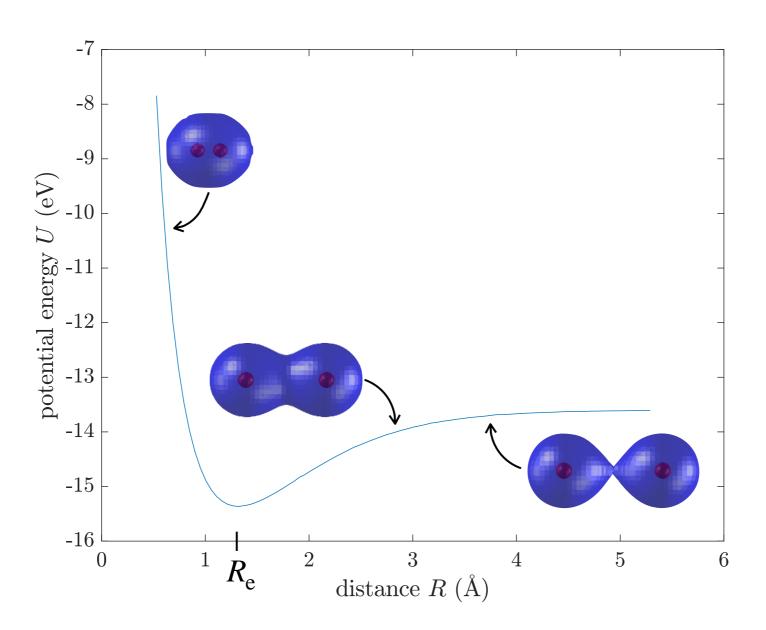
# **Born-Oppenheimer**

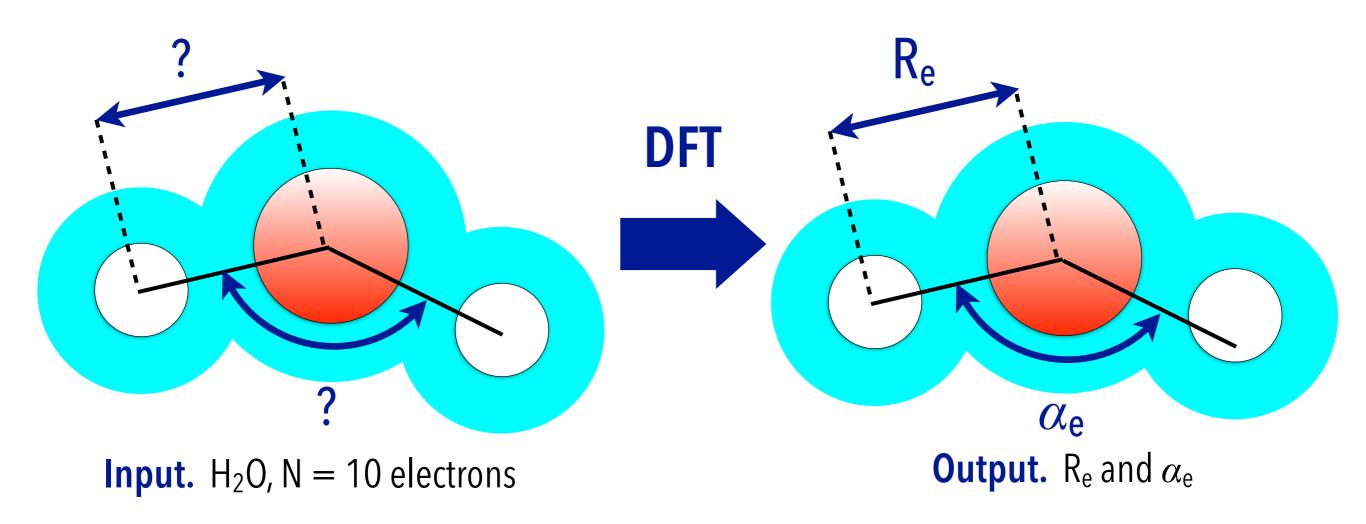




Max Born (1882–1970)



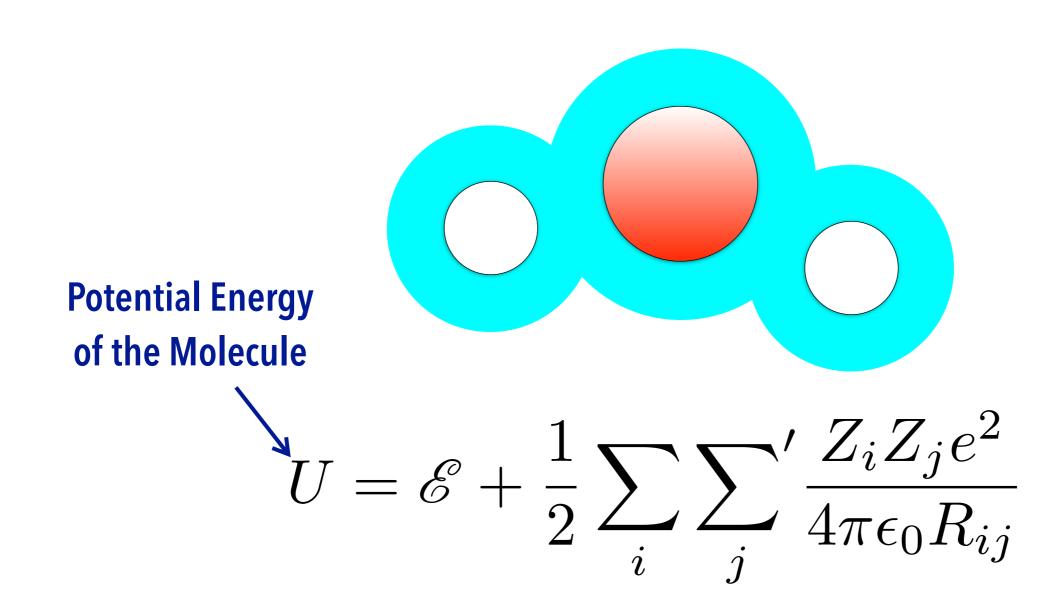
Julius Robert Oppenheimer (1904–1967)



**Density-Functional Theory (DFT)** is a quantum method widely used in materials science and chemistry to calculate the properties of systems of atoms (energy, geometry, reactivity, phase stability) using limited experimental data.

