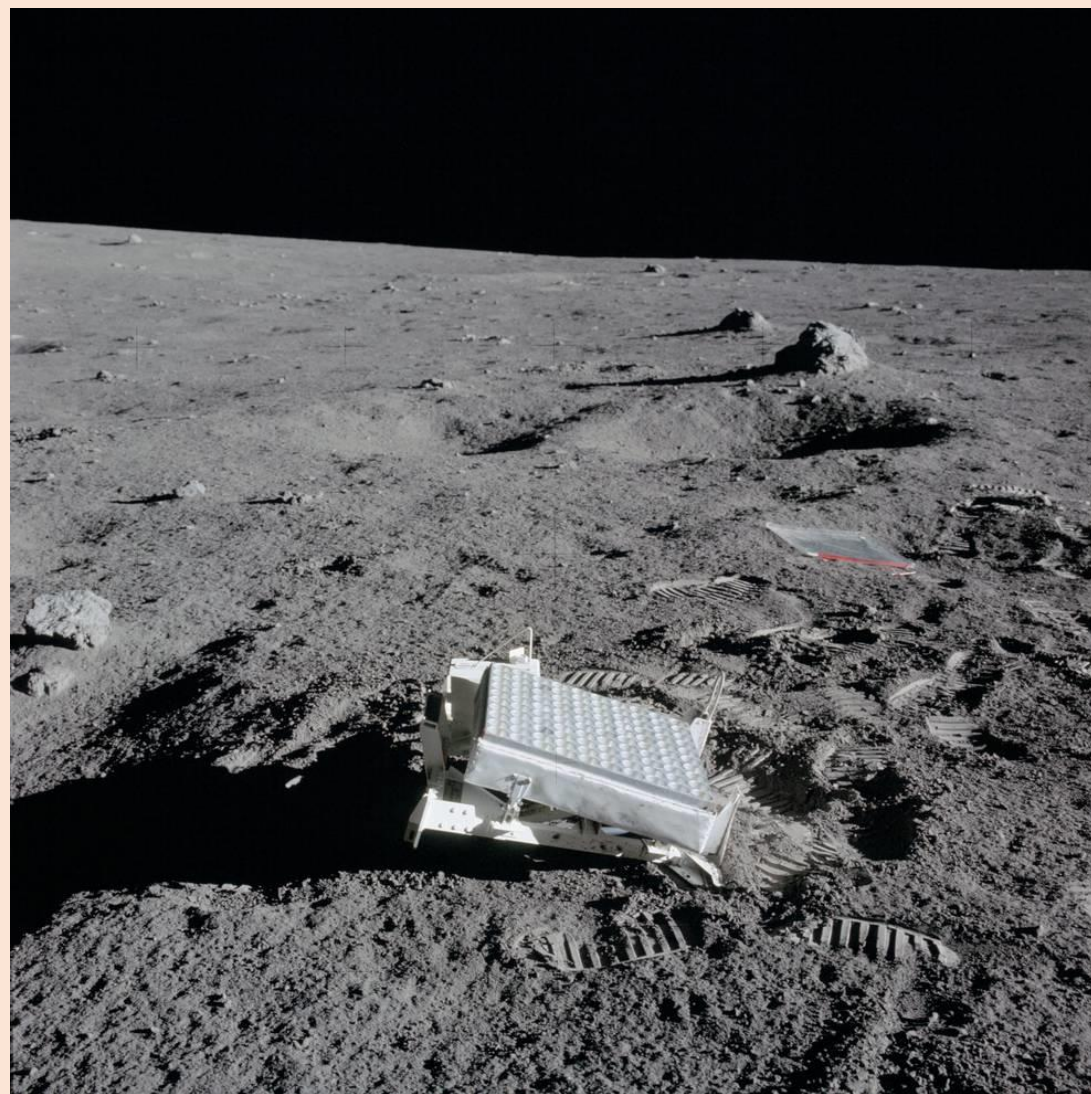


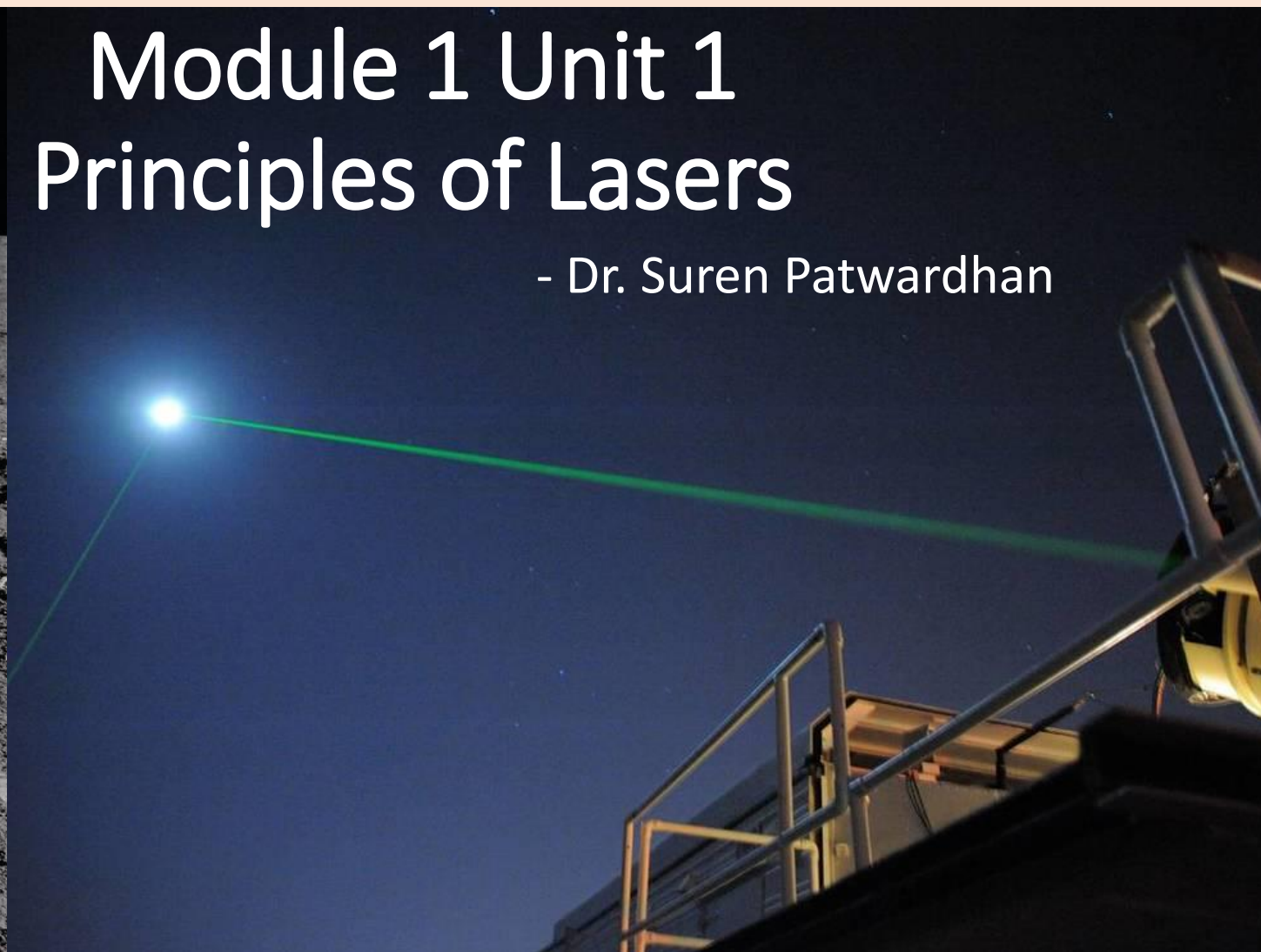
Module 1 Unit 1

Principles of Lasers

- Dr. Suren Patwardhan



Lunar surface and reflector by Apollo Astronauts



Geophysical and Astronomical Observatory at NASA's Goddard Space Flight Center in Greenbelt, Md., USA

Image Courtesy: <https://www.nasa.gov/multimedia/imagegallery>

Ordinary Light v/s LASER Light

- Examples of ordinary Light
 - Bulbs (Incandescent, CFL, LED), Tube lights, Decorative lights
 - Halogen lamps, Flood Lights
 - Photographic Light, Xenon Lamp
 - Neon Lamp, Street Lamps (yellow)
 - Burner flame
 - Petromax Light, Kerosene Lamp
 - Candle, Matchstick
 - Sunlight
- Laser Light?
 - Laser Pointer (semiconductor Laser)

Properties of LASER

Light Amplification by Stimulated Emission of Radiation

- Monochromatic
 - Unique wavelength
 - Coherent
 - All waves are In phase
 - Directional
 - Light emitted along same path
