

# PROVINCIAL EXAMINATION JUNE 2022 GRADE 9

## **NATURAL SCIENCES**

NAME OF	LEARNER:
GRADE 9:	
TIME:	2 hours
MARKS:	100

19 pages + 1 data sheet

QUESTION	1	2	3	4	5	6	7	TOTAL
LEARNER'S MARK								
MARKS	22	11	12	10	11	22	12	100

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#### INSTRUCTIONS AND INFORMATION

- 1. Write your name and grade on the cover page of this question paper that serves as an ANSWER BOOK.
- 2. Answer ALL questions in the spaces provided.
- 3. This question paper consists of SECTION A and SECTION B based on the prescribed content framework in the CAPS Document.
- 4. Allocation of marks:

SECTION A: 22 SECTION B: 78

- 5. This question paper consists of SEVEN questions.
- 6. All drawings should be done in pencil and must be labelled in blue or black ink.
- 7. Write neatly and legibly.

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## **SECTION A**

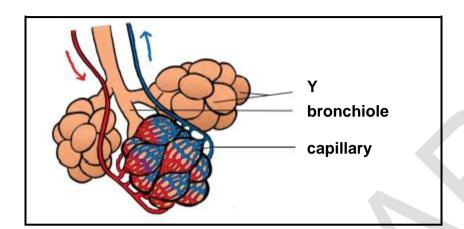
## **QUESTION 1**

## **MULTIPLE-CHOICE QUESTIONS**

1.1		the cor	is are provided as possible answers to the following questions. The option by writing the correct letter $(A - D)$ in the blocks	
	1.1.1	When	substances enter or leave any kind of cell, it must pass through	
		A B C D	Golgi apparatus. ribosome. cell wall. cell membrane.	(1)
	1.1.2	When	an organism grows, it is because of	
		A B C D	an uptake of water. an increase in mass. cells that are continually dividing. an uptake of food.	(1)
	1.1.3	Which	n of the following is an unspecialised cell?	
		A B C D	Sperm cell Nerve cell Stem cell Red blood cell	(1)
	1.1.4	Which	of the following is NOT a function of the tongue?	
		A B C	Secretes enzymes that digest starch in food Chewing of food and mixing it with saliva Forms a food bolus Helps with the swallowing process	(1)

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1.1.5 Study the diagram below and choose the correct name for the part labelled **Y**.



- A Blood
- B Vein
- C Alveoli
- D Urine

(1)

1.1.6 Study the following statements.

Exhaled air differs from inhaled air in the following respects:

- 1 Contains more carbon dioxide
- 2 Contains less oxygen
- 3 Contains more water vapour

Choose the correct combination.

- A 1 and 2 only.
- B 2 and 3 only.
- C 1 and 3 only.
- D 1, 2 and 3.

(1)

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1.1.7	What is the function of the placenta?			
	A B C D	Provides nutrition to the foetus and removes waste products Transmits hereditary material between the mother and foetus Helps in identifying the gender of an unborn child An unborn child holds onto it to prevent being born prematurely	(1)	
1.1.8	Why a	re the elements Mg and Ca in the same group?		
	A B C D	They have similar chemical properties. Their atomic numbers are the same. Both are non-metals. Both are gases.	(1)	
1.1.9		tio in which the atoms of sulphur trioxide combine to form a ule can be represented by:		
	A B C D	2:1:3 3:1 1:3 3:3	(1)	
1.1.10	Why is	a universal indicator the most recommended indicator?		
	A B C D	It does not change colour at all. It only works in an acid. It is homemade. It functions in a wide range of pH values.	(1) <b>[10]</b>	

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## **TERMINOLOGY**

.2		e correct scientific term for each of the following descriptions. Write only n in the spaces provided.
	1.2.1	The living, outermost membrane which surrounds the cytoplasm of cells
	1.2.2	The process whereby food is broken down into smaller particles by enzymes and acids in the mouth, stomach and intestines
	1.2.3	The blood vessel which carries deoxygenated blood away from the heart
	1.2.4	The cord which carries blood between the foetus and the placenta in humans
	1.2.5	A type of chemical reaction where a substance and oxygen react during burning to form a new product
	1.2.6	A non-metal that will not react with oxygen

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#### **MATCHING ITEMS**

1.3 Choose an item from COLUMN B that matches a statement in COLUMN A. Write only the letter (A – H) next to the question numbers (1.3.1 – 1.3.6) in the spaces provided.

