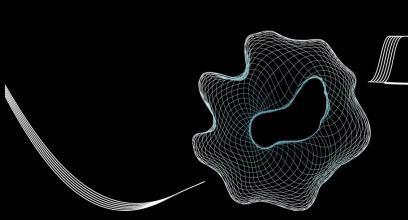
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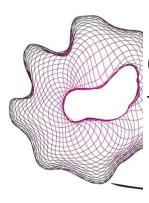


Multiphase Reactor Technology HC 4: Fluidization

Fausto Gallucci

Sascha Kersten

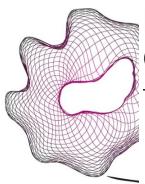




Contents

- Dense fluidized beds
 - ✓ Gas distributor design
 - ✓ Gas bubbles behaviour
 - ✓ Flow models
 - ✓ Entrainment and elutriation
 - ✓ Mixing and segregation of solids





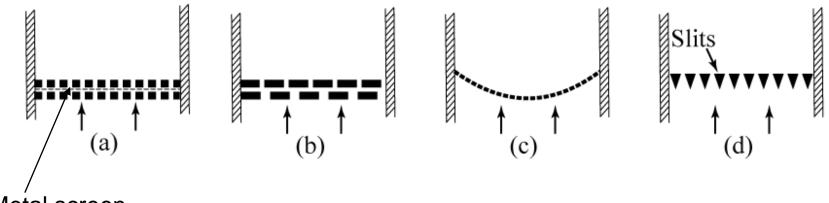
 Gas distributors are required to properly feed the gas through the bed.

Ideally they need to have the following characteristics

- ✓ Low pressure drop
- ✓ High strength
- ✓ High resistance against thermal stresses
- ✓ Resistance to clogging
- ✓ Low cost



Plate and grate distributors



Metal screen (in some cases not preferred)

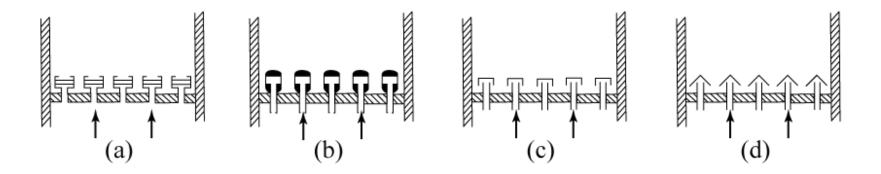
Cheap but low resistance for heavy loads

Higher resistance for heavy loads

Orifice diameters:

1-2 mm small scale up to 50 mm large scale

Tuyere distributors (bottom)



Used in high temperature or highly reactive conditions

Much more expensive than perforated plates Problems with settling and sintering of particles on the distribution plate

- Summary of experimental findings (d=distributor; b=bed)
 - Even gas distribution for u close to u_{mf} : $\Delta p_d >= 0.15 \Delta p_b$
 - Ratio of Δp_d and Δp_b decreases with increasing u/u_{mf} ratio
 - $\Delta p_d/\Delta p_b$ roughly independent of bed height

difference decreases to zero for u>>u_{mf}

■ Practical "rule of thumb": $\Delta p_d = (0.2 \text{ to } 0.4) \Delta p_b$ UNIVERSITEIT TWENTE.

Initial bubble diameter (cm) for distributor with N_{or} orifices per unit area (cm⁻²) of plate where distance between orifices equals I_{or}:

$$d_{bo} = \frac{1.30}{g^{0.2}} \left[\frac{u_0 - u_{mf}}{N_{or}} \right]^{0.4}$$

$$d_{bo} = \frac{2.78}{g} \big(u_0 - u_{mf} \big)^2$$

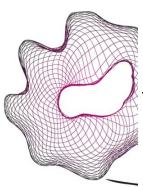
$$\begin{pmatrix}
0 & 0 & 0 \\
0 & 0 & 0 \\
0 & 0 & 0
\end{pmatrix}$$

$$N_{or} = \frac{1}{l_{or}^2}$$

$$\begin{array}{c} \text{low gas flow rate} \\ \text{non-touching bubbles} \\ \text{d}_{bo} < \text{l}_{or} \end{array}$$

high gas flow rate touching bubbles $d_{bo}>l_{or}$

$$\begin{pmatrix}
\bigcirc & \bigcirc & \bigcirc \\
\bigcirc & \bigcirc & \bigcirc \\
\bigcirc & \bigcirc & \bigcirc
\end{pmatrix}
\qquad
N_{or} = \frac{2}{\sqrt{3}l_{o}^{2}}$$