 - Focused
 - Low spreading
 - Bright
 - High intensity
 - Polarized (optional)
 - E-M vibrations in same plane
- Most of the lasers are in the IR to Visible range
 - Radio and microwave frequency lasers are called MASERS
 - Difficult to get Ultraviolet lasers
 - May not be possible to get X-ray and γ -ray lasers

Laser Beam Parameters

- Monochromaticity
 - Linewidth of radiation
- Coherence length
 - Distance up to which waves are in phase
- Directionality
 - Unique direction of emission
- Divergence
 - Amount of spread
- Intensity
 - Optical power emitted per unit area

Parameter	Ordinary sources	Laser sources
Monochromaticity	10^5 nm	10^{-4} nm
Coherence length	mm to cm	m to km
Directionality	$30-360^\circ$	$< 2^\circ$
Divergence	$> 10^{-2}$ strad	$< 10^{-4}$ strad
Intensity	mW to W	mW to kW

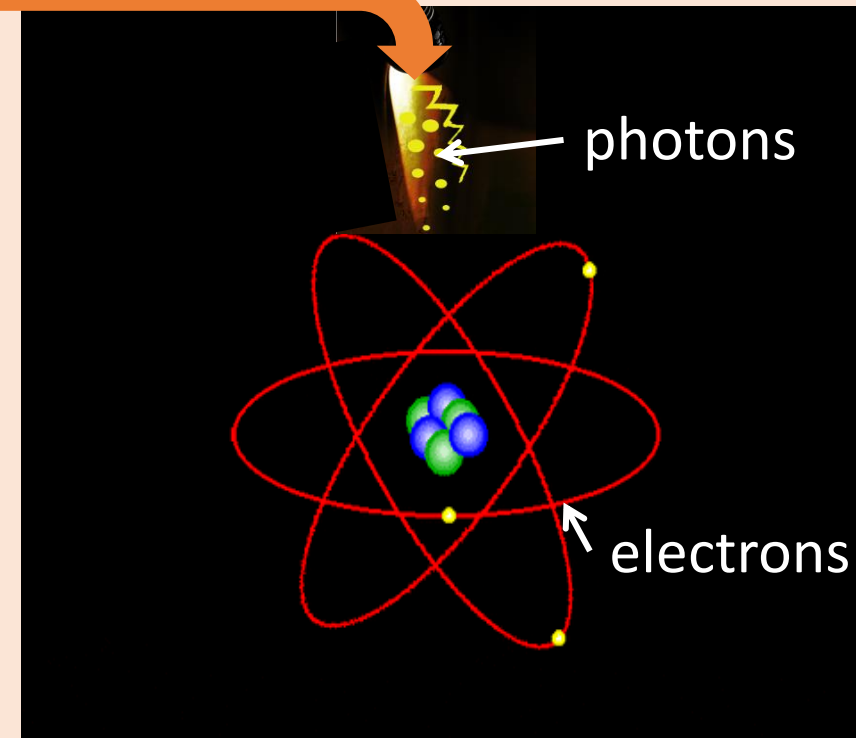
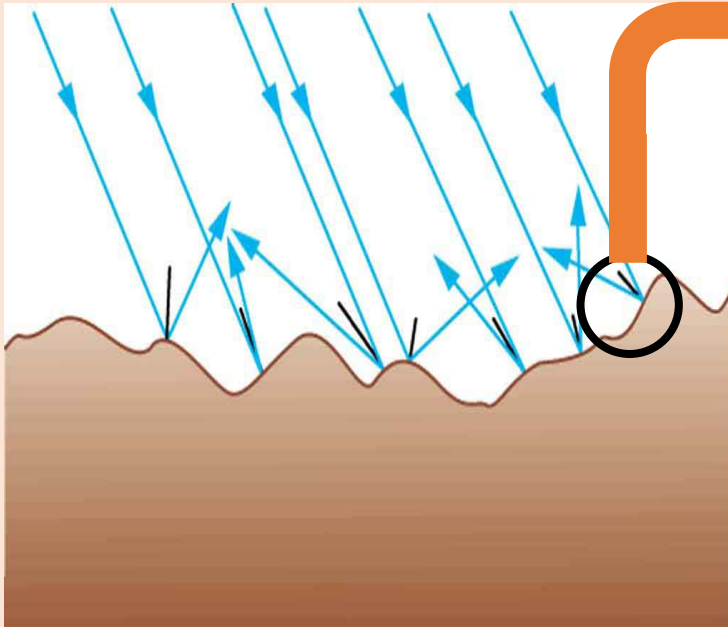
LASER are Everywhere

