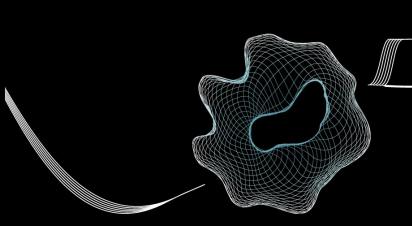
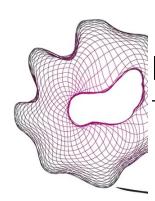
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Multiphase Reactor Technology HC 5: Fluidization

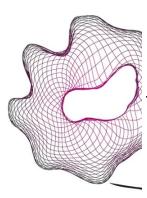
Fausto Gallucci





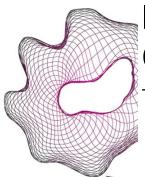
Resume

- Different types of distributors
- Distributor determines the initial bubble size
- Bubble velocity/dimensions change depending on the position and on the characteristics of solids
- Bubbles have wakes which are responsible of solid movements/mixing
- Entrainment of solids should be taken into account when designing
 FB
- Mixing and segregation of solids can be important for chemical reactions in FB



Contents

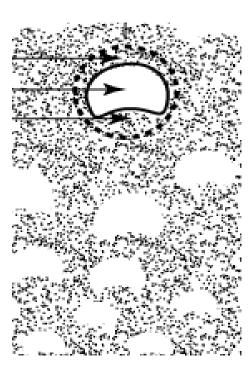
- ✓ Gas dispersion and gas exchange
- ✓ Particle-to-gas mass and heat transfer
- ✓ Heat transfer between fluidized beds and surfaces
- ✓ Conversion of gas due to catalytic reactions
- Recent developments
 - ✓ Applications of fluidized bed chemical reactors
 - ✓ Computational fluid dynamics (CFD) based modeling









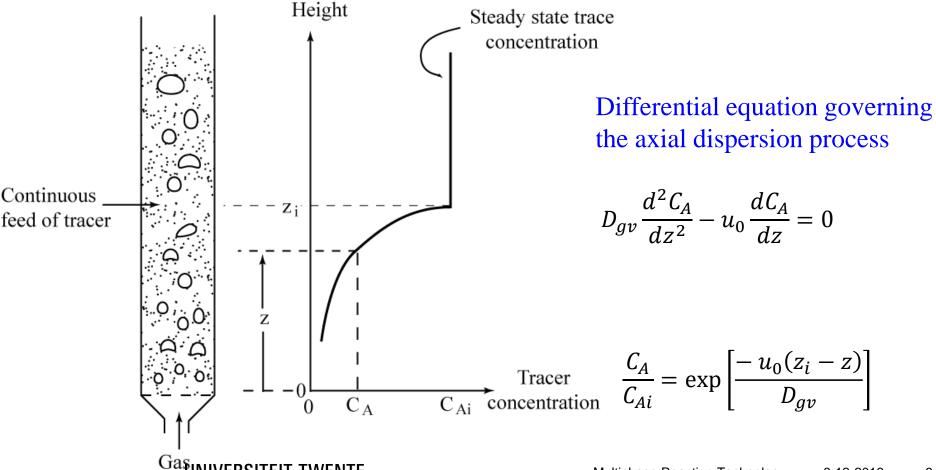


- Relevance: especially for gas-solid contacting (catalytic reactions) gas dispersion and bubble-emulsion exchange of gas are very important
- Dispersion of gas in fluidized beds: both vertical and horizontal dispersion of gas occurs. General experimental finding: D_{gv}>D_{gh}
- Experimental techniques for dispersion coefficients:

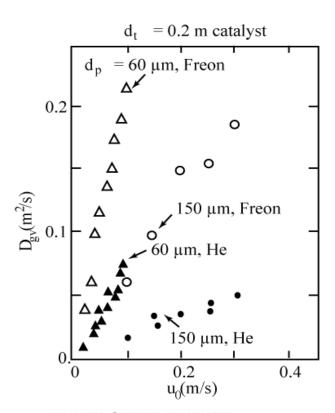
from steady state gas tracer experiments + fitting experimental results to diffusion equation

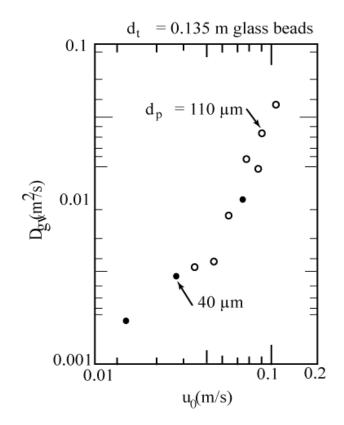
from stimulus-reponse experiments + fitting experimental results to diffusion equation

Steady state experiment for D_{qv} in a gas-fluidized bed



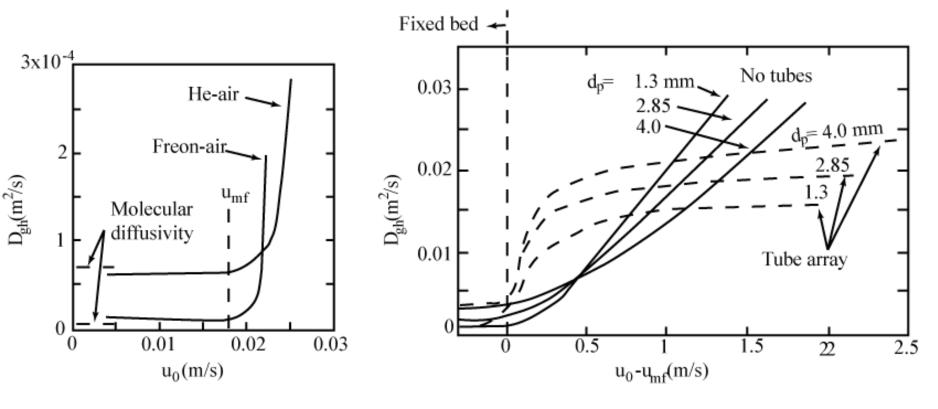
Experimental results for D_{gv} (steady state backmixing (left) and stimulus response (right)
 experiments) for microspherical catalyst d_p=150 μm; triangular points for FCC catalyst 60 μm





