

# Module 4- Thermal Engineering

**NOV 20**

- 1 A brick wall 250mm thick is faced with concrete 50mm thick. The wall is 10m long and 5m high. The temperature of the exposed brick face is  $300^{\circ}\text{C}$  and that of concrete is  $50^{\circ}\text{C}$ . Thermal conductivities of brick and concrete is  $0.69\text{W/m}^{\circ}\text{K}$  and  $0.93\text{W/m}^{\circ}\text{K}$ . Determine: (i) Heat loss per hour  
(ii) Interface temperature.

**Oct 19**

- 2 (a) A brick wall 250 mm thick is faced with concrete 50 mm thick. If the temperature of the exposed brick face is  $30^{\circ}\text{C}$  and that of concrete is  $5^{\circ}\text{C}$ , determine the heat lost per hour, through a wall 10m long and 5m high. Also determine the interface temperature (Given  $K_{\text{brick}} = 0.69 \text{ W/m}^{\circ}\text{K}$ ,  $K_{\text{concrete}} = 0.93 \text{ W/m}^{\circ}\text{K}$ ).
- 3 (b) A single acting air compressor has a cylinder of 200 mm diameter and 300 mm stroke and runs at 150 rpm. Suction pressure and temperature are 1 bar absolute and  $15^{\circ}\text{C}$  respectively. Delivery pressure is 10 bar absolute. Calculate work done per cycle and power required to drive the compressor. Assume the law of compression to be  $PV^{1.2} = \text{constant}$ . Neglect clearance volume.

**APR 19**

- 4 (b) The inside and outside surfaces of a window glass are at  $20^{\circ}\text{C}$  and  $-5^{\circ}\text{C}$  respectively. If the glass is  $1000\text{mm} \times 500\text{mm}$  in size and 15 mm thick with thermal conductivity of  $0.78\text{W/mK}$ . Determine the heat loss through the glass over a period of 2 hours.
- 5 (b) Find the amount of work required to compress and discharge  $1\text{m}^3$  of air at  $15^{\circ}\text{C}$  and 1 bar to 7 bars absolute. When compression is isothermal. Take  $R = 0.29\text{kJ/kgK}$ .

## OCT 18

- 6 (a) A single cylinder, single acting reciprocating air compressor has a cylinder diameter 150 mm and a stroke of 250 mm. Air is drawn in the cylinder at a pressure of 1 bar and a temperature of 15°C. It is then compressed to 6 bar. If the compressor speed is 120 rpm calculate,
- (i) mass of air compressed per cycle
  - (ii) Work required per cycle
  - (iii) Power required to drive the compressor, if compression is adiabatic
  - (iv) Volumetric efficiency

Assume  $\gamma = 1.4$  and  $R = 0.290 \text{ KJ/KgK}$

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## APR 18

- 7 (a) A single stage reciprocating compressor is required to compress 1 kg of air from 1 bar to 4 bar. Initial temperature is 27°C. Compare the work requirement on following cases.

- (i) Isothermal compression
- (ii) Compression with  $PV^{1.2} = \text{constant}$
- (iii) Isentropic compression (Assume  $R = 287 \text{ KJ/KgK}$ ).

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## OCT 17

- 8 (b) A single acting single cylinder air compressor is required to compress 1 kg of air from 100 KPa to 400 KPa isothermally. The initial temperature of air is 27°C. Calculate the power required to drive the compressor, if speed of compressor is 100 rpm. Take characteristic gas constant for air as 0.287 KJ/Kgk.

- 9 X (a) Heat is conducted through a composite plate composed of two parallel plates of different materials A and B of thermal conductivities 134 w/m-k and 60 w/m-k and thickness 36mm and 42mm respectively. The temperature of outer surface of slab A and that of B are 96°C and 8°C respectively. Find the rate of heat transfer and inter face temperature, if the cross sectional area of plate across direction of heat flow is 10 m<sup>2</sup>.

