

Properties of Electric Charges.

i) Like charge repeal each other whereas, unlike charge attract eachother. il electric charges follow the law of conservation of charges. iii electric charges are quantized that is q = + ne. The electric charges are integral multiple of charges of electron. iv) Electric charges are scalar quantities that is they doesn't follow vetor law.

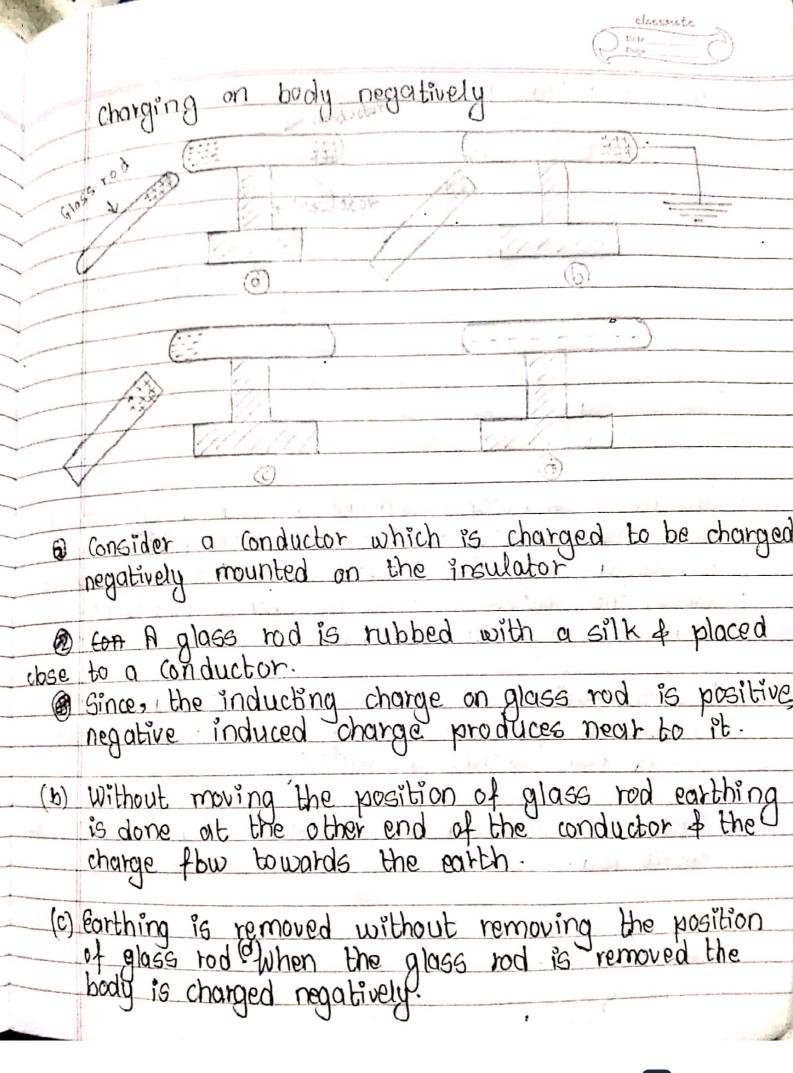
v) The magnitude of charges are not affected by the speed of the body. Electrification (charging a body)

