NMRlipids IV: Headgroup & glycerol backbone structures, and cation binding in bilayers with PE and PG lipids

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Primarily measured but also simulated NMR order parameters will be collected also for other than phophatidylcholine (these are discussed in NMRlipids I) headgroup. The information will be used to understand structural differences between different lipid molecules in bilayers.

INTRODUCTION

In NMRlipids I and II project we were looking for a MD model which would correctly reproduce headgroup and glycerol backbone structures and cation binding for PC lipid bilayers [1, 2]. Here we extend the same goal for other than PC lipids. Currently the focus is on PE, PG and PS bilayers and their mixtures with PC. Experimental data with different amounts of added salt is now collected and presented in this manuscript.

Experimental order parameters for different lipid head-groups are collected in Fig. 1.

Based on superficial reading, the conclusions in the literature are roughly

- 1) glycerol backbone structures are largely similar irrespectively of the headroup [3],
- 2) glycerol backbone and headgroup structure and behaviour are similar in model membranes and in bacteria [3–5],
- 3) headgroup structures are similar in PC, PE and PG lipids, while headgroup is more rigid in PS lipids [6, 7].

Extensive discussion about structural details of PE, PG or PS headgroups do not exists (as far as I know), In contrast to PC lipids (see [1] and references therein).

Several simulations containing PE, PG and PS lipids have been published [?], 1.List should be completed however, glycerol backbone and headgroup order parameters are not compared to the experiments (based on superficial reading of literature).

METHODS

Experimental C-H bond order parameters

The headgroup and glycerol backbone C–H bond order parameter magnitudes and signs of POPE and POPG were determined by measuring the chemical-shift resolved dipolar splittings with a R-type Proton Detected Local Field (R-PDLF) experiment [8] and S-DROSS experiments [9] using natural abundance ¹³C solid state NMR spectroscopy as described previously [10–12]. 2.The rest of the details to be written. I am not sure how much we need to repeat the NMRlipidsIVps paper.

Molecular dynamics simulations

Molecular dynamics simulation data were collected using the Open Collaboration method [1], with the NMR-lipids Project blog (nmrlipids.blogspot.fi) and GitHub repository (github.com/NMRlipids/NMRlipidsIVotherHGs) as the communication platforms. The simulated systems are listed in Tables I (pure PE and PG bilayers without additional ions). and III (mixtures and systems with additional ions). Further simulation details are given in the SI, and the simulation data are indexed in a searchable database available at www.nmrlipids.fi, and in the NMRlipids/MATCH repository (github.com/NMRlipids/MATCH).

The C–H bond order parameters were calculated directly from the carbon and hydrogen positions using the definition

$$S_{\rm CH} = \frac{1}{2} \langle 3\cos^2 \theta - 1 \rangle,\tag{1}$$

where θ is the angle between the C–H bond and the membrane normal (taken to align with z, with bilayer periodicity in the xy-plane). Angular brackets denote average over all sampled configurations. The order parameters were first calculated averaging over time separately for each lipid in the system. The average and the standard error of the mean were then calculated over different lipids. Python programs that use the MDAnalysis library [13, 14] used for all atom simulations is available in Ref. 15 (scripts/calcOrderParameters.py). For united atom simulations, the trajectories with hydrogens having ideal geometry were constructed first using either buildH program [?] or (scratch/opAAUA_prod.py) in Ref. 15, and the order parameters were then calculated from these trajectories. This approach has been tested against trajectories with explicit hydrogens and the deviations in order parameters are small [??]. The ion number density profiles were calculated using the gmx density tool of the Gromacs sofware package [16].