## APR 17

- 10 (b) A wall is made up of two layers of bricks each 150mm thick with a 40mm air space between them. Coefficients of thermal conductivities are :  
(i) inside brick =  $0.69 \text{ W/mK}$ , (ii) Air =  $0.0605 \text{ W/mK}$ , (iii) Outside brick =  $1.038 \text{ W/mK}$ . The wall is 6.15m long and 5.5m high. Determine the heat loss/hour through the wall, if inside face temperature is  $24^\circ\text{C}$  and outside temperature is  $7^\circ\text{C}$ . 8
- 11 (b) A single stage single acting air compressor has a cylinder of 200mm diameter and 300mm stroke and runs at 150 RPM. If the suction pressure and temperature are 1 bar absolute and  $15^\circ\text{C}$  respectively and the delivery pressure 10 bar absolute. Calculate the power required to drive the compressor. Assume the law of compression to be  $pV^{1.2} = \text{constant}$ .

# Module 4 - Thermal Engineering

①

NOV 20

1. Given Data

$$L_b = 250 \text{ mm} = 0.25 \text{ m}$$

$$L_c = 50 \text{ mm} = 0.05 \text{ m}$$

$$k_b = 0.69 \text{ W/mK}$$

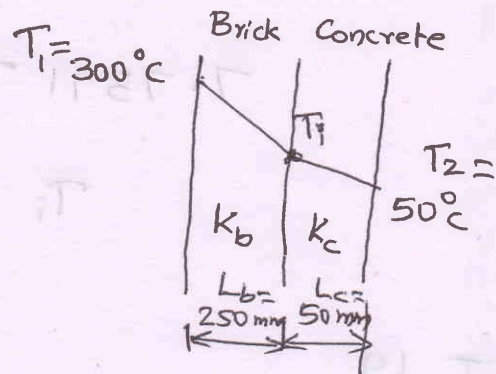
$$k_c = 0.93 \text{ W/mK}$$

$$A = 10 \text{ m} \times 5 \text{ m} = 50 \text{ m}^2$$

$$T_1 = 300^\circ\text{C}$$

$$T_2 = 50^\circ\text{C}$$

$$\dot{Q} = ? \quad T_i = ?$$



$$\dot{Q} = \frac{T_1 - T_2}{\frac{L_b}{k_b A} + \frac{L_c}{k_c A}} = \frac{300 - 50}{\frac{0.25}{0.69 \times 50} + \frac{0.05}{0.93 \times 50}}$$

$$= 30042 \text{ W} = 30.042 \text{ kW}$$

$$= 30.042 \text{ kJ/s} = 108151 \text{ kJ/h}$$

To get interface temperature

$$\frac{T_1 - T_i}{\frac{L_b}{k_b A}} = \frac{T_i - T_2}{\frac{L_c}{k_c A}}$$

$$\Rightarrow \frac{300 - T_i}{\left(\frac{0.25}{0.69}\right)} = \frac{T_i - 50}{\left(\frac{0.05}{0.93}\right)}$$



$$300 - T_i = 6.7391 \quad T_i - 336.955$$

$$7.7391 T_i = 636.955$$

$$T_i = \underline{\underline{82.3^\circ\text{C}}}$$

OCT 19

②

Given Data

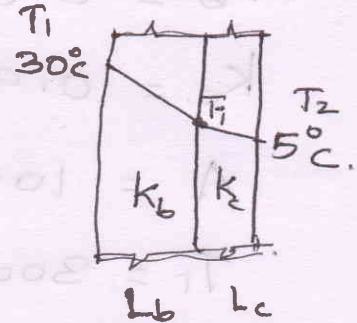
$$A = 10\text{m} \times 5\text{m} = 50\text{ m}^2$$

$$T_1 = 30^\circ\text{C} \quad T_2 = 5^\circ\text{C}$$

$$L_b = 0.25\text{m} \quad L_c = 0.05\text{m}$$

$$k_b = 0.69\text{ W/mK}$$

$$k_c = 0.93\text{ W/mK}$$



$$\dot{Q} = \frac{T_1 - T_2}{\frac{L_b}{k_b A} + \frac{L_c}{k_c A}} = \frac{30 - 5}{\frac{0.25}{0.69 \times 50} + \frac{0.05}{0.93 \times 50}}$$

$$= 3004\text{ W} = 3.004\text{ kW}$$

$$= 3.004\text{ kJ/s} = \underline{\underline{10814.4\text{ kJ/h}}}$$

To get interface temperature.

