Physics 2610H: Assignment II

Jeremy Favro

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Problem 1. Write out the total wave function $\Psi(x,t)$ for an electron in the n=3 state of a 10 nm wide infinite well. Other than the symbols x and t, the function should include only numerical values.

Solution 1. We can split $\Psi(x,t) = \psi(x)\phi(t) = \psi(x)\exp\left(\frac{-iEt}{\hbar}\right)$ then for a particle in an infinite potential well $\psi(x) = \sqrt{\frac{2}{L}}\sin\left(\sqrt{\frac{2mE}{\hbar^2}}x\right)$ and $E = \frac{n^2\pi^2\hbar^2}{2mL^2}$ So,

$$\begin{split} \Psi(x,t) &= \sqrt{\frac{2}{L}} \sin \left(\sqrt{\frac{2m \frac{n^2 \pi^2 \hbar^2}{2mL^2}}{\hbar^2}} x \right) \exp \left(\frac{-i \frac{n^2 \pi^2 \hbar^2}{2mL^2} t}{\hbar} \right) \\ &= \sqrt{\frac{2}{L}} \sin \left(\frac{n \pi x}{L} \right) \exp \left(-i \frac{n^2 \pi^2 \hbar t}{2mL^2} \right) \\ &= \sqrt{\frac{2}{10 \times 10^{-9}}} \sin \left(\frac{3 \pi x}{10 \times 10^{-9}} \right) \exp \left(-i \frac{3^2 \pi^2 \hbar t}{2 \cdot 9.109 \times 10^{-31} \cdot (10 \times 10^{-9})^2} \right) \\ &\approx 1.414 \times 10^3 \, \mathrm{m}^{-0.5} \sin \left(9.425 \times 10^{17} \, \mathrm{m}^{-1} x \right) \exp \left(-i \cdot 5.142 \times 10^{13} \, \mathrm{s}^{-1} \right) \end{split}$$

Problem 2. An electron in the n = 4 state of a 5 nm wide infinite well makes a transition to the ground state, giving off energy in the form of a photon. What is the photon's wavelength?

Solution 2. By equation 1.11 from the formula sheet the energy of a particle in state n in an infinite well is

$$E = \frac{n^2 h^2}{8mL^2}$$

Here we're looking at ΔE of an electron between n=4 and n=1 so,

$$\Delta E = \left. \frac{n^2 h^2}{8mL^2} \right|_4^1 = \frac{15h^2}{8m_e L^2}$$

Then,
$$E = \frac{hc}{\lambda} \implies \lambda = hc \frac{8m_e(5 \text{ nm})^2}{15h^2} = \frac{8cm_e(5 \text{ nm})^2}{15h} \approx 5500 \text{ nm}$$

Problem 3. What is the probability that a particle in the first excited (n = 2) state of an infinite well would be found in the middle third of the well? How does this compare with the classical expectation? Why?

Solution 3.

$$= \int_{\frac{L}{3}}^{\frac{2L}{3}} \left[\sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right) \right]^2 dx = \frac{2}{L} \int_{\frac{L}{3}}^{\frac{2L}{3}} \sin^2\left(\frac{n\pi x}{L}\right) dx$$

$$= \frac{2}{L} \left[\frac{x}{2} - \frac{\sin\left(\frac{2n\pi x}{L}\right)}{\frac{4n\pi}{L}} \right]_{\frac{L}{3}}^{\frac{2L}{3}}$$
 (From formula sheet)
$$= \frac{2}{L} \left[\frac{x}{2} - \frac{L}{4n\pi} \sin\left(\frac{2n\pi x}{L}\right) \right]_{\frac{L}{3}}^{\frac{2L}{3}}$$

$$= \frac{2}{L} \left[\frac{\frac{2L}{3}}{2} - \frac{L}{4n\pi} \sin\left(\frac{2n\pi \frac{2L}{3}}{L}\right) - \frac{\frac{L}{3}}{2} + \frac{L}{4n\pi} \sin\left(\frac{2n\pi \frac{L}{3}}{L}\right) \right]$$

$$= \frac{2}{3} - \frac{2}{6} + \frac{2}{4n\pi} \sin\left(\frac{2n\pi}{3}\right) - \frac{2}{4n\pi} \sin\left(\frac{4n\pi}{3}\right)$$

$$= \frac{1}{3} + \frac{1}{2n\pi} \left[\sin\left(\frac{2n\pi}{3}\right) - \sin\left(\frac{4n\pi}{3}\right) \right] \approx 0.196$$

Classically I think we'd expect that the particle has an even probability density over the entire well and therefore the probability that it is inside one third of the well is $\frac{1}{3}$. The difference between these two arises from the fact that in the n=2 state we have an antinode in the middle third which means that the probability is lesser than the outer two thirds.