- Davidson model for isolated bubbles
 - ✓ Postulate 1: A gas bubble is solid-free and circular (2D) or spherical (3D) in shape.
 - Postulate 2: The emulsion phase flowing in the vicinity of the bubble behaves like an incompressible, inviscid fluid with macroscopic density $\rho_s(1-\epsilon_{mf})$.
 - ✓ Postulate 3: The relative motion between the gas phase and the emulsion phase is given by Darcy's law:

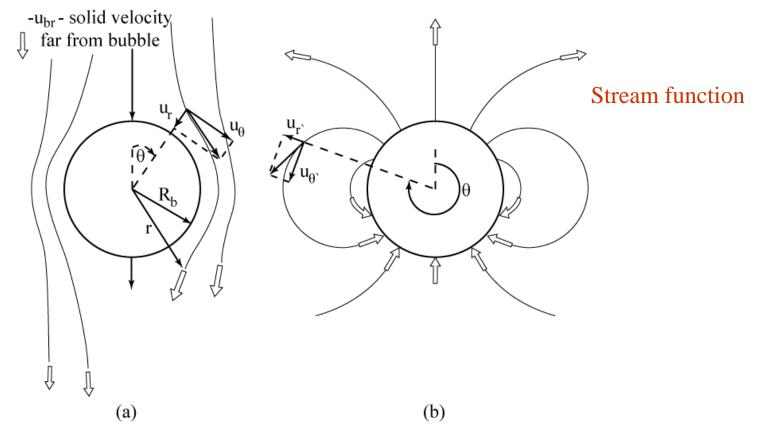
gas phase velocity

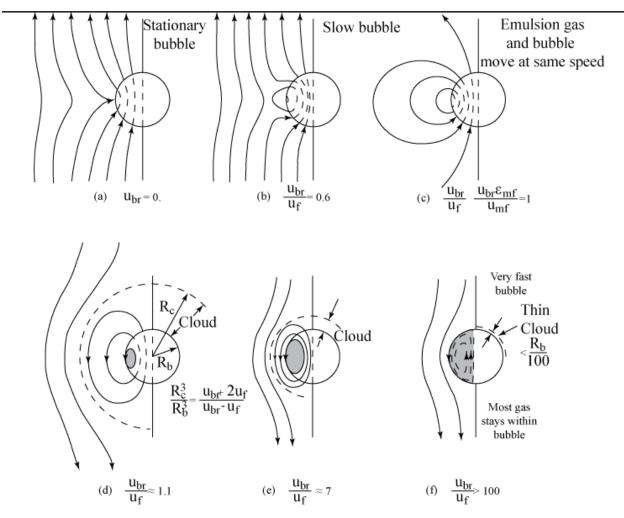
Boundary conditions

- Far from bubbles the pressure gradient in undisturbed
- The pressure in a bubble is constant

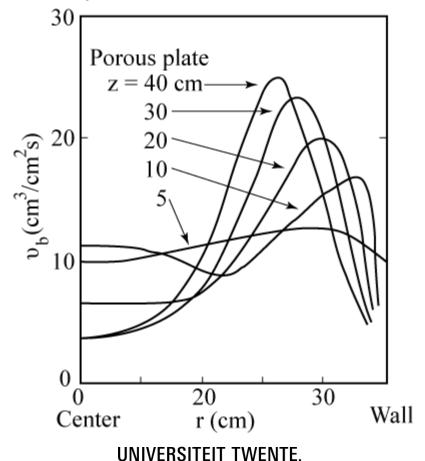
Motion of solids in the vicinity of a rising bubble viewed by observer moving with the bubble (a)
 and stationary observer (b)

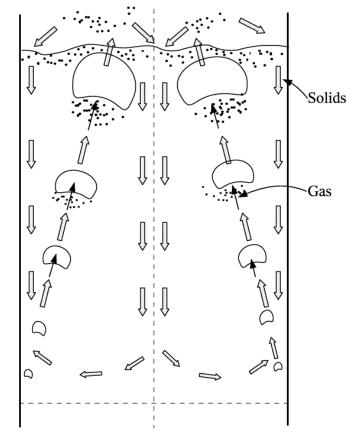
Velocity profile





• Visible bubble flow pattern for Geldart B solids ($d_t=1m$, $d_p=103 \mu m$, $u_{mf}=1.35 cm/s$ and $u_0=20 cm/s$) and general emulsion flow pattern