- First discovered by Maiman 1960 (kind of MASER – M for Microwaves)
- Once teased as discovery in search of applications, today has hundreds of applications

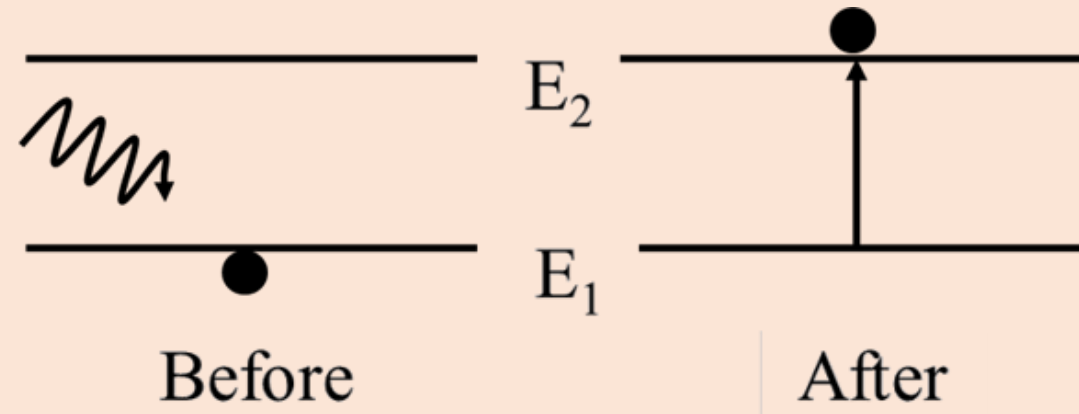
Name of laser source	Emission wavelength	Applications
He-Ne laser	Red, green	Laboratory purpose
Ar laser	Blue-green	Eye surgery
Nd:YAG laser	Invisible (IR)	Machining, surgery
Ruby laser	Extreme Red	High speed photography
GaAs/GaN lasers	Red to Blue	Fibre optic communications
CO ₂ laser	Invisible (IR)	Dermatology
Dye lasers	Visible to IR	Spectroscopy
Nd:Glass	Invisible (IR)	Nuclear reactors

Interaction of Radiation with Matter

1. Absorption - Energy supplied
2. Emission - Energy emitted
 - Spontaneous emission
 - Stimulated emission



Stimulated Absorption (or just Absorption)



Rate of transition

$$\left. \frac{dN}{dt} \right|_{ab} = B_{12} N_1 Q$$

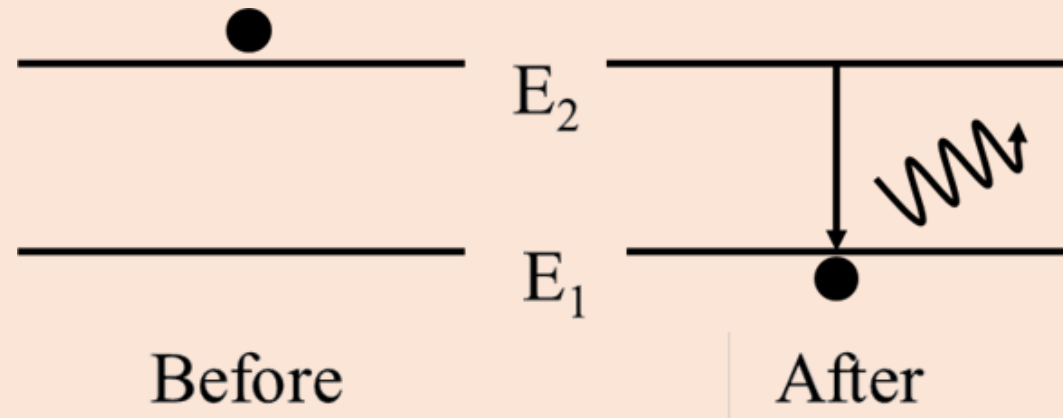
Where,

B_{12} : probability of absorption process

N_1 : number of atoms in lower energy level E_1

Q : energy density of incident radiation per unit frequency (J-s/m^3)

Spontaneous Emission (i.e. Natural Emission)