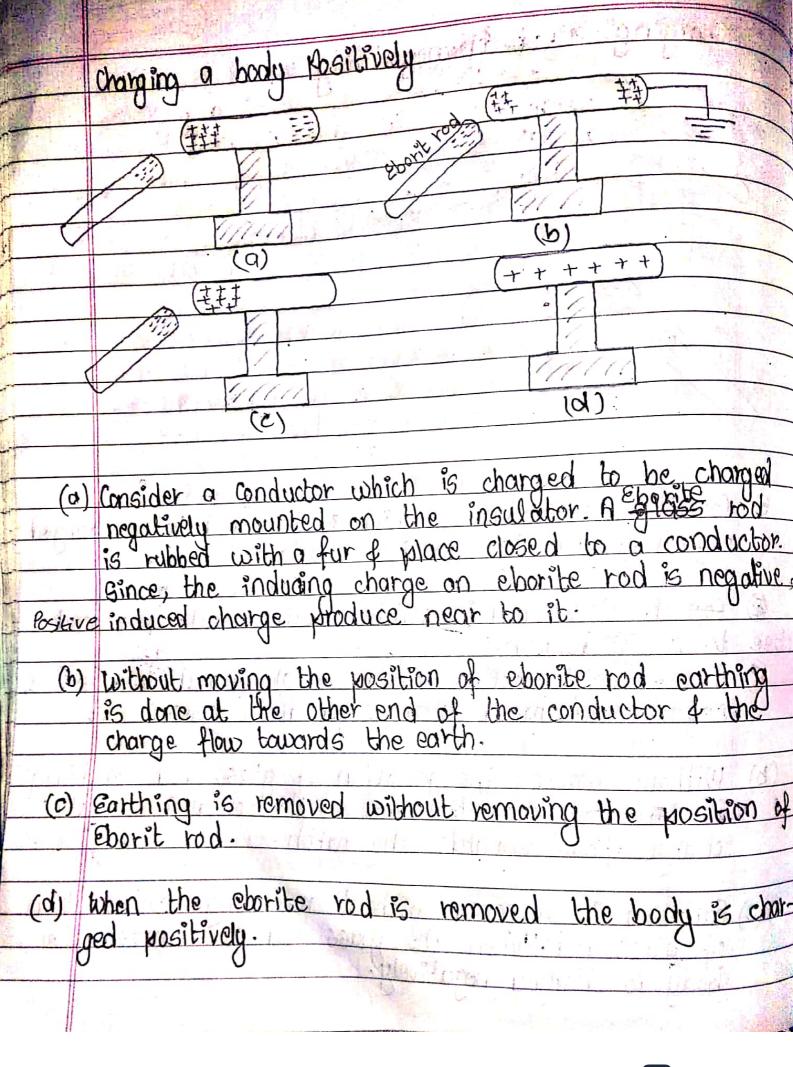
> The phenomenon of the production of electric charge in body is called electrification. In general, there are three methods of electrification. i) By fiction (ii) By conductor iii) By Induction 1. Charging a body by friction: When two bodies are trobbed with eachother & charges are produced then it is known as charging a body by friction. For eq:

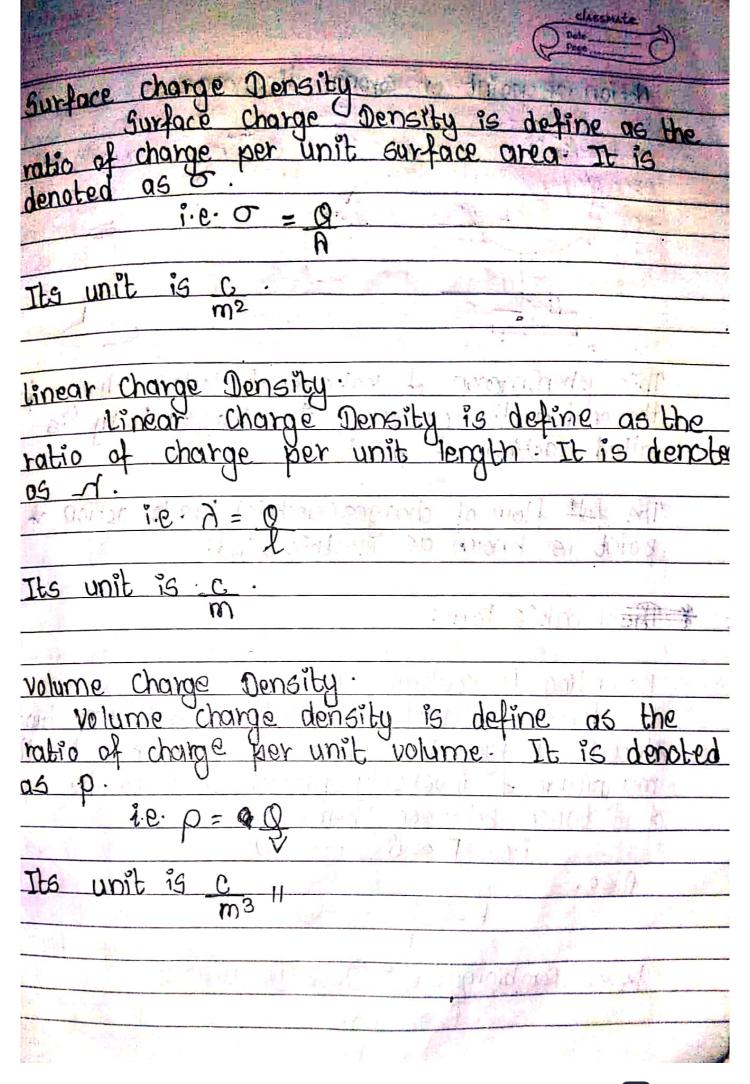
when silk is robbed with Glass rod glass rod is charged positively & silk negatively.