	COLUMN A	COLUMN	В
1.3.1	Lasts for 28 days	A Abortion	1.3.1
1.3.2	Method of pregnancy prevention	B Oxygen	1.3.2
1.3.3	The inflammation of the lining of the bronchial tubes, which carry air to and from the lungs	C Menstrual cycle	1.3.3
1.3.4	A gas released when an acid reacts with a metal	D NaOH+HCI	1.3.4
1.3.5	Ignites with a popping sound	E Abstinence	1.3.5
1.3.6	Neutralisation process	F Asthma	1.3.6
		G Bronchitis	
		H H <sub>2</sub>	

(1 x 6) **[6]** 

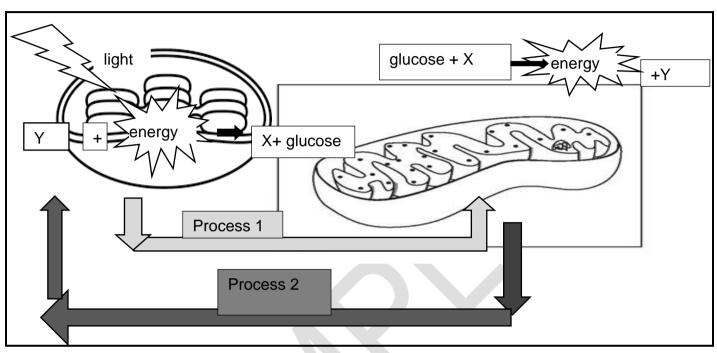
TOTAL SECTION A: 22

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#### **SECTION B**

#### **QUESTION 2**

The diagram below shows the link between two important organelles and the processes that they perform. Study the diagram below and answer the questions that follow.



2.1	Name	the organelles in which the following takes place:
	2.1.1	Process 1(1)
	2.1.2	Process 2 (1)
2.2	Identify	y the processes represented by:
	2.2.1	Process 1(1)
	2.2.2	Process 2(1)
	2.2.3	The glucose produced in Process 1 is used in Process 2. What is the gas represented by <b>Y</b> ? (1)
	2.2.4	Energy is released in Process 2. What is the gas represented by <b>X</b> ? (1)
	2.2.5	Why is this gas mentioned in QUESTION 2.2.4 important to humans? (1)

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2.3 Complete the table below by indicating with a CROSS (X) whether the organelle or component will be present in a PLANT CELL ONLY or an ANIMAL CELL ONLY or in BOTH PLANT AND ANIMAL CELLS.

	Present in:			
Organelle or component	PLANT CELL ONLY	ANIMAL CELL ONLY	BOTH PLANT AND ANIMAL CELLS	
Cell wall				
Cell membrane				
Cytoplasm				
A few small vacuoles or completely absent				

(4) **[11]** 

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#### **QUESTION 3**

A group of Grade 9 learners investigated the effect different types of exercises have on the heart rate. They measured the heart rate of 5 learners after each of the following activities:

- Rest
- Brisk walking for 5 minutes
- Jogging for 2 minutes
- Jumping with a skipping rope for 2 minutes.

They recorded their results in the table shown below.

	Heart rate (beats per minute) after each activity					
	Rest Brisk walking Jogging Jumping					
Learner 1	66	72	80	98		
Learner 2	72	72	86	100		
Learner 3	72	74	82	108		
Learner 4	68	74	84	96		
Learner 5	78	86	100	120		
AVERAGE	71,2	75,6	86,4	Х		

AV	ERAGE	71,2	75,6	86,4	X	
3.1	Identify th	e dependant varia	able.			(1
3.2	Calculate skipping r		the table (average l	neart rate for jump	ing with a	
						(2)

		1
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the differen	types of exercis	ses. 		

Criteria	Marks
Heading/Title	1
Axes labelled	2
Plotting of data	3

3.4	Use the graph to identify the type of exercise that is the most effective in
	increasing the heart rate.

3.5 Suggest ONE reason why a person's heart rate changes during exercise.

Suggest ONE reason why a person's heart rate changes during exercise.				
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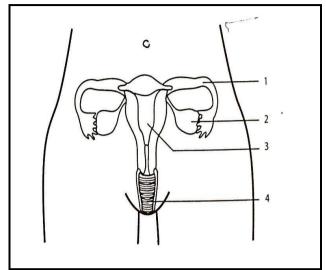
(2) **[12]** 

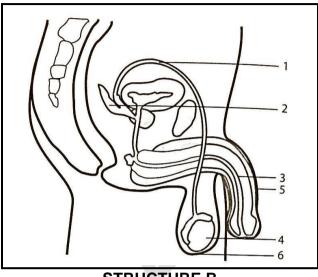
(6)

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#### **QUESTION 4**

Study the diagrams below and answer the questions that follow.





