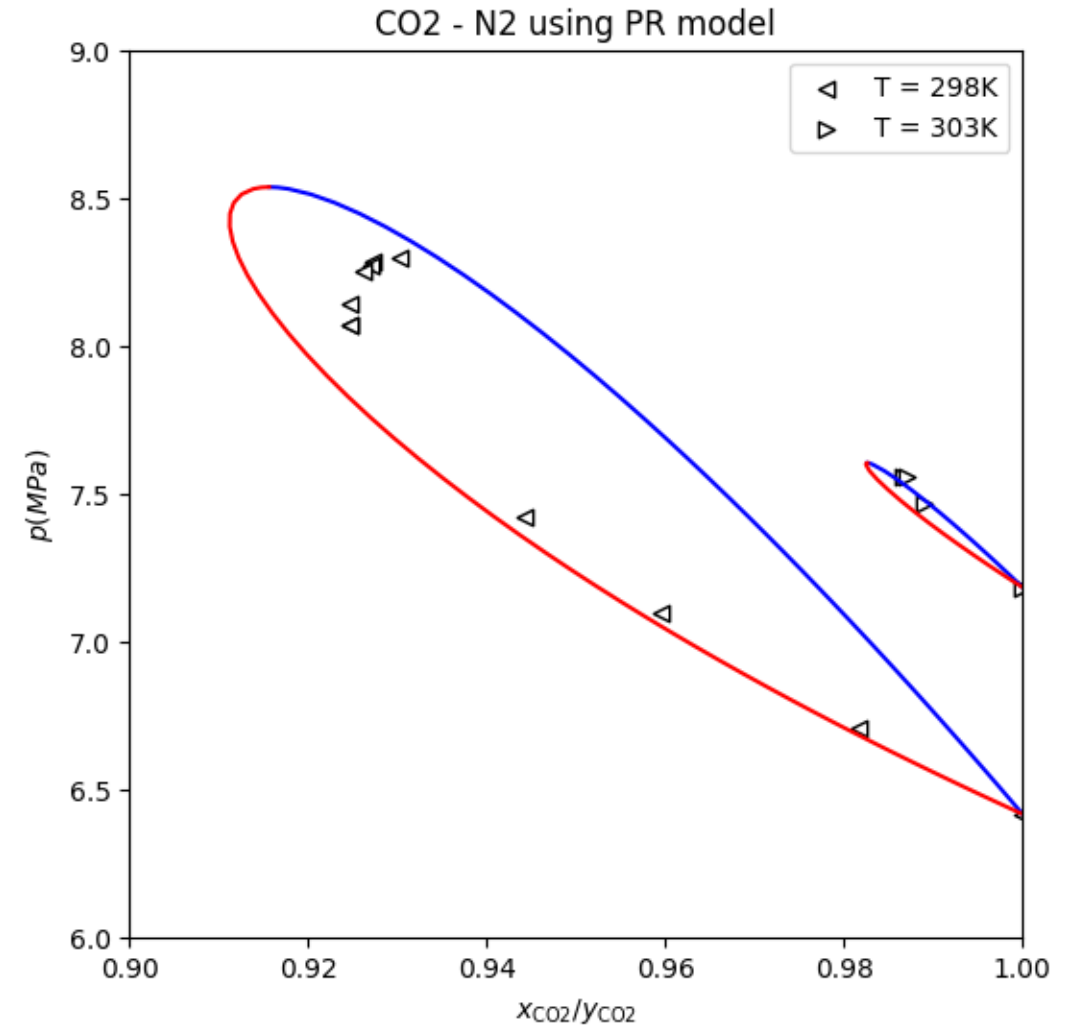
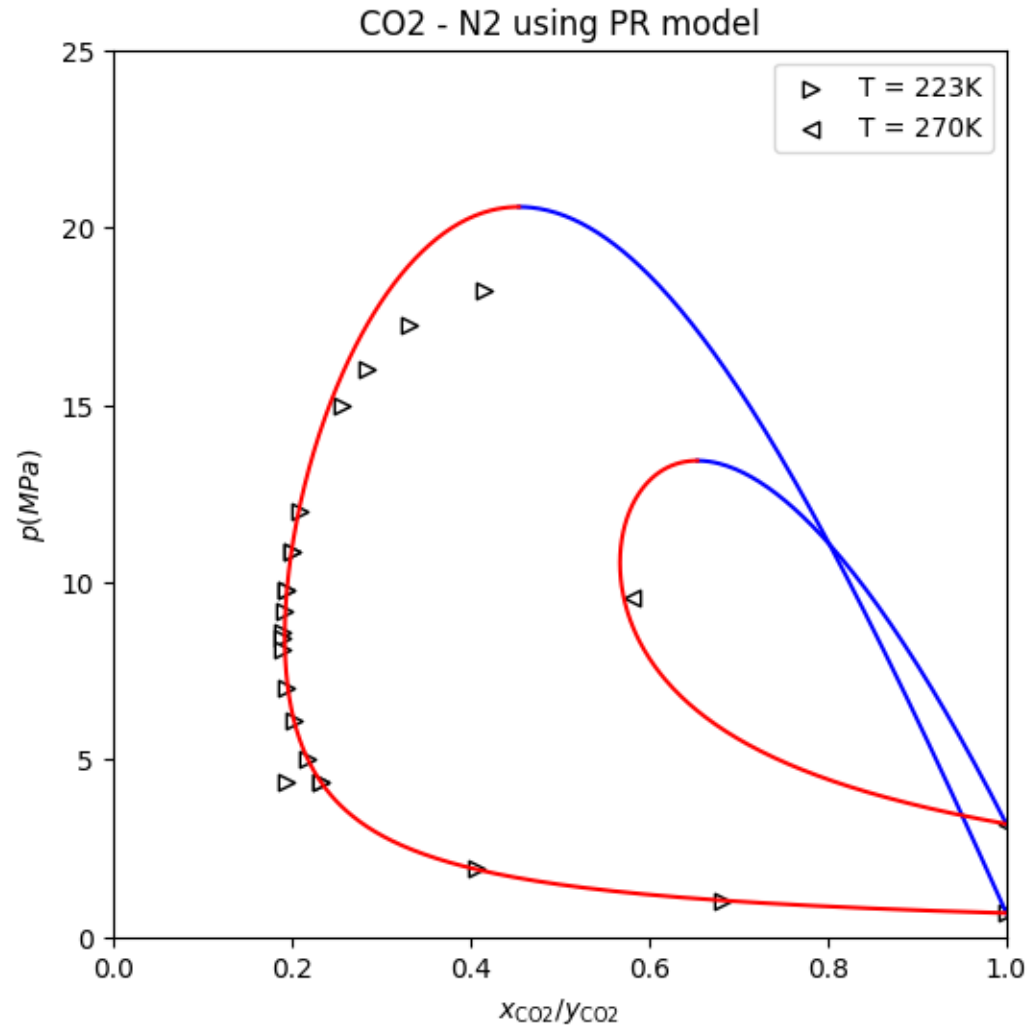
85% CO<sub>2</sub> + 15% N<sub>2</sub> phase diagram