Emulsion gas flow and voidage

$$\left(\frac{\varepsilon_e}{\varepsilon_{mf}}\right)^3 = \frac{1 - \varepsilon_{mf}}{1 - \varepsilon_e} = \left(\frac{u_e}{u_{mf}}\right)^{0.7}$$

Bubble size correlation due to Mori and Wen (size in cm):

$$\frac{d_{bm} - d_b}{d_{bm} - d_{bo}} = e^{-0.3 \left(\frac{Z}{d_t}\right)} \qquad \text{with} \qquad d_{bm} = 0.65 \left[\frac{\pi}{4} d_t^2 (u_0 - u_{mf})\right]^{0.4}$$

range of experimental conditions:

$$d_t <= 1.3 \text{ m}$$
 0.5<= $u_{mf} <= 20 \text{ cm/s}$
 $60 <= d_p <= 450 \text{ } \mu\text{m}$ $u_o - u_{mf} <= 48 \text{ } cm/s$

Bubble size correlation due to Werther (size in cm):

$$d_b = 0.853 \left[1 + 0.272 \left(u_0 - u_{mf} \right) \right]^{\frac{1}{3}} (1 + 0.0684z)^{1.21}$$

range of experimental conditions (valid for porous plate)

$$d_t>=20 \text{ cm}$$
 1<= $u_{mf}<=8 \text{ cm/s}$
100<= $d_p<=350 \mu m$ 5<= u_0 - $u_{mf}<=30 \text{ cm/s}$

 Werther correlation can be adapted for gas distributor with orifices (initial bubble size not zero in this case):

strategy: fit initial bubble size d_{b0} at initial height of bubble formation z_0

- Bubble rise velocity correlations
 - Geldart A solids with d_t<=1 m (velocity in m/s):</p>

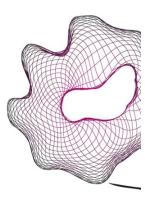
$$u_b = 1.55\{(u_0 - u_{mf}) + 14.1(d_b + 0.005)\}d_t^{0.32} + u_{br}$$

■ Geldart B solids with d_t<= 1m (velocity in m/s):</p>

$$u_b = 1.6\{(u_0 - u_{mf}) + 1.13d_b^{0.5}\}d_t^{1.35} + u_{br}$$

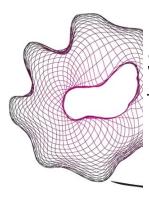
Note that u_{br} denotes the rise velocity of single bubbles, given by:

$$u_{br} = 0.711 \sqrt{gd_b}$$

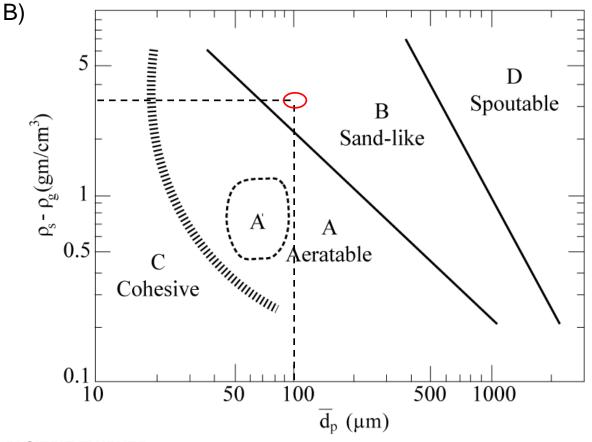


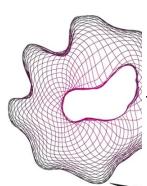
Exercise (10 minutes)

■ Estimate bubble diameter and bubble velocity at height z = 0.5 m in a bed ($d_t = 0.5$ m) of sand (ρ_s - $\rho_g = 3$ g/cm₃, d_p =100 μ m, $u_{mf} = 1$ cm/s, $u_0 = 0.45$ m/s) supported by a perforated plate distibutor (triangular arrangement, $d_{or} = 2$ mm, $l_{or} = 30$ mm)

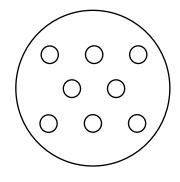


Check which kind of particle we have (from the graph they are Geldart





Calculate the bubble size.