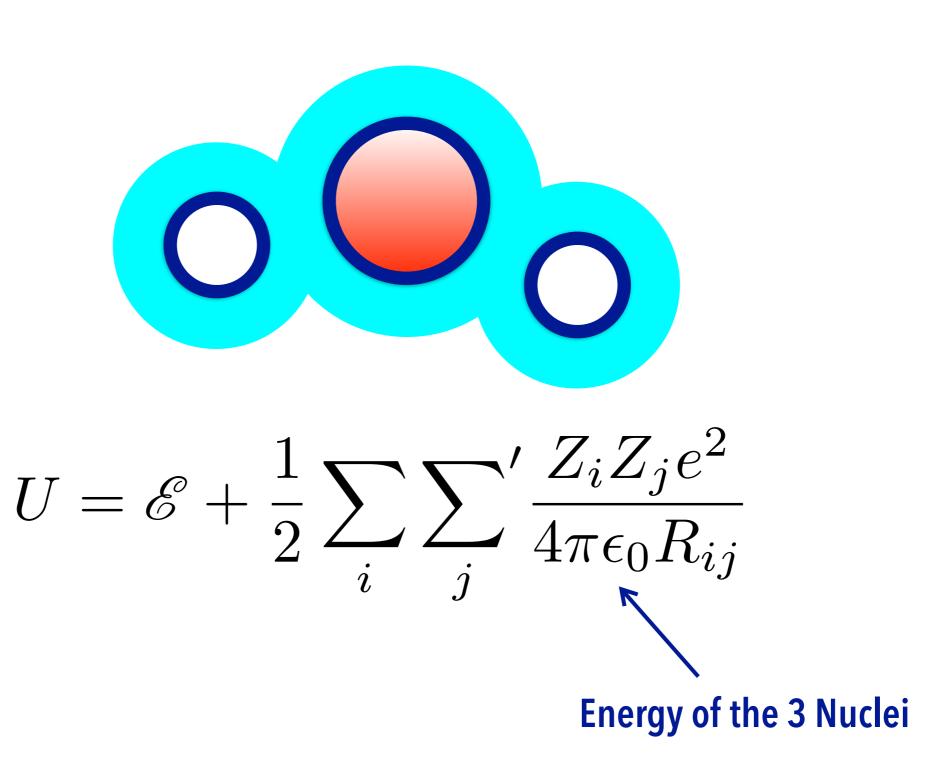
## **Density-functional theory**

lacktriangle Now, let's study the potential energy U of the molecule

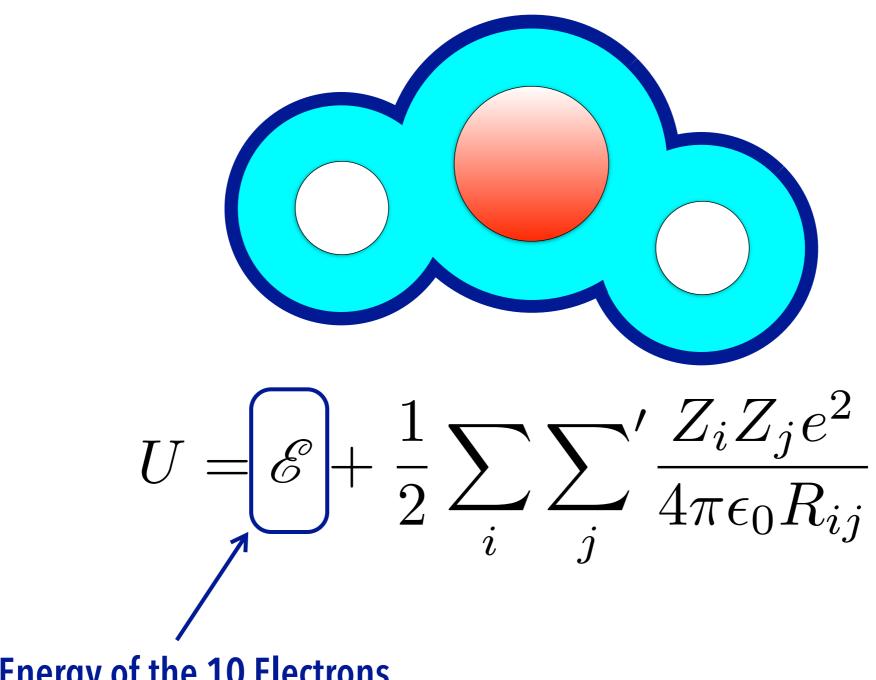


**Recall.** We have seen this equation. It is the **Born-Oppenheimer Equation**.

lacktriangle Now, let's study the potential energy U of the molecule

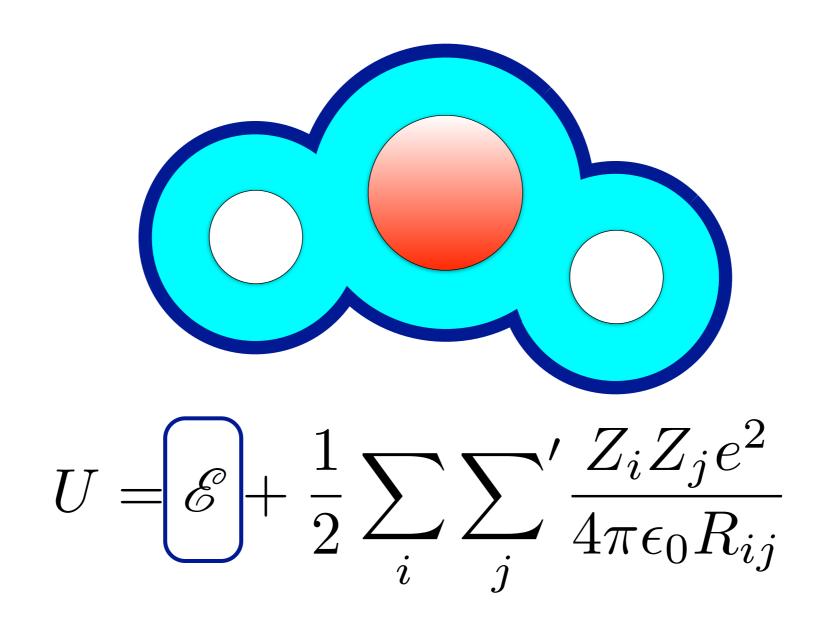


 $\blacktriangleright$  Now, let's study the potential energy U of the molecule



**Energy of the 10 Electrons** 

lacktriangle Now, let's study the potential energy U of the molecule

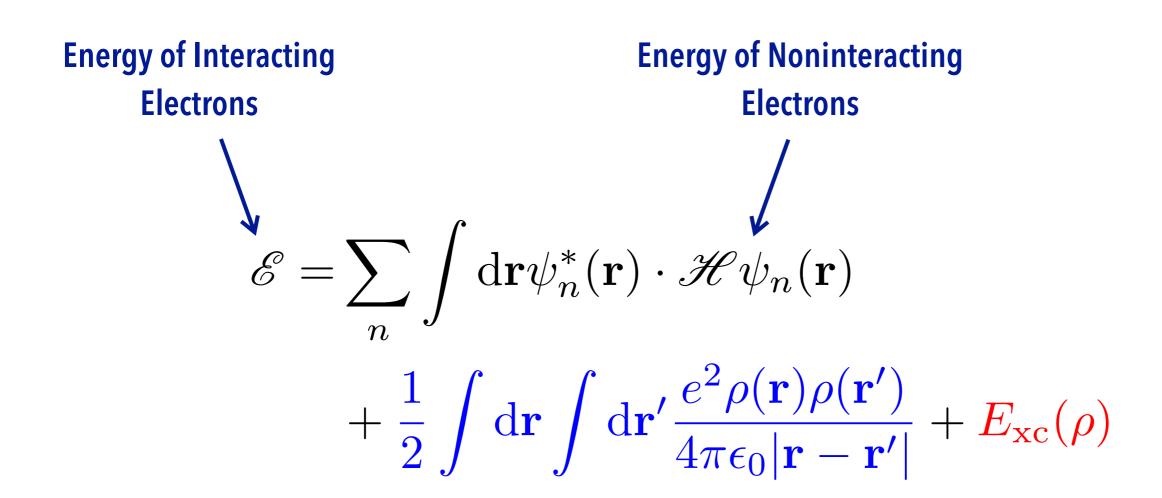


lacktriangle DFT provides a way to calculate  $\begin{center} \mathcal{E} \end{center}$ 

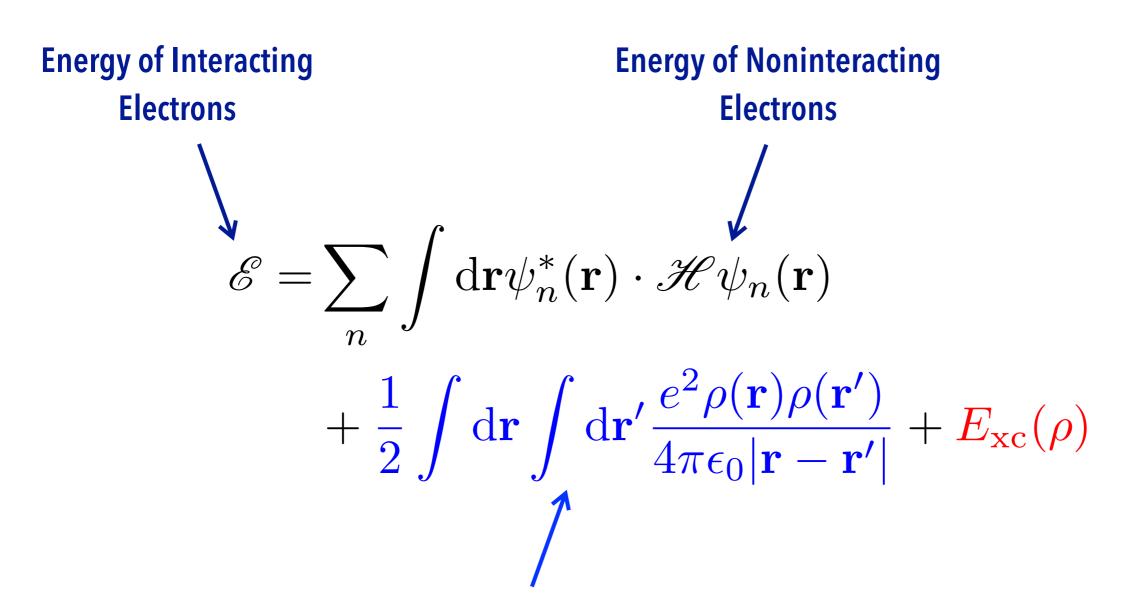
▶ Here is how the energy of the interacting electrons is written within DFT:

Energy of Interacting Electrons 
$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathscr{H} \psi_{n}(\mathbf{r}) + \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi\epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{\mathrm{xc}}(\rho)$$

▶ Here is how the energy of the interacting electrons is written within DFT:

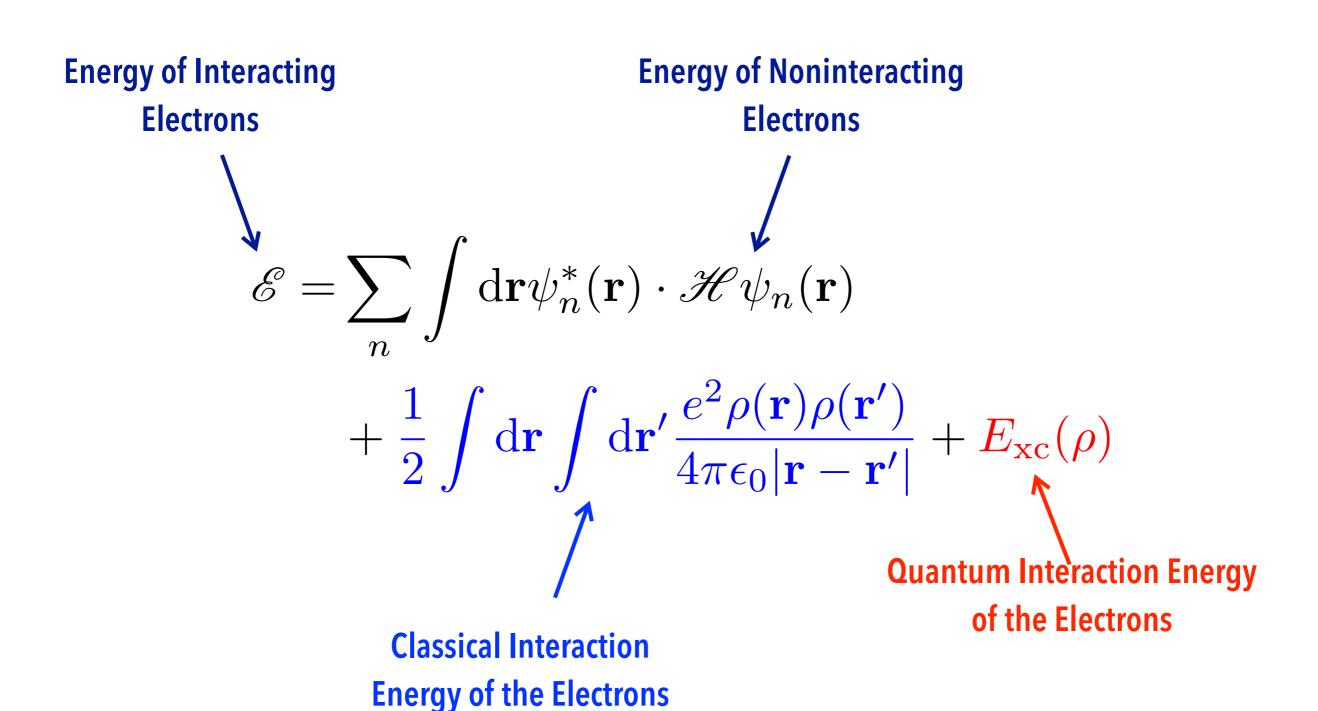


▶ Here is how the energy of the interacting electrons is written within DFT:



