

Physics 2610H: Assignment III

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March 6, 2025

Problem 1. A 50 eV electron is trapped between electrostatic walls 200 eV high. How far does its wave function extend beyond the walls?

Solution 1. The penetration depth, given in the text as equation 24, is

$$\delta = \frac{\hbar}{\sqrt{2m_e(U_0 - E)}} = \frac{\hbar}{\sqrt{2m_e(200 \text{ eV} - 50 \text{ eV}) \cdot 1.602 \times 10^{-19} \text{ J eV}^{-1}}} \approx 1.59 \times 10^{-11} \text{ m}$$

Problem 2. To a good approximation, the hydrogen chloride molecule, HCl, behaves vibrationally as a quantum harmonic oscillator of spring constant 480 N m^{-1} and with effective oscillating mass just that of the lighter atom, hydrogen. If it were in its ground vibrational state, what wavelength photon would be just right to bump this molecule up to its next-higher vibrational energy state?

Solution 2. The energy of a quantum harmonic oscillator at state n is given by $E = (\frac{1}{2} + n) \hbar \sqrt{\frac{\kappa}{m}}$ here $\kappa = 480 \text{ N m}^{-1}$. ΔE for this situation is then $(\frac{1}{2} + 1 - \frac{1}{2} - 0) \hbar \sqrt{\frac{\kappa}{m}} = \hbar \sqrt{\frac{\kappa}{m_e + m_p}}$ then the wavelength is given by $\frac{hc}{E} = \lambda \approx 3.52 \mu\text{m}$

Problem 3. Calculate the reflection probability for a 5 eV electron encountering a step in which the potential drops by 2 eV.

Solution 3. Reflection probability is given by

$$\left(\frac{\sqrt{E} - \sqrt{E - U_0}}{\sqrt{E} + \sqrt{E - U_0}} \right)^2$$

where E is the energy of the particle and U_0 the energy of the step. For a potential drop U_0 will be negative provided we set zero so that (in this case) U_0 is 2 eV below zero potential. So the probability is

$$\left(\frac{\sqrt{5 \text{ eV}} - \sqrt{5 \text{ eV} + 2 \text{ eV}}}{\sqrt{5 \text{ eV}} + \sqrt{5 \text{ eV} + 2 \text{ eV}}} \right)^2 \approx 0.704\%$$

Problem 4. A beam of particles of energy E and incident upon a potential step of $U_0 = \frac{5}{4}E$ is described by the wave function

$$\psi_{inc}(x) = e^{ikx}$$

- Determine completely the reflected wave and the wave inside the step by enforcing the required continuity conditions to obtain their (possibly complex) amplitudes.
- Verify by explicit calculation that the ratio of reflected probability density to the incident probability density is 1.

Solution 4.

- Here I'll skip a few steps because they are direct from the text and say that because of the condition requiring that the incident, reflected, and transmitted wavefunctions must be continuous and so must their first derivatives we obtain that before the step $\psi(x) = Ae^{ikx} + Be^{-ikx}$ and "inside" the step $\psi(x) = Ce^{-\alpha x}$ which gives us that $A + B = C$ and $ik(A - B) = -\alpha C$. Here A is the amplitude of the incident wave, B that of the reflected

wave, and C that of the transmitted wave. Now, by the requirement that total probability never exceeds 1 we assume that the incident wave is normalized so $A = 1$. So,

$$1 + B = C \implies ik(1 - B) = -\alpha(1 + B) \implies B = \frac{-\alpha - ik}{\alpha - ik}$$

which we then substitute the given values for α and k into,

$$\begin{aligned} B &= \frac{-\frac{\sqrt{2m(U_0 - E)}}{\hbar} - i\frac{\sqrt{2mE}}{\hbar}}{\frac{\sqrt{2m(U_0 - E)}}{\hbar} - i\frac{\sqrt{2mE}}{\hbar}} \\ &= \frac{-\sqrt{\frac{5}{4}E - E} - i\sqrt{E}}{\sqrt{\frac{5}{4}E - E} - i\sqrt{E}} \\ &= \frac{-\frac{1}{2} - i}{\frac{1}{2} - i} \\ &= \frac{3}{5} - \frac{4}{5}i \end{aligned}$$

And $C = 1 + B = \frac{8}{5} - \frac{4}{5}i$

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$$\begin{aligned} R &= \frac{AA^*}{BB^*} \\ &= \frac{1}{\left(\frac{3}{5} - \frac{4}{5}i\right)\left(\frac{3}{5} + \frac{4}{5}i\right)} \\ &= \frac{1}{\frac{9}{25} + \frac{16}{25}} = 1 \end{aligned}$$

Problem 5. What fraction of a beam of 50 eV electrons would get through a 200 V, 1 nm wide electrostatic barrier?

Solution 5. The transmission probability T for electrons of energy E for potential step of height U_0 and “width” L is given by

$$T = \frac{4(E/U_0)(1 - E/U_0)}{\sinh^2\left(\sqrt{2m(U_0 - E)} \cdot L/\hbar\right) + 4(E/U_0)(1 - E/U_0)}$$

here we have $U_0 = 200$ eV as an eV is defined as the change in potential energy experienced by an electron across 1 V. This yields $T \approx 9.55 \times 10^{-55}$ which means that 1 in 9.55×10^{55} electrons would make it through the barrier. Note that this answer is a bit far from what is given in the back of the book as I used the constants pre-programmed into my calculator to perform the calculation which I assume are more accurate than those used in the book’s answers section.