



Introduction

- Chemically all living organisms have three basic types of macromolecules, which are polymers of simple subunits called
 - monosaccharide (A)
- (B) dipeptide
- (C) monomers
- (D) nucleotides
- Sugar, amino acids and nucleotides unite to their 2. respective subunits to form
 - (A) bio-elements
- (B) micromolecules
- (C) macromolecules (D)
 - all of these



Biomolecules in the Cell 6.1

Carbohydrates

- Carbohydrates are composed of 3.
 - (A) carbon
- (B) hydrogen
- (C) oxygen
- (D) all of these
- In which of the following, the ratio of hydrogen 4. and oxygen atoms is 2:1?
 - (A) Proteins
- (B) Fats
- (C) Oil
- (D) Carbohydrates
- Which of the following do not give smaller 5. sugar units on hydrolysis?
 - Monosaccharides (B)
- Disaccharides
 - (C) Polysaccharides
- (D) Glycogen
- 6. Carbohydrates are biomolecules made from just three elements: carbon, hydrogen and oxygen with the general formula
 - (A) $C_x(HO)_y$
- (B) C,(H2O),
- (C) $C_x(H_2)_x$
- (D) $C_3(O_2)_v$
- 7. Match the following and select the correct option.

| | Column I (Sugars) | | Column II (No. of Carbons) |
|------|----------------------|----|-------------------------------|
| i. | Erythrose | а. | Five |
| ii. | Glucose | b. | Four |
| iii. | Ribose | c. | Seven |
| iv. | Sedoheptulose | d. | Six |

- (A) i-a, ii-c, iii-d, iv-b
- (B) i-a, ii-b, iii-c, iv-d
- (C) i-b, ii-c, iii-d, iv-a
- (D) i-b, ii-d, iii-a, iv-c
- 8. Disaccharide when two is formed monosaccharides react by condensation reaction molecule. releasing a
 - (A) CO₂
- water (B)
- (C) hydrogen
- (D) CO

- Which of the following bond holds the two monosaccharide units together?
 - Sulphide bond
 - (B) Glycosidic bond
 - (C) Peptide bond
 - (D) Disulphide bond
- Identify the INCORRECT pair from the 10. following.
 - Sucrose (cane sugar): On hydrolysis, it produces Glucose and Fructose
 - (B) Lactose (milk sugar): On hydrolysis, it produces Glucose and Galactose
 - Maltose (malt sugar). On hydrolysis, it (C) produces two units of Glucose
 - Fructose (Fruit sugar): On hydrolysis, it (D) produces Glucose and Lactose
- Which of the following is a stored food in the 11. plants?
- (B) Starch
- (A) Glucose (C) Cellulose
- (D) Glycogen
- 12. Starch and cellulose are the compounds made up of many units of
 - (A) Simple sugar
- (B) Fatty acid
- (C) Glycerol
- (D) Amino acid
- 13. Identify the two types of glucose polymers present in starch.
 - (A) Amylose and glycogen
 - Amylose and amylopectin (B)
 - Amylopectin and glycogen (C)
 - Cellulose and amylopectin
- 14. Which of the following correctly describes the two types of polysaccharides?
 - (A) Homopolysaccharides: It contains same type of amino acids.

Heteropolysaccharides: It contains two or more different monosaccharides.

(B) Homopolysaccharides: It contains same type of monosaccharides.

> Heteropolysaccharides: It contains two or more different monosaccharides.

Homopolysaccharides: It contains same (C) type of monosaccharides.

Heteropolysaccharides: It contains two or more different nucleotides.

Homopolysaccharides: It contains same type of nucleotides.

> Heteropolysaccharides: It contains two or more different amino acids.