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Experimental results for D_{gh} : near u_{mf} in beds of fine solids (d_t =0.2 m, microspherical catalyst, d_p =150 μm), the effect of a tube array in beds of coarse particles, u_{mf} =0.73-1.83 m/s

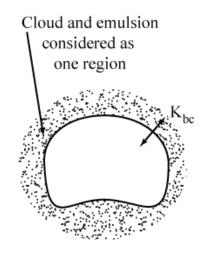


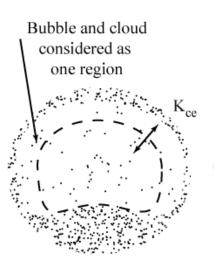
- Gas exchange between bubble and emulsion phase
 - Definitions of gas exchange:

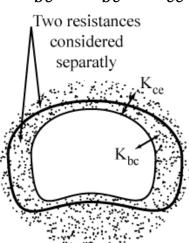
$$-\frac{1}{V_{b}}\frac{dN_{Ab}}{dt} = -u_{b}\frac{dC_{ab}}{dz} = K_{be}(C_{Ab} - C_{Ae}) = K_{bc}(C_{Ab} - C_{Ac}) = K_{ce}(C_{Ac} - C_{Ae})$$

Relationship between interchange coefficients:

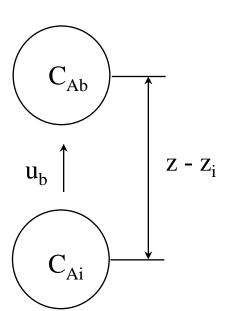
$$\frac{1}{K_{be}} = \frac{1}{K_{bc}} + \frac{1}{K_{ce}}$$







 Experimental methods for K_{be}: single bubble method (injection of a single bubble with tracer at z=zi and initial tracer concentration C_{Ai}



measurement of concentrations in the bubble at two z-levels yields K_{he} (bubble rise velocity known)

$$-u_{b} \frac{dC_{Ab}}{dz} = K_{be}(C_{Ab} - C_{Ae})$$

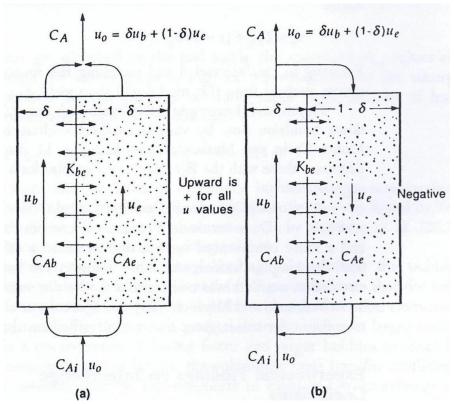
$$z - z_{i}$$

$$\ln(C_{Ab} - C_{Ae})|_{zi}^{z} = -\frac{K_{be}(z - z_{i})}{u_{b}}$$

$$\ln(C_{Ab} - C_{Ae})|_{zi}^{z} = -\frac{K_{be}(z - z_{i})}{u_{b}}$$

$$\frac{C_{Ab} - C_{Ae}}{C_{Ai} - C_{Ae}} = exp \left[-\frac{K_{be}(z - z_i)}{u_b} \right]$$

- Experimental methods for Kbe
 - Bubbling bed method (pulse- or step-response of tracer + fitting experimental concentration profiles against two-zone model)



- Mass balances for two-zone model
 - bubble phase:

$$\frac{\partial C_{Ab}}{\partial t} + u_b \frac{\partial C_{Ab}}{\partial z} = K_{be}(C_{Ab} - C_{Ae})$$

• emulsion phase: $\frac{\partial C_{Ae}}{\partial t} + \frac{u_e}{c} \frac{\partial C_{Ae}}{\partial z} = \frac{\delta}{1 - \delta} K_{be} (C_{Ab} - C_{Ae})$

integral balance: $u_0 = \delta u_b + (1 - \delta)u_e$

empirical information on bubble velocity u_b , emulsion velocity u_e , emulsion voidage ε_e and bubble holdup δ required (from integral balance one of the quantities can be computed from the others)

- Estimation of gas exchange coefficients (Davidson model + Higbie penetration model)
 - bubble to cloud exchange coefficient K_{bc}:

$$K_{bc} = 4.5 \left(\frac{u_{mf}}{d_b}\right) + 5.85 \left(\frac{D^{\frac{1}{2}}g^{\frac{1}{4}}}{\frac{5}{d_b^{\frac{1}{4}}}}\right)$$

cloud to emulsion exchange coefficient K_{ce}:

$$K_{ce} = 6.77 \left(\frac{D\varepsilon_{mf} (0.711) (gd_b)^{\frac{1}{2}}}{d_b^3} \right)^{\frac{1}{2}} = 6.77 \left(\frac{D\varepsilon_{mf} u_{br}}{d_b^3} \right)^{\frac{1}{2}}$$

note: K_{bc} contains both convective and diffusive contributions

 Relevance: mass and/or heat transfer between fluidized particles and fluidizing agent occur frequently in a great variety of processes such as adsorption, drying, granulation, gas phase polymerization of C₂H₄

accurate prediction of transfer rates of mass and/or heat is required for design purposes