**Classical Interaction Energy of the Electrons** 

▶ Here is how the energy of the interacting electrons is written within DFT:



▶ Here is how the energy of the interacting electrons is written within DFT:

$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{xc}(\rho)$$
Quantum Interaction Energy of the Electrons

▶ Question 1. Why is this method called densityfunctional theory?

▶ Here is how the energy of the interacting electrons is written within DFT:

$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{xc}(\rho)$$
Quantum Interaction Energy of the Electrons

▶ Answer 1. Because the quantum interaction energy  $E_{xc}$  is a function of the electron density  $\rho$ .

▶ Here is how the energy of the interacting electrons is written within DFT:

$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{xc}(\rho)$$
Quantum Interaction Energy of the Electrons

▶ Question 2. Do you remember the expression of the electron density  $\rho$ ?

▶ Here is how the energy of the interacting electrons is written within DFT:

$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{\mathrm{xc}}(\rho)$$
Quantum Interaction Energy of the Electrons

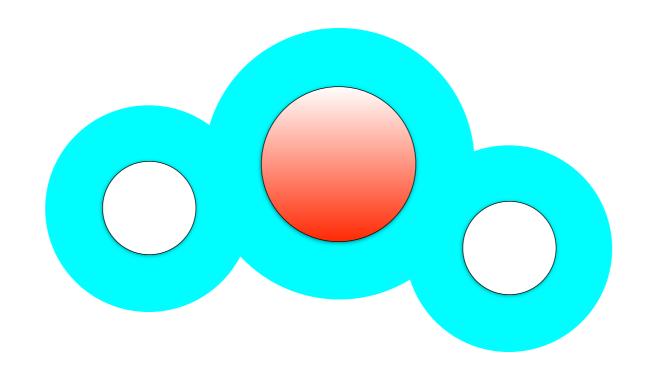
▶ Answer 2. The electron density is  $\rho(\mathbf{r}) = \sum_{n=1}^{\infty} |\psi_n(\mathbf{r})|^2$ 

- Now, we define DFT.
- ▶ Definition | Density-functional theory. Density-functional theory is a method to calculate the energy of interacting electrons. Electron interactions include two contributions: a classical (electrostatic) contribution and a quantum (exchange-correlation) contribution that depends on the electron density  $\rho(\mathbf{r}) = \sum_{n=1}^{N} |\psi_n(\mathbf{r})|^2$ .

$$\mathcal{E} = \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

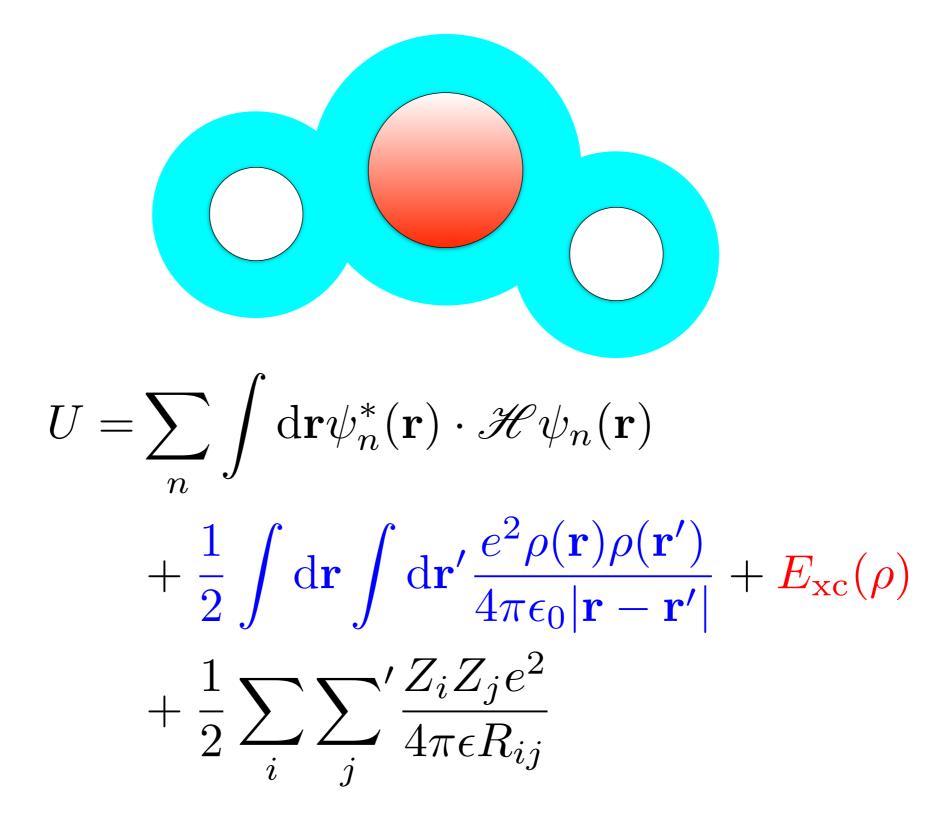
$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{xc}(\rho)$$

▶ This is the energy of the molecule:

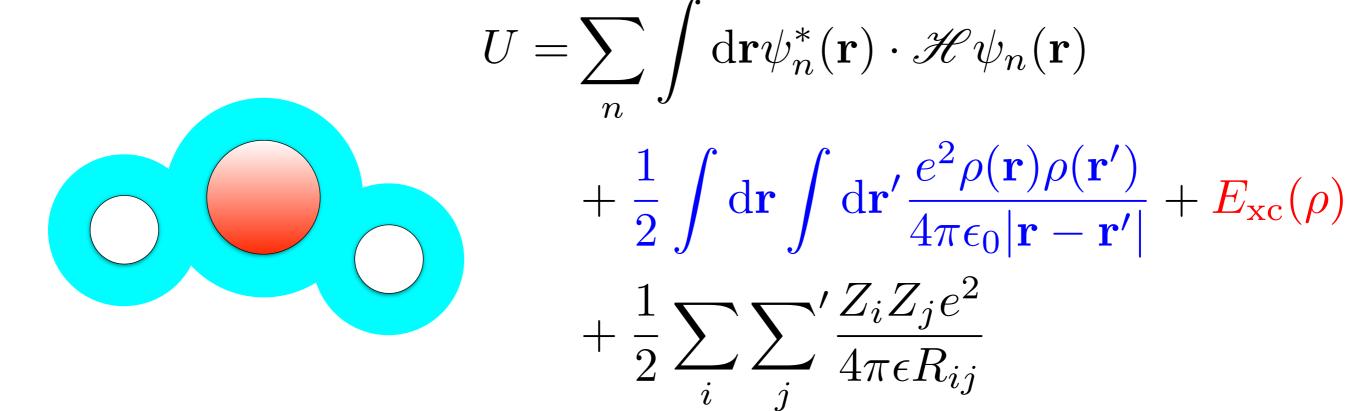


$$U = \mathcal{E} + \frac{1}{2} \sum_{i} \sum_{j}' \frac{Z_i Z_j e^2}{4\pi \epsilon_0 R_{ij}}$$

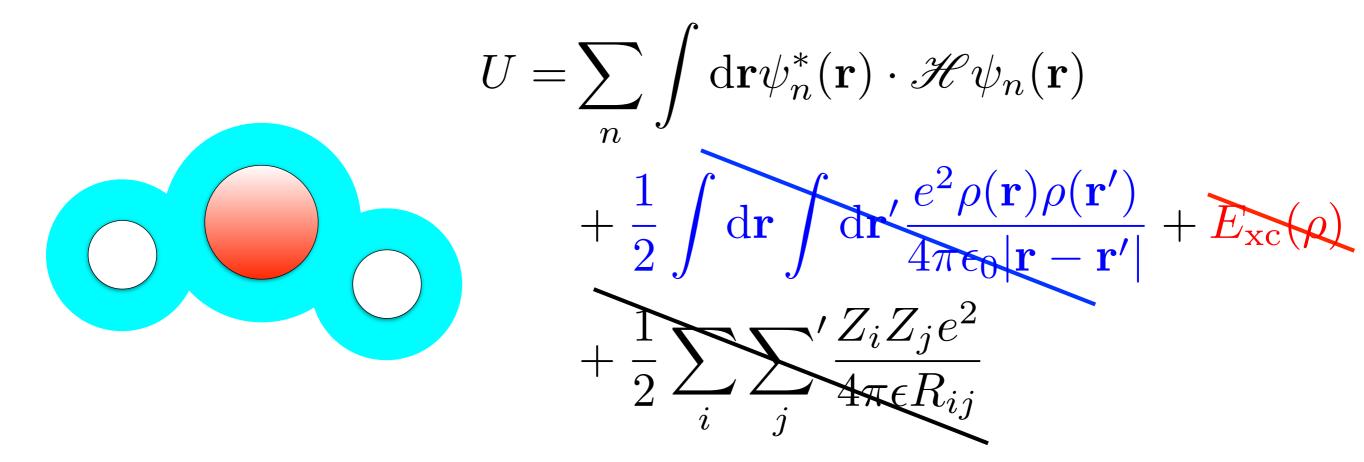
▶ Substituting the DFT energy, we obtain:



#### **Minimization**



- ▶ Objective. We must find the minimum of *U* to get the geometry.
- ▶ Let's break down this complexity.