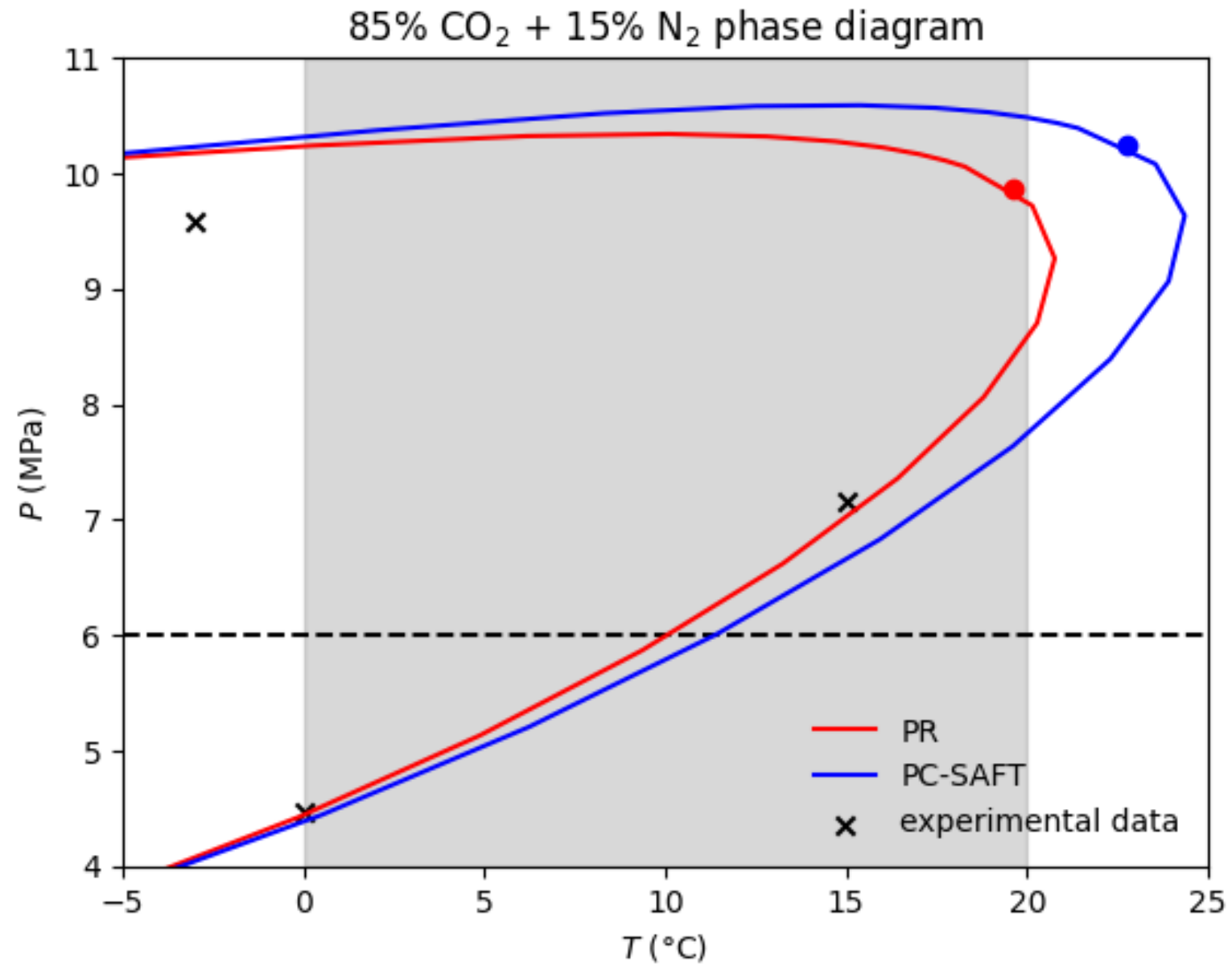
Difference between PR and PC-SAFT



# Real data in the vicinity of our domain



To sum up