- Topics covered:
 - ✓ mass transfer: experimental
 - ✓ interpretation of mass transfer coefficients
 - ✓ heat transfer: experimental
 - ✓ interpretation of heat transfer coefficients

Mass transfer: experimental results for single spheres (Froessling):

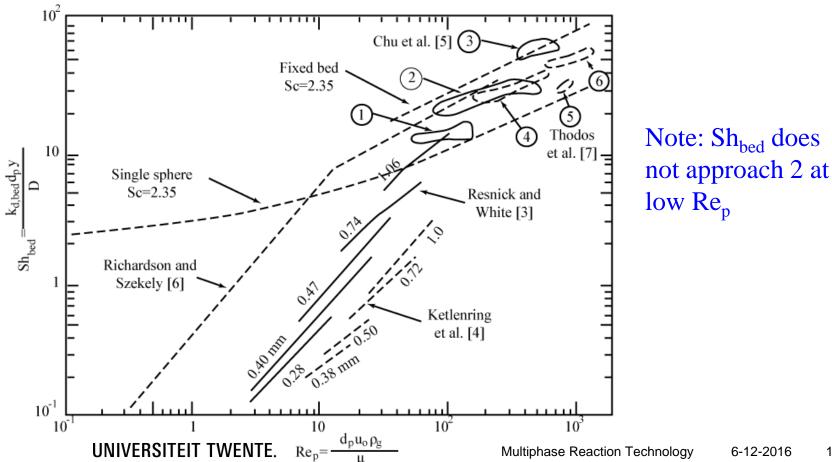
$$Sh^* = \frac{k_d^* d_{sph} y}{D} = 2 + 0.6 \left(\frac{\rho u_0 d_{sph}}{\mu}\right)^{\frac{1}{2}} \left(\frac{\mu}{\rho D}\right)^{\frac{1}{3}}$$
$$= 2 + 0.6 \left(Re_{sph}\right)^{\frac{1}{2}} (Sc)^{\frac{1}{3}}$$

with y the logarithmic mean fraction of inert non-diffusing component

$$Sh^* = 2 + 1.8(Re_p)^{\frac{1}{2}}(Sc)^{\frac{1}{3}}$$

- Mass transfer: experimental results for fixed beds (Ranz):
- Mass transfer: experimental results for fluidized beds: difficult to measure due to experimental problems (high volumetric mass transfer rate + proper driving force)

Experimental findings on mass transfer in fluidized beds



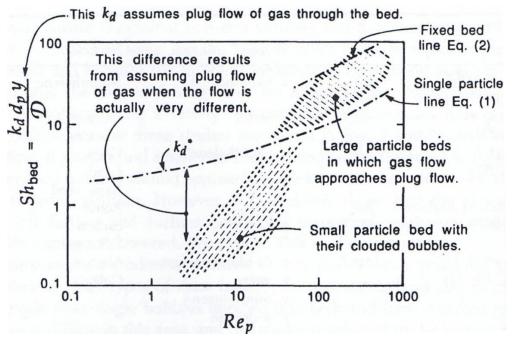
- Interpretation of mass transfer coefficients: distinction between single particle mass transfer coefficient k_{d,p} and bed mass transfer coefficient k_{d,bed}
- Single particle mass transfer coefficient k_{d,p}:

Consider a single particle containing A immersed in a bed of other particles free of component A. In this case we measure the mass coefficient of a single particle

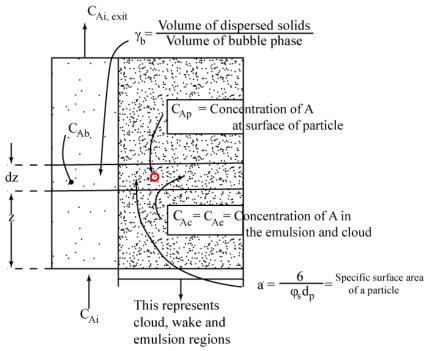
$$-\frac{1}{S_{particle}}\frac{dN_{A,particle}}{dt} = k_{d,p}(C_{A,p} - C_{A,bed})$$

 Bed mass transfer coefficient k_{d,bed}: especially for fine particle systems single particle method impractical and therefore whole bed of particles is studied instead

literature results suffer from incorrect interpretation (incorrect flow models)



Mass transfer rate from bubbling bed model:



Differential mass balance for (transferred) component A):

$$\frac{1}{S_{particles}} \frac{dN_A}{dz} = k_{d,bed} (C_{A,p} - C_{A,b})$$

Differential mass balance for bubble phase:

specific particle surface a'

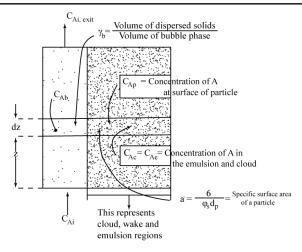
$$\frac{dC_{A,b}}{dt} = u_b \frac{dC_{A,b}}{dz} = \frac{k_{d,bed}(1 - \epsilon_f)a'}{\delta}(C_{A,p} - C_{A,b}) \quad a' = \left(\frac{A_{particle}}{V_{particle}}\right) = \frac{6}{\phi_s d_{sph}}$$