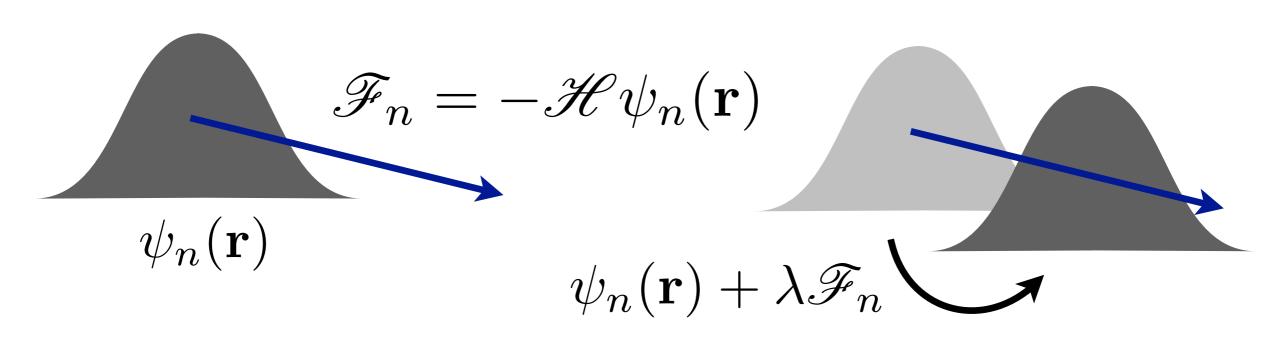


- ▶ Objective. We must find the minimum of U to get the geometry.
- Let's break down this complexity.
- ▶ We ignore the last 3 terms (!).

$$\sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

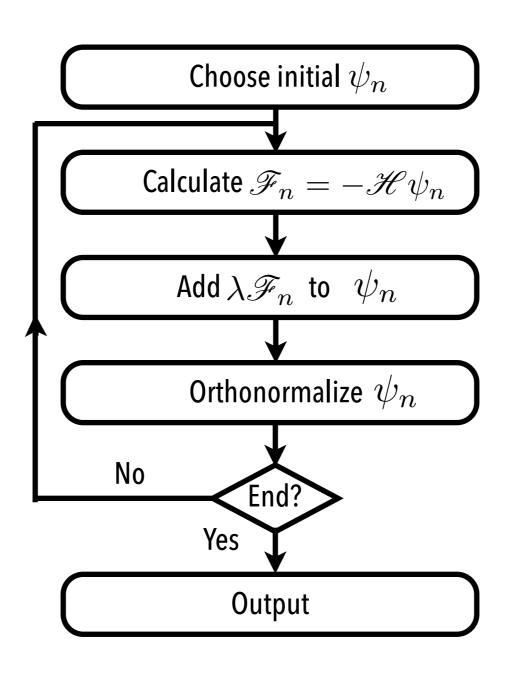
- ▶ We know how to minimize this expression.
- ▶ We can use steepest descent

#### **Steepest-Descent Method in Quantum Mechanics**



$$\sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

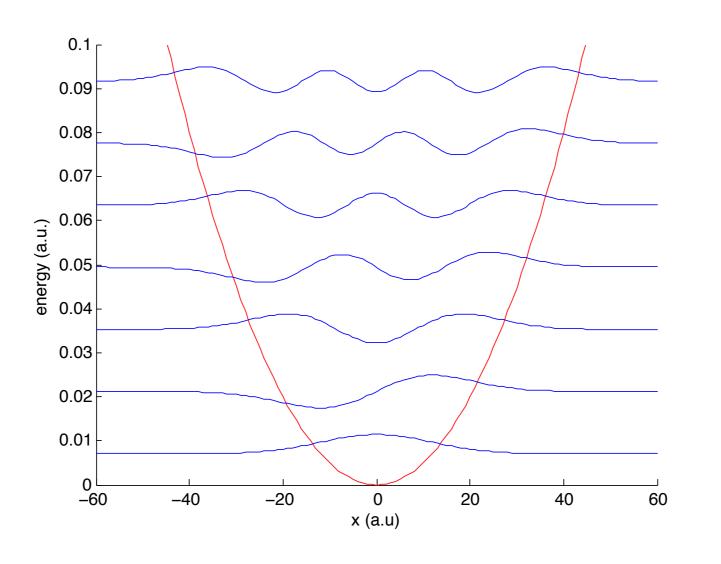
#### ▶ Steepest descent runs as follows



#### Simulation

▶ Let's test the steepest-descent algorithm

▶ We've already coded steepest descent:



We coded steepest descent to find electronic states in a harmonic well.

▶ The code was many\_electrons.m

#### Simulation (cont'd)

▶ Question. How would you modify this code to show that the algorithm works (i.e., the energy goes down)?

$$\sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

▶ Hint. If and psi are already defined. We can compute the integral in the sum as

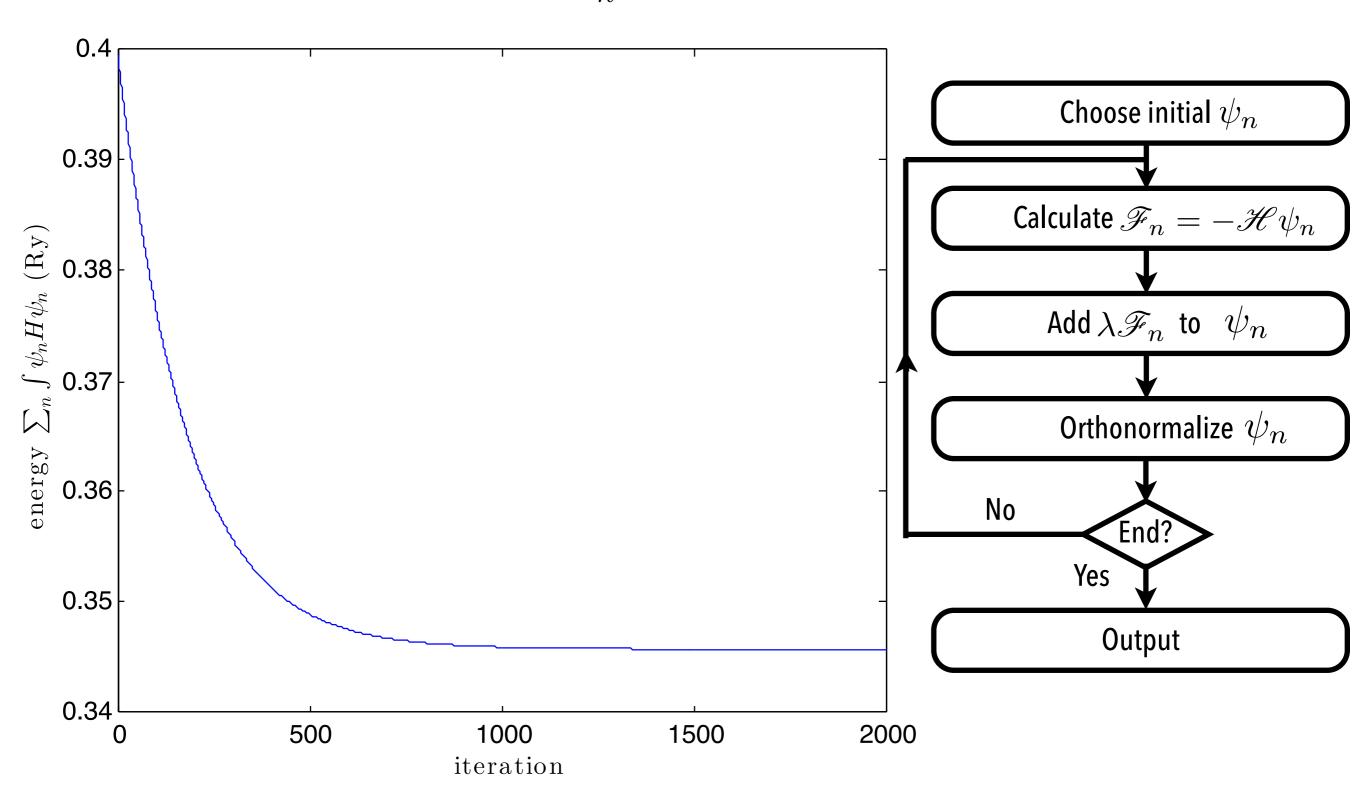
sum(dx\*conj(psi(:,n)).\*(H\*psi(:,n)))