Rate of transition

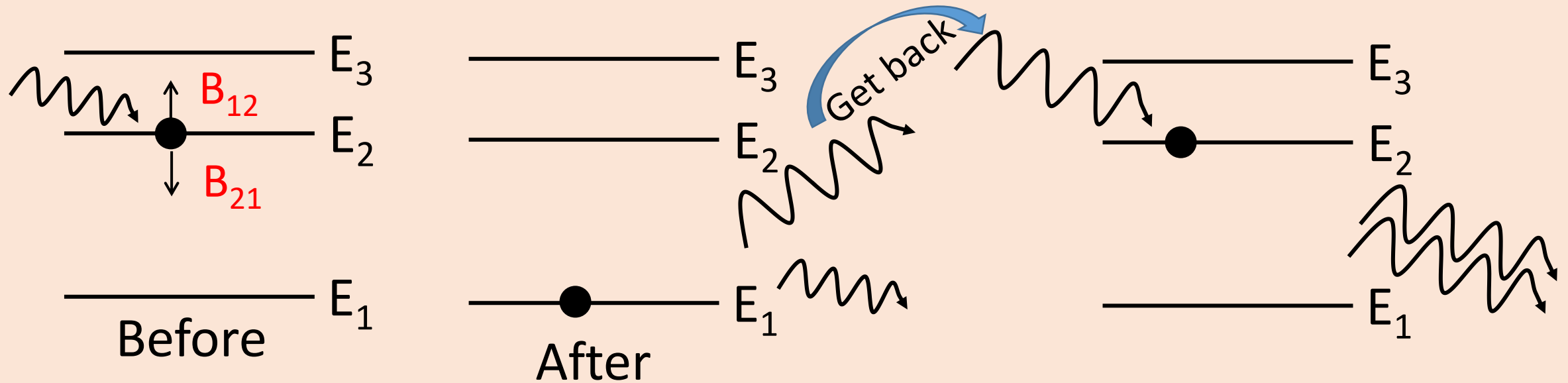
$$\left. \frac{dN}{dt} \right|_{sp} = A_{21} N_2$$

Where,

A_{21} : probability of spontaneous emission process

N_2 : number of atoms in higher energy level

Stimulated Emission (i.e. Triggered Emission)



Einstein showed $B_{12} = B_{21}$

$$\text{Rate of transition } \left. \frac{dN}{dt} \right|_{st} = B_{21} N_2 Q$$

Where,

B_{21} : probability of stimulated emission process

N_2 : number of atoms in higher energy level E_2

Q : energy density of incident radiation per unit frequency

Advantages

Emitted radiation has:

- Same wavelength
- Same direction
- Same phase

Laser Emission is Not Possible in Ordinary Conditions

Reason:

Spontaneous emission dominates at normal temperatures

Rate of absorption: $\left. \frac{dN}{dt} \right|_{ab} = B_{12} N_1 Q$

Rate of spontaneous emission: $\left. \frac{dN}{dt} \right|_{sp} = A_{21} N_2$

Rate of stimulated emission: $\left. \frac{dN}{dt} \right|_{st} = B_{21} N_2 Q$

How to Lase?

Answer: Break the laws

Laws:

1. “Equilibrium” condition – more atoms in lower states than upper states
 - At equilibrium, $N_1 \gg N_2$
2. Any medium is “Lossy” – on passage, optical intensity decreases
 - For lossy medium, loss (α) > gain (γ)

How to break them?

1. Make $N_2 > N_1$ i.e. populate upper states E_2 as compared to Lower states E_1
(population invasion)
2. so make $\gamma > \alpha$ i. e. trap light to achieve optical gain in a region
(light amplification)

Agents to Break the Laws

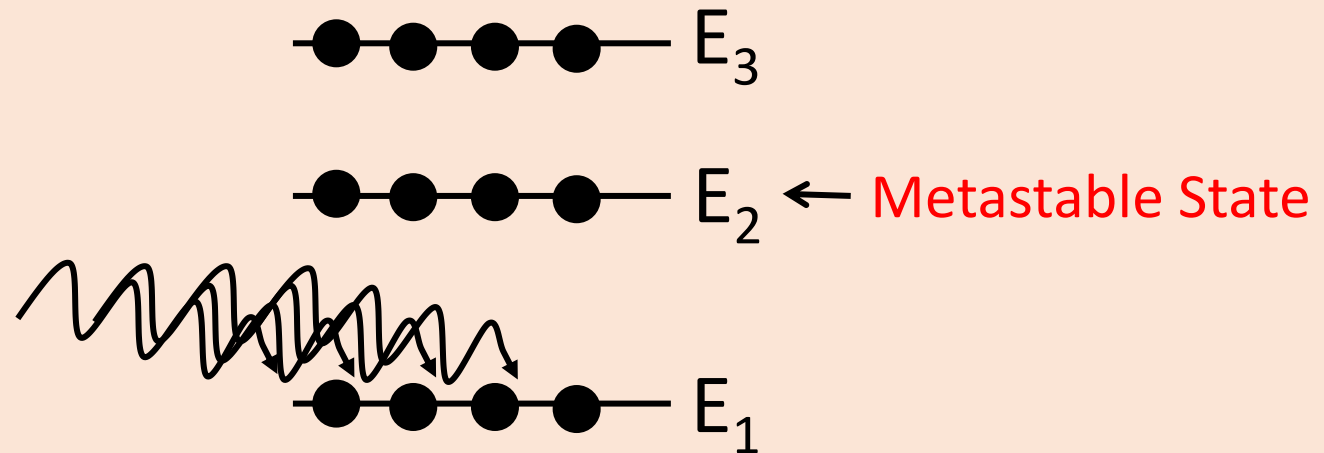
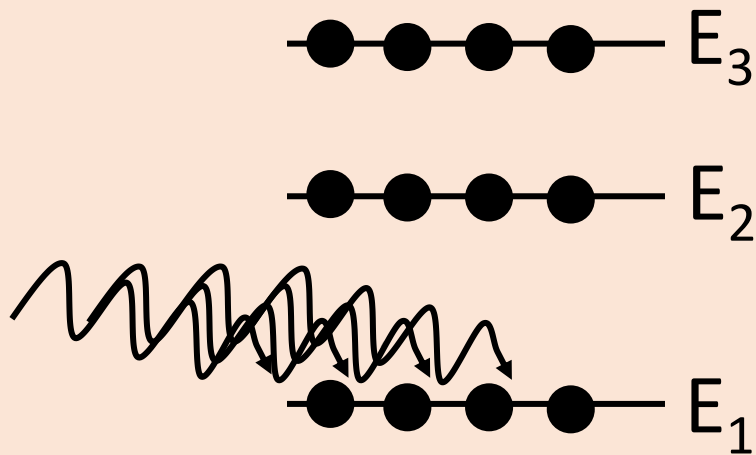
Achieved by two important concepts:

1. **Metastable states** – ensures $N_2 > N_1$
 - Achieves **population inversion** along with pumping
 2. **Optical Resonator** – ensures $\gamma > \alpha$
 - Achieves **light amplification***
- 1 and 2 together ensure stimulated emission is promoted for laser emission