Mass transfer in terms of interchange coefficient K_d:

$$u_b \frac{dC_{A,b}}{dz} = K_d(C_{A,p} - C_{A,b})$$

Comparison of above expressions leads to:

$$\frac{k_{d,bed}(1-\epsilon_f)a'}{\delta} \longrightarrow Sh_{bed} = \frac{k_{d,bed}d_py}{D} = \frac{y\phi_s d_p^2\delta}{6D(1-\epsilon_f)}K_d$$



$$K_d = \begin{pmatrix} \text{added from particles} \\ \text{dispersed in the bubbles} \end{pmatrix} + \begin{pmatrix} \text{transfer across the} \\ \text{bubble-cloud boudary} \end{pmatrix}$$

$$K_d = \gamma a' k_d^* + K_{bc} = \gamma_b \frac{6}{\phi_s d_p} k_d^* + K_{bc}$$
$$= \gamma_b \frac{6 Sh^* D}{\phi_s d_p^2 y} + K_{bc}$$

K_{bc} given earlier

Combining previous equations leads to final expression for Sh_{bed}:

$$Sh_{bed} = \frac{\delta}{1 - \varepsilon_f} \left[\gamma_b Sh^* + \frac{y\phi_s d_p^2}{6D} K_{bc} \right]$$

From Kunii-Levenspiel model:

$$\frac{\delta}{1-\varepsilon_f} = \frac{u_0 - u_{mf}}{u_{br}(1-\varepsilon_{mf})}$$

For a given bed of solids and constant bubble size above equation can be written as (using equation from Kunii-Levenspiel model):

$$Sh_{bed} = A Re_p - B$$

Equation can fit large portion of experimental data for fluidized beds

Heat transfer: experimental results for single spheres (Ranz):

$$Nu^* = \frac{h^* d_{sph}}{k_g} = 2 + 0.6 \left(\frac{\rho u_0 d_{sph}}{\mu}\right)^{\frac{1}{2}} \left(\frac{C_p \mu}{k_g}\right)^{\frac{1}{3}} = 2 + 0.6 \left(Re_p\right)^{\frac{1}{2}} (Pr)^{\frac{1}{3}}$$

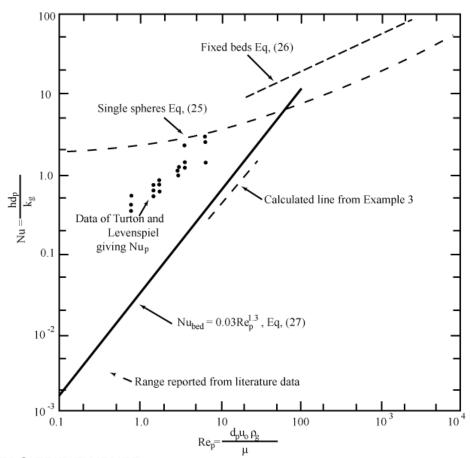
Heat transfer: experimental results for fixed beds (Ranz):

$$Nu^* = 2 + 1.8 \left(Re_p \right)^{\frac{1}{2}} (Pr)^{\frac{1}{3}}$$

Heat transfer: experimental techniques for gas-fluidized beds: both steady state techniques and unsteady state techniques have been used by many investigators !!!

General problem: flow model used for interpretation of the experimental data

Experimental findings on heat transfer in fluidized beds



• Nusselt number at low particle Reynolds number (Re_p=0.1 to 100):

$$Nu_{bed} = \frac{hd_p}{k_q} = 0.03 \left(\frac{\rho u_0 d_p}{\mu}\right)^{\frac{1}{3}} = 0.03 (Re_p)^{\frac{1}{3}}$$

- Note: data of Turton and Levenspiel correspond to heat transfer coefficients for individual particles and not to bed average quantities !!!
- Expression for bed average Nusselt number from bubbling bed model:

$$Nu_{bed} = \frac{hd_p}{k_g} = \frac{\delta}{1 - \varepsilon_f} \left[\gamma_b N u^* + \frac{\phi_s d_p^2}{6k_g} H_{bc} \right]$$

$$H_{bc} = 4.5 \left(\frac{u_{mf} \rho_g C_{p,g}}{d_b} \right) + 5.85 \frac{\left(k_g \rho_g C_{p,g} \right)^{\frac{1}{2}} g^{\frac{1}{4}}}{d_b^{\frac{5}{4}}}$$

■ Relevance: temperature control required in many physical and chemical operations —>for design purposes prediction of heat transfer rates between fluidized beds and surfaces is required

- Topics covered:
 - Definition of heat transfer coefficient h
 - Experimental findings vertical tubes
 - Experimental findings horizontal tubes
 - Bed conductivity models
 - General expression for h at a heat exchanger surface

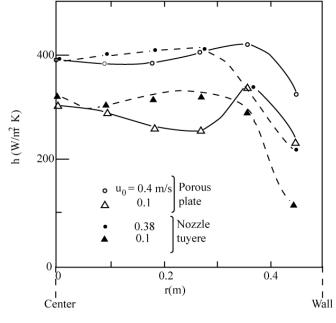
Definition of heat transfer coefficent h:

$$q = A_w h \Delta T$$

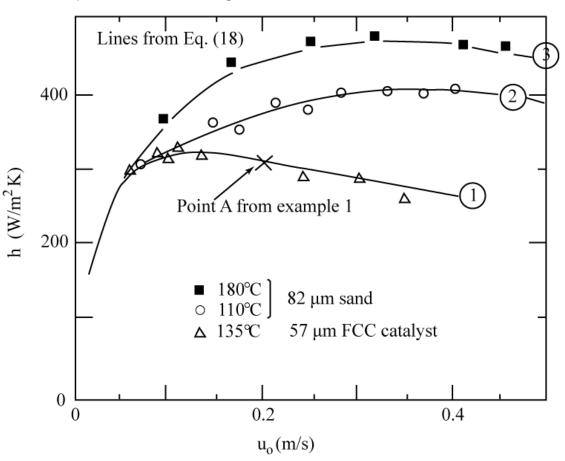
h for fluidized beds is 1-2 orders of magnitude higher than h for gas