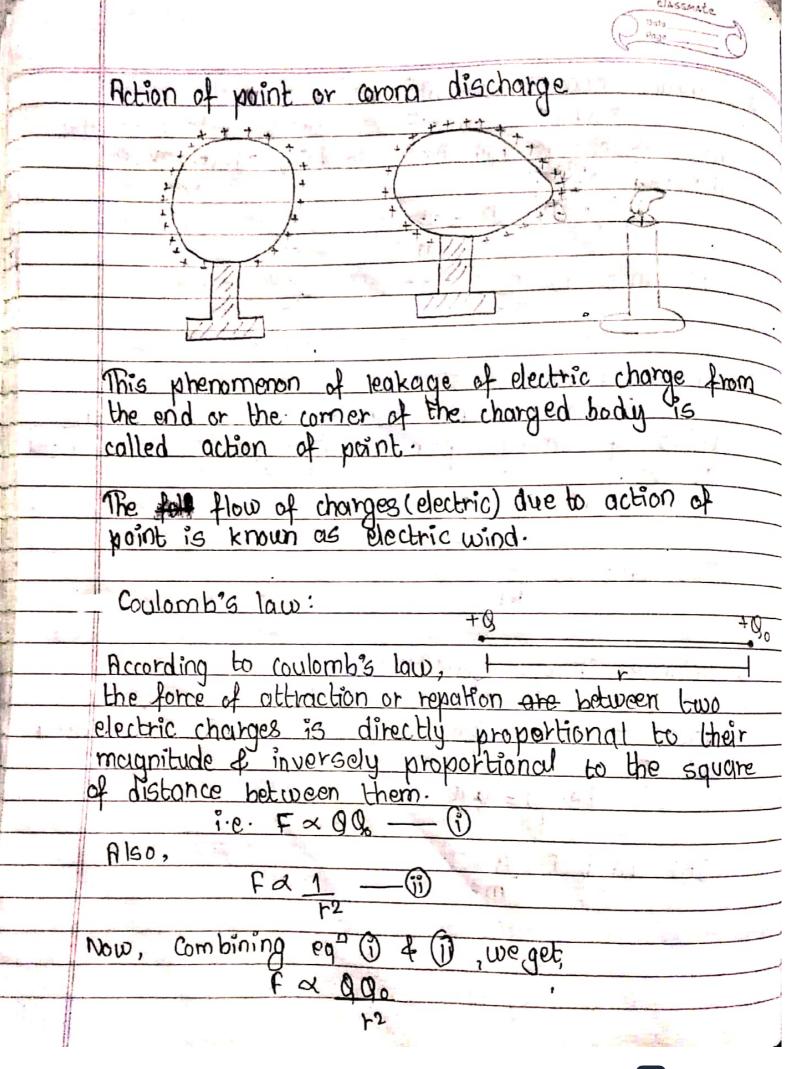
2. Charging a body by conductor: When a charged body is placed in contant with uncharged body then it

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	and the second section
Elg.	Electrification by Induction: The temperary electrification
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100	Inducing Charge or bound charge
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1.	Inducing Charge Induced Bound charge stree charge
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	to charge another body when kept stand.
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	fig: Electrosatic induction of mal
2.	Hound charge charge
->	The charge produce on a body when inducing charge is brought close to it is called Bound Charge
W. A.	brought close to it is called Bound Change
• 3-	Free Charge
->1	he charge produce on a body of the other and is
	he charge produce on a body at the other end is known as free charge. They are free to flow.
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or, P = K 990 or, F = 1 x 000 - 1 x 100 where en is the permitivity of vacuum or free where k is a proportionality constant where ese value depends upon volume or media.