$$\frac{T_1 - T_i}{\frac{L_b}{k_b A}} = \frac{T_i - T_2}{\left(\frac{L_c}{k_c A}\right)} \Rightarrow \frac{30 - T_i}{\left(\frac{0.25}{0.69}\right)} = \frac{T_i - 5}{\left(\frac{0.05}{0.93}\right)}$$

$$30 - T_i = 6.7391 \quad T_i = 33.6955$$

$$7.7391 T_i = 63.6955$$

$$T_i = \frac{63.6955}{7.7391} = \underline{\underline{8.23^\circ\text{C}}}$$

OCT 19

3. Given Data

$$D = 200 \text{ mm}$$

$$L = 300 \text{ mm}$$

$$N = 150 \text{ rpm}$$

$$P_1 = 1 \text{ bar} = 100 \text{ kPa}$$

$$T_1 = 15^\circ\text{C}$$

$$P_2 = 10 \text{ bar} = 10^3 \text{ kPa}$$

$$W = ? \quad P = ? \quad n = 1.2$$

Theoretical Swept Volume of cylinder

$$V_1 = \frac{\pi}{4} D^2 L = \frac{\pi}{4} \times 0.2^2 \times 0.3 = 0.00942 \text{ m}^3$$

$$W = \frac{n}{n-1} \cdot P_1 V_1 \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right]$$

$$= \frac{1.2}{1.2-1} \times 100 \times 0.00942 \left[ \left( \frac{10}{1} \right)^{\frac{1.2-1}{1.2}} - 1 \right]$$

$$= 2.644 \text{ kJ}$$

$$\text{Power required, } P = \frac{W \times N}{60} = \frac{2.644 \times 150}{60}$$

$$= \underline{\underline{6.61 \text{ kW}}}$$

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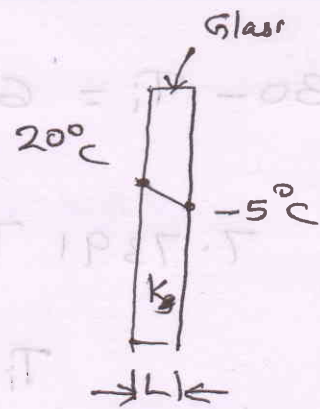
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Given Data

$$k_g = 0.78 \text{ W/mK}$$

$$L = 15 \text{ mm} = 0.015 \text{ m}$$

$$A = 1000 \text{ mm} \times 500 \text{ mm} \\ = 1 \text{ m} \times 0.5 \text{ m} = 0.5 \text{ m}^2$$



$$\text{Time } t = 2 \text{ hours. } T_1 = 20^\circ\text{C} \quad T_2 = -5^\circ\text{C}$$

$$\text{Heat transfer rate } \dot{Q} = \frac{T_1 - T_2}{\frac{L}{kA}}$$

$$= \frac{20 - (-5)}{\frac{0.015}{(0.78 \times 0.5)}} = \frac{25}{\left(\frac{0.015}{0.78 \times 0.5}\right)}$$

$$= 650 \text{ W} = 0.65 \text{ kW} = 0.65 \text{ kJ/s}$$

$\therefore$  Heat loss in 2 hrs.

$$Q = \dot{Q} \times t = 0.65 \text{ kJ/s} \times 2 \text{ hrs.}$$

$$= 0.65 \times 2 \times 3600 = \underline{\underline{4680 \text{ kJ}}}$$

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Given Data

$$P_1 = 1 \text{ bar} \quad V_1 = 1 \text{ m}^3$$

$$T_1 = 15^\circ\text{C}$$

$$P_2 = 7 \text{ bar}$$

$$W = P_1 V_1 \ln\left(\frac{P_2}{P_1}\right) = 100 \times 1 \times \ln\left(\frac{7}{1}\right) \\ = \underline{\underline{194.59 \text{ kJ}}}$$