Experimental findings vertical tubes (z=0.85 m, d_t=1 m, Lm=1.36 m, bed of quartz sand

 $d_p=96 \text{ mm}$) + effect of gas distributor plate



Experimental findings horizontal tubes in a 0.3 x 0.3 m bed:

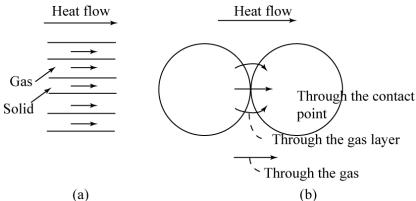


Note: with increasing superficial gas velocity u_o we have two opposing effects:

Decreasing ε_s : lower h

Increasing bubble frequency: higher h

Bed conductivity models



Case (a): heat flows in parallel paths:

$$k_e^0 = \varepsilon_{mf} k_g + (1 - \varepsilon_{mf}) k_s$$

Case (b): modification of parallel path model:

ŀ

superscript "o" refers to stagnant gas conditions

$$k_e^0 = \varepsilon_{mf} k_g + \left(1 - \varepsilon_{mf}\right) k_s \left[\frac{1}{\phi_b \left(\frac{k_s}{k_g}\right) + \frac{2}{3}} \right]$$

φ_b=d_{eqv}/d_p with d_{eqv} the equivalent thickness of the gas film near the contact points which aids the interparticle heat transport

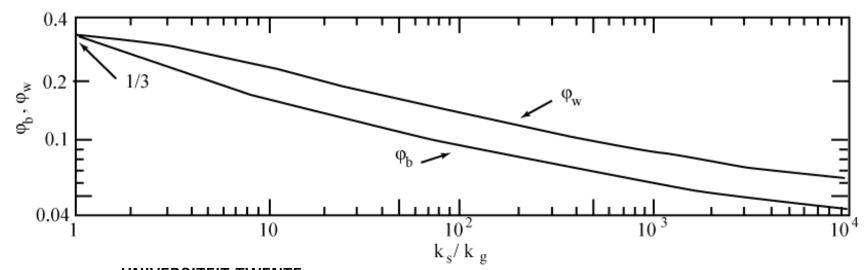
Modification of parallel path model for wall layer:

$$k_{e,w}^{0} = \varepsilon_{w} k_{g} + (1 - \varepsilon_{w}) k_{s} \left[\frac{1}{\phi_{w} \left(\frac{k_{s}}{k_{g}} \right) + \frac{1}{3}} \right]$$

 $\varepsilon_{\rm w}$ represents the mean void fraction for the wall layer

• Figure for determination of ϕ_b and ϕ_w :

$$h_w^0 = \frac{2k_{e,w}^0}{d_p}$$



 Conceptual picture of heat transfer between gas-fluidized beds and immersed surfaces (container walls or submerged heat exchange tubes)

wall at temperature T_w

parallel heat transfer to fraction $\delta_{\rm w}$ of wall covered with bubbles and remaining fraction $(1-\delta_{\rm w})$ covered with particles bubble

bubble fraction at surface δ_w

gas convection and radiation through bubble to solids at temperature T_s

emulsion packet swept to the wall due to bubble passage

heat transfer due to radiation, conduction and gas convection in first particle layer with effective thickness of 1/2d_p and subsequent heat transport to emulsion packet arriving at wall layer

General expression for h at a heat exchanger surface:

$$h = \left[\delta_w(h_r + h_g)\right] + \left[\frac{1 - \delta_w}{\frac{1}{h_r + h_g} + \frac{1}{h_{packet}}}\right]$$

first contribution: bubble at surface; second contribution: emulsion at surface gas convection contribution in bubble h_g can often be neglected

Additional equations for general h expression

Contribution due to radiation h_r

$$h_r = \frac{\sigma(T_s^4 - T_w^4)}{\left(\frac{1}{e_s} + \frac{1}{e_w} - 1\right)(T_s - T_w)}$$
 Stefan Boltzmann constant
$$\sigma = \frac{5.67 \cdot 10^{-8} \text{ W/(m}^2 \cdot \text{K}^4)}{\sigma = 5.67 \cdot 10^{-8} \text{ W/(m}^2 \cdot \text{K}^4)}$$

contribution due to gas convection h_w:

$$h_w = h_w^0 + \alpha_w (C_{p,g} \rho_g u_0) = \frac{2k_{e,w}^0}{d_p} + \alpha_w (C_{p,g} \rho_g u_0) \quad \alpha_w = 0.05 \text{ (fitted constant)}$$

contribution due to packet renewal h_{packet}:

$$h_{packet} = 1.13 \left[\frac{k_e^0 \rho_s (1 - \varepsilon_{mf}) C_{p,s} n_w}{(1 - \delta_w)} \right]^{\frac{1}{2}}$$

$$n_w: \text{ bubble frequency}$$

Extreme of fine particles at low temperatures:

$$h = 1.13 \left[k_e^0 \rho_s (1 - \varepsilon_{mf}) C_{p,s} n_w (1 - \delta_w) \right]^{\frac{1}{2}}$$
 surface renewal of packets emulsion phase