- serves to form the cell walls in plant 15. cells.
 - Cellulose (A)
- (B) Starch
- Glycogen (C)
- Amylopectin (D)

16. Match the columns and select the correct option.

| | Column I | 1 . 3 | Column II |
|------|---------------------------|-------|------------------------------|
| i. | Starch | a. | Animal storage molecule |
| ii. | Cellulose | b. | Plant storage molecule |
| iii. | Glycogen | c. | Heparin |
| iv. | Heteropoly- saccharide | d. | Plant cell wall component |

- (A) i-a, ii-c, iii-d, iv-b
- (B) i - a, ii - b, iii - c, iv - d
- i b, ii c, iii d, iv a(C)
- (D) i - b, ii - d, iii - a, iv - c
- 17. The exoskeleton of insects is made up of chitin which is a
 - (A) mucoprotein
- (B)
- (C) lipoprotein
- (D) polysaccharide

Lipids

- 18. In lipids, ratio is greater than 2:1
 - carbon to oxygen (A)
 - (B) hydrogen to oxygen
 - (C) oxygen to nitrogen
 - (D) sulphur to oxygen
- 19. are water insoluble and small molecular weight compounds as compared to macromolecules.
 - (A) Lipids
- (B) Proteins
- (C) Carbohydrates
- Nucleic acids (D)
- 20. Fats are esters of fatty acids with
 - (A) glycerol (CH2OH-CHOH-CH5OH)
 - glycerol (CH3OH-CHOH-CH3OH) (B)
 - (C) glycerol (CH2OH-CHOH-CHOOH)
 - glycerol (CH2OH-COH-CH2OH) (D)
- 21. Unsaturated fats are liquid at room temperature and are called
 - (A) oils
- (B) glycolipids
- phospholipids (C)
- (D) waxes
- 22. Phospholipids have
 - only hydrophilic polar groups (A)
 - only hydrophobic non-polar groups (B)
 - both hydrophilic polar groups (C) and hydrophobic non-polar groups
 - phytosterols (D)
- Match the columns and select the correct option.

| | Column I | | Column II |
|------|-----------------|----|----------------|
| i. | Simple lipids | a. | Cholesterol |
| ii. | Compound lipids | b. | Vanaspati ghee |
| iii. | Sterols | c. | Glycerol |
| | | d. | Lecithin |

- i a, ii c, iii d(A)
- i a, ii b, iii c(B)
- i b, ii c, iii d(C)
- i b, ii d, iii a (D)

- Fatty acids which do not contain double bond 24. between carbon atoms are
 - saturated fatty acids (A)
 - unsaturated fatty acids (B)
 - esters of glycerol (C)
 - polymers of triglycerides (D)
- Which type of fatty acid is represented in the 25. following structure?



- (A) Saturated fatty acid
- Amphipathic fatty acid (B)
- Unsaturated fatty acid (C)
- (D) Glycerol
- Large amounts of ____ 26. have been found in the brain white matter and myelin sheath.
 - adrenocorticoids (B)
 - cerebrosides
- (C) oleic acids (D) mucoproteins
- 27. Which of the following are the examples of unsaturated fatty acids?
 - (A) Oleic acid
- (B) Linoleic acid
- (C) Glycerol
- (D) Both (A) and (B)
- Yam Plant (Dioscorea) produces a steroid 28. compound called
 - (A) diosgenin
- (B) adrenocorticoid
- (C) progesterone
- (D) estrogen

Proteins

- The term 'Protein' was coined by 29.
 - Schleiden (A)
- (B) M. Wilkins
- (C) Berzelius
- (D) Luca Ghini
- 30. **Proteins**
 - are micromolecules (A)
 - (B) are macromolecules
 - have low molecular weight (C)
 - are found in small quantity in the cell (D)
- 31. Proteins consist of
 - carbon, hydrogen, chlorine, sulphur (A) (B)
 - carbon, hydrogen, oxygen, nitrogen
 - carbon, manganese, phosphorus, nitrogen (C)
 - carbon, iodine, oxygen and inorganic (D) phosphate
- In proteins, amino acids are linked together by 32. bonds which join the carboxyl group of one amino acid residue to the amino group of another residue.
 - (A) glycosidie
- (B)
- (C) ester
- peptide (D) phosphodiester