[K: 9×103 Nm2s-2] A130, or,  $\epsilon_0 = 8.85 \times 10^{-12} c^2 N^{-1} m^{-2}$ or, cq6 System, k = 1

1 Coulomb's : 1 coulomb is that electric charge with a force which repeals on equal electric charge with a force of 9×10<sup>19</sup> N when placed of a distance of 1m in our or vaccum.

Relative Permittivity or Dielectric Constant:

Mostive Permittivity is defined as the ration of Permittivity of medium to permittivity of vacuum. It is denoted by K or Er.

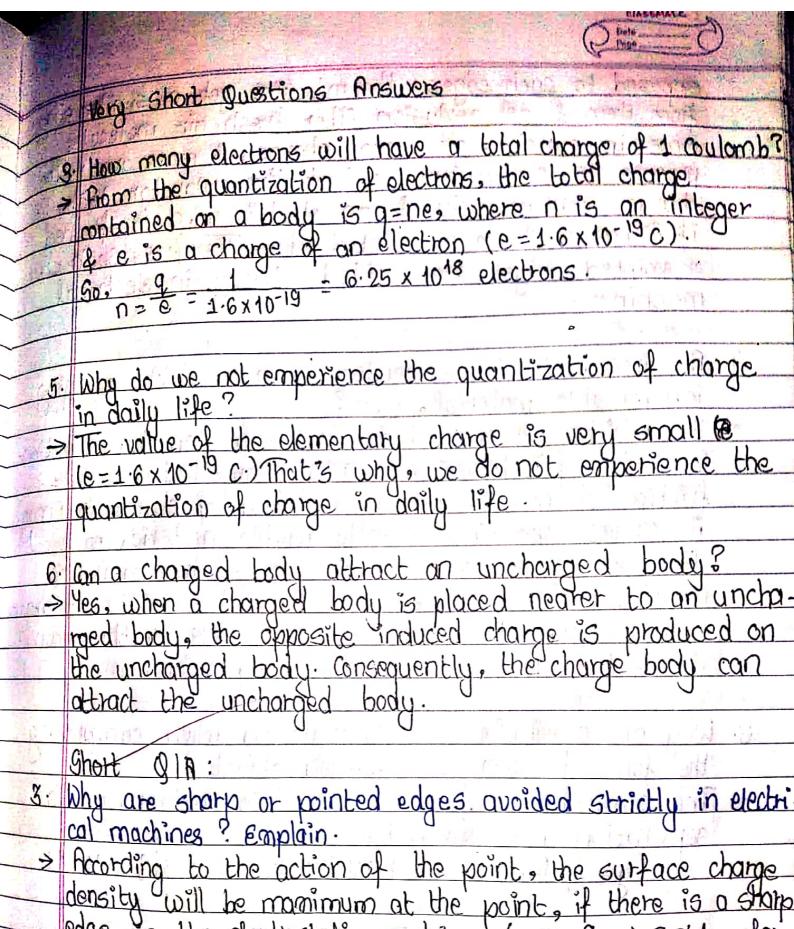
Mothematically,

K = Er = Permittivity of medium

Permittivity of vacuum

trto=t or, Er= E we know, force between two charges when placed at vacum, Fr 1 990 --- (3) 160 100 --- (ii) Fm = 1 Force botween two charges when placed at medium 4,50 Now, Dividing egr (i) by (ii) we get,