Dense Fluidized Beds Conversion of gas due to catalytic reactions

- Relevance: for rational research and the development of new chemical processes tools to predict the performance of fluidized bed reactors are required
- Topics covered:
 - measures of reaction rate and reactor performance
 - experimental findings for fine particle bubbling beds
 - reactor model for fine particle bubbling beds
- Restriction to fine particle bubbling beds due to (relative) significance in large number of industrial processes

Dense Fluidized Beds Conversion of gas due to catalytic reactions

Measures of reaction rate and reactor performance

- ✓ conversion rate for solid-catalyzed reaction:
- Integration of conversion rate equation

$$-\frac{1}{V_b}\frac{dN_A}{dt} = K_r C_A$$

valid for non-porous solid and independent of bed voidage and particle size

$$\checkmark$$
 plug flow:
$$1 - X_A = \frac{C_{A,0}}{C_{A,i}} = \exp(-K_r \tau) \qquad \tau = \frac{V_S}{\phi_g} \qquad \text{i=m, f or mf}$$

$$\checkmark$$
 mixed flow:
$$1 - X_A = \frac{C_{A,0}}{C_{A,i}} = \frac{K_r \tau}{1 + K_r \tau} \qquad K_r \tau = K_r \frac{K_i (1 - \varepsilon_i)}{u_0}$$

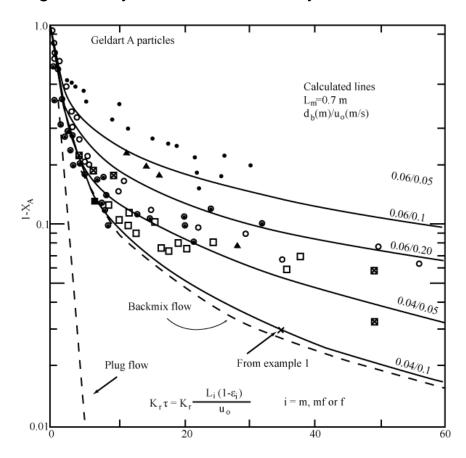
 Equations for plug flow and mixed flow are useful for reference purposes to interprete experimental data and more advanced reactor models

Dense Fluidized Beds Conversion of gas due to catalytic reactions

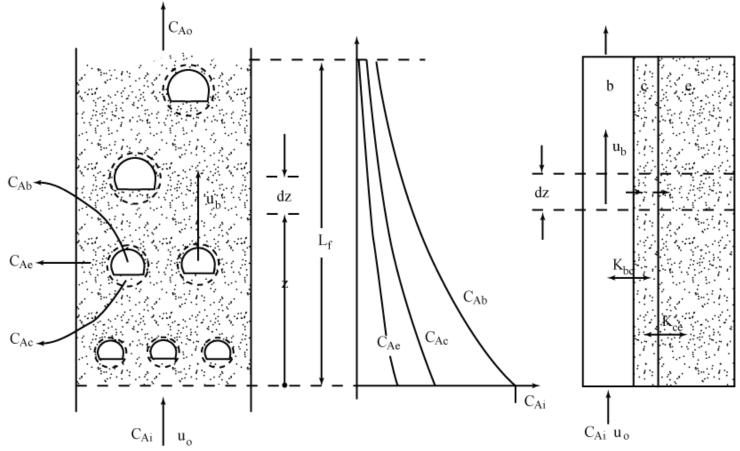
- Experimental findings
 - In general fluidized bed reactors require more catalyst than a fixed bed to achieve a given degree of conversion due to less effective gas-solid contacting
- Behaviour has been studied extensively in literature (often using beds with a too small diameter: hydrodynamics sensitive to bed diameter) using model reactions:
- + Ozone decomposition
- + Oxidation of carbon monoxide

fast irreversible reactions first order kinetics simple reaction scheme

Experimental findings for very fine Geldart A catalyst



Reactor model for fine particle bubbling beds (vigorously bubbling)



Assumptions for "fine particle model"

$$u_{o}/u_{mf} >> 1$$
 and $u_{b}/u_{mf} >> 1$

- ✓ Fresh entering feed gas containing reactant A enters bed and on contact with
 the fine catalyst powder reacts according to a first order reaction
- ✓ Bed consists of three regions: bubble [b], cloud [c] and emulsion [e] region (bubble wake is part of the cloud)
- ✓ All feed gas passes through the bed as bubbles (u_o>>u_{mf})
- ✓ gas interchange rate between bubble and cloud and between cloud and emulsion are given by K_{bc} and K_{ce} respectively

Recall definition of solids distribution in bubble, cloud and emulsion:

 γ_b , γ_c and γ_e denote respectively volume of solids dispersed in bubble, cloud and emulsion divided by volume of bubble

Balance formulation for reactant A:

```
transfer to
overall disappearance
                          reaction
                           in bubble
      in bubble
                                               cloud-wake
                          reaction in
                                               transfer to
      transfer to
                          cloud-wake
                                               emulsion
      cloud-wake
     transfer to
                          reaction in
     emulsion
                           emulsion
```

■ Mass balance for reactant A:
$$-\frac{dC_{A,b}}{dt} = -u_b \frac{dC_{A,b}}{dz} = \gamma_b K_r C_{A,b} + K_{b,c} (C_{A,b} - C_{A,c})$$

additional equations

$$K_{b,c}(C_{A,b} - C_{A,c}) = \gamma_c K_r C_{A,c} + K_{c,e}(C_{A,c} - C_{A,e})$$

$$K_{c,e}(C_{A,c} - C_{A,e}) = \gamma_e K_r C_{A,e}$$

Upon eliminating concentrations of A in emulsion and cloud we get:

$$-u_{b} \frac{dC_{A,b}}{dz} = K_{f}C_{A,b}$$

$$K_{f} = \gamma_{b}K_{r} + \frac{1}{\frac{1}{K_{b,c}} + \frac{1}{\gamma_{c}K_{r} + \frac{1}{\frac{1}{K_{c,e}} + \frac{1}{\gamma_{e}K_{r}}}}$$