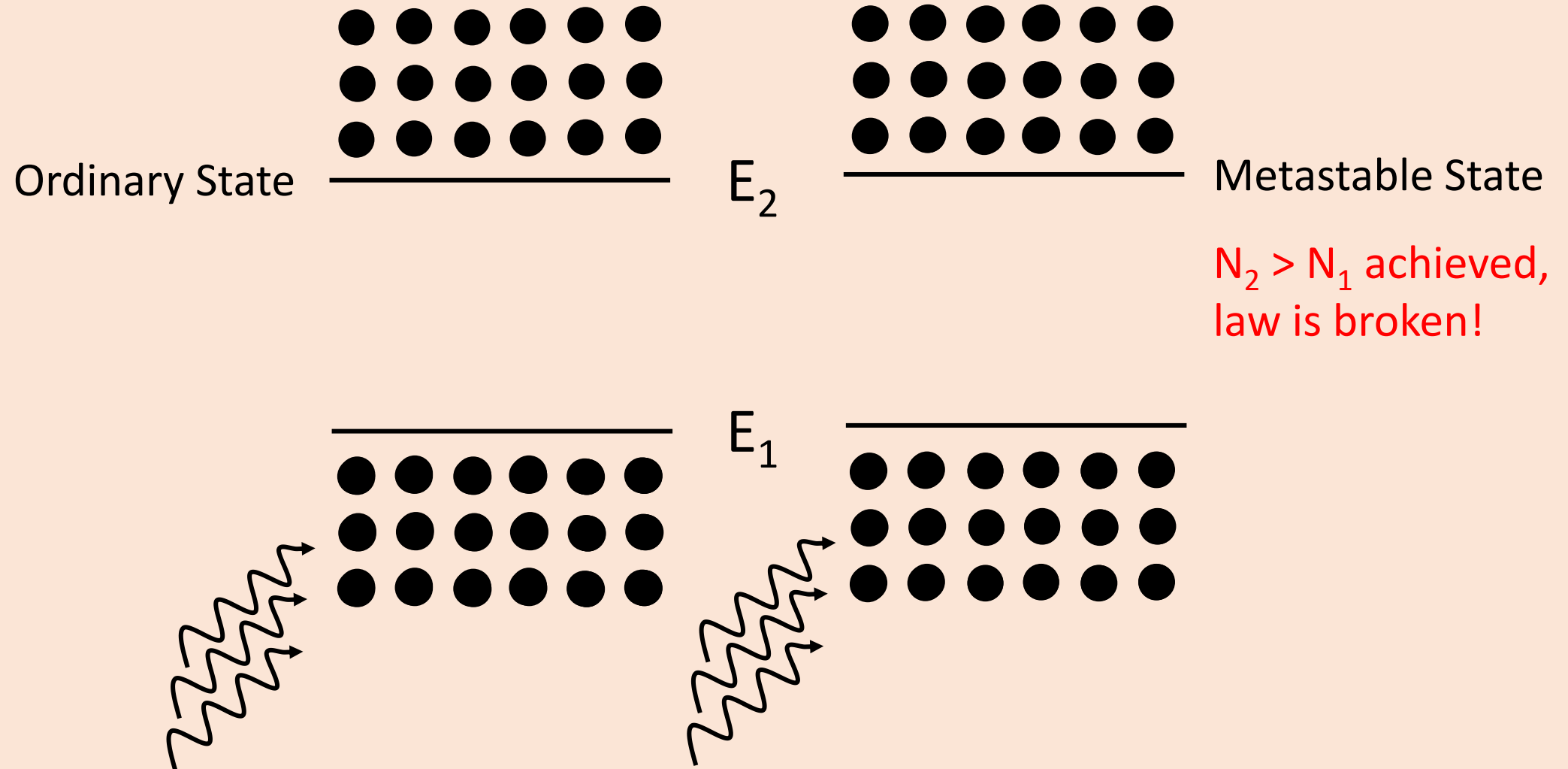
*Does it violate laws of thermodynamics?

Metastable State

- Typical lifetime of an excited state in any atom is $\approx 10 \text{ ns}$ (10^{-8} s)
- But there are certain special excited state in some atoms
- For these states, the lifetime is $\approx 1 \text{ ms}$ (10^{-3} s); 100000 times longer!
- Such states are called metastable states
- Materials having metastable states is necessary condition for laser