n: index of the wave function

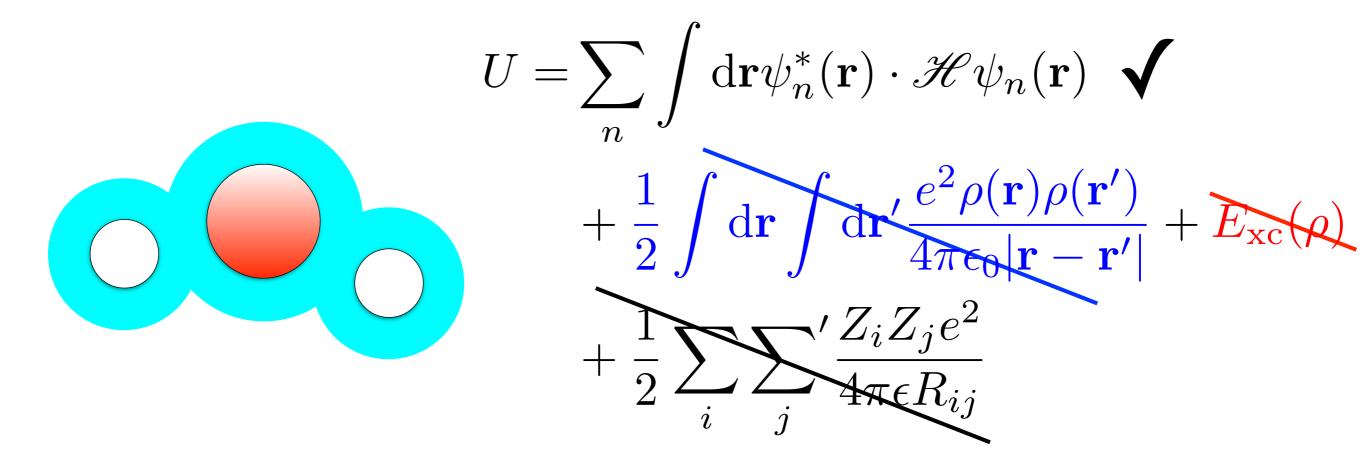
dx: space step.

#### Result

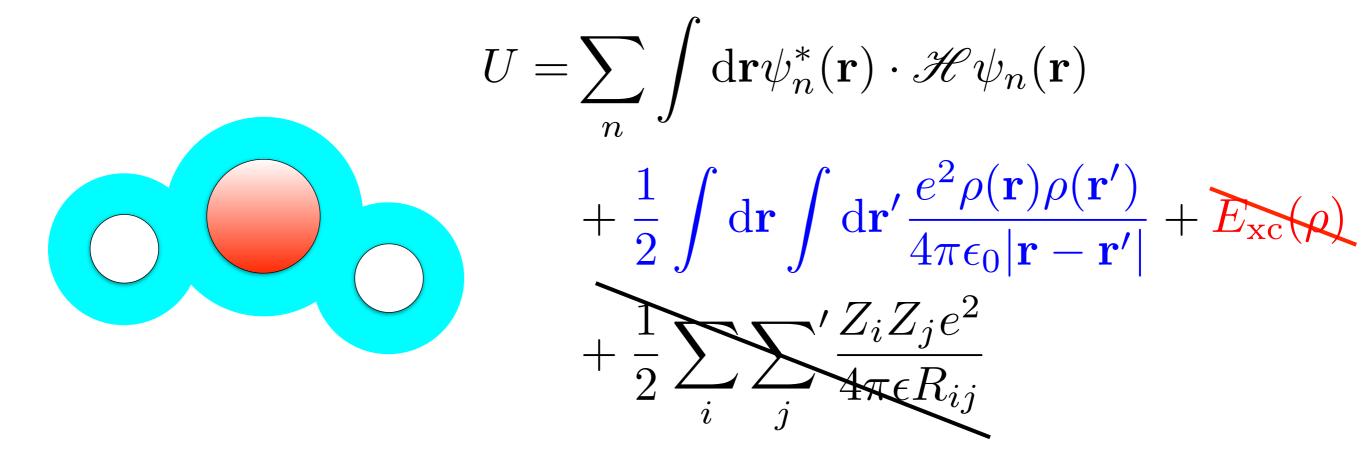
▶ The algorithm works (  $\sum_{\mathbf{r}} \int d\mathbf{r} \psi_n^*(\mathbf{r}) \cdot \mathcal{H} \psi_n(\mathbf{r})$  goes down)



#### Minimization



▶ So, we know how to handle the first term.



- ▶ So, we know how to handle the first term.
- Now, let's study the second term.

#### Electrostatic potential

$$\sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|}$$

- ▶ We've an electrostatic term in the energy.
- ▶ Therefore, the force  $\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r})$

changes into 
$$\mathscr{F}_n = -\,\mathscr{H}\psi_n(\mathbf{r}) - v_{\mathrm{ele}}(\mathbf{r})\psi_n(\mathbf{r})$$

This is the potential acting on the electrons due to the electrostatic interactions.

$$\sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|}$$

- ▶ We've an electrostatic term in the energy.
- ▶ Therefore, the force  $\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r})$

changes into 
$$\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r}) - v_{\mathrm{ele}}(\mathbf{r})\psi_n(\mathbf{r})$$

**Question.** Can you guess the expression of the electrostatic potential v<sub>ele</sub>?

**Answer.** From classical electrostatics, we know that the electrostatic potential due to the charge density  $\rho(\mathbf{r})$  is

$$v_{\text{ele}}(\mathbf{r}) = \int d\mathbf{r}' \frac{e^2}{4\pi\epsilon_0} \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}$$

**Answer.** From classical electrostatics, we know that the electrostatic potential due to the charge density  $\rho(\mathbf{r})$  is

$$v_{\text{ele}}(\mathbf{r}) = \int d\mathbf{r}' \frac{e^2}{4\pi\epsilon_0} \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|}$$

**But let's assume** that you have forgotten this result. You could still recover it by applying a mathematical procedure called **functional derivative**.

$$E_{\mathrm{ele}}$$
  $v_{\mathrm{ele}}(\mathbf{r})$ 
Functional Derivative

This procedure is explained in the next slide.

**Computing the Functional Derivative of E**<sub>ele</sub>. Consider the electrostatic energy

$$\mathsf{E}_{\mathsf{ele}}(\rho) = \mathsf{e}^2/(8\pi\varepsilon_0) \int \mathsf{d}\mathbf{r} \int \mathsf{d}\mathbf{r}' \, \rho(\mathbf{r}) \rho(\mathbf{r}') / |\mathbf{r} - \mathbf{r}'|$$

Then expand this energy to first order to obtain an expression of the form

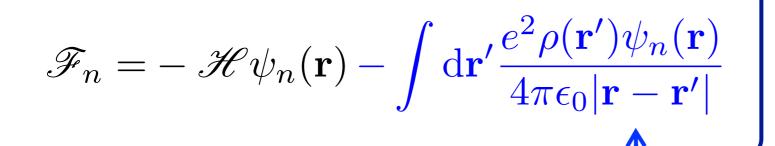
$$E_{ele}(\rho + \delta \rho) = E_{ele}(\rho) + \int d\mathbf{r} \, v_{ele}(\mathbf{r}) \, \delta \rho(\mathbf{r}) + \dots$$

The term  $v_{ele}(\mathbf{r})$  in front of  $\delta \rho(\mathbf{r})$  in the integral is called the functional derivative of  $E_{ele}$ .

**Note.** If you are interested you can follow the derivation in the **Appendix**. You can also skip it if you are not so interested, but make sure that you understand the general idea.

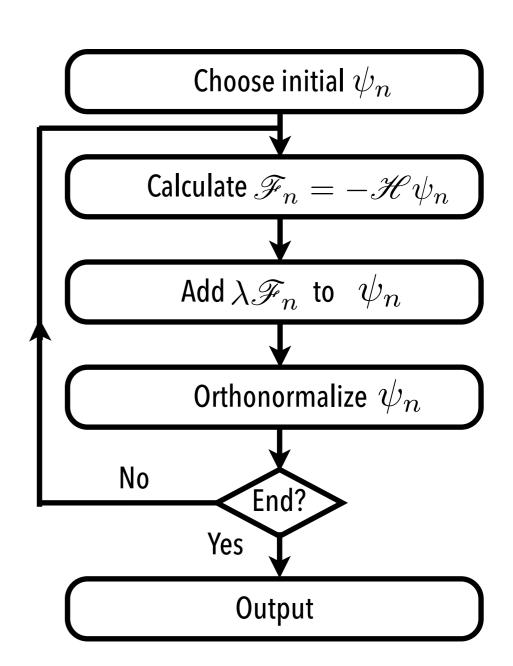
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▶ Question. Now, how would you change the algorithm?



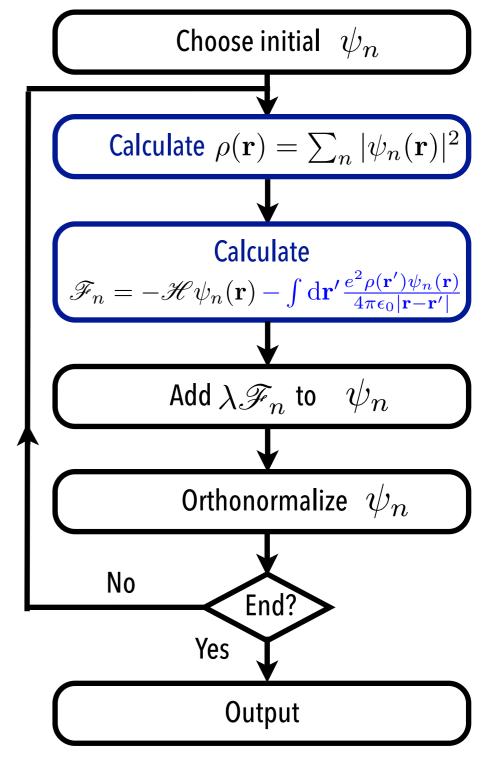
We've the new term –  $v_{ele}(\mathbf{r})\psi_n(\mathbf{r})$  in the force.