| | | | Chapter 06: Biomolecules |
|-----|--|-----|--|
| 33. | Which of the following are the two types of | 43. | Which one of the following pairs of nitrogenous |
| | secondary structure of proteins? | | bases of nucleic acids, is wrongly matched with |
| | (A) δ-helix and β-pleated sheets | | the category mentioned against it? |
| | (B) α-helix and β-pleated sheets | | (A) Guanine, Adenine – Purines |
| | (C) β-helix and α-pleated sheets | | (B) Adenine, Thymine - Purines |
| | (D) β-helix and δ-pleated sheets | | (C) Thymine, Uracil - Pyrimidines |
| 34. | Complete the analogy and select the correct | | (D) Uracil, Cytosine – Pyrimidines |
| 34. | option. | | |
| | α-helix structure : Keratin :: β-pleated sheet : | 44. | The base pairs of DNA are correctly shown as |
| | | | (A) $A = T$ and $C = G$ |
| | (A) Phospholipids (B) Silk fibres | | (B) $A = T$ and $C = G$ |
| | (C) Palmitic acid (D) Lecithin | | (C) $A = T$ and $C = G$ |
| 35. | Proteins are in nature. | | (D) A ≡ T and C ≡ G |
| 33. | (A) hydrophilic (B) hydrophobic | 45. | Complementary strands of DNA molecule |
| | (C) amphoteric (D) polar | 13. | are (i) and bound by (ii). |
| | (-) point | | (A) i – parallel, ii – peptide bond |
| 36. | A protein consisting of more basic amino acids such as exists as a cation at the | 1 | (B) i – antiparallel, ii – hydrogen bond |
| | such as, exists as a cation at the physiological pH of 7.4. Such proteins are | | (C) i - linear, ii - glycosidic bond |
| | called basic proteins. | | |
| | (A) lysine and aspartic acid | | (D) i – cyclic, ii – peptide bond |
| | (B) methionine and arginine | 46. | The amount of purines is equal to the amount of |
| | (C) lysine and arginine | | pyrimidines was suggested by |
| | (D) proline and cysteine | | (A) Robert Brown (B) Miescher |
| 37. | Histories of | | (C) Chargaff (D) Khorana |
| 37. | Histones of are basic proteins. (A) chromoproteins (B) nucleoproteins | 47. | In an experiment, DNA was found to have 31% |
| | (C) phytoproteins (D) mucoproteins | | adenine and 19% guanine. The percentage of |
| | | 1 | cytosine shall be |
| 38. | Histones are involved in packaging of DNA into | | (A) 38% (B) 31% |
| | structural units called | | (C) 19% (D) 62% |
| | (A) nucleotides (B) nucleosides (C) nucleosomes (D) chromatin | 40 | |
| 1 | A A | 48. | general are underlied by |
| 39. | Conjugated proteins consist of a simple protein | 1 | (A) peptide bonds |
| | united with some non-protein substance. The | | (B) phosphodiester bonds |
| | non-protein group is called | | (C) hydrogen bonds |
| | (A) mucoprotein (B) lipoprotein (C) prosthetic group (D) globin | | (D) glycosidic bonds |
| | S groun | 49. | . In a DNA strand, the nucleotides are linked |
| 40. | Identify the examples of derived proteins. | | together by |
| | (A) Metaproteins, haemoglobin | | (A) glycosidic bonds |
| | (B) Haemoglobin, peptones | | (B) phosphodiester bonds |
| A | (C) Metaproteins, peptones | | (C) peptide bonds |
| 1 | (D) Mucoproteins, peptones | 1 | (D) hydrogen bonds |
| Nuc | eleic Acids | 50 | The transfer of the transfer o |
| 41. | In 1024 F 1 | 30 | The two strands of DNA complete a turn at vertical distance of |
| 3 | In 1924, Feulgen showed that contain DNA. | | (4) 20 1 |
| | 145 | | (0) 215 |
| | (C) thomosomes | | (C) 34 Å (D) 3.4 Å |
| 40 | (E) mixenes | 51 | Distance between two successive base pairs of |
| 42. | A nucleotide contains | | DNA molecule is |
| | (A) sugar + phosphate | | (A) 6.8 Å or 0.034 nm |
| | (B) N-base + phosphate | | (B) 4.3 Å or 3.4 nm |
| | (C) sugar + nitrogenous base | | (C) 3.4 Å or 0.34 nm |
| _ | (D) sugar + N-base + phosphate | | (D) 4.4 Å or 44 nm |
| | | | |