**Theoretical  
model**

**Chiller setup**

**Temperature  
acquisition**

# The use of the chiller and preparation of the cooling liquid



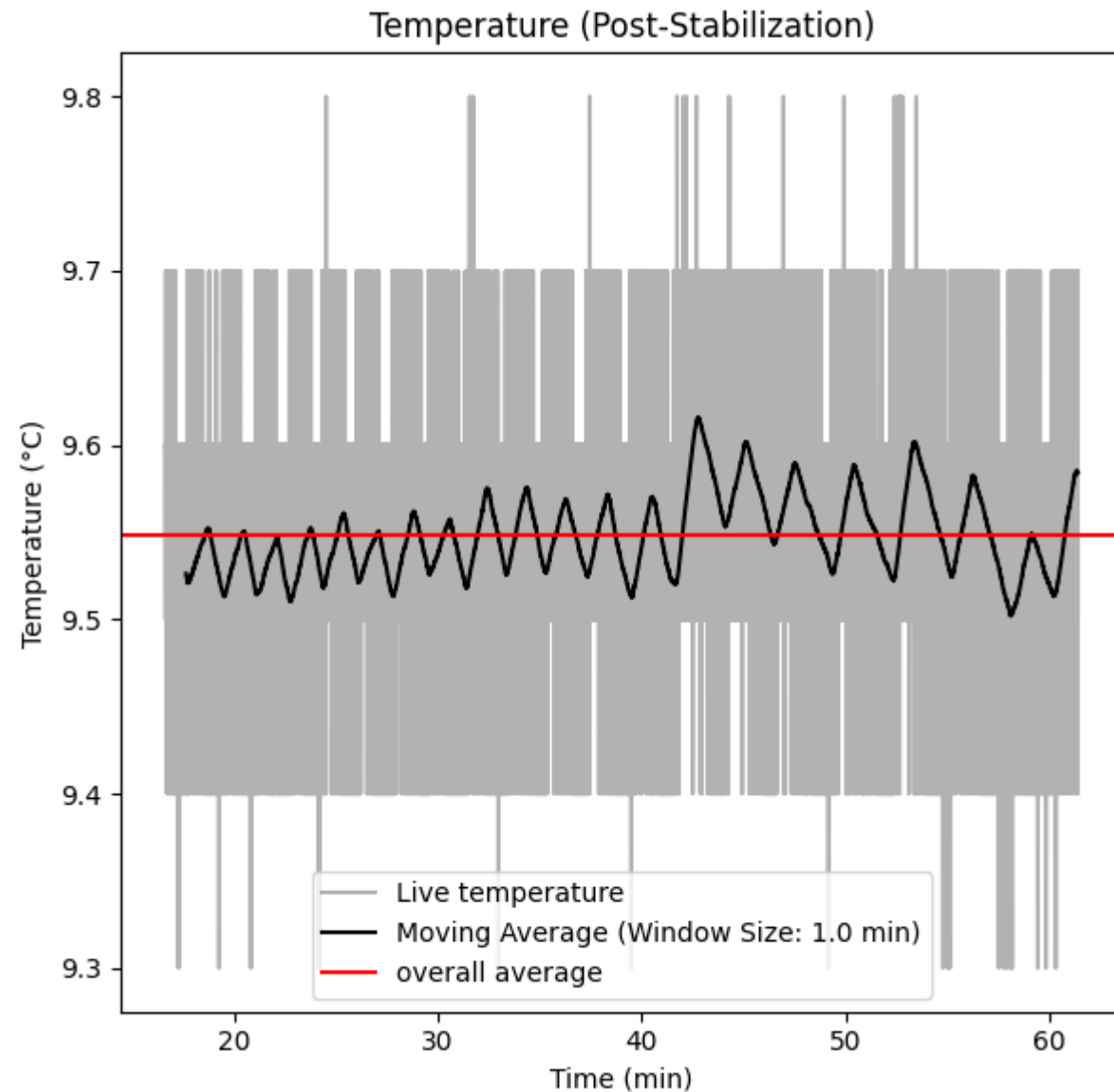
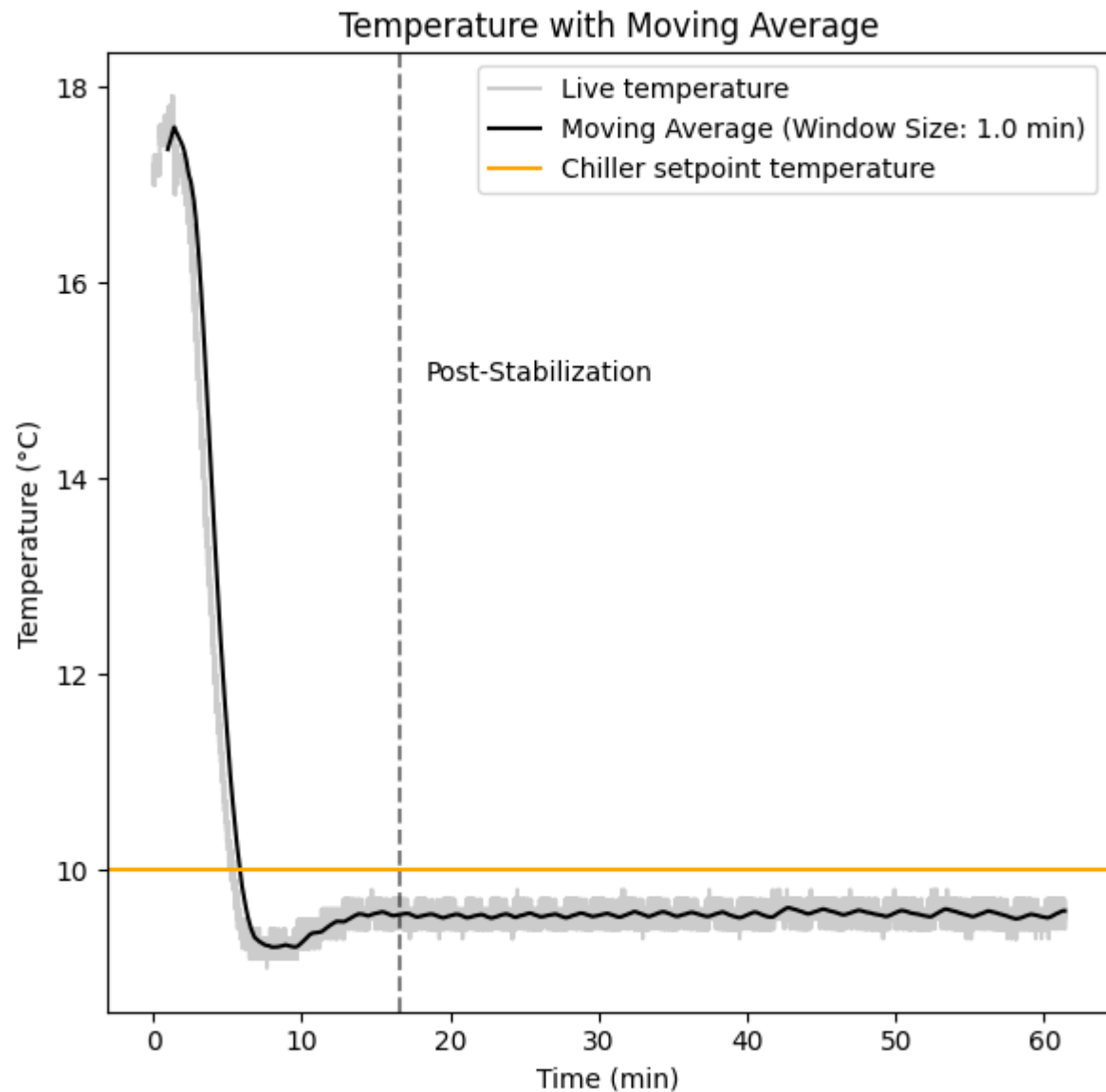
For temperatures less than 20°C chiller needs a special cooling fluid