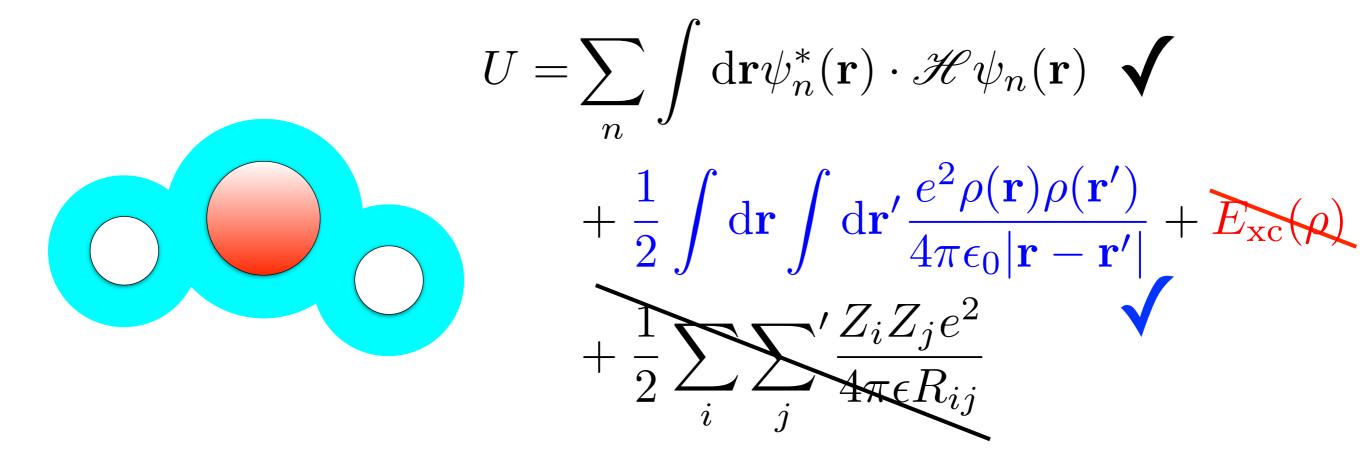
We need to change the algorithm accordingly.



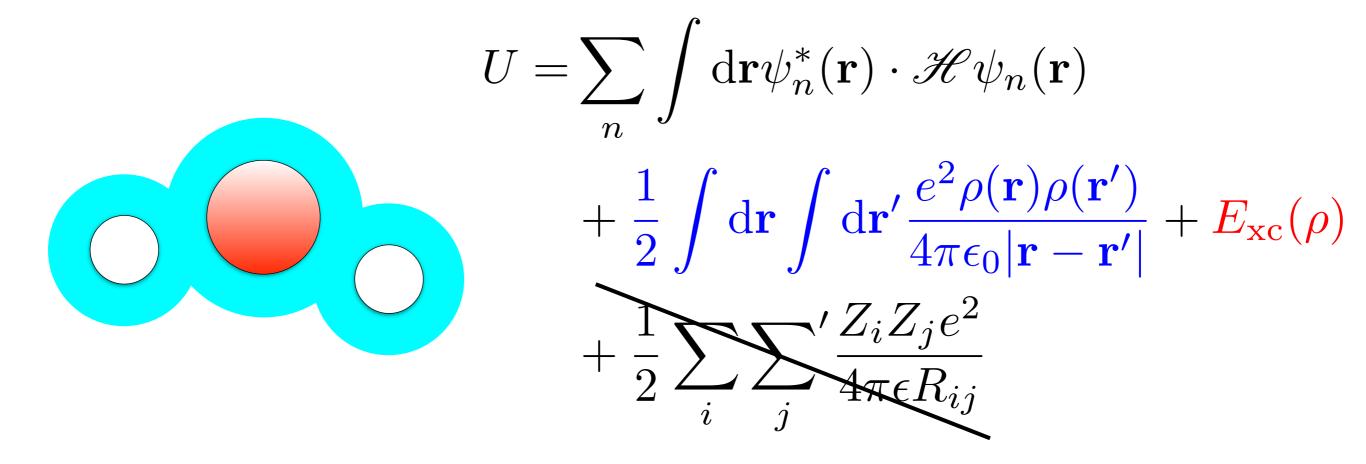
▶ **Answer.** We need to calculate the density  $\rho$  and the electrostatic potential  $\nu_{\text{ele}}$ .

$$\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r}) - \int d\mathbf{r}' \frac{e^2 \rho(\mathbf{r}')\psi_n(\mathbf{r})}{4\pi\epsilon_0 |\mathbf{r} - \mathbf{r}'|}$$





▶ So, we also know how to handle the second term.



- ▶ So, we also know how to handle the second term.
- Now, let's study the third term.

#### Hohenberg-Kohn theorem

#### Property | Hohenberg-Kohn theorem

There exists a one-to-one correspondence between the ground-state wave function  $\Psi$  and charge density  $\rho$  of a (nondegenerate) system of N electrons in a potential v.

In other words, it is impossible for two ground-state wave functions  $\Psi_1$  and  $\Psi_2$  corresponding to two potentials  $v_1$  and  $v_2$  to have the same density  $\rho_1 = \rho_2$ .



Pierre Hohenberg (1934-2017)



Walter Kohn (1923-2016)

### Hohenberg-Kohn theorem (cont'd)

#### Property | Hohenberg-Kohn theorem

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In other words, it is impossible for two ground-state wave functions  $\Psi_1$  and  $\Psi_2$  corresponding to two potentials  $v_1$  and  $v_2$  to have the same density  $\rho_1 = \rho_2$ .

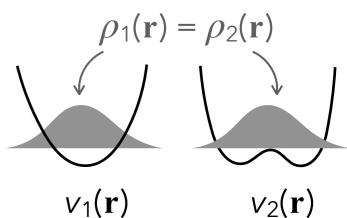


Pierre Hohenberg (1934-2017)



Walter Kohn (1923-2016)

**Proof (optional).** To prove this theorem, take two potentials  $v_1$  and  $v_2$  and write the energy  $E_i[\Psi_j]$  of  $\Psi_j$  in  $v_i$  as  $E_i[\Psi_j] = \int \Psi_j * (T+W)\Psi_j + \int v_i \rho_j$ , where  $T = -\sum_i (\hbar^2/2m)\nabla_i^2$  and  $W = \frac{1}{2}\sum_i \sum_{j\neq i} e^2/(4\pi\epsilon_0 |\mathbf{r}_i - \mathbf{r}_j|)$ . Show that if the groundstate wave functions  $\Psi_1$  and  $\Psi_2$  are not degenerate, the equality  $\rho_1 = \rho_2$  gives a contradiction.



### Hohenberg-Kohn theorem (cont'd)

#### Property | Hohenberg-Kohn theorem (cont'd)

Furthermore, the ground-state energy  $E[\rho]$  of N electrons in the potential v can be found by solving

$$\min_{\rho} F[\rho] + \int v(\mathbf{r})\rho(\mathbf{r})d\mathbf{r}$$
$$\int \rho(\mathbf{r})d\mathbf{r} = N$$

where  $F[\rho] = E[\rho] - \int v(\mathbf{r})\rho(\mathbf{r})d\mathbf{r} = \int \Psi^*[\rho](T+W)\Psi[\rho]$ 



Pierre Hohenberg (1934-2017)



Walter Kohn (1923-2016)

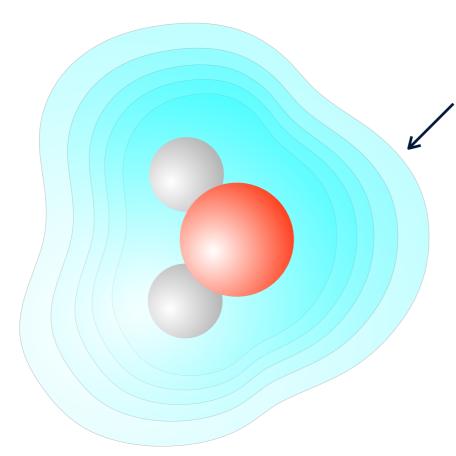
In other words, instead of solving for the ground-state wave function, one can solve for the ground-state electron density.