TABLE I: List of MD simulations with PE lipids.

lipid/counter-ions	force field for lipids / ions	NaCl (M)	$^{a}\mathrm{N}_{\mathrm{l}}$	$^b\mathrm{N_w}$	$^c\mathrm{N_c}$	$^{d}T(K)$	$^{e}t_{\mathrm{sim}}(\mathrm{ns})$	ft _{anal} (ns) gfiles
POPE	CHARMM36 [?]	0	144	5760	0	310	500	400 [17]
POPE	CHARMM36 [?]	0	500	25000	0	310	500	100 [18]
POPE	CHARMM36 [?]	0.11	500	25000	50	310	500	100 [19]
POPE	CHARMM36ua [?]	0	336	15254	0	310	2×200	2×100 [20]
DPPE	Slipids [21]	0	288	9386	0	336	200	100 [22]
POPE	Slipids [21?]	0	336	?	0	310	2×200	2×100 [23]
POPE	Slipids [?]	0	500	25000	0	310	500	100 [24]
POPE	Slipids [?] 3.	0.11	500	25000	50	310	500	100 [25]
DPPE	GROMOS-CKP [?]	0	128	3655	0	342	2×500	2×400 [26]
POPE	GROMOS-CKP[?]	0	128	3552	0	313	2×500	2×400 [27]
POPE	GROMOS-CKP [?]	0	500	25000	0	310	500	100 [28]
DOPE	GROMOS-CKP [?]	0	128	4789	0	271	2×500	2×400 [29]
POPE	GROMOS 43A1-S3 [?]	0	128	3552	0	313	2×200	2×100 [30]
POPE	OPLS-UA vdW on H [?]	0	128	3328	0	303	2×200	2×100 [31]
POPE	OPLS-UA[?]	0	128	3328	0	303	2×200	2×100 [32]
POPE	Berger-Vries [?]	0	128	3552	0	303	2×200	2×100 [33]
POPE	Berger-largeH [?]	0	128	3552	0	303	2×200	2×100 [34]
DOPE	Berger-Vries [?]	0	128	4789	0	271	2×200	2×100 [35]
DOPE	Berger-largeH [?]	0	128	4789	0	271	2×300	2×100 [36]
POPE	LIPID17 [?]	0	500	25000	50	310	500	100 [?]
POPE	LIPID17 [?]	0.11	500	25000	50	310	500	100 [37]

^aNumber of lipid molecules with largest mole fraction ^bNumber of water molecules ^cNumber of additional cations

TABLE II: List of MD simulations with PG lipids.

lipid/counter-ions	force field for lipids / ions	NaCl (M)	$^{a}N_{l}$	$^b\mathrm{N_w}$	$^c\mathrm{N}_\mathrm{c}$	$^{d}T\left(K\right)$	$^{e}t_{\mathrm{sim}}(\mathrm{ns})^{f}$	t _{anal} (ns) ^g files
POPG/K ⁺	CHARMM36 [?] 4.	0	118	4110	0	298	100	100 [38]
POPG	CHARMM36 [?]	0.11	500	25000	49	310	500	100 [39]
POPG	CHARMM36 [?]	0	500	25000	0	310	500	100 [40]
POPG/Na ⁺	Slipids [41]	0	288	10664	0	298	250	100 [42]
DPPG/Na ⁺	Slipids [41]	0	288	11232	0	314	200	100 [43]
DPPG/Na+	Slipids [41]	0	288	11232	0	298	400	100 [44]
POPG	Slipids [?] 5.	0	500	25000	0	310	500	100 [45]
POPG	Slipids [?] 6.	0.11	500	25000	49	310	500	100 [46]
POPG	LIPID17 [?]	0	500	25000	0	310	500	100 [47]
POPG	LIPID17 [?]	0.11	500	25000	49	310	500	100 [48]
POPG	GROMOS-CKP [?]	0	500	25000	0	310	500	100 [49]
POPG	GROMOS-CKP[?]	0.11	500	25000	49	310	500	100 [50]

^aNumber of lipid molecules with largest mole fraction

^dSimulation temperature ^eTotal simulation time

fTime used for analysis gReference for simulation files

bNumber of water molecules Number of additional cations dSimulation temperature Total simulation time

fTime used for analysis gReference for simulation files

TABLE III: List of MD simulations with PE and PG lipids mixed with PC.