NATION Triumph Blokagy (MCQs)

- 15. Which of the following is a derived lipid with four interlocking rings?
 - (A) Cholesterol
- (B) Estrogen
- (C) Testosterone
- (D) All of these
- Match the items in Column I with items in Column II and choose the correct answer.

| | Column | 12 | Column II |
|------|----------------|----|---------------------|
| i. | Triglyceride | a | Animal hormones |
| ri. | Membrane lipid | b. | Feathers and leaves |
| toi. | Steroid | C. | Phospholipids |
| iv. | Wax | d. | 3 FA + 1 Glycerol |

- (A) i-d, ii-c, iii-a, iv-b
- (B) i b, ii c, iii d, iv a
- (C) i-c, ii-d, iii-a, iv-b
- (D) i-d, ii-a, iii-b, iv-c

Proteins

- Study the following statements and select the correct option.
 - In quaternary structure, the peptide chains are much looped, twisted and folded back on themselves due to formation of hydrogen bonds.
 - When a protein has more than two polypeptide subunits their arrangement in space is called secondary structure.
 - (A) Statement (i) is correct.
 - (B) Statement (ii) is correct.
 - (C) Both the statements (i) and (ii) are correct.
 - (D) Both the statements (i) and (ii) are incorrect.
- 18. Which of the following correctly describes the amphoteric nature of proteins?
 - (A) Proteins can act as only acids.
 - (B) Proteins can act as only bases.
 - (C) Proteins can act as both acids and bases.
 - (D) Proteins only show presence of non-polar amino acids.

Read the following statements with respect to albumin and histones and select the correct option.

- Histones are insoluble in water but they get coagulated on heating.
- ii. Albumins are insoluble in water.
- (A) Statement (i) is correct.
- (B) Statement (ii) is correct.
- (C) Both the statements (i) and (ii) are correct.
- (D) Both the statements (i) and (ii) are incorrect.

20. Identify 'X' and 'Y' in the following table.

| - t-las | Role |
|----------------------------|----------------|
| Proteins Immunoglobulin | X |
| | Blood clotting |

- (A) X: Transport of Oxygen
 - Y: Haemoglobin
- (B) X: Resistance against diseases
 - Y: Fibrinogen
- (C) X: Muscle contraction
 - Y: Fibrinogen
- (D) X: Structural stability of cell
 - Y: Fibrinogen

Nucleic Acids

- 21. DNA consists of two complementary nucleotide chains. If the sequence of nucleotide in one of the chains is 5'AGCTTCGA3', then the nucleotide sequence in the other chain shall be
 - (A) 5'TAGCATAT3' (B) 5'GATCCTAG3'
 - (C) 3'TCGAAGCT5' (D) 3'GCTAAGCT5'
- 22. Chargaff's rules are applied to
 - (A) ssRNA
- (B) ssDNA
- (C) dsDNA
- (D) mRNA
- A DNA molecule measuring 680 Å contains nucleotides.
 - (A) 1360
- (B) 510
- (C) 340
- (D) 400
- 24. DNA differs from RNA in
 - (A) absence of OH group at the 2'- position
 - (B) presence of OH group at the 2'- position
 - (C) absence of phosphate group at the 2'-position
 - (D) presence of phosphate group at 2'- position.
- Read the following statements and select the correct option.
 - m-RNA carries genetic information from DNA to ribosomes, which are the sites of protein synthesis.
 - r-RNA provides proper binding site for m-RNA during protein synthesis.
 - iii. t-RNA helps in elongation of polypeptide chain during the process called translation.
 - (A) Statements i and ii are correct.
 - (B) Statements ii and iii are correct.
 - (C) Statements i and iii are correct.
 - (D) Statements i, ii and iii are correct.