

COMMON UNIT CONVERSIONS (C1)

<i>Unit</i>	Unit Symbol	Quantity	Equation Expressed in Terms of SI Base Units	Equation Expressed in Terms of Other Units
<i>Degree Celsius</i>	°C	Celsius Temperature	$x\text{ }^{\circ}\text{C} = x\text{ K,}$ where $-273.15\text{ }^{\circ}\text{C} \equiv 0\text{ K}$	$x\text{ }^{\circ}\text{C} \equiv (x - 32) \times \frac{5}{9}\text{ }^{\circ}\text{F}$
<i>Degree Fahrenheit</i>	°F	Fahrenheit Temperature	$x\text{ }^{\circ}\text{F} \equiv \left(x \times \frac{9}{5} - 459.67\right)\text{ K}$	$x\text{ }^{\circ}\text{F} \equiv \left(x \times \frac{9}{5} + 32\right)\text{ }^{\circ}\text{C}$
<i>Atomic Mass Unit</i>	u Da	Atomic Mass	$x\text{ u} \equiv 1.660\,539\,066\,60(50) \times 10^{-27}x\text{ kg}$	-
<i>Atmospheric Pressure</i>	atm	Atmospheric Pressure of Earth	$x\text{ atm} \equiv 101\,325x\text{ Pa}$	-
<i>Electronvolt (Mass)</i>	eV/c ²	Mass	$x\text{ eV}/c^2 \equiv 1.782\,661\,92 \times 10^{-36}x\text{ kg}$	$x\text{ eV}/c^2 \equiv 96\,385\,542.168\,675x\text{ u}$
<i>Electronvolt (Energy)</i>	eV	Energy	$x\text{ eV} \equiv 1.602\,176\,634 \times 10^{-19}x\text{ J}$	-
<i>Elementary Charge</i>	<i>e</i>	Charge	$x\text{ }e \equiv 1.602\,176\,634 \times 10^{-19}x\text{ C}$	-

Sources:

- Unit
- Unit Symbol
- Quantity
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