lipid/counter-ions	force field for lipids / ions	NaCl (M)	$CaCl_{2}\left(M\right)$	$^{\it a}N_{\it l}$	$^b\mathrm{N_w}$	$^c\mathrm{N_c}$	$^{d}T(K)$	$e_{t_{sim}(ns)}$	$f_{\rm tanal}$ (ns)	g files
POPC	CHARMM36 [?]	0.11	0	500	25000	48	310	500	100	[51]
POPC:POPG (7:3)	CHARMM36 [?]	0.11	0	350	?	?	310	500	100	[52]
POPC:POPG (1:1)/K ⁺	CHARMM36 [?]	0	0	?	?	?	?	?	?	[?] 7.
POPC:POPG (1:1)/K ⁺	CHARMM36 [?]	0	0.15 8.	?	?	?	?	?	?	[?] 9.
POPC:POPG (1:1)/K ⁺	CHARMM36 [?]	0	1.0 10.	?	?	?	?	?	?	[?] 11.
POPC:POPG (4:1)/K+	CHARMM36 [?]	0	0	?	?	?	?	?	?	[?] 12.
POPC:POPG (4:1)/K ⁺	CHARMM36 [?]	0	0.15 13.	?	?	?	?	?	?	[?] 14.
POPC:POPG (4:1)/K+	CHARMM36 [?]	0	1.0 15.	?	?	?	?	?	?	[?] 16.
POPC	CHARMM36 [?]	0	0	256	8704	0	300	300	250	[53]
POPC:POPE (1:1)	CHARMM36 [?]	0	0	128	8704	0	300	300	250	[54]
POPC	Slipid [?]	0.11	0	500	25000	48	310	500	100	[55]
POPC:POPG (7:3)	Slipid [?]	?	0	?	?	?	310	500	100	[?] 17.
POPC	Berger [?] 18.	0	0	256	10240	0	300	300	200	[56]
POPC:POPE (1:1)	Berger [?] 19.	0	0	128	11008	0	300	300	200	[57]
POPC:DOPE (1:1)	Berger [?] 20.	0	0	128	10240	0	300	300	200	[58]
DOPC	Berger [?] 21.	0	0	256	11008	0	300	300	200	[59]
DOPC:DOPE (1:1)	Berger [?] 22.	0	0	128	11008	0	300	300	200	[60]

^aNumber of lipid molecules with largest mole fraction ^bNumber of water molecules ^cNumber of additional cations ^dSimulation temperature ^eTotal simulation time ^fTime used for analysis ^gReference for simulation files

RESULTS AND DISCUSSION

Headgroup and glycerol backbone order parameters of POPE and POPG from 13 C NMR

The glycerol backbone and α -carbon peaks in INEPT spectra of POPE were assigned based on previously measured POPC spectra (Fig. S1) [10]. The β -carbon peak was assigned based on 13 C chemical shift table for amines available at https://www.chem.wisc.edu/areas/reich/nmr/c13-data/cdata.htm. The order parameters for the glycerol backbone and headgroup C-H bonds were determined from 2D-RPDLF and S-DROSS experiments (Fig. S1), as described previously [12]. The POPE experiments were recorded at 310 K, where the bilayer is in liquid disordered phase [?]. 23.Details to be checked by Tiago.

24. Figure and discussion about POPG experiments to be addded.

The headgroup and glycerol backbone order parameters of PE lipids are similar with different acyl chains and also close the values for POPC, althought PE gives systematically slightly more positive values (Fig. 1). These could be explained with slightly larger temperature in PE measurements, except for the α -carbon with the positive sign, for which the more positive value is farther away from zero. For PG lipids, the glycerol backbone order parameters are more positive than for other lipids. The headgroup α -carbon gives value close to PE, while the value of β -carbon is distinct from other lipid being only one which has positive sign, suggesting distinct conformation of PG lipids in this region. This was not observed in previous 2 H NMR study, where sign was not measured and β -carbon order parameter was apparently similar to the value for PE and PC results.

