

Assignment-1

Year: 2017, Semester: Fall, Course: PHY-105, Title: Physics

1. Calculate the mean free path and the collision frequency for air molecules at 0°C and 1 atm pressure. Given the effective diameter of air molecule = 2\AA , the r.m.s speed of air molecules at N.T.P is about 1×10^5 cm/sec and the number of air molecules per c.c. = 3×10^{19} .
2. Estimate the size of a helium atom, assuming its mean free path at N.T.P to be 28.5×10^{-6} cm. Given that the density of helium at N.T.P is 0.178 g/liter and the mass of the helium atom is 6×10^{-24} g.
3. Calculate the r.m.s speed of hydrogen and air at 0°C and atmospheric pressure. Given at N.T.P. $\rho_H = 0.000089$ gm/cm³ and $\rho_{air} = 0.001293$ gm/cm³.
4. At what temperature, pressure remaining constant, will the r.m.s. velocity of a gas molecule be half its value at 0°C?
5. Calculate the mean translational kinetic energy per molecule of a gas at 727°C, given $R = 8.32$ J/mol-k.
6. Calculate the r.m.s velocity of a molecule of mercury vapour ($M_{Hg} = 221$ gm) at 300 K.
7. A particle of mass 6.2×10^{-27} kg is suspended in a liquid at 27°C and the r.m.s speed is found to be 1.4×10^{-2} m/s. If $R = 8.31$ J/K-mol, find the Avogadro's number from the equipartition theorem.
8. Calculate the values of the molar heat capacities C_v and C_p of a gas, if the ratio of heat capacities is 1.33. What is the degrees of freedom/atomicity of the gas? Given $R = 8.31$ J/k-mol.
9. The Brownian motion of some suspended particles inside a cylinder of radius is 30 cm is moving randomly at 20°C. The time needed to diffuse the particles is 2 sec. If viscosity co-efficient of the suspended fluid is 0.01 dynes-sec/cm² and $R = 8.31 \times 10^7$ erg/k-mol, calculate the diffusive coefficient using Einstein's Brownian motion theory. [Hint: Here, first calculate $\Delta^2 = \frac{RT}{N} \left(\frac{1}{3\pi\eta a} \right) \tau$ where, η = co-efficient of viscosity. Then find diffusive coefficient.]
10. Calculate the diffusivity of the Brownian gas following 5 sec requires to diffuse the gas particles along a cylinder of length 35 cm.