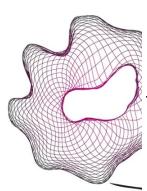
$$N_{or} = \frac{2}{\sqrt{3}l_{or}^2} = 1.3 \cdot 10^3 m^{-2}$$

high gas flow rate touching bubbles $d_{bo}>l_{or}$

$$d_{bo} = \frac{2.78}{g} (u_0 - u_{mf})^2$$

$$= \frac{2.78}{9.81} (0.45 - 0.01)^2 = 0.055m$$

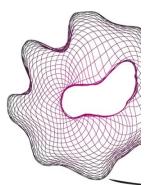
$$= 5.5cm$$



Calculate the bubble size.

$$d_{bm} = 0.65 \left[\frac{\pi}{4} d_t^2 (u_0 - u_{mf}) \right]^{0.4} = 0.65 \left[\frac{\pi}{4} 50^2 (45 - 1) \right]^{0.4}$$
$$= 61.3cm$$

$$\frac{d_{bm} - d_b}{d_{bm} - d_{b0}} = e^{-0.3\left(\frac{Z}{d_t}\right)} = \frac{61.3 - d_b}{61.3 - 5.5}$$
$$= e^{-0.3\left(\frac{50}{50}\right)}$$
$$d_b = 20cm$$



Calculate the bubble velocity.

Geldart B solids with dt<= 1m (velocity in m/s):

$$u_b = 1.6\{(u_0 - u_{mf}) + 1.13d_b^{0.5}\}d_t^{1.35} + u_{br}$$

$$u_b = 1.6\{(0.45 - 0.01) + 1.13 \cdot 0.2^{0.5}\}0.5^{1.35} + 0.711(9.81 \cdot 0.2)^{0.5}$$

$$u_b = 1.59m/s$$



Necessity for flow models:

provide conceptual framework to estimate a.o. volume fractions and phase velocities insight in gas-solids contacting predict performance of bubbling beds in physical and chemical applications

- Many types of models have been proposed:
 - Toomey and Johnstone simple two-phase model
 - Kunii and Levenspiel model

Simple two-phase model:

$$u_{br} = 0.711 \sqrt{gd_b}$$

$$u_e = \frac{u_{mf}}{\varepsilon_{mf}}$$

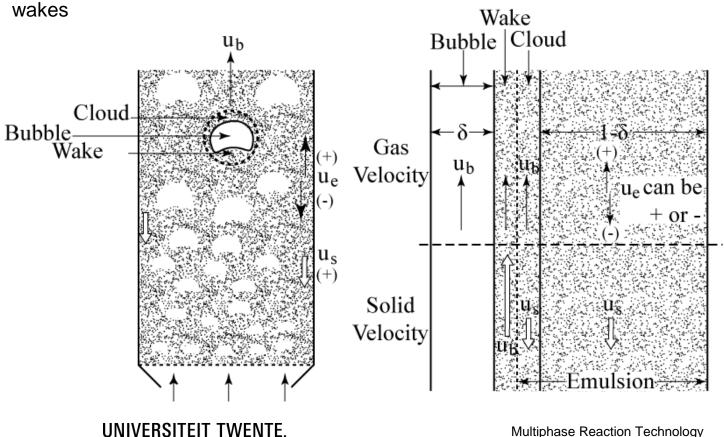
• Superficial rise velocity of emulsion gas:
$$u_{mf}$$

$$u_{s} = u_{s,up} = u_{s,down} = 0$$

$$\delta = \frac{u_0 - u_{mf}}{u_b - u_{mf}}$$

$$1 - \delta = \frac{u_b - u_0}{u_b - u_{mf}}$$

- Kunii and Levenspiel (K-L) model with Davidson bubbles and wakes
- Distinction between slow cloudless bubbles and fast clouded bubbles and accounts for bubble



- Bubble rise velocities
- ✓ small laboratory beds of Geldart A and B solids and any size bed of Geldart D solids use:

$$u_b = u_0 - u_{mf} + u_{br} = u_0 - u_{mf} + 0.711\sqrt{gd_b}$$

✓ large diameter beds:

$$u_b = 1.55\{(u_0 - u_{mf}) + 14.1(d_b + 0.005)\}d_t^{0.32} + u_{br}$$
 Geldart A solids

$$u_b = 1.6\{(u_0 - u_{mf}) + 1.13d_b^{0.5}\}d_t^{1.35} + u_{br}$$
 Geldart B solids

with
$$u_{br}$$
 given by: $u_{br} = 0.711 \sqrt{g d_b}$

- Fraction of bubbles in the bed δ:
 - ✓ Slow bubbles without clouds, or $u_b < u_{mf}/e_{mf}$:

