# Practice Problems - Solution

# Questions

#### Question 1

A large well-mixed tank used to allow settling of large particles from brine usually operates at steady-state with constant input (10 kg/min of 0.20 mass fraction salt in water) and output flowrates (10 kg/min, at the salt concentration in the tank). Due to operator error, a valve is opened and an additional 10 kg/min of pure water is introduced to the tank. Material balances on the total mass (M, in kg) and mass of salt (S, in kg) in the tank yield the following pair of equations:

$$\frac{dM}{dt} = 10$$

$$\frac{dS}{dt} = 10 \times 0.2 - 10 \left(\frac{S}{M}\right)$$

Where S/M represents the mass fraction of salt in the tank.

At t=0, the mass of brine M=1000 kg at 0.20 mass fraction, so  $S=(0.20/\mathrm{M})=200$  kg.

- Using solve\_ivp and the initial conditions provided, solve the system above for a time long enough that tank overflow occurs, estimated to be at M = 2000kg.
- Plot the total mass and the mass fraction of salt in the tank as a function of time.

#### Solution

This problem should be relatively straightforward – just a straight use of the builtin solvers. Remember that you need to solve the problem simultaneous, so the dependent variable is a vector.

Let x = [M, S] then

$$\frac{\mathrm{d}x}{\mathrm{d}t} = \begin{bmatrix} \frac{\mathrm{d}M}{\mathrm{d}t} \\ \frac{\mathrm{d}S}{\mathrm{d}t} \end{bmatrix} = \begin{bmatrix} 10 \\ 2 - \frac{10S}{M} \end{bmatrix}$$

With the IC x(0) = [M(0), S(0)] = [1000, 200].

An irreversible first-order reaction  $A \rightarrow B$  occurs in a catalyst of thickness L = 0.001 m; assume that the catalyst is wide in the other directions. At steady state, diffusion and reaction in the slab are described by,

$$D\frac{\mathrm{d}^2 C_A}{\mathrm{d}x^2} = kC_A$$

Where the diffusivity  $D = 1.2 \times 10^{-9} \ m^2/s$ , and the rate constant  $k = 0.001 \ s^{-1}$ .

The reactant at the surface of the catalyst (x = 0) is at concentration  $C_{A0} = 0.2$  mol/L. The catalyst is supported by an impermeable wall, so there is no flux of either species through the lower surface of the catalyst:

$$\frac{\mathrm{d}C_A}{\mathrm{d}x}\Big|_{x=L} = 0$$

- a) Use the shooting method (with solve\_ivp) to obtain a solution for the concentration profile  $C_A(x)$ .
- b) Use solve\_bvp to obtain the same solution.
- c) Present your results.

#### Solution

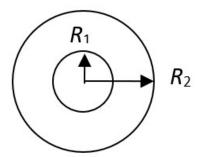
This problem is again setting up an ode, but here it is a boundary value problem. There are some choices to make when shooting: first, you can elect to solve in either direction (so your IVP starts either at x = 0 or at x = L), and, second, you can either do linear interpolation to get the unknown initial condition (since this is a linear ode) or you can rely on in-built solvers.

For solve\_bvp, the main thing is understanding how to set up the initial conditions and express the boundary conditions as a residual. First, express system as a pair of first-order equations.

Let 
$$u = \frac{\mathrm{d}C_A}{\mathrm{d}x}$$
 then  $\frac{\mathrm{d}u}{\mathrm{d}x} = \frac{\mathrm{d}^2 C_A}{\mathrm{d}x^2}$  and the ode becomes  $D\frac{\mathrm{d}u}{\mathrm{d}x} = kC_A$ .

Hence 
$$\frac{d}{dx}\begin{bmatrix} C_A \\ u \end{bmatrix} = \begin{bmatrix} u \\ \frac{k}{D}C_A \end{bmatrix}$$
 subject to  $C_A(x=0)=0.2$  and  $u(x=L)=0$ . Now we write the code!

You wish to solve for the steady-state temperature distribution in a long annular cylinder:



With  $R_1 = 5$  cm and  $R_2 = 10$  cm. In cylindrical coordinates, the governing equation is:

$$\frac{\mathrm{d}^2 T}{\mathrm{d}r^2} + \frac{1}{r} \frac{\mathrm{d}T}{\mathrm{d}r} = 0$$

The temperature of the inside wall is constant,  $T(R_1) = 120$ °C. The heat loss through the outside wall has been measured as  $\dot{Q} = 1$  kW per m of pipe length (along the axis), so

$$-k\frac{\mathrm{d}T}{\mathrm{d}r}\Big|_{r=R_2} = \frac{\dot{Q}}{2\pi R_2}$$

The pipe is steel, with  $k = 14 W m^{-1} K^{-1}$ .