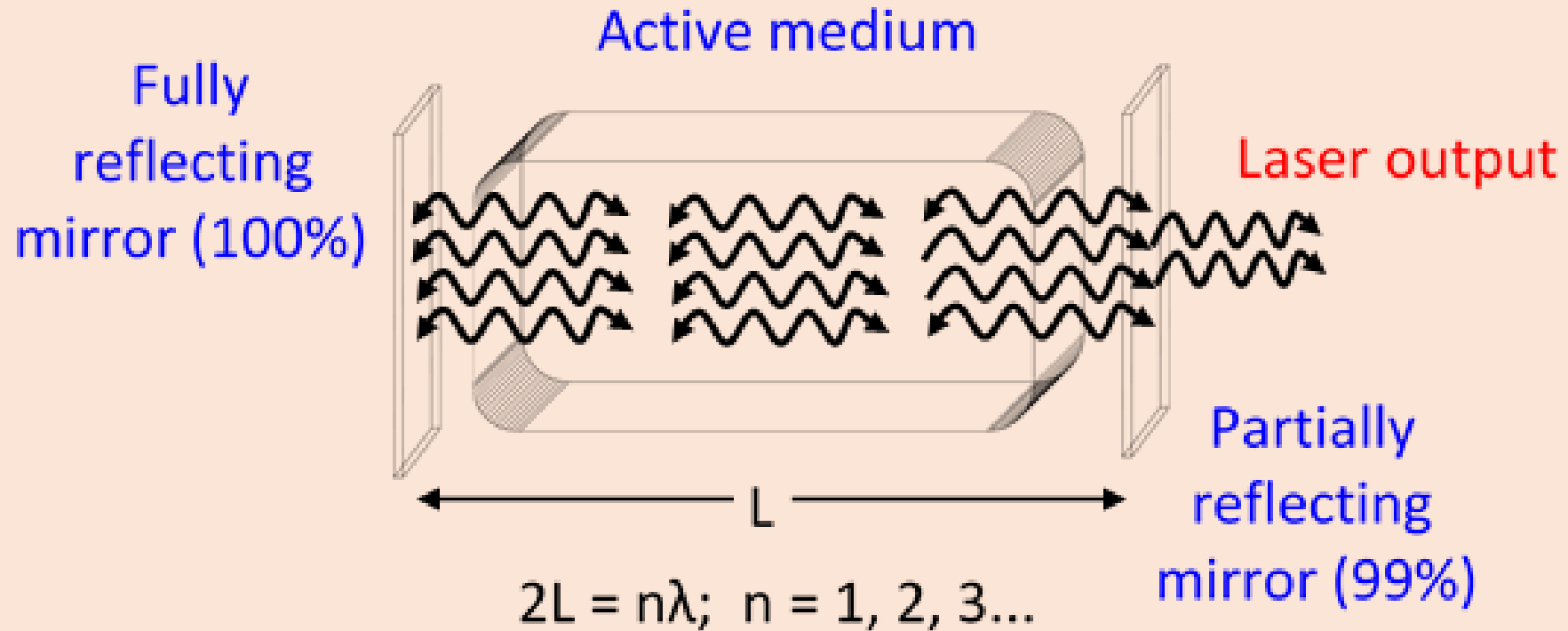


Action of Metastable State

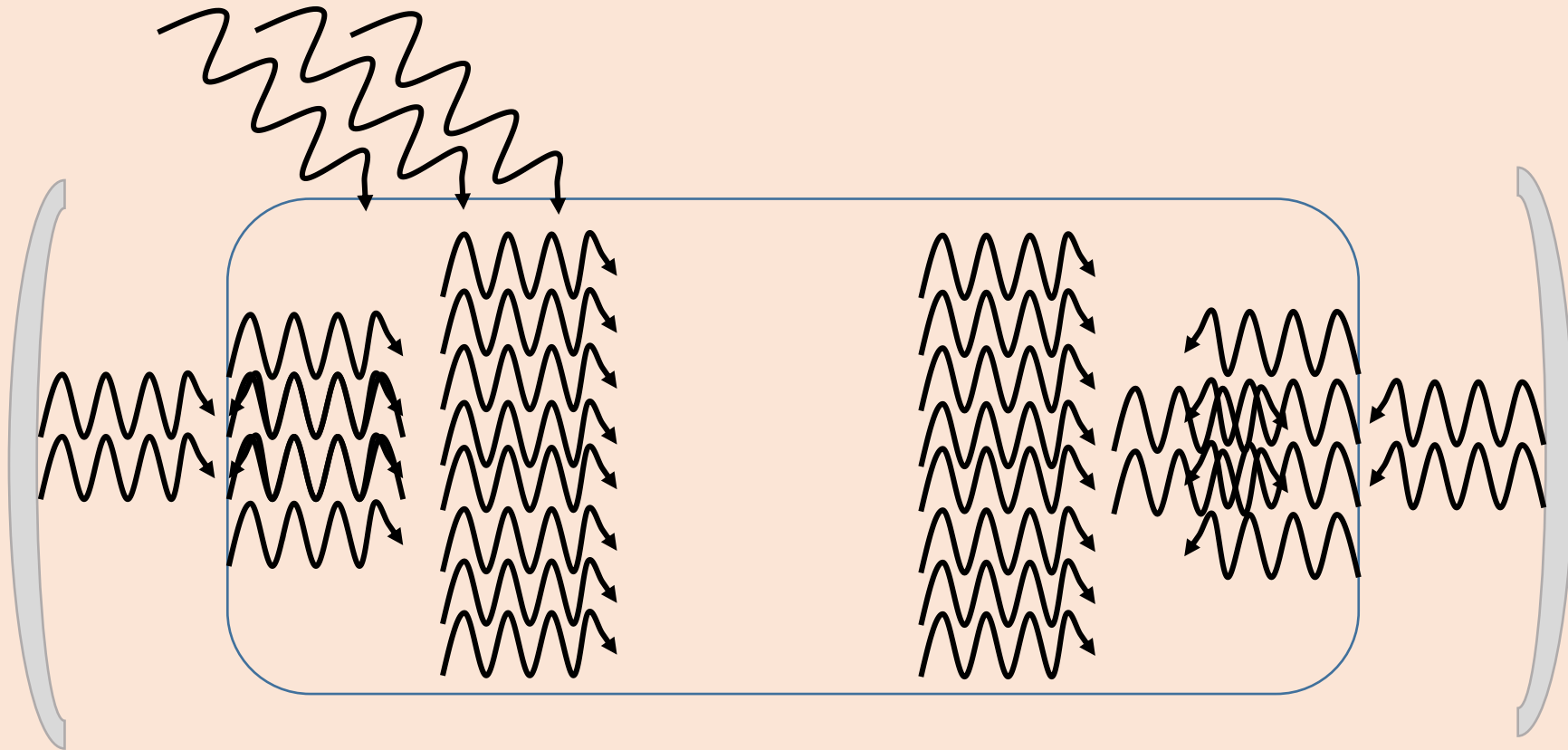


Optical Resonator/Active Medium