$$\delta = \frac{u_0 - u_{mf}}{u_b + 2u_{mf}}$$

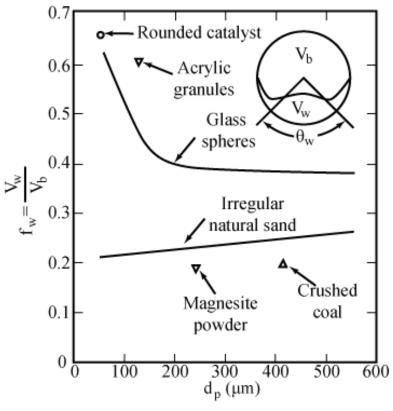
✓ Intermediate bubbles with thick clouds $u_{mf}/e_{mf} < u_b < 5u_{mf}/e_{mf}$:

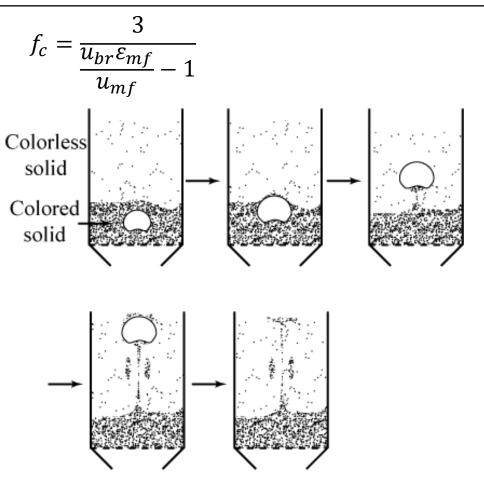
$$\delta = \frac{u_0 - u_{mf}}{u_b + u_{mf}} \qquad \text{(lower boundary)} \qquad \delta = \frac{u_0 - u_{mf}}{u_b} \quad \text{(upper boundary)}$$

 \checkmark Fast bubbles with thin clouds, or $u_b > 5u_{mf}/e_{mf}$:

$$\delta = \frac{u_0 - u_{mf}}{u_b - u_{mf}}$$

- Cloud volume to bubble volume f_c:
- Wake volume to bubble volume f_w:





Fraction of bed in emulsion:

$$f_e = 1 - \delta - f_w \delta$$

Definition of solids distribution:

 γ_b , γ_c and γ_e denote respectively volume of solids dispersed in bubble, cloud and emulsion divided by volume of bubble

$$\gamma_b$$
, γ_c and $\gamma_e = \frac{volume\ of\ solids\ dispersed\ in\ b, c, e}{volume\ of\ bubble}$

Volume of solids in emulsion divided by bubble volume γ e:

$$\gamma_e = \frac{(1 - \varepsilon_{mf})(1 - \delta)}{\delta} - \gamma_b - \gamma_c$$

Volume of solids in cloud (+ wake) divided by bubble volume γ_c :

$$\gamma_c = (1 - \varepsilon_{mf})(f_c + f_w)$$

$$= (1 - \varepsilon_{mf}) \left[\frac{3}{u_{br} \varepsilon_{mf}} + f_w \right]$$

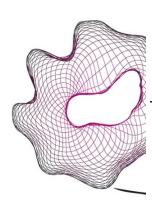
• Volume of solids in bubble divided by bubble volume γ_b :

$$\gamma_b = 0.005$$
 from experimental observations

- Rise velocity of wake solids $u_{s,wake}$: $u_{s,wake} = u_b$
- Downflow velocity of emulsion solids $u_{s,down}$: $u_{s,down} = \frac{f_w \delta}{1 \delta f_w \delta} u_b$

Rise velocity of emulsion gas through bed u_e:

$$u_e = \frac{u_{mf}}{\varepsilon_{mf}} - u_{s,down}$$
 criterion for downflow of emulsion gas $\frac{u_b}{u_{mf}} > \frac{1 - \delta - f_w \delta}{f_w \varepsilon_{mf} \delta}$

