23.3 Coulomb's Law

Point charge -> of zero volume.

. The magnitude of the electric force between two point charges is given by:

$$\int_{e} = k_{e} \frac{19,119,1}{r^{2}}$$

r = distance between 9, and 92

Ke = Coulomb's constant ~ 9×109 N.m²

$$k_e = Coulomb's$$
 constant
 $k_e = \frac{1}{4\pi \epsilon_0}$, $\epsilon_o = permitivity of free space
 $\approx 8.85 \times 10^{12} \frac{C^2}{N \cdot m^2}$$

See Table 23.1 [electron us proton us neutron]

$$|9| = |9p| = 1.6 \times 10^{19} C$$

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$$m_p = 1.67 \times 10^{27} \text{ kg}, m_p = 9.11 \times 10^{31} \text{ kg}$$

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$$m_{\rho} = 1.67 \times 10^{-13}$$
 $E_{\chi} = 1.67 \times 10^{-13}$
 $F_{e} = ?, F_{o} = ?$
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$$F_{e} = \frac{19e|19p|}{r^{2}} = \cdots \Rightarrow F_{e} = \frac{8.2 \times 10^{8} \text{ N}}{10^{2} \text{ N}}$$

$$F = Ke$$

$$F_g = G \frac{m_e m_p}{V^2} = \dots \Rightarrow F_g = 3.6 \times 10^7 N$$

$$F_g = G \cdot 674 \times 10^{11} \frac{N \cdot m^2}{k_g^2}$$
Note that $F_e \approx 2 \times 10^3 9$!!

$$F_g = \frac{1000}{100} \times 10^{11} = \frac{1000}{100} \times 10^{11} = \frac{1000}{100} = \frac{1000$$

Force is a Grector !!

Fig. by 9, on 92

Fig. by 92 on 91

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Fig. coulomb's law

fig. anit vector directed from 91 to 92:

Note that
$$F_{21} = -F_{12}$$

Point charges!!

La For instance, if you have 4 poron charges ⇒ Net force on 9 =?

$$4 \vec{F_1} = \vec{F_{21}} + \vec{F_{31}} + \vec{F_{41}}$$

Question: If the force between Q, and Question: If the force between Q, and Q, and Q, has a magnitude of Fwhen the charges is r. The distance between the charges is r. Now if $r \rightarrow 3r \Rightarrow F_{rew}$?

$$F_{\text{New}} = Ke \frac{Q_1 Q_2}{r^2} = Ke \frac{Q_1 Q_2}{(3r)^3} = Ke \frac{Q_1 Q_2}{r^2} = F_{\text{new}} = \frac{F}{9}$$

 $\frac{QQ}{Q} = \frac{23.3}{Page} = \frac{696}{96} = \frac{9}{96} = \frac{42 MC}{96} = \frac{46 MC}{96}$

Answer is (b)
$$F_{AB} = -F_{BA}$$