

# **Gamma-Spectroscopy**

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# **Agenda**

What is gamma spectroscopy? Interactions in the Spectrum Detectorsystems Applications

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### What is gamma spectroscopy?

### studies of energy spectra of gamma rays

spectrum used to determine quantity and identity of  $\gamma$ -emitters gamma ray sources emitters on earth, astrophysical sources

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# Interaction of y-rays with matter

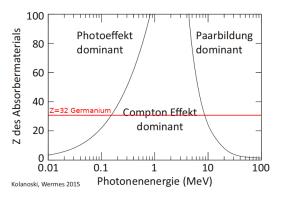


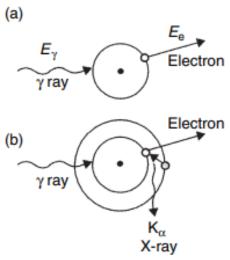
Abbildung: atomic number Z against photon energy E.

processes above ionization threshold Gamma ray absorption  $\rightarrow$  intensity loss Material thickness dependend intensity:  $N(D) = N_0 e^{-\mu D}$ 

D: thickness,  $\mu$ : absoption coefficient,  $N_0$ : initial intensity

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#### **Photoeffect**



 $E_{\gamma}$  < several 100 keV ionizing bound electron (K-shell)

 $\gamma$  + atom  $\rightarrow$  atom<sup>+</sup> + e<sup>-</sup>

hole is filled with electrons from higher shells recursively

energy diff. release as x-rays characteristic

rarely: photon leaves absorber. often excite more electrons inside

K-L-M-absorption edge: Quantumenergy enough to release bound electron from

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# **Compton scattering**

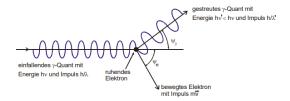
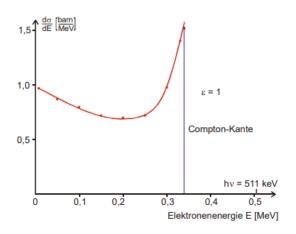


Abbildung: compton continuum

main interaction (100 keV < E < 5 MeV) inelastic scattering photons only transfers an energy fraction cannot view full spectrum ©

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### **Compton scattering**



non-isotropic angular distribution

$$E_{\gamma}' = \frac{E_{\gamma}}{1 + \left(\frac{E_{\gamma}}{511 \text{keV}} (1 - \cos \theta)\right)}$$

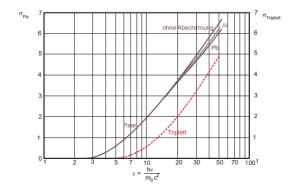
$$= \int \frac{E_{\gamma}}{\frac{E_{\gamma}}{511 \text{keV}} (1 - \cos \theta)}$$

$$E_{e^{-}} = E_{\gamma} \left( \frac{\frac{E_{\gamma}}{511keV} (1 - cos\theta)}{1 + \frac{E_{\gamma}}{511keV} (1 - cos\theta)} \right)$$

extreme cases: backward scattering, light graze

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### **Pair production**



photon produces e+e- pair if E is high enough (5 MeV < E < 10 MeV) occurs in proximity of nucleus/scattering partner

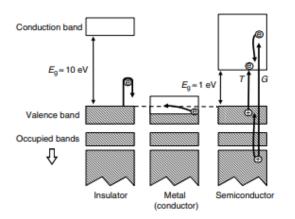
photon line visible if both leptons are absorbed;

annihilation peak : 511 keV (e- mass) or doubled for both

single- and double-escape peaks

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#### **Band structure**



electrons in discrete/precise energy bands

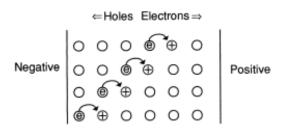
valence band: outer band for chemical reactions; most inhibited

conduction band: migration of electrons

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### Mobility of "Holes"



positive charge ≡ hole in the band measuring the energy relies on separating the charge carriers electrons from valence band filling holes -> effective moving -> conductivity

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### **Creation of charge carriers**

excite electrons from low bands through high energies ( $E > E_{therm}$ ) redistribution of electrons-holes throughout energy-bands holes: top of valence band electrons: bottom of conduction band external field: charge carriers migrate towards respective electrode

number of electron-hole pairs  $n = E_{\rm abs}/\epsilon$  $\epsilon$ : average energy needed to create electron hole pair

 $E_{abs}$ : absorbed gamma ray energy

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### resolution and suitable semiconductors

resolution ∝ number of pairs produced

large absorption coefficient (high atomic number Z)

low  $\epsilon$ : to produce many electron-hole pairs

allow good Mobility (trapping inside semiconductor lattice)

pure crystal structure (traps for charge carriers)

cannot be too expensive

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### Why Germanium as detector material?