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6 Given Data  $d = 150 \text{ mm} = 0.15 \text{ m}$   $L = 0.25 \text{ m}$ 

$$P_1 = 1 \text{ bar} \quad T_1 = 15^\circ\text{C} \quad P_2 = 6 \text{ bar}$$

$$N = 120 \text{ rpm.} \quad \gamma = 1.4 \quad R = 0.29 \text{ kJ/kg}\cdot\text{K}$$

$$m = ? \quad W = ? \quad P = ? \quad \eta = ?$$

Swept Volume,  $V_s = \frac{\pi d^2 L}{4}$

$$= \frac{\pi}{4} \times 0.15^2 \times 0.25$$

$$= \cancel{0.00} \quad 4.418 \times 10^{-3}$$

(i) mass of air  $m = \frac{P_1 V_1}{RT_1}$

$$= \frac{1 \times 10^5 \times 4.418 \times 10^{-3}}{290 \times (15 + 273)}$$

$$= 5.29 \times 10^{-3} \text{ kg.}$$

(ii) Work required  $W = \frac{\gamma}{\gamma - 1} \times m R T_1 \left[ \left( \frac{P_2}{P_1} \right)^{\frac{\gamma - 1}{\gamma}} - 1 \right]$

$$= \frac{1.4}{1.4 - 1} \times 5.29 \times 10^{-3} \times 290 \times 288 \left[ \left( \frac{6}{1} \right)^{\frac{1.4 - 1}{1.4}} - 1 \right]$$

$$= 1033.8 \text{ J / cycle.}$$

$$= 1.0338 \text{ kJ / cycle}$$



(iii) Power required

$$P = \frac{W \times N}{60}$$

$$= 1.0338 \times \frac{120}{60} = \underline{\underline{2.068 \text{ kW}}}$$

(ii) Volumetric efficiency.

Since Free air conditions are not given it is assumed that  $P_a = 1.01325 \times 10^5 \text{ N/m}^2$  and  $T_a = 15^\circ\text{C}$ .

$$\frac{P_a V_a}{T_a} = \frac{P_1 V_1}{T_1}$$

$$\therefore V_a = V_1 \times \frac{P_1}{P_a} \times \frac{T_a}{T_1}$$

$$= 4.418 \times 10^{-3} \times \frac{1 \times 10^5}{1.01325 \times 10^5} \times \frac{288}{288}$$

$$= 4.36 \times 10^{-3} \text{ m}^3$$

$$\eta_v = \frac{V_a}{V_s} = \frac{4.36 \times 10^{-3} \text{ m}^3}{4.418 \times 10^{-3}} = \underline{\underline{98.69\%}}$$

APR 18

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Given Data  $m = 1 \text{ kg}$   $P_1 = 1 \text{ bar} = 100 \text{ kPa}$

$P_2 = 4 \text{ bar} = 400 \text{ kPa}$   $T_1 = 27^\circ\text{C} = 300 \text{ K}$

$R = 287 \text{ J/kg} \cdot \text{K}$   $n = 1.2$

(i) isothermal Compression

$$\begin{aligned}
 W_{\text{Isothermal}} &= P_1 V_1 \ln \frac{P_2}{P_1} = m R T_1 \ln \frac{P_2}{P_1} \\
 &= 1 \times 287 \times 300 \times \ln \frac{4}{1} \\
 &= 119360 \text{ J} = \underline{\underline{119.36 \text{ kJ}}}
 \end{aligned}$$

(ii) Compression with  $PV^{1.2} = \text{Constant}$ .

$$\begin{aligned}
 W_{\text{polytropic } n=1.2} &= \frac{n}{n-1} \times m R T_1 \times \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right] \\
 &= \frac{1.2}{1.2-1} \times 1 \times 287 \times 300 \left[ \left( \frac{4}{1} \right)^{\frac{1.2-1}{1.2}} - 1 \right] \\
 &= 134275 \text{ J} = \underline{\underline{134.28 \text{ kJ}}}
 \end{aligned}$$

(iii) Isentropic Compression.

$$\begin{aligned}
 W_{\text{isentropic}} &= \frac{\gamma}{\gamma-1} \times m R T_1 \left[ \left( \frac{P_2}{P_1} \right)^{\frac{\gamma-1}{\gamma}} - 1 \right] \\
 &= \frac{1.4}{1.4-1} \times 1 \times 287 \times 300 \left[ \left( \frac{4}{1} \right)^{\frac{1.4-1}{1.4}} - 1 \right] \\
 &