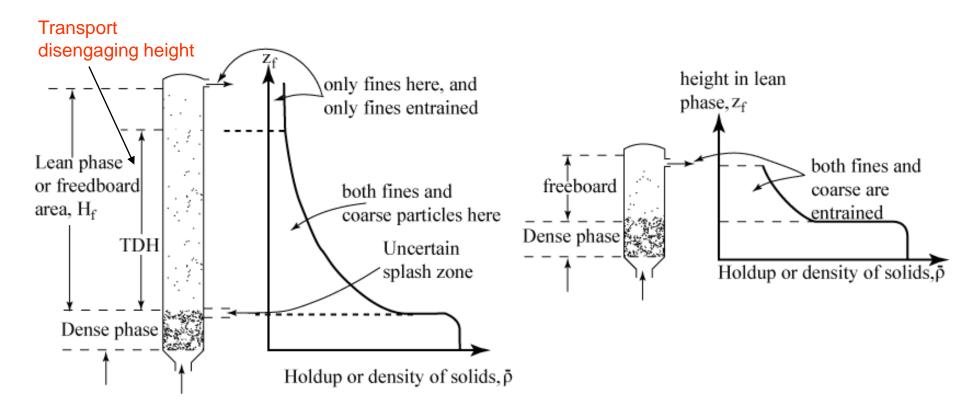


Entrainment and Elutriation

✓ Entrainment: solids flux Gs (kg/m².s) carried by gas at exit of vessel

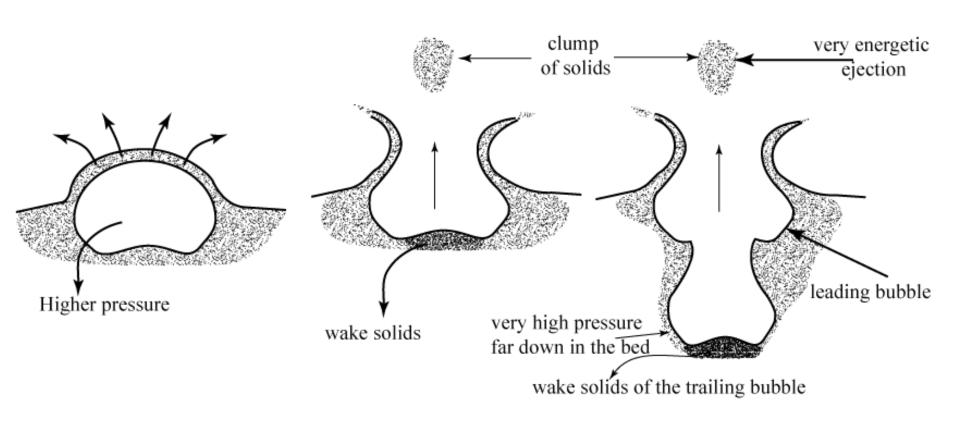
✓ Elutriation: separation or removal of fines from mixture

Dense Fluidized Beds Entrainment and elutriation



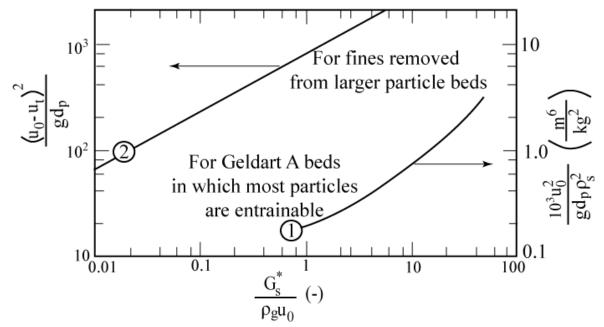
Dense Fluidized Beds Entrainment and elutriation

Mechanism of solids ejection into freeboard



Dense Fluidized Beds Entrainment and elutriation

- Procedure for calculating the entrainment flux G_s (kg/m².s) from tall vessels (H_f>TDH)
 - Divide PSD in narrow size interval and find size intervals with: u_t<u_o
 - Find G^{*}_{s,i} from figure below for relevant system



Dense Fluidized Beds Entrainment and elutriation

■ Total entrainment flux G_s (kg/m².s):

$$G_{S} = \sum_{\substack{elutriable \\ intervals}} x_{i} G_{S,i}^{*}$$
 $G_{S} = \int_{\substack{elutriable \\ intervals}} G_{S}^{*}(d_{p}) p d(d_{p})$

discrete PSD

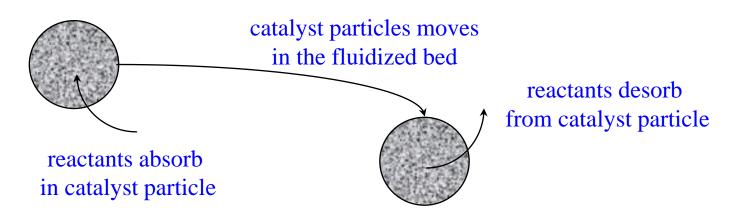
continuous PSD

x_i: mass fraction for size interval i

Entrainment flux G_s (kg/m².s) from short vessels (H_f< TDH)