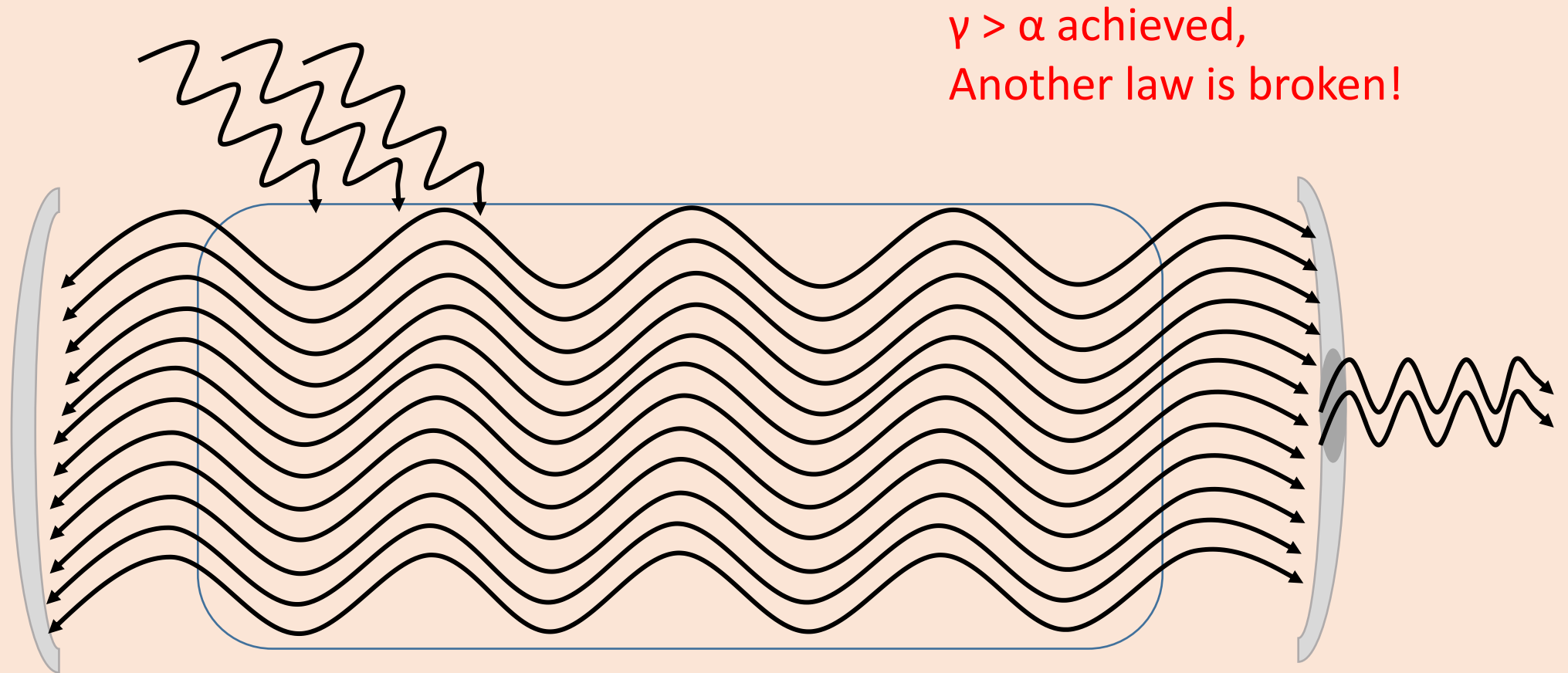
Active medium: *Region of the laser source where optical gain takes place*