K_f represents an overall rate constant for chemical reaction accounting for all relevant mass transfer resistances in fine particle gas-fluidized beds

Inspection: combination of resistances in series/parallel

• Integration of mass balance between inlet and certain position in bed:

$$\frac{C_{A,b}}{C_{A,inlet}} = \frac{C_{A,b}}{C_{A,i}} = \exp\left[-K_f \frac{z}{u_b}\right]$$
 assumption: bubble size remains approximately constant in bed

Since "all" gas fed to the bottom of the bed passes in the form of bubbles we can write for the reactor as a whole:

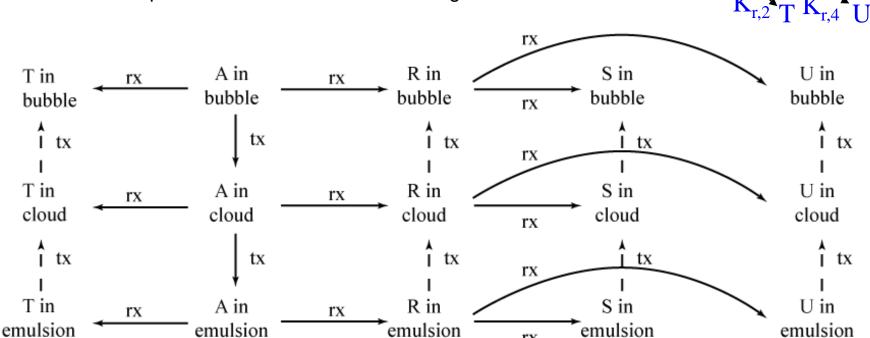
$$1 - X_A = \frac{C_{A,0}}{C_{A,i}} = \exp\left[-K_f \frac{z}{u_b}\right]$$
 contribution of emulsion is negligible (not valid for coarse particles)

Note that all parameters (except Kr of course) can be obtained from the Kunii and Levenspiel (K-L) model discussed before in detail !!!

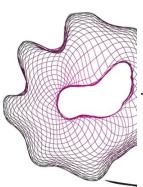
estimates of extent of conversion can be made for fine particle beds

Extension to multiple reactions is straightforward although the algebra becomes (much) more involved !!!

Schematic representation for a so-called Denbigh reaction



rx



Recent Developments

- Applications
 - Fluidization of fine particles (including C-powders)
 - Multi-scale modeling



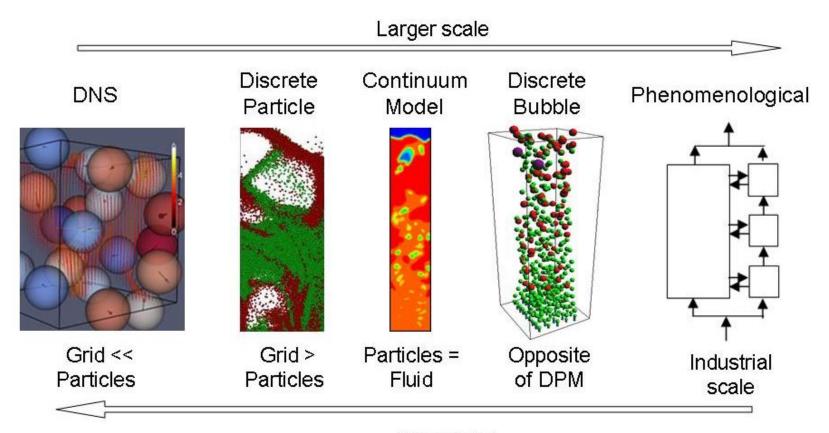
Recent Developments Experimental validation Humidity meter -20 Adjustable camera set-up -10Pseudo-2-D fluidized bed 10 Lights -30 -40 ₫ -50 High speed Water vessel -60 camera -70 -80 Mass flow controllers -90 100 110 120 PIV Computer -140 -120 -100 -40-20 UNIVERSITEIT TWENTE. iviuitipnase Reaction Technology 6-12-2016 45

Recent Developments

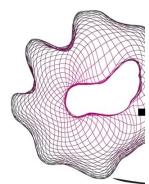
CFD models versus GLOBAL SYSTEM MODELS

CFD Models	Global System Models
Advantage	Advantage
More exact solution available	Simple models and simple solutions
	facilitate understanding
Phenomena follow from	After adjustment of parameters accurate
calculation a priori	macro scale behavior prediction
Disadvantage	Disadvantage
Detailed knowledge required	Experimental validation and adjustment of
about the elementary processes	parameters necessary
Macroscopic behavior not	Meaning of parameters sometimes unclear
always accurately predicted	due to lumping

