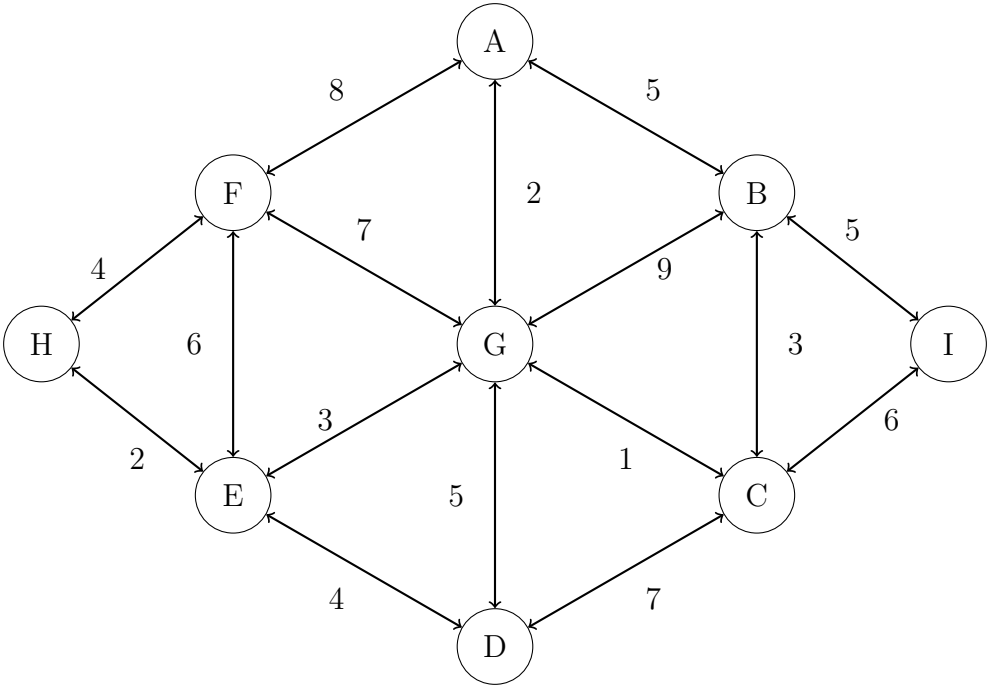
A
B
C
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6
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2
1
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7

$$\frac{\mathrm{d}}{\mathrm{d}t}\left[c\left(\mathrm{O_3}\right)\right] = -(2.3 \times 10^{-11}) \cdot c\left(\mathrm{CFC}\right) \cdot c\left(\mathrm{UV}\right)$$

where $k$ is reaction rate constant.

$$c\left(\mathrm{O_3}\right) \propto \frac{1}{c\left(UV\right)}$$

$$\frac{dP}{dt} = k \cdot P \cdot \frac{G}{Q}$$

$$\frac{\mathrm{d}}{\mathrm{d}t}G = a \cdot G \cdot \left(1 - \frac{P}{K}\right)$$

where $k$ and $a$ represent proportional sensitivity.