• F is called the universal functional. It is often written in terms of the kinetic, electrostatic, and exchange-correlation contributions

$$F[\rho] = T[\rho] + \int \int e^2 \rho(\mathbf{r}) \rho(\mathbf{r}') / (8\pi\epsilon_0 |\mathbf{r} - \mathbf{r}'|) d\mathbf{r} d\mathbf{r}' + E_x[\rho] + E_c[\rho]$$

### **Uniform electron gas**

Now, let's see how to approximate  $T[\rho]$ ,  $E_x[\rho]$ , and  $E_c[\rho]$  using a uniform electron gas as a model:

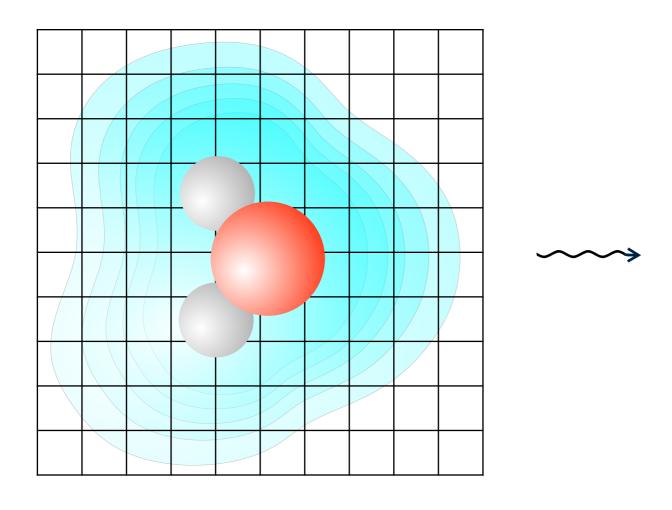


$$\rho(\mathbf{r}) = \sum_{\sigma=\uparrow,\downarrow} \sum_{n=1}^{N} |\phi_{n\sigma}(\mathbf{r})|^2$$

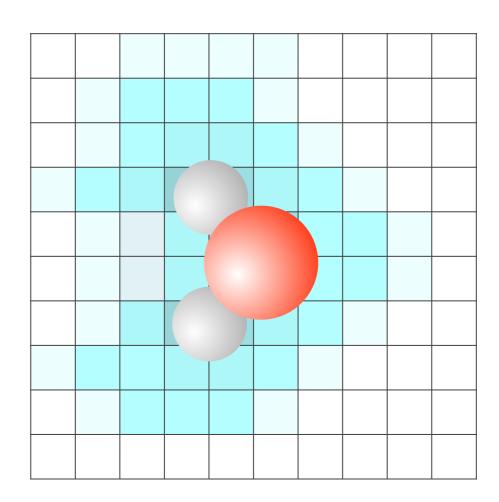
Start from the electronic density...

### Uniform electron gas (cont'd)

Now, let's see how to approximate  $T[\rho]$ ,  $E_x[\rho]$ , and  $E_c[\rho]$  using a uniform electron gas as a model:



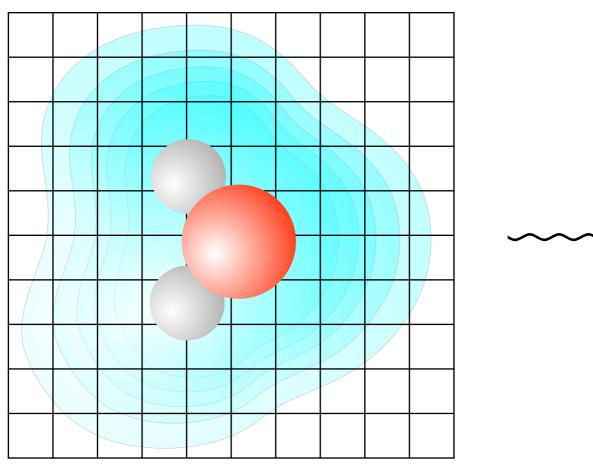
... Divide space into small volumes



... In each of these small volumes, approximate the density as uniform

### Uniform electron gas (cont'd)

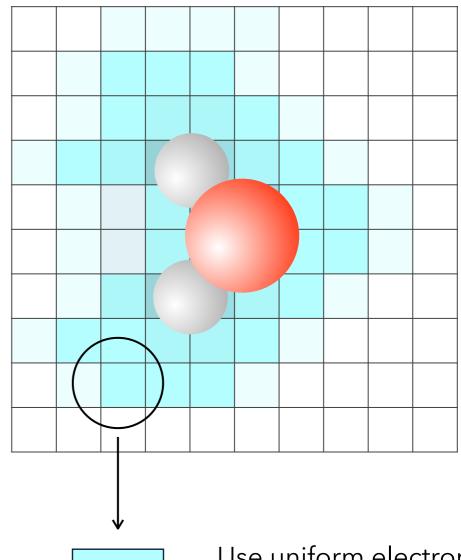
Now, let's see how to approximate  $T[\rho]$ ,  $E_x[\rho]$ , and  $E_c[\rho]$  using a uniform electron gas as a model:



Kinetic energy per volume in uniform gas of density  $\rho$ 

$$dT = t(\rho(\mathbf{r})) d\mathbf{r} \qquad \Longleftrightarrow \qquad$$

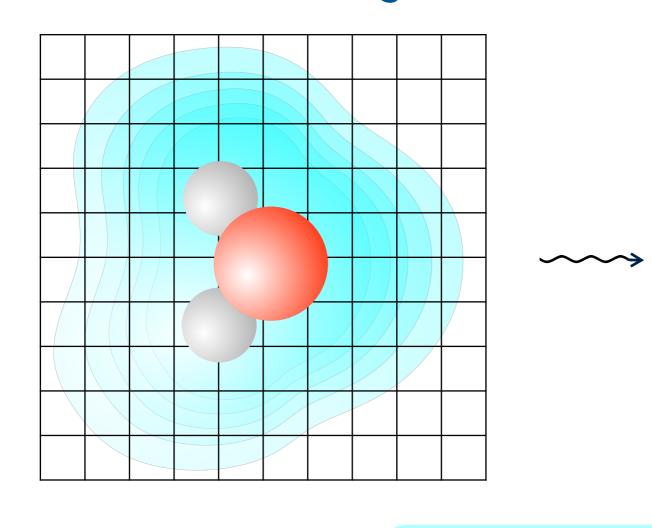
Kinetic energy contribution from the small volume

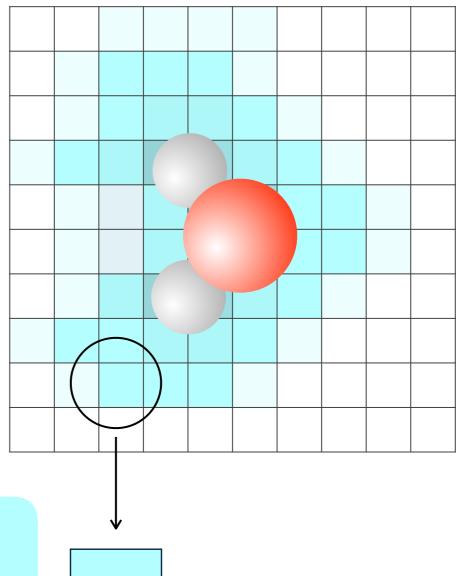


... Use uniform electron gas approximation to estimate  $T[\rho]$  in small volume

### Uniform electron gas (cont'd)

Now, let's see how to approximate  $T[\rho]$ ,  $E_x[\rho]$ , and  $E_c[\rho]$  using a uniform electron gas as a model:





... And the same procedure can be applied to  $E_x[\rho]$  and  $E_c[\rho]$ 

$$dT = t(\rho(\mathbf{r})) d\mathbf{r}$$

$$dE_x = e_x(\rho(\mathbf{r})) d\mathbf{r}$$

$$dE_c = e_c(\rho(\mathbf{r})) d\mathbf{r}$$

$$= \sum_{n} \int d\mathbf{r} \psi_{n}^{*}(\mathbf{r}) \cdot \mathcal{H} \psi_{n}(\mathbf{r})$$

$$+ \frac{1}{2} \int d\mathbf{r} \int d\mathbf{r}' \frac{e^{2} \rho(\mathbf{r}) \rho(\mathbf{r}')}{4\pi \epsilon_{0} |\mathbf{r} - \mathbf{r}'|} + E_{xc}(\rho)$$

- ▶ The red term changes the force.
- ▶ The new force is

$$\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r}) - \int d\mathbf{r}' \frac{e^2 \rho(\mathbf{r}')\psi_n(\mathbf{r})}{4\pi\epsilon_0 |\mathbf{r} - \mathbf{r}'|} - v_{xc}(\mathbf{r})\psi_n(\mathbf{r})$$

 $\triangleright v_{xc}$  is called the exchange-correlation potential.

# Exchange-correlation potential (cont'd)

Computing the Functional Derivative of Exc. Slater proposed the following

expression for the XC energy

$$E_{xc}(\rho) = -C_{\alpha} \int d\mathbf{r} \, \rho^{4/3}(\mathbf{r})$$



John C. Slater (1900–1976)

By applying a functional derivative (see previous slides), we get

$$E_{xc}(\rho + \delta \rho) = E_{xc}(\rho) + \int d\mathbf{r} \left[ -4C_{\alpha}/3 \rho^{1/3}(\mathbf{r}) \right] \delta \rho(\mathbf{r}) + ....$$

As a result,

Slater's Exchange-Correlation Potential 
$$v_{xc}(\mathbf{r}) = -4C_{\alpha}/3 \rho^{1/3}(\mathbf{r})$$

**Note.** This is just one example. There exist many other types of XC energies.

# **Exchange-correlation approximations**

▶ Two popular families of XC energies are LDA and GGA:

▶ Definition | Local density approximation (LDA) and generalized-gradient approximation (GGA). LDA and GGA are two families of approximations to describe quantum electron interactions within DFT.