Kunii and Levenspiel freeboard-entrainment model (domain of Fast Fluidization)

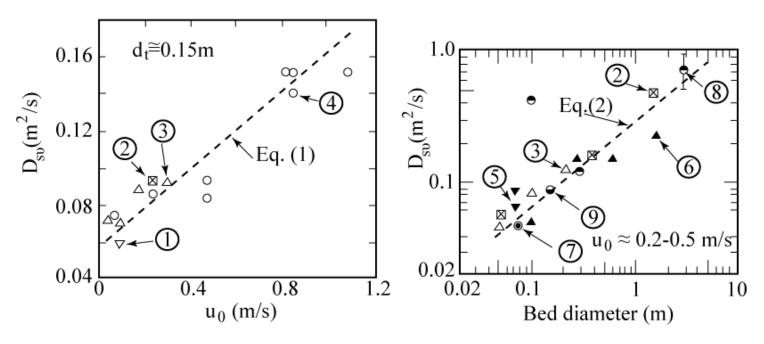
- Topics covered:
 - ✓ Vertical mixing and segregation of solids
 - ✓ Horizontal mixing and dispersion of solids
 - ✓ Mixing-segregation equilibrium
 - Relevance of solid movement/mixing for catalytic reactors



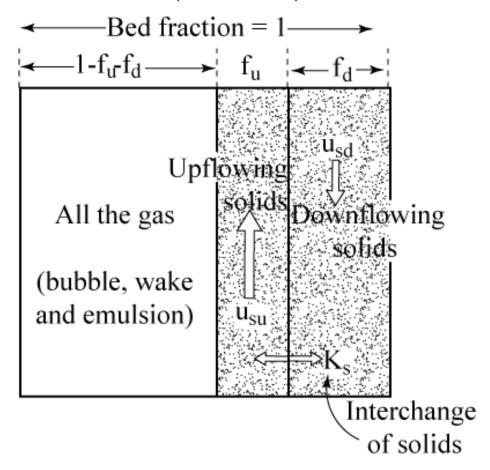
Vertical solids mixing: often reasonably well described by the (axial) dispersion model:

$$\frac{\partial}{\partial t}C_s = D_{sv}\frac{\partial^2}{\partial x^2}C_s$$

Experimental results for axial dispersion coefficient for solids D_{sv}:



Counterflow solid circulation model (van Deemter):



- Counterflow solid circulation model (van Deemter):
 - Mass balance downflowing solids (fraction f_d):

$$f_d \frac{\partial}{\partial t} C_{sd} + f_d u_{sd} \frac{\partial}{\partial z} C_{sd} + K_s (C_{sd} - C_{su}) = 0$$

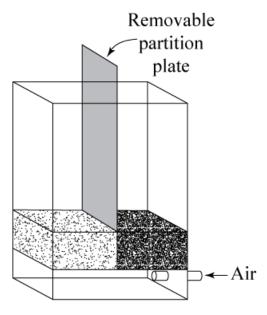
Mass balance upflowing solids (fraction f_u):

$$f_u \frac{\partial}{\partial t} C_{su} + f_u u_{su} \frac{\partial}{\partial z} C_{su} + K_s (C_{su} - C_{sd}) = 0$$

Effective dispersion coefficient for tall beds of fine particles:

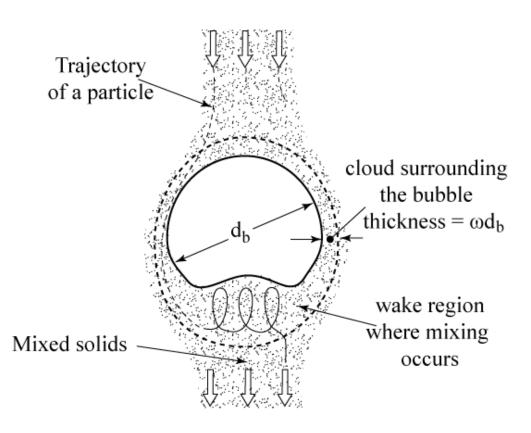
$$D_{sv} = \frac{f_d^2 u_{sd}^2}{K_s (f_d + f_u)} = \frac{f_d^2 u_{sd}^2}{K_s (1 - \delta)(1 - \varepsilon_f)}$$