In conclusion, the results suggests that the glycerol backbone conformations in all lipids are relatively similar. Also, the headgroup conformations are similar for PC and PE lipids, while PS and PG are singnificantly different. For PS lipids, the differences are discussed previously [?].

Headgroup and glycerol backbone of POPE and POPG in MD simulations

The glycerol backbone order parameters in Slipid simulations are off from experiments for both PE and PG lipids (Figs. 2 and 3), as already reported previously for PC lipids [1] and PS lipids [12]. The Slipid simulations were able to capture the essential differences between PC and PS lipid headgroups [12], but this is not the case for PE and PG lipids.

26.Berger results are not here yet, but we should mention the ring like structures pointed out by T. Piggot in the blog: The poor performance of headgroup order parameters in Berger model can be probably explained by ring like structures seen in Fig. 6 in Ref. 64. These ring like structures are a widespread feature of typical Berger based lipid force fields containing explicit hydrogen atoms in the head group [65–67].

CHARMM36 simulations seems to capture all the essential differences between headgroup order parameters of PC

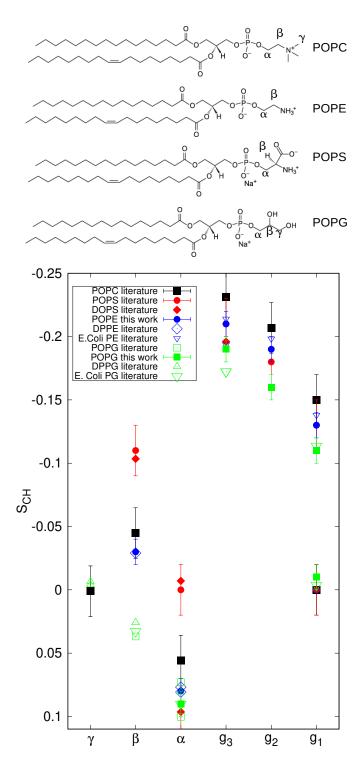


FIG. 1: (top) Chemical structure of different lipids (bottom) Headgroup and glycerol backbone order parameters measured from lipids with different headgroups in lamellar liquid disordered phase. The values and signs for POPE (310 K), POPG (298 K). POPS (298 K) [12] and POPC (300 K) [10, 11] are measured using ¹³C NMR. The literature values for DOPS with 0.1M of NaCl (303 K) [61], POPG with 10nM PIPES (298 K) [62], DPPG with 10nM PIPES and 100mM NaCl (314 K) [6], DPPE (341 K) [63], E.coliPE and E.coliPG (310 K) [3] are measured using ²H NMR. The signs from ¹³C NMR are used also for the literature values.