They can be written as:

$$E_{\rm xc}^{\rm LDA}(\rho) = \int \epsilon_{\rm xc}(\rho(\mathbf{r})) d\mathbf{r}$$

$$E_{\rm xc}^{\rm GGA}(\rho) = \int \epsilon_{\rm xc}(\rho(\mathbf{r}), \nabla \rho(\mathbf{r})) d\mathbf{r}$$

# Exchange-correlation approximations (cont'd)

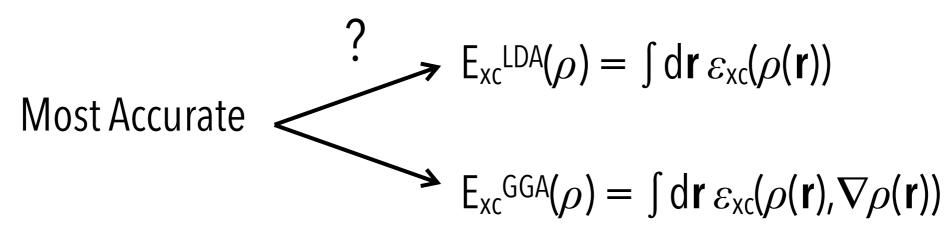
▶ Question 1. To which family does the Slater's exchange-correlation energy belong?

$$\frac{?}{\mathsf{E}_{\mathsf{xc}}\mathsf{LDA}(\rho)} = \int \mathsf{d}\mathbf{r} \, \varepsilon_{\mathsf{xc}}(\rho(\mathbf{r}))$$

$$\mathsf{E}_{\mathsf{xc}}\mathsf{GGA}(\rho) = \int \mathsf{d}\mathbf{r} \, \varepsilon_{\mathsf{xc}}(\rho(\mathbf{r}), \nabla \rho(\mathbf{r}))$$

$$\mathsf{E}_{\mathsf{xc}}\mathsf{GGA}(\rho) = \int \mathsf{d}\mathbf{r} \, \varepsilon_{\mathsf{xc}}(\rho(\mathbf{r}), \nabla \rho(\mathbf{r}))$$

▶ Question 2. Which one of the two families is more accurate in your opinion?



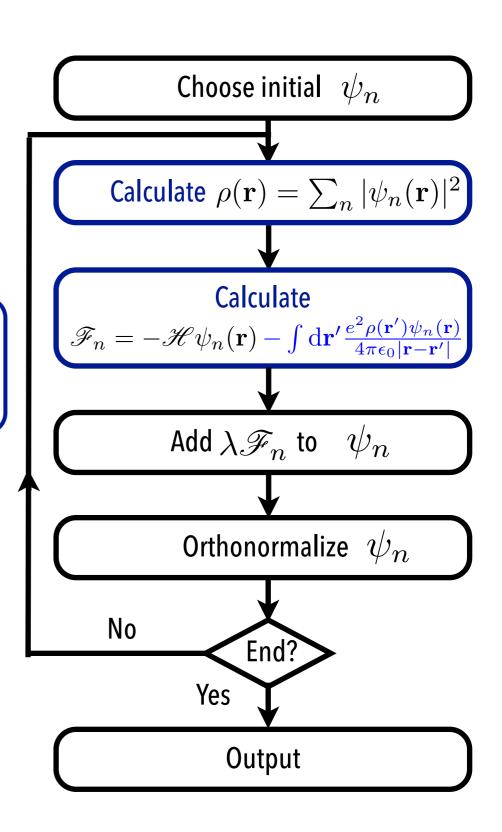
### **Minimization**

▶ Question. How would you change the algorithm?

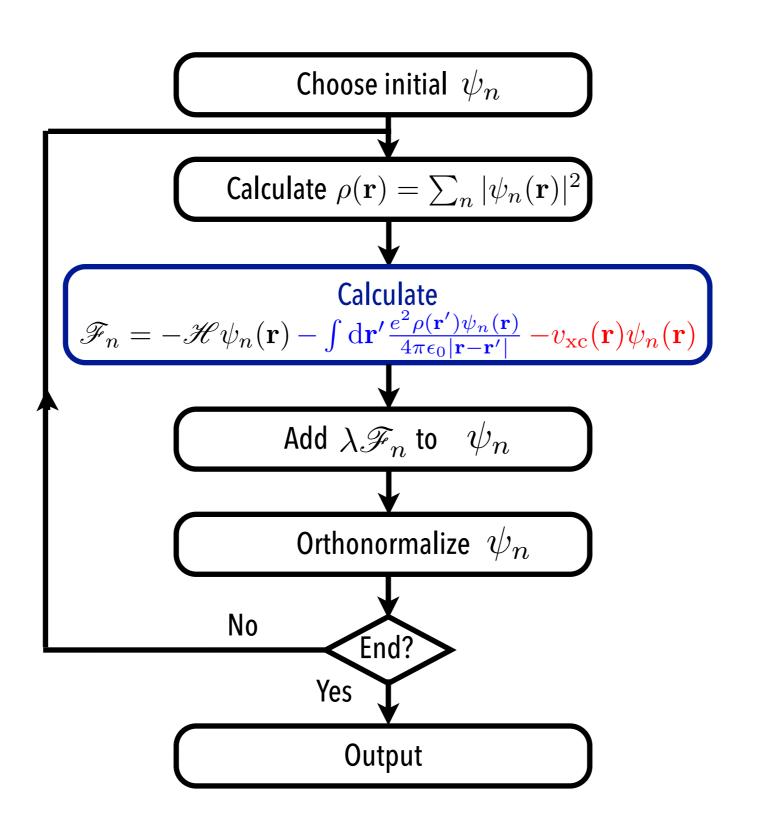
$$\mathscr{F}_n = -\mathscr{H}\psi_n(\mathbf{r}) - \int d\mathbf{r}' \frac{e^2 \rho(\mathbf{r}')\psi_n(\mathbf{r})}{4\pi\epsilon_0 |\mathbf{r} - \mathbf{r}'|} - v_{xc}(\mathbf{r})\psi_n(\mathbf{r})$$

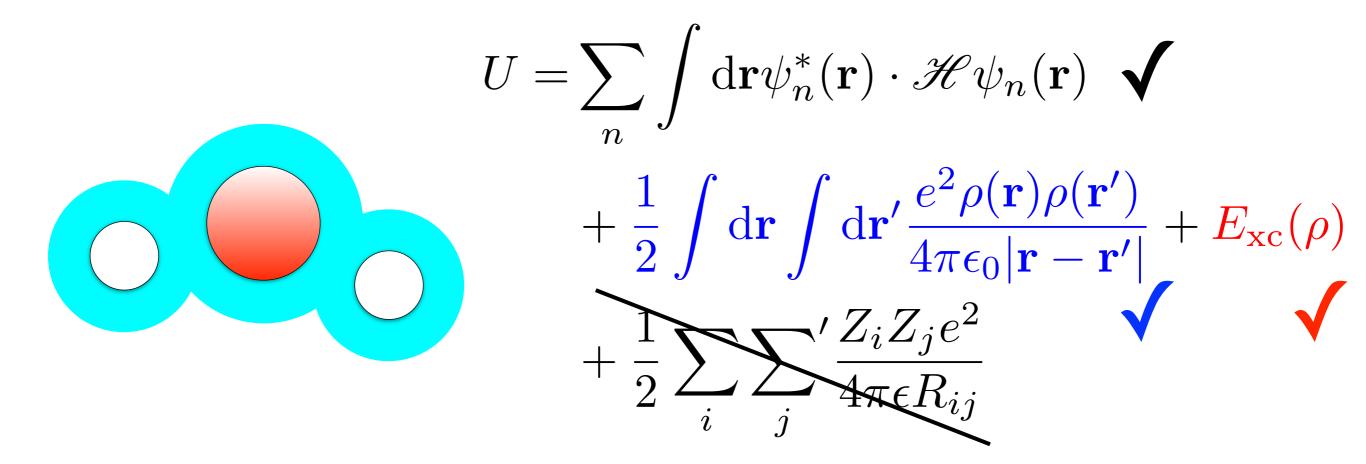
We've now 3 contributions:

- (1) Noninteracting Electronic Contribution;
- (2) Classical Interaction Contribution;
- (3) Quantum Interaction Contribution.

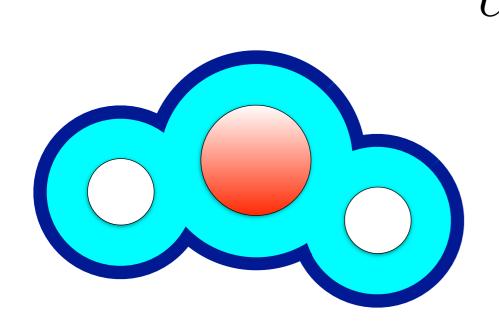


Answer. Add the new contribution to the force.

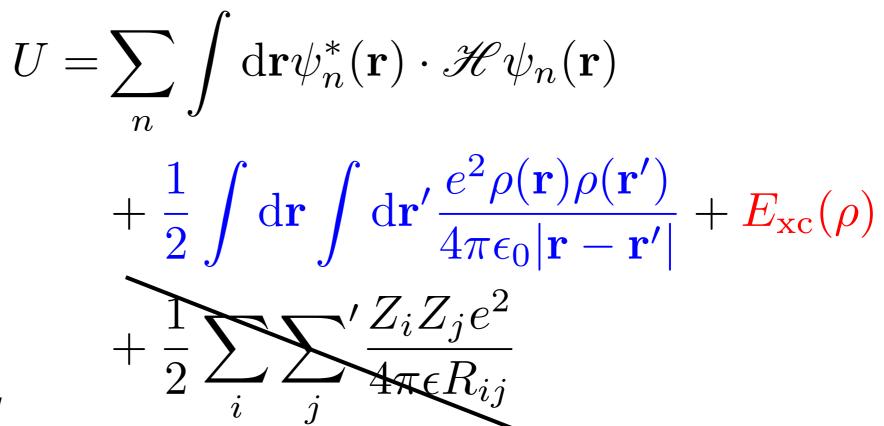




▶ We're done with the third term.



By minimizing these 3 terms (the only ones that depend on  $\psi_n$  and  $\rho$ ), we've found the electronic ground state.



- ▶ We're done with the third term.
- ▶ This means we've found the electronic ground state.