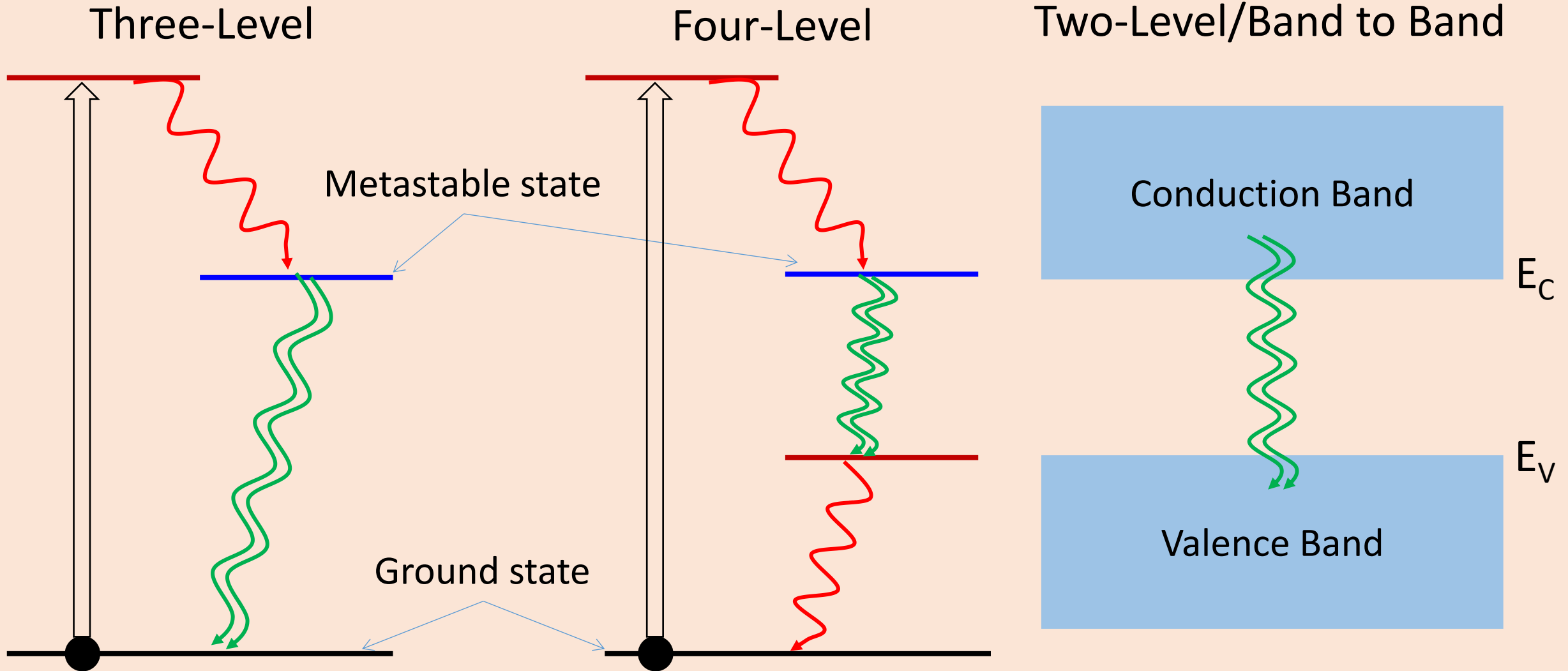
Action of Optical Resonator



Action of Optical Resonator



Pumping Schemes



Quick Poll



Types of Pumping

Pumping: *Process of supplying energy to trigger population inversion*

- **Optical pumping**

A broad and bright source of light (photons) is used to supply energy

Example: Xenon flash lamp in Ruby laser, Nd:YAG laser, Dye lasers

- **Electrical Pumping**

High electric field is set up by a pair of electrodes

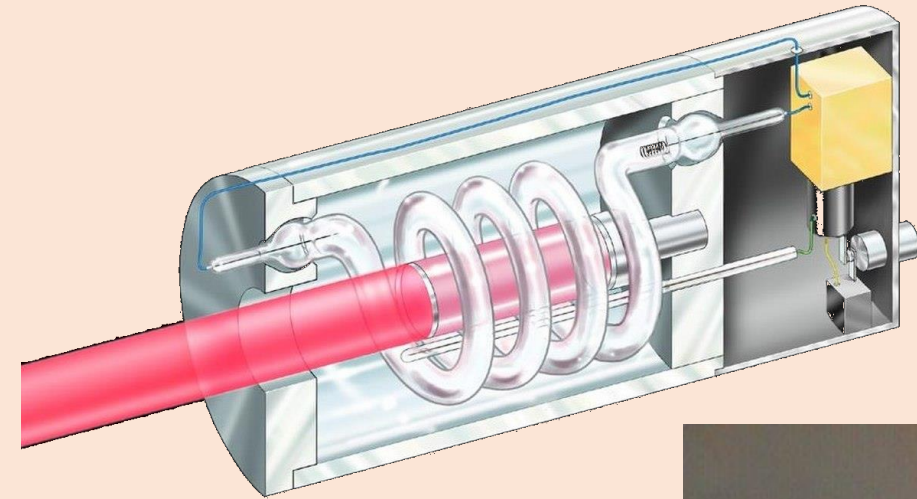
Example: He-Ne laser, CO₂ laser, Ar laser

- **Direct conversion**

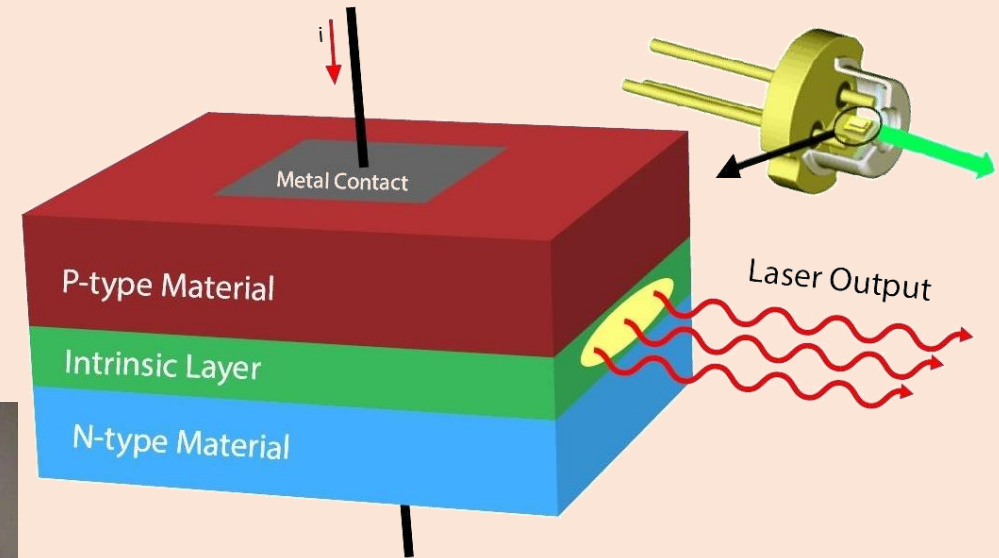
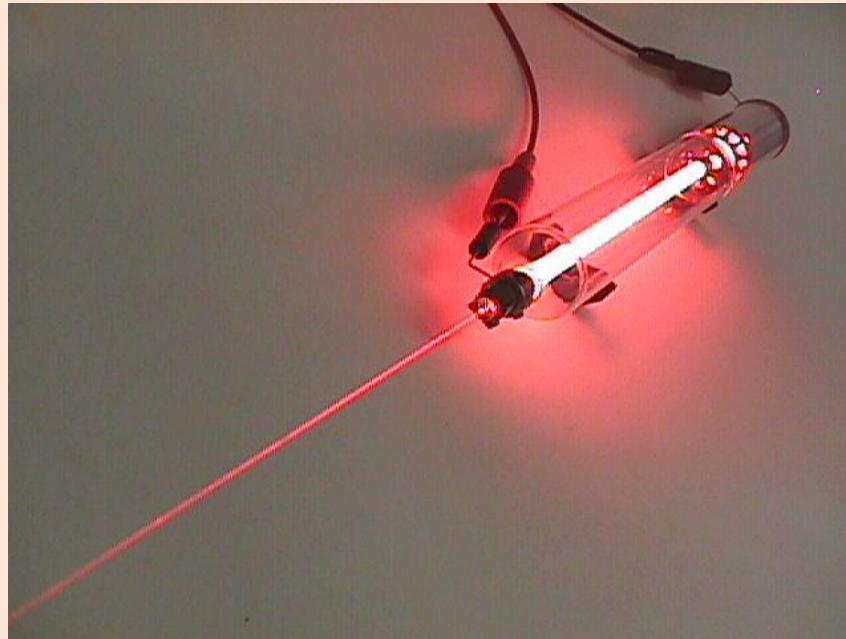
By passing an electric current

Example: Semiconductor diode lasers

Types of Pumping - Schematics



Optical



Direct conversion

Electrical

Important Laser Physics Terms

- Population

The number of active atoms occupying a particular energy state

- Population Inversion

Creating a non-equilibrium state with more atoms in excited states

- Pumping

Process of supplying energy to achieve population inversion

- Metastable State

Special energy levels having unusually high lifetime than normal excited states

- Active medium

Region of the laser source where optical gain takes place

Condition for Lasing



$$I(x) = I_0 e^{(\gamma - \alpha)x}$$

Thanks!