Material	Atomic number	Operating temperature	Band gap (eV) <sup>a</sup>	$(eV)^{a,b}$	Density (g cm <sup>-3</sup> )	Mobility(cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> ) <sup>a</sup>	
						Electrons	Holes
Si	14	RT	1.106	3.62	2.33	1350	480
Ge	32	Liquid N <sub>2</sub> (77 K)	0.67	2.96	5.32	$3.6 \times 10^{4}$	$4.2 \times 10^{4}$
CdTe	48, 52	RT	1.47	4.43	6.06	1000	80
CdZnTe	48, 30, 52	RT	1.57	4.64	5.78	1000	50-80
HgI,	80, 53	RT	2.13	4.22	6.30	100	4
GaAs	31, 33	RT	1.45	4.51	5.35	8000	400
TlBr	81, 35	-20°C	2.68	?	7.56	_	_
PbI <sub>2</sub>	82, 53	_	2.6	7.68	6.16	8	2
GaSe	31, 34	_	2.03	6.3	4.55	_	_
AlSb	13, 51	_	1.62	5.05	4.26	_	_
CdSe	48, 34	_	1.75	?	5.74	_	_

<sup>&</sup>lt;sup>a</sup> Values are given at 77 K for Ge and 300 K otherwise.
<sup>b</sup> Electron-hole creation energy.

Silicon: highly pure, low-priced, low atomic number (low energy photons)

Germanium: higher Z -> good for higher energy gamma radiation

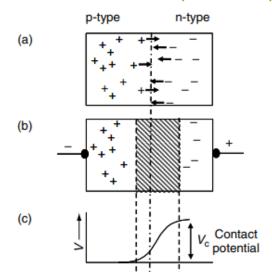
low temperature -> reduce leakage current

improvements reduced resolution to ≈ 1.8 keV

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### Semiconductor detector (Germanium)



p-type, n-type germanium (3 and 5 valent impurities)

electrons and holes recombine by diffusion -> depletion zone

-> only local donator and acceptor hulls remain

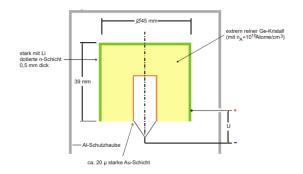
goal: seperation via external field inside depletion zone

maximising the depletion zone -> hinder recombination -> reconstruct energy of the event

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#### The Ge-detector



Aluminium casing as VETO region (40 - 50 keV)

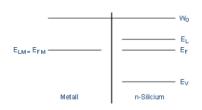
Lithium doped outside: connecting rreverse voltage

inside steamed with gold (M-S junction) electronic signal: spectrum-histogram from Multichannel analyzer (MCA)

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# Metal-Semiconductor junction (n-Si)



 $W_0$ : vacuum level,  $E_L$ : conduction band,  $E_F$ : fermi energy

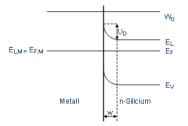
 $E_V$ : valence band,  $E_{LM}$ : energy level in metal,  $E_{FM}$ : fermi energy in metal electrons can migrate from Semiconductor to metal since E-level is lower

probability of finding electrons in conduction band gets lower

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# Metal-Semiconductor junction (n-Si)



migrated charge carriers form depletion zone and lower fermi level additional  $e^-$  must overcome the Schottky-barrier to flow into Metal fermi levels in metal and semiconductor equalize via diffusion process

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# Why is this junction used over p-n?

Si-Schottky diodes are substantially faster changing from forward bias to reverse bias

-> switching action: 10 - 100 GHz possible because no moving "holes" in metal

low forward voltage drop (0.15 - 0.45V) p-n: 0.6 - 0.75 V

But: higher reverse leakage current

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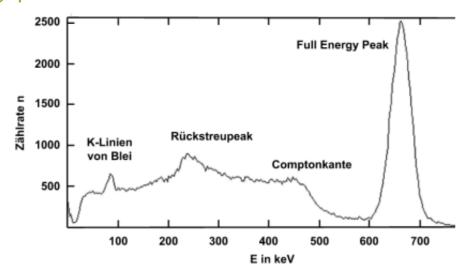
# **Separation efficiency**

minimal distance of 2 signals to register both Full Width at Half Maximum  $\Delta E_{1/2} \approx 2.35 \sqrt{FE_{\gamma}\epsilon}$  Fano factor of germanium F = 0.1

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### **Energy spectra**



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### **Evaluation of gamma spectroscopy**

#### PRO:

quite cheap in material costs

relatively fast result evaluation

multinuclide analysis (distinct lines

visible for all nuclides)

non-destructive for emitter (radiation

hardness of detector given)

remote measurement

#### CONTRA:

often less sensitive

require large sample masses (if not gamma rays from space)

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### **Applications**

medical application: positron emission tomography (PET) [tumor therapy]

-> positron-emitting pharmaceutical injection near tumor -> imaging through 511 keV lines security scans for explosives

monitoring nuclear waste

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### Thank you for listening!

Questions?

Comments?

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### Quellen

https://onlinelibrary.wiley.com/doi/book/10.1002/9780470861981

V18\_Anleitung.pdf

https://www.nrc.gov/docs/ML1122/ML11229A699.pdf

https://www.electrical4u.com/schottky-diode/

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### **Backup**

Attenuation egde for caesium iodine (CsI)

2 K-lines and 2 L-lines