STRUCTURE A

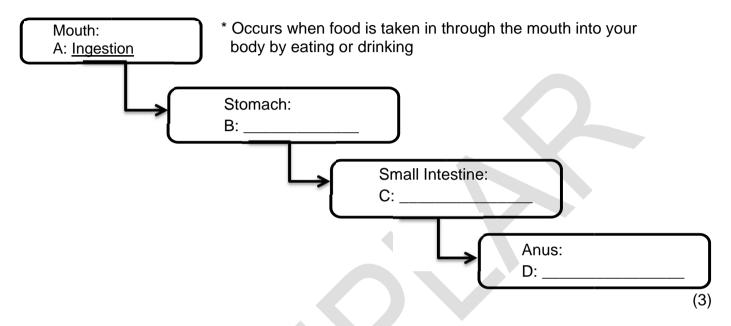
STRUCTURE B

4.1.1 Identify the heading for structure **A** and structure **B**.

	Structure A	(1)
	Structure B	(1)
4.1.2	Label the following parts in structure <b>B</b> :	
	(a) Part 1	(1)
	(b) Part 6	(1)
4.1.3	Give the number and the name of the part in structure <b>B</b> that:	
	(a) Transports both semen and urine	(2)
	(b) Produces testosterone	(2)
4.1.4	Label the part marked <b>3</b> in structure <b>A</b> .  Part 3:	(1)
4.1.5	Name the part of structure <b>A</b> where fertilisation take place.	(1 <u>)</u> <b>[10</b> ]
4.1.4	(a) Transports both semen and urine  (b) Produces testosterone  Label the part marked 3 in structure A.  Part 3:	

#### **QUESTION 5**

5.1 Below is a flow diagram of the processes within the digestive system. Study the flow diagram and complete it by writing the missing processes next to each letter which indicates where they take place.



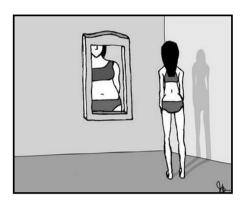
5.2	Describe what happens	in each of the	processes from	m <b>B</b> to <b>D</b> .
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:	
	(;
	_
	(:
	`

(2)

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5.3 The picture below shows a person with an eating disorder that is associated with the digestive system.



5.3.1 Name the disorder that is shown in the picture above.

(1)

5.3.2 Why should you always consult a doctor or healthcare worker before going on a weight loss diet?

(1)

[11]

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## **QUESTION 6**

17 CI 35,45

6.1	What i	is atomic mass of the element above?	(1)
6.2	Give th	he element's atomic number.	(1)
6.3	Is the	element a metal, a non-metal or a semi-metal?	(1)
6.4	Write	down the group number of the element in the periodic table.	(1)
6.5	6.5.1	Write down the name and the chemical formula of the product formed when the element used in QUESTION 6 reacts with magnesium.	(2)
	6.5.2	Write down the formula for the diatomic molecule formed by the element mentioned above.	(1)

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6.6 Answer the following questions.

6.6.1	Differentiate between a pure substance and a	mixture.

(2)

6.6.2 Balance the following chemical equations.

(a) Na + O<sub>2</sub> 
$$\rightarrow$$
 Na<sub>2</sub>O

(2)

(b) 
$$H_2 + O_2 \rightarrow H_2O$$

(2)

Rust is a process that occurs naturally and its effects can be quite beautiful but also very harmful and dangerous.



6.7.1 What is rusting and how is it undesirable for us?

(3)

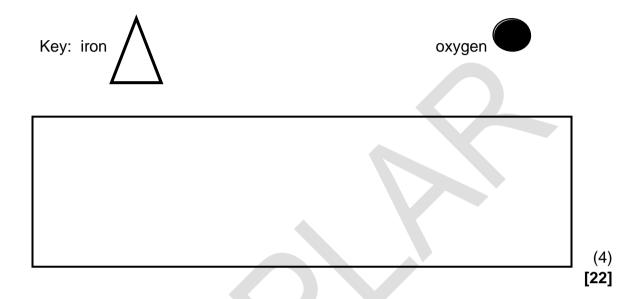
6.7.2 List TWO ways to prevent iron from rusting.