- Write code to implement the finite difference approach for this problem; clearly state the finite difference approximation you are using as a comment within the code. Make n configurable (i.e., there is a line  $n = \dots$  near the top of the file; you can decide whether n represents internal nodes or all nodes as convenient).
- Note: You need to consider the boundary nodes separately to the interior nodes. You can use in-built methods to solve the resulting matrix equation.
- Make suitable plot(s) to illustrate your results. What is the temperature at  $R_2$  and the temperature gradient at  $R_1$ ?

#### Solution

See hand-written notes.

The governing equation for the steady-state temperature distribution in a pin fin (pictured below) is:

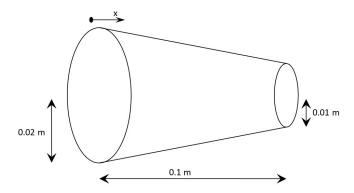
$$\frac{\mathrm{d}}{\mathrm{d}x} \left( [r(x)]^2 \frac{\mathrm{d}T}{\mathrm{d}x} \right) - \frac{2h}{k} r(x) (T - T_{\mathrm{air}}) = 0$$

Where the boundary conditions are given by:

$$T(0) = 100^{\circ}C$$
 
$$-k \frac{\mathrm{d}T}{\mathrm{d}x}\Big|_{x=L} = h\left(T(L) - T_{\mathrm{air}}\right)$$

The radius r of the fin varies with distance according to r(x) = 0.02–0.1x [in m] and the length L = 0.10 m. The air temperature  $T_{\text{air}} = 20^{\circ}C$  and the pin is steel, with  $h = 20Wm^{-2}K^{-1}$  and  $k = 14Wm^{-1}K^{-1}$ .

- Convert the system into a pair of first-order differential equations. Please write your system of first-order equations as a comment in your ode file, together with the boundary conditions you are using.
- Solve the system as a boundary-value problem with solve\_bvp. [If you prefer, you can solve the problem via shooting with solve\_ivp].
- Make suitable plot(s) to illustrate your results.



#### Solution

Again, a second-order byp. First, convert to a pair of first-order equations. There are two choices that should give the same result.

Let  $u = r^2 \frac{dT}{dx}$  then the ode becomes :

$$\frac{\mathrm{d}u}{\mathrm{d}x} - \frac{2h}{k}r(x)(T - T_{\mathrm{air}}) = 0$$

Giving a pair of equations,

$$\frac{\mathrm{d}}{\mathrm{d}x} \begin{bmatrix} T \\ u \end{bmatrix} = \begin{bmatrix} \frac{u}{r^2} \\ \frac{2h}{k} r \left( T - T_{\mathrm{air}} \right) \end{bmatrix}$$

subject to boundary conditions T(0) = 100 and  $-k \frac{dT}{dx}\Big|_{x=L} = h(T(L) - T_{air})$ .

We need to re-express the second condition in terms of **u** :

$$-k\frac{u(L)}{r(L)^2} = h\left(T(L) - T_{\text{air}}\right)$$

#### Alternative,

Note, we could have defined u differently. For clarity, I will call it v.

Let  $v = \frac{dT}{dx}$  then the ode becomes :

$$\frac{\mathrm{d}}{\mathrm{d}x} \left( r^2 v \right) - \frac{2h}{k} r(x) (T - T_{\mathrm{air}}) = 0$$

This definition forces us to do a little more work:

$$v\frac{\mathrm{d}}{\mathrm{d}x}(r^2) + r^2\frac{\mathrm{d}}{\mathrm{d}x}(v) - \frac{2h}{k}r(x)(T - T_{\mathrm{air}}) = 0$$
$$v2r\frac{\mathrm{d}r}{\mathrm{d}x} + r^2\frac{\mathrm{d}v}{\mathrm{d}x} - \frac{2h}{k}r(x)(T - T_{\mathrm{air}}) = 0$$
$$-0.2rv + r^2\frac{\mathrm{d}v}{\mathrm{d}x} - \frac{2h}{k}r(x)(T - T_{\mathrm{air}}) = 0$$

Hence,

$$\frac{\mathrm{d}}{\mathrm{d}x} \begin{bmatrix} T \\ v \end{bmatrix} = \begin{bmatrix} v \\ \frac{2h}{kr^2} r \left( T - T_{\mathrm{air}} \right) + \frac{0.2rv}{r^2} \end{bmatrix} = \begin{bmatrix} v \\ \frac{2h}{kr} \left( T - T_{\mathrm{air}} \right) + \frac{0.2v}{r} \end{bmatrix}$$

subject to boundary conditions T(0) = 100 and  $-kv(L) = h(T(L) - T_{air})$ . Now we write the code and either method should give the same result.