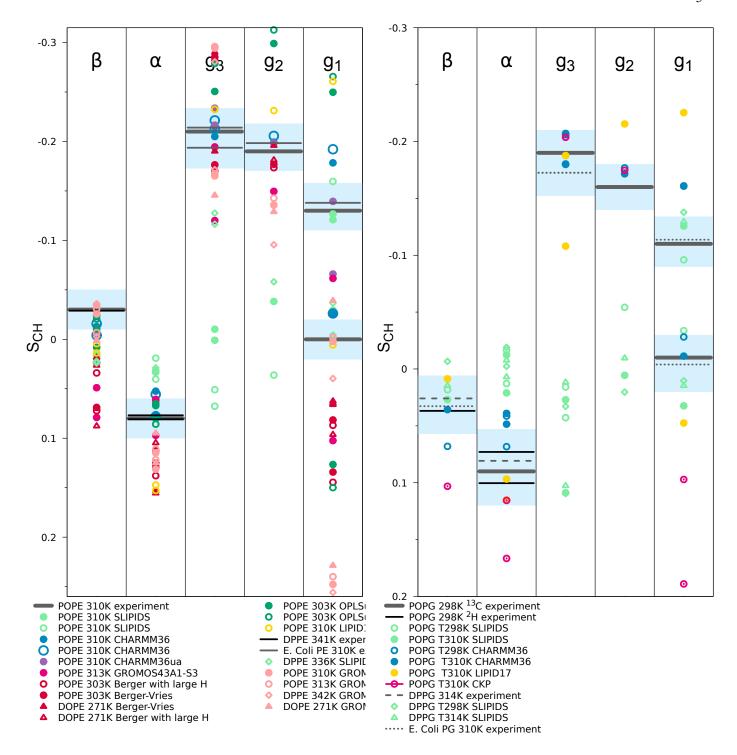


FIG. 2: The headgroup and glycerol backbone order parameters of PE lipids from experiments (POPE and signs this work, DPPE from Ref. 63 and E.coliPE from Ref. 3) and simulations with different force fields.

 $25. This should be clarified as in NMR lipids I and error bars should be added. \\ Probably larger error bars for united atom models based on the report by Fuchs et$

FIG. 3: The headgroup and glycerol backbone order parameters of PG lipids from experiments (POPG and signs from this work and from Ref. 62, DPPG with 100mM NaCl from Ref. 6,and E.Coli PG results from Ref. 3). and simulations with different force fields.

PE, PG (Figs. 2 and 3), as previously observed also for PS lipids [12]. Therefore, the CHARMM36 simulations can be used to analyze the structural differences between headgroups (Fig. 4).

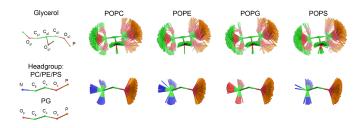


FIG. 4: Overlayed snapshots from CHARMM36 simulations of different lipids which give the best agreement with experiments.

Lipid headgroup interactions in PC:PE and PC:PG mixtures

According to the electrometer concept, the headgroup order parameters increase with the addition of negatively charged PG lipids, but are not affected by the addition of zwitterionic PE lipids (Fig. 5) [4, 68]. This is roughly observed also in simulations here, althouht the changes in PC headgroup with PE are slightly overestimated and changes with PG lipid are slightly underestimated. The underestimated increase of PC headgroup order parameters with the addition of negaticely charged lipids was previously interepreted to indicate overestimated counterion binding affinity, which overcompensates the effect of negative charge [12]. Thus, the counterion binding to PG lipid bilayers (Fig. 6) may be slightly overestimated also in simulations here.

28. This is text by P. Fuchs, copied from the blog. Area results in nm 2 , the error is <=0.003 nm 2

- pure POPCCHARMM36: 0.624Berger: 0.649- POPC/POPE 50:50

CHARMM36: POPC 0.609, POPE 0.557 Berger-hacked: POPC 0.637, POPE 0.632

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One can see that CHARMM 36 predicts a drop in the area on going from pure POPC to POPC/POPE 50:50. This means that POPC pack tightly to POPE. In contrast, the values for Berger are not that changed. The POPE value predicted by CHARMM 36 (in the mixture POPC/POPE 50:50) is much smaller than that predicted by Berger.

The experimental acyl chain order parameters for POPE [71] seem larger than reported for POPC [10], which supports the more condensed PE bilayer. In principle, this is beyond the scope of this work (lipid headgroups), but we can consider mentioning this.

The order parameter of headgroup β -carbon of PG slightly increases with the addition of PC [69], althought also smaller changes have been observed [62] (Fig. ??). The increase in PG headgroup order parameter is not observed in simulations, which may arise from overestimated counterion binding overcompensating the changes in electrostatic environment as suggested for PC headgroup above. However, the changes are quite small and not fully consistent in experiments.