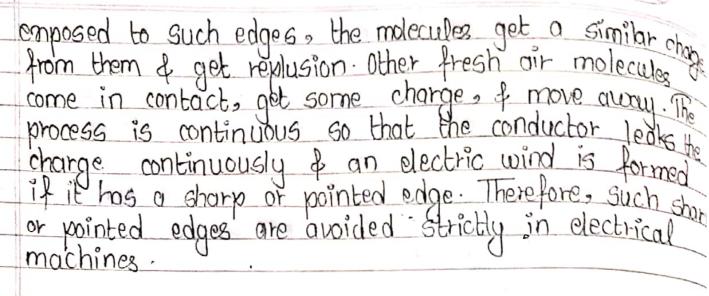
Fr = 4760 900



edge in the electrostatic machine! (:6=9 >001, for

constant charge.) Due to this, when air molecules are

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4. An iron chain is suspended from a vehicle carrying inflammable material, why?

The vehicles get charged due to the friction during motion. The body of vehicles also gets charged due to the friction of oir. If the accumulation of charge is enough to causes speaking, of inflammable material may attribute to the fire of any analysis may become like an inspect fire, of an emplosion may happen. When an iton of metallic chain is suspended from the vehicle, it drags charges from the body of the vehicle to the ground. Hence, the iton chain prevents emplosion.

6. Why are gravitational forces neglected when computing the force between the charged object?

The gravitational force between two electrons with

separation r is given by  $fg = G Mm = 6.67 \times 10^{-11} (9.1 \times 10^{-31}) \times (9.1 \times 10^{-31})$ 

= 5.52 × 10-71 N.

Similarly, the electrostatic force between two electronic

- charges separated by the distance ris

Fe = 1 9192 - 9×109 (1.6×10-19)× (1.6×10-19)

4x60 r2 r2 - 2.3 × 10-28 N On dividing a ux get,  $f_{9} = 2.4 \times 10^{-43} = 10^{-43}$ from the above discussion it is clear that the anatolism force is very small than that of the electrostatic time.
Therefore, gravitational forces are neglected while comparting the force between charged objects. 7. How for apart should two electrons be placed if the force emerted on each is equal to the weight of the election in free space? -> Suppose d'be the seperation between two electrons suppose the control of each is equal to the most of the electron in free space, i.e. weight of electron = force between two electrons in some or,  $mg = k \frac{q^2}{d^2}$  (:  $k = 1 - \frac{q \times 10^9}{4 \pi \epsilon_0}$ where m & a are mass & charge on an electron respectively. Thus, the required value of separation between two electrons is 5 m.

2. What is the total charge of 1 kg of electrons? Electronic charge (e) = 1.6×10 - 19c Moss of an electron (me) = 9.1 × 10-31 kg Mass (m) = 1 kg charge of 1 kg electrons (q) = ? Gince the no. of electrons in 1 kg is

n = m = 1 1 x 10<sup>30</sup> electrons

me 9.1 x 10<sup>-31</sup> Go the charge of 1 kg of electrons becomes  $q = ne = 1.1 \times 10^{30} \times 1.6 \times 10^{-19} = 1.76 \times 10^{11} \text{ C}$ Hence, the total charge of 1 kg of electrons is 1.76 x 104c. 3. ratculate the value of two equal charges if they reveal by a force equal to the weight of 45 kg person when situated 1m a part in a vacum (Eo = 8.85 x 10<sup>12</sup> C<sup>2</sup>m<sup>-2</sup>N<sup>-1</sup>) Solo: Givens Force between two charge (F) = 45 kg = 45 x 9.8 N = 441N First charge = 9.1 Second Charge = 9.2 Eo = 8.85 x 10<sup>-12</sup> c<sup>2</sup>m<sup>-2</sup>N<sup>1</sup> Separation (r) = 1 m q1 = q2 = q = ? From. Coulomb's law in dectrostatics, we have or,  $441 = 9 \times 10^{8} \frac{q_{1} \cdot q_{2}}{1^{2}}$  or,  $q^{2} = 441 - 4.9 \times 10^{-4}$  $\therefore Q = 2.2 \times 10^{-4} \text{ C}.$ Thus, the value of the given two equal charges is 22x10

