

INTRODUCTION

Anything which can flow like liquid & gases
Known as Fluids.

Pressure

Force acting per unit Area. $P = \frac{F_{\perp}}{A}$

Gauge Pressure

Excess pressure over the atmospheric pressure ($P_o - P_{atm}$) measured with instrument. $P_{gauge} = P - P_o = \rho gh$

Atmospheric Pressure

Force exerted by atmospheric column on unit area at mean sea level.

$$[P_o = 1.013 \times 10^5 \text{ N/m}^2]$$

Hydraulic Paradox

Water is filled to a height H behind a dam of width w . The resultant Pressure on dam will be $-P_{net} = \rho gH$

9. MECHANICAL PROPERTIES OF FLUIDS

Characteristics of Ideal Fluids

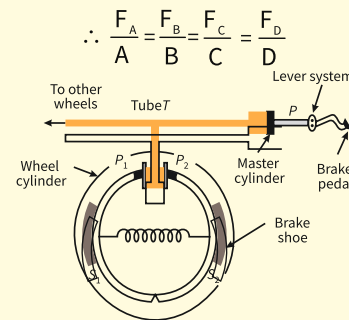
- ✦ Incompressible
- ✦ Non - Viscous
- ✦ Irrotational
- ✦ Steady (Laminar) flow

Pascal Law

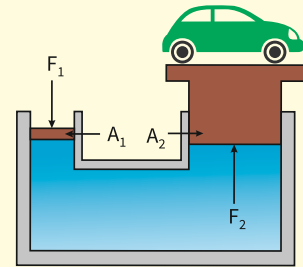
Whenever external Pressure is applied on any part of fluid contained in a Vessel, it is transmitted undiminished and equally in all direction is known as Pascal Law.

Hydraulic Machine

Hydraulic Brakes



Hydraulic lift



Hydrodynamics

Equation of Continuity

$$A_1 V_1 = A_2 V_2$$

Bernoulli's Theorem

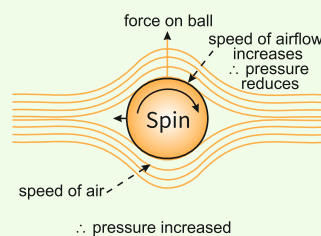
$$P + \rho gh + \frac{1}{2} \rho v^2 = \text{Constant}$$

P = Pressure; v = Velocity

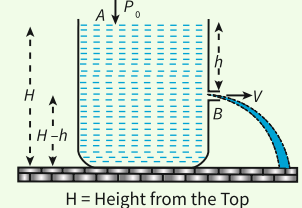
ρ = density; h = height; g = gravity

Applications

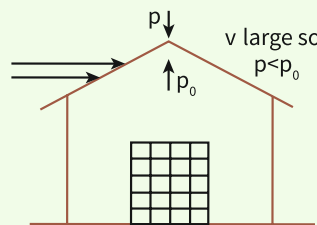
(1) Magnus Effect:



(3) Speed of Efflux: $V_B = \sqrt{2gh}$

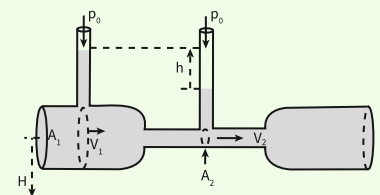


(2) Blowing off of thin Roof in storm:



(4) Venturi meter : The entering Velocity of fluids is given by

$$V_1 = \sqrt{\frac{2\rho mgh}{\rho} \left[\left(\frac{A_1}{A_2} \right)^2 - 1 \right]^{-\frac{1}{2}}}$$



VISCOSITY

Newton's Law of Viscosity

Viscous Force, $f = -\eta A \frac{dV}{dx}$

A = Area

velocity Gradient = $\frac{dV}{dx}$

Stoke's Law

When a small sphere of radius r is moving with velocity v through a homogeneous Fluid, then viscous force acting on sphere – $F_v = 6 \pi \eta r v$;

Where η = Coefficient of viscosity; Unit of η = Poise.

Terminal Velocity

Constant Velocity achieved Before net force on a body becomes Zero.

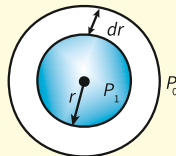
Surface energy

Additional potential energy exhibited by liquid molecules present at the surface of the molecules.

Excess Pressure Inside a Curved Liquid Surface

Excess pressure inside the drop

$$P_{ex} = (P_i - P_o) = \frac{2S}{r}$$

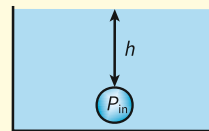


Excess pressure inside a cavity or air bubble in liquid

$$P_{ex} = \frac{2S}{r} + \rho gh$$

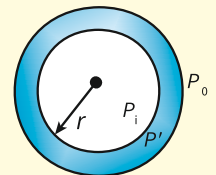
$$P_{inside} = P_{atm} + \frac{2S}{r} + \rho gh$$

$$P_{out} = P_{atm}$$



Excess pressure inside a soap bubble

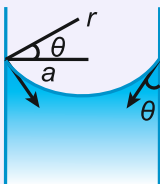
$$P_{ex} = P_i - P_o = \frac{4S}{r}$$



Capillarity

It is Property due to which liquid elevates & depressed in a capillary Tube. The Rise in height of liquid in capillary tube is given by

$$-h = \frac{2s \cos \theta}{r \rho g}$$



Cohesive Force and Adhesive Force

Cohesive Force:- Attractive Force between the molecules of same materials.

Adhesive Force:- Attractive Force between the molecules of different Materials.

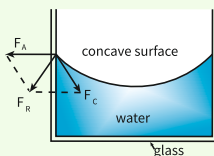
Angle of Contact

Angle between tangent Plane at the liquid surface and tangent plane at point of contact of solid.

Shape of Meniscus

Relation between cohesive and adhesive force

$$F_A > \frac{F_C}{\sqrt{2}}$$



Angle of contact

$\theta < 90^\circ$ (Acute angle)

Shape of meniscus

Concave

Wetting property

Liquid wets the solid surface

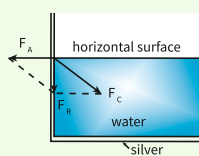
Level in capillary tube

Liquid rises up

Example

Glass-Water

$$F_A = \frac{F_C}{\sqrt{2}}$$



$\theta = 90^\circ$ (Right angle)

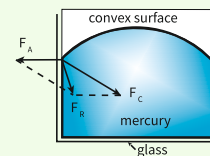
Plane

Liquid does not wet the solid surface

Liquid neither rises nor falls

Silver-Water

$$F_A < \frac{F_C}{\sqrt{2}}$$



$\theta > 90^\circ$ (Obtuse angle)

Convex

Liquid does not wet the solid surface

Liquid falls (or) depressed

Glass-Mercury