Recent Developments Computational Fluid Dynamics (CFD) approach



More detail



DPM

Basic idea in discrete particle models

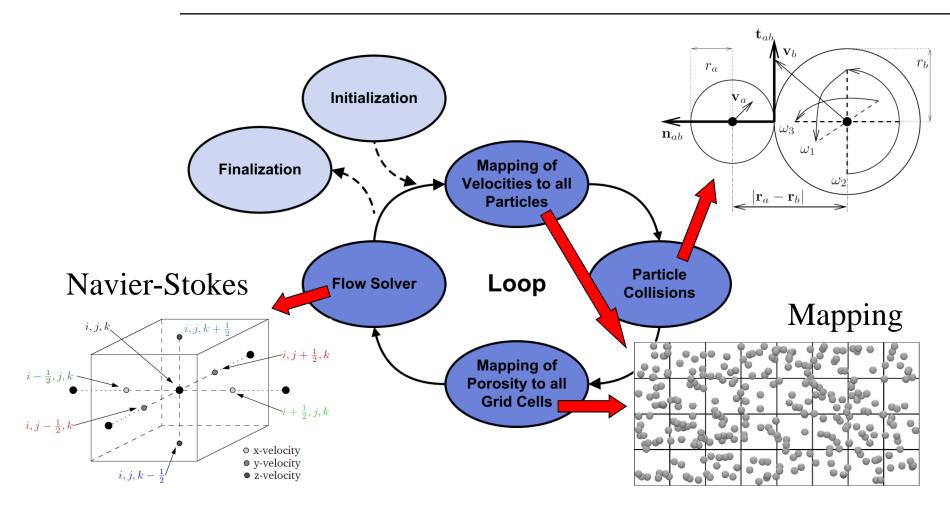
gravity drag

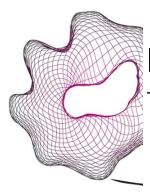
| drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag | drag |

particle moves due to external forces while collisions with other particles and/or confining walls may occur

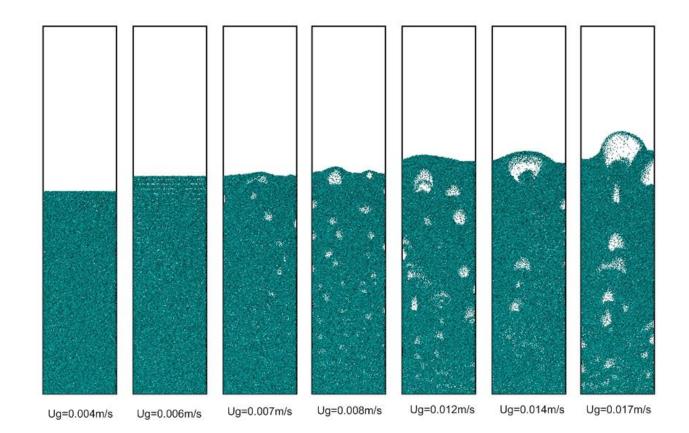
- + Eulerian grid size is larger than the particle size
- + empirical correlation for inter-phase interaction force is needed

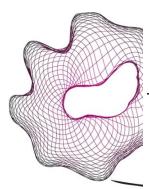
DPM



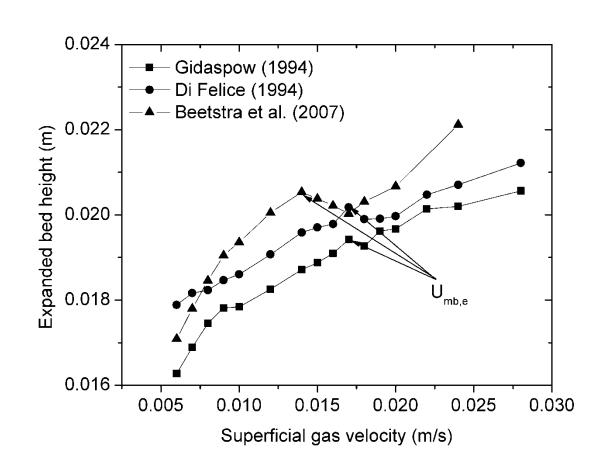


Bed contraction phenomenon

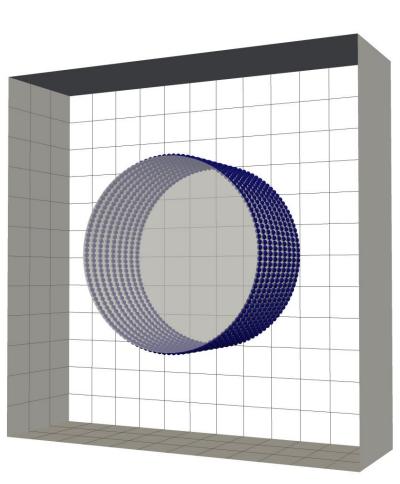




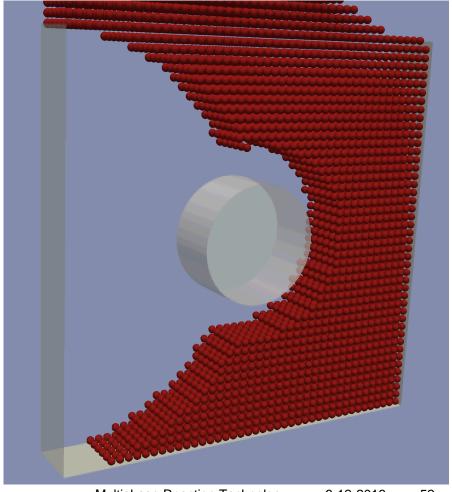
Bed contraction phenomenon



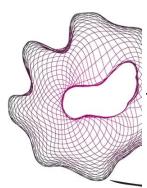
Immersed Boundary Method



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Multiphase Reaction Technology



Resume

- What is a fluidized bed
- How to define a particle (dimension, classification)
- Particles (and fluid) determine u_{mf}, u_t
- Different types of distributors
- Distributor determines the initial bubble size. Solids and fluid determine bubble velocity/dimensions at a given position
- Bubbles have wakes which are responsible of solid movements/mixing, gas exchange, solid segregation
- A model can be easily solved providing that Kf can be defined (and it depends on all the exchanges already discussed)