# 16

# Green Chemistry and Nanochemistry

# Subtopics

- 16.1 Introduction
- 16.2 Sustainable development
- 16.3 Principles of green chemistry
- 16.4 The role of green chemistry
- 16.5 Introduction to nanochemistry
- 16.6 Characteristic features of nanoparticles
- 16.7 Synthesis of nanomaterials
- 16.8 History of nanotechnology
- 16.9 Applications of nanomaterials
- 16.10 Nanoparticles and nanotechnology

#### The toxicity of Benzene

Benzene (C<sub>6</sub>H<sub>6</sub>) is widely used as a solvent in synthesis of various organic compounds. It is present in gasoline, automobile emissions as well as cigarette smoke. It has been categorised as air pollutant and a known carcinogen. Chronic exposure to benzene leads to leukemia and aplastic anemia (involves bone marrow damage).

Industrial workers are more prone to develop leukemia when are exposed to benzene (around 1 ppm) in atmosphere for long time. Thus, study of Green chemistry is important in designing safer chemicals.



chemicals

## Quick Review

### > The 12 principles of green chemistry:

- i. Prevention of waste or by products
- iii. Less hazardous chemical synthesis
- v. Use of safer solvent and auxiliaries
- vii. Use of renewable feed stocks
- ix. Use of catalysis
- xi. Real-time analysis pollution prevention
- ii. Atom economy
- iv. Designing safer chemicals
- vi. Design for energy efficiency
- viii. Reduce derivatives (Minimization of steps)
- x. Design for degradation
- xii. Safer chemistry for accident prevention

### Formula of atom economy:

% atom economy =  $\frac{\text{Formula weight of the desired product}}{\text{Sum of formula weight of all the reactants used in the reaction}} \times 100$ 

### Types of nanomaterials according to dimensions:

Nanomaterial Dimension	Nanomaterial Type	Example
All three dimensions < 100 nm	Nanoparticles, Quantum dots, nanoshells, nanorings, microcapsules	Example
Two dimensions < 100 nm	Nanotubes, fibres, nanowires	- 4
One dimension < 100 nm	Thin films, layers and coatings	