(2)

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6.7.3 Study the key below and draw a picture of a balanced equation for the reaction between iron and oxygen in the space provided.

$$Fe + O_2 \rightarrow Fe_2O_3$$

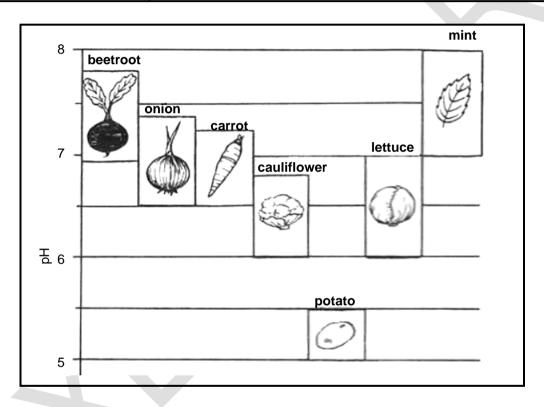


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#### **QUESTION 7**

Read the case study below and answer the questions that follow.

Different soils have different pH levels. Some soils are acidic and others are alkaline. The pH of a soil depends on the rock from which the soil was formed, and the plant remains that are in the soil. Soils from limestone are alkaline with a pH of about 8. Clay soils with decomposing plant material may have a more acidic pH of about 4 or 5. Some plants grow better in acidic soils and others grow better in neutral or alkaline soils. Most plants grow well in soils with a pH of 6,5. If the pH of the soil is not right, the plant will not grow very well. For this reason, farmers change the pH of the soil so that their crops will grow well. Acidic soil can be neutralised by adding powdered limestone or lime to the soil. Lime is a metal hydroxide and its chemical name is calcium hydroxide.



- 7.1.1 In what pH range will potatoes grow well?
- 7.1.2 If the soil had a pH of 7,6, which crop (plant) could be grown in that soil?
- 7.1.3 If the soil was found to have a pH of 6,5, which TWO kinds of plants could a farmer grow in that soil?

  (2)

7.1.5		e down the general word equation to show the reaction that happens n soil is neutralised.
7.1.6	Write study	e down the chemical formula for the hydroxide mentioned in the case
7.1.7	Iden	tify the colour of the universal indicator in the soil samples of:
	(a)	Limestone
	(b)	Clay soil
	(c)	Neutral soil
		TOTAL SECTION E
		TOTAL

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#### TABLE 3: THE PERIODIC TABLE OF ELEMENTS/TABEL 3: DIE PERIODIEKE TABEL VAN ELEMENTE

	1 (l)		2 (II)		3		4	5	6	7	8	9	10	11	12	13 (III)	14 (IV)	15 (V)	16 (VI)	17 (VII)	18 (VIII)
2,1	1 H 1							KEY/SLE	UTEL	<b>A</b>	tomic nu Atoomg ↓										2 He 4
1,0	3 Li 7	1,5	4 Be 9						onegativ negatiw		29 Cu 63,5		nbol nbool			5'0 B 11	2,5 C 12	7 0: N 14	8 9. 0 16	0.4 19 6 8	10 Ne 20
6'0	23	1,2	12 Mg 24						Benad	erde rela	elative a	oomma	ssa			27	14 © Si 28	15 P 31	32 32 32	17 Cl 35,5	18 Ar 40
8,0	19 K 39	1,0	20 Ca 40	1,3	21 Sc 45	1,5	22 Ti 48	9. V 51	9. Cr 52	25 Mn 55	26 Fe 56	27 © Co 59	28 Ni 59	29 Cu 63,5	9 Zn 65	31 9 Ga 70	73	33 O As 75	79	35 87 80	36 Kr 84
8,0	37 Rb 86	1,0	38 Sr 88	1,2	39 Y 89	1,4	40 Zr 91	92	% Mo 96		101	103	106	47 Ag 108	112	49 In 115	119	122	128	53 127	54 Xe 131
2,0	55 Cs 133	6,0	56 Ba 137		57 La 139	1,6	72 Hf 179	73 Ta 181	74 W 184	75 Re 186	76 Os 190	77 Ir 192	78 Pt 195	79 Au 197	80 Hg 201	81 ∞ Tℓ 204	82 ∞ Pb 207	83 ල Bi 209	84 0.0 Po	85 95 At	86 Rn
2,0	87 Fr	6,0	88 Ra 226		89 Ac			58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
								140 90 Th 232	91 Pa	144 92 U 238	93 Np	150 94 Pu	152 95 Am	157 96 Cm	159 97 Bk	163 98 Cf	165 99 Es	167 100 Fm	169 101 Md	173 102 No	175 103 Lr