Carbon dioxide is being absorbed into the surface of an alkaline solution containing a catalyst. The concentration, C(t,x), of  $CO_2$  in solution is described by the following equation,

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2} - kC$$

Where t is time, x is the distance from the surface, D is the diffusivity of  $CO_2$  in the alkaline solution and k is a reaction rate constant.

Initially, there is no  $CO_2$  dissolved in the solution. At time 0, the partial pressure of  $CO_2$  in the gas at the solution's surface is suddenly increased to 1 atm, which gives a concentration at the surface of C(t,0) = 0.030 mol/L. Assume that the solution is deep, so that the dissolved  $CO_2$  concentration far from the surface is 0. Take  $D = 1.5 \times 10^{-9}$  m<sup>2</sup>/s and k = 35 s<sup>-1</sup>.

- a) State suitable boundary and initial conditions for this problem.
- b) Solve this PDE at steady state first. So set  $\frac{\partial C}{\partial t} = 0$  and solve the resulting ODE.
- c) Establish a set of equations to solve the PDE using the method of lines; include equations to implement the boundary conditions. Use second-order approximations wherever difference equations are required.
- d) Implement the solution; you should obtain a solution that approaches steady-state (start using a 0.01 s timeframe). The boundary layer is quite thin; try using a depth of  $10^{-5}$  m and node spacing of  $10^{-6}$  m. Does the steady state solution match the one you first obtained?
- e) Check that your solution is reasonable by halving the step size  $\Delta x$ .
- f) Make suitable plots to illustrate the solution.

# Solution

a) We want to solve for the concentration C(t,x)

IC: C(0,x) = 0 mol/L

BCs: C(t, 0) = 0.030 mol/L

$$C(t, L) = 0 \text{ mol/L or we could try } \frac{\partial C}{\partial x}\Big|_{x=L} = 0$$

where L is "deep enough" to be equivalent to infinity; here, it is suggested that  $10^{-5}$  m will do! In fact, we really expect both of these BCs to be satisfied and one check on our numerical solution should be to test the one we don't implement.

- b) Please see file. Note that this is a shooting method problem.
- c) We want to apply finite difference approximations in the x-direction only. This will leave us with a set of ode IVPs (one per node) in place of a PDE. See notes in PDF.
- d) This is a case of implementing the equation and then playing a little with the values to get what you expect. I solved it using solve\_ivp and then looked at the graph.

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If you increase the time frame to 0.1 s then it looks like steady-state has been reached. You need a bit more depth to get to 0 concentration, maybe 50  $\mu$ m rather than 10  $\mu$ m.

The second interesting plot is concentration vs. depth at a given time. The solution shows this changing with time.

### Question 6

Heat transfer, at unsteady state and in one-dimension, is governed by the equation:

$$\frac{\partial T}{\partial t} = \alpha \frac{\partial^2 T}{\partial x^2}$$

You wish to model the temperature T(t,x) at various depths in soil. Assume that the temperature at the soil surface T(t,0) varies over the year according to:

$$T(t,0) = 10 - 14\cos\left[\frac{2\pi}{365}(t - 37)\right]$$

Where t is the time in days and T is in Celsius. The temperature gradient is zero at large depth (assume 20 m is "large"),

$$\frac{\partial T}{\partial x} = 0$$
 at  $x = 20$  m

The thermal diffusivity of soil  $\alpha = 0.081 \text{ m}^2/\text{day}$ .

- Solve this problem using the method of lines. Use a node spacing of  $\Delta x = 2$  m, model the soil temperature to a depth of 20 m over a 4-year (1460 day) period. Assume that the soil is initially at 10°C.
- Make suitable plot(s) to illustrate your results.
- Why was it necessary to solve over more than one year? [write a brief comment in the file]

Hint: in developing your solution, you may find it convenient to first obtain a solution for a constant surface temperature T(t,0) of, say,  $20^{\circ}C$  (note: do not choose  $10^{\circ}C$ , since this will just give T(t,x) = constant!) before substituting the actual boundary condition above.

#### Solution

Please see code file.