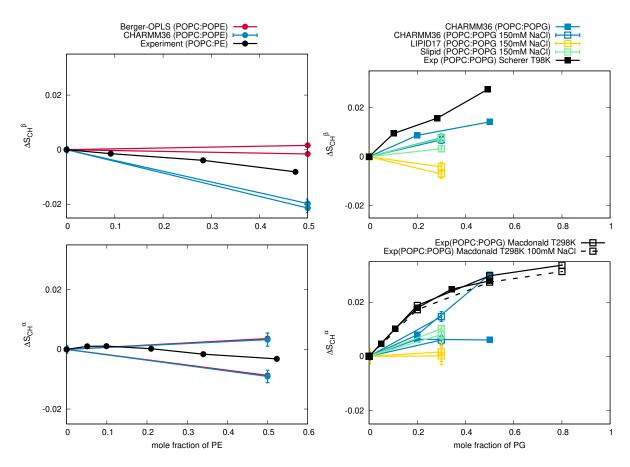
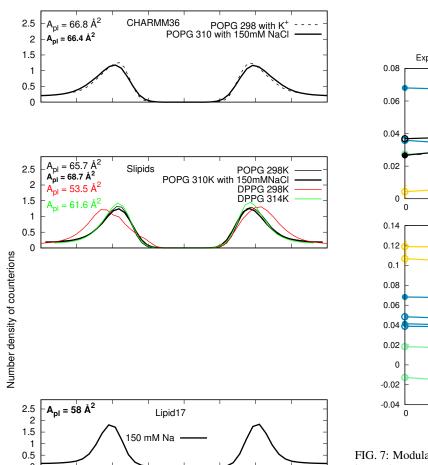


FIG. 5: Modulation of PC headgroup order parameters with increasing amount of PE (left) and PG (right) in bilayer from experiments [4, 69] and simulations with different force fields. Signs are determined as discussed in [1, 70].

 ${\bf 27. Simulation\ of\ CHARMM36\ at\ 298K\ should\ be\ maybe\ rerun\ with\ Gromacs\ 5.}$



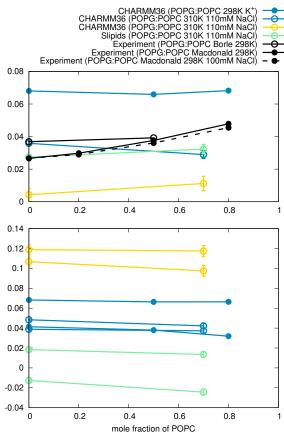


FIG. 7: Modulation of PG lipid headgroup order parameters with the increasing amount of PC in lipid bilayer from experiments [62, 69] and simulations with different force fields.

FIG. 6: Counterion densities from simulations with PG lipids.

Cation binding to PE and PG lipid bilayers

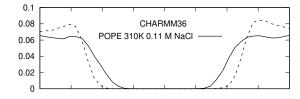
The headgroup order parameters of PC lipids can used to measure ion binding affinity to lipid bilayers, because their magnitude is linearly proportional to the amount of bound charge in bilayer according to the molecular electrometer concept [2, 68]. The molecular electrometer concept can be used also for bilayers containing PC lipids mixed with charged lipids [12, 62, 69, 72]. Based on the electrometer concept and other data is has been suggested that [5]

- "(i) Ca^{2+} binds to neutral lipids (phosphatidylcholine, phosphatidylethanolamine) and negatively charged lipids (phosphatidylglycerol) with approximately the same binding constant of $K = 10-20 \ M^{-1}$;
- (ii) the free Ca²⁺ concentration at the membrane interface is distinctly enhanced if the membrane carries a negative surface charge, either due to protein or to lipid;
- (iii) increased inter-facial Ca^{2+} also means increased amounts of bound Ca^{2+} at neutral and charged lipids;
- (iv) the actual binding step can be described by a Langmuir adsorption isotherm with a 1 lipid:1 Ca^{2+} stoichiometry, provided the interfacial concentration C_M , is used to describe the chemical binding equilibrium."