- Horizontal mixing and dispersion of solids
 - ✓ Experimental setup for horizontal solids dispersion coefficient D_{sh}



• General finding: vertical mixing rate considerably faster than horizontal mixing rate $D_{sv}(0.1-0.4 \text{ m}^2/\text{s}) - D_{sh}(10-30 \text{ cm}^2/\text{s})$

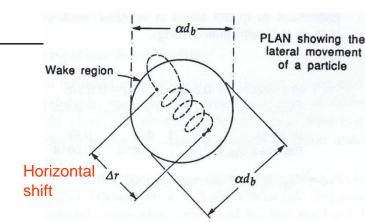
- Mechanistic model for D_{sh} based on the Davidson bubble
 - Particles moving in the cloud
 region are caught in the bubble wake
 - Uniform mixing of these particles in the bubble wake
 - Finally particles leave
 bubble wake at random positions



Probability theory:

$$\overline{\Delta r^2} = \frac{(\alpha d_b)^2}{4}$$

αd_b represents the effective wake diameter



Final expression for D_{sh} for both fast and intermediate bubbles:

$$D_{sh} = \frac{3}{16} \frac{\delta}{1 - \delta} \frac{\alpha^2 u_{mf} d_b}{\varepsilon_{mf}}$$

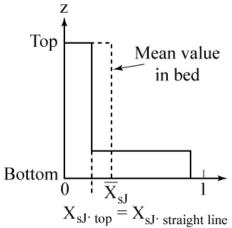
- Comparison with experimental data:
 - ✓ For Geldart A and AB solids: good fit of experiments with α =1
 - ✓ For Geldart BD solids: good fit of experiments with α =0.77

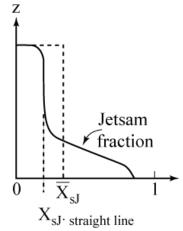
- Mixing-segregation equilibrium
 - Relevant for many industrial applications involving (large) size and/or density differences in the fluidized bed
 - The large and/or more dense particles tend to reside at the bottom of the fluidized bed (especially near umf of these large/more dense particles)
- Terminology:
 - Jetsam: component that ultimately sinks
 - Flotsam: component that floats to the top of the bed

Definition of solids mixing index M (ratio of jetsam in top portion of the bed and (average) jetsam fraction in well-mixed bed):

$$M = \frac{X_{SJ,top}}{\bar{X}_{SJ}}$$

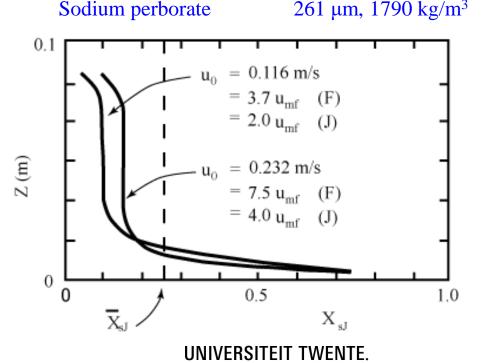
- Solids distribution in a binary mixture:
 - left: perfect segregation (at low u)
 - right: segregation under vigorous bubbling conditions (at high u)





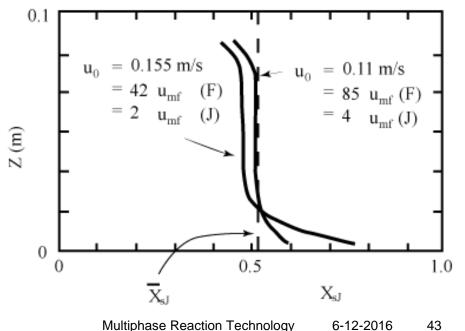
Vertical segregation of commercial solids (d_t=0.141 m, L_m=0.10-0.15 m). left: density difference; right: size difference

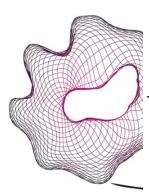
Persi washing powder base 237 μm, 1020 kg/m³ Sodium perborate 261 μm, 1790 kg/m³



Cracking catalyst 60.4 μm, 2198 kg/m³

Alumina catalyst 210 μm, 2100 kg/m3





Resume

- Different types of distributors
- Distributor determines the initial bubble size
- Bubble velocity/dimensions change depending on the position and on the characteristics of solids
- Bubbles have wakes which are responsible of solid movements/mixing
- Entrainment of solids should be taken into account when designing
 FB
- Mixing and segregation of solids can be important for chemical reactions in FB