50%

50%

# Stability of the Chiller





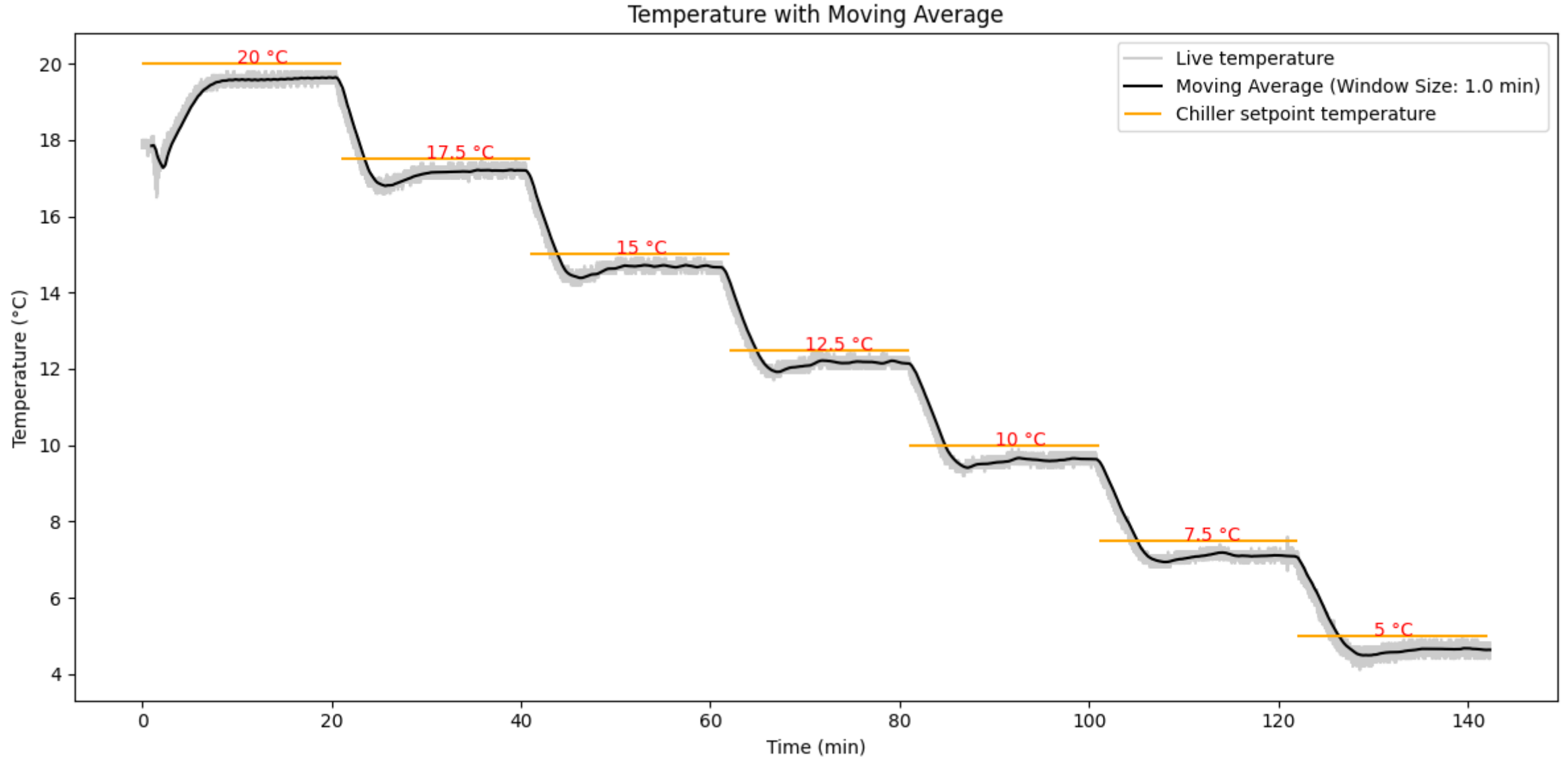
**Theoretical  
model**

**Chiller setup**

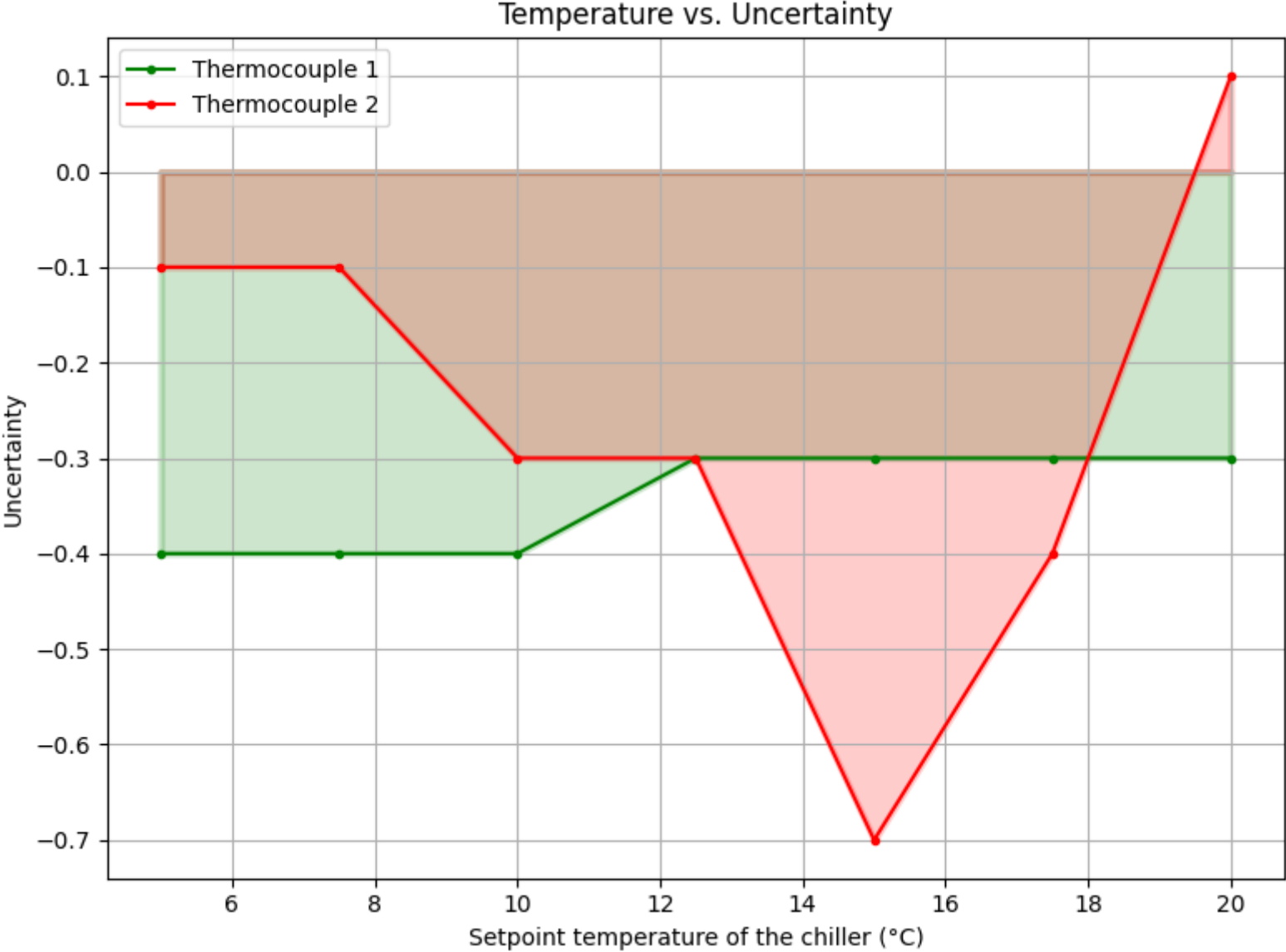
**Temperature  
acquisition**



# Uncertainty interval of thermocouples



# Uncertainty interval of thermocouples



# What's next?

**Measure the  
temperature  
of the cooling  
lines**

**Viscometry**

**Preparing the  
gas mixture**