The electrometer concept has been very useful in evaluating ion binding affinity in simulations against experiments, because the headgroup order parameter changes as a function of ion concentration can be directly compared with experiments [2, 12, 73].

Sodium binding to PC lipid bilayers is significantly overestimated by most simulations models [2]. Also, sodium binding to PS lipids seems to be overestimated, although the presence of counterions complicate the comparison for negatively charged lipids [12]. Sodium binding to PE lipids is weaker than to PC lipids in previous work [2] in both CHARMM36 and Slipids simulations (Fig. 8). This cannot be evaluated using electrometer concept because experimental data with PE lipids is not available 29.0ther experimental data should be checked and discussed. The difference in Slipids may also be due to different ion model used 30.The used ion model here should be checked.

Calcium binding affinity to PC and PS lipid bilayers was not correctly described by any of the standard MD simulation forced fields [2, 12], while recently introduced force field with electronic continuum correction (ECC) performed better [73]. The decrease of α -carbon order parameter of PC lipids in PC:PG mixtures as a function of calcium concentration is close to experiments CHARMM36 simulations (Fig. 9), but the decrease of β -carbon order parameter seems to be overestimated. However, the β -carbon order parameter was not actually measured from these samples, but they are calculated from empirical relation $\Delta S_{\beta} = 0.43 \Delta S_{\alpha}$ [74]. The result is



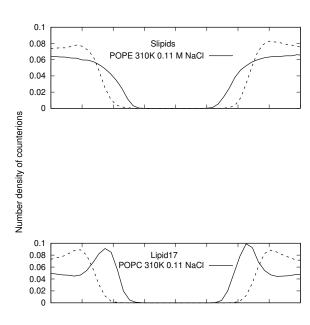


FIG. 8: Sodium (solid line) and choride ion density profiles along membrane normal from different simulations with PE lipids.

similar to the \sim 200 ns simulations with PC lipids in previous work [2]. However, when simulation was continued for μ s, the binding affinity substantially increased and interpretation was that calcium overbinds to PC lipid in CHARMM36. Therefore, the conclusion seems to be similar here, although the new NBfix parameters may complicate the situation 31.The status of NBfix parameters in these simulations should be checked.

The β -carbon order parameter of PG exhibits a rapid decrease with small CaCl₂ concentrations and a more modest decrease with larger concentrations in experiments [62] (Fig. ??). The rapid decrease with CaCl₂ is observed but overestimated in CHARMM36 simulation with POPC:POPG 1:1 mixture, but not in 4:1 mixture 32.This is little bit weird, should be checked.

33.We need PC:PG simulations with CaCl₂ from different force fields to finish the discussion.

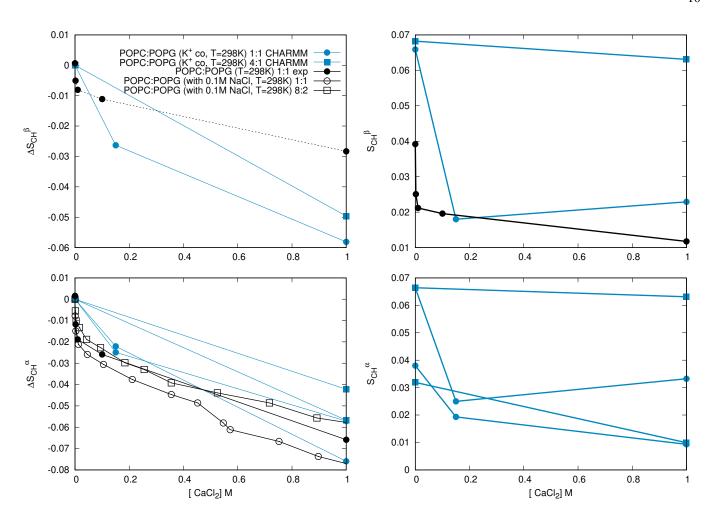


FIG. 9: (left) The headgroup order parameters of PC from PC:PG mixtures as a function CaCl₂ concentration from experiments [62, 69] and CHARMM36 simulations. Note that beta order parameter is calculated from empirical relation $\Delta S_{\beta}=0.43\Delta S_{\alpha}$ [74], not actually measured. (right) The headgroup order parameters of PG from PC:PG mixtures as a function CaCl₂ concentration from experiments [62] and CHARMM36 simulations.

CONCLUSIONS

SUPPLEMENTARY INFORMATION

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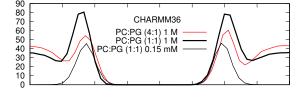


FIG. 10: Calcium ion density profiles along membrane normal from different simulations with PG lipids.

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ToDo

	P
1. List should be completed	1
2. The rest of the details to be written. I am not sure	
how much we need to repeat the NMRlipidsIVps paper.	1
3. Ion parameters?	2
4. Correct citation for CHARMM POPG	
5. Ion parameters?	
6. Ion parameters?	2

7. Data to be uploaded by J. Madsen. Details to be	25. This should be clarified as in NMRlipidsI and error
filled once we have the data	bars should be added. Probably larger error bars for
8. Concentration to be checked	3 united atom models based on the report by Fuchs et al. 5
9. Data to be uploaded by J. Madsen. Details to be	28. This is text by P. Fuchs, copied from the blog.
filled once we have the data	Δ real results in nm ² the error is $\langle -0.003 \text{ nm}^2 \rangle$
10. Concentration to be checked	or a pure POPC
11. Data to be uploaded by J. Madsen. Details to be	CHARMM36· 0.624
filled once we have the data	Berger : 0.649
12. Data to be uploaded by J. Madsen. Details to be	- POPC/POPE 50:50
filled once we have the data	3 CHARMM36 · POPC 0.609 POPE 0.557
13. Concentration to be checked	Berger-hacked: POPC 0.637, POPE 0.632
14. Data to be uploaded by J. Madsen. Details to be	<u> </u>
filled once we have the data	One can see that CHARMM 36 predicts a drop in
15. Concentration to be checked	the area on going from pure POPC to POPC/POPE
16. Data to be uploaded by J. Madsen. Details to be	50:50 This means that POPC pack tightly to
filled once we have the data	POPE In contrast the values for Rerger are
17. Zenodo entry unclear	not that changed. The POPE value predicted
18. This is probable not plain berger, correct force filed	by CHARMM 36 (in the mixture POPC/POPE
should be described	3 50:50) is much smaller than that predicted by Berger.
19. This is probable not plain berger, correct force filed	
should be described	The experimental acyl chain order parameters for POPE
20. This is probable not plain berger, correct force filed	[71] seem larger than reported for POPC [10], which
should be described	supports the more condensed PE bilayer. In principle,
21. This is probable not plain berger, correct force filed	this is beyond the scope of this work (lipid headgroups),
should be described	but we can consider mentioning this 6
22. This is probable not plain berger, correct force filed	
should be described	3 27. Simulation of CHARMM36 at 298K should be
23. Details to be checked by Tiago	4 maybe rerun with Gromacs 5
24. Figure and discussion about POPG experiments to	29. Other experimental data should be checked and dis-
be addded	4 cussed
26. Berger results are not here yet, but we should men-	
tion the ring like structures pointed out by T. Piggot in	30. The used ion model here should be checked 9
the blog: The poor performance of headgroup order pa-	31. The status of NBfix parameters in these simulations
rameters in Berger model can be probably explained by	should be checked
ring like structures seen in Fig. 6 in Ref. 64. These	32. This is little bit weird, should be checked 9
ring like structures are a widespread feature of typical	
Berger based lipid force fields containing explicit hy-	33. We need PC:PG simulations with CaCl ₂ from dif-
drogen atoms in the head group [65–67]	4 ferent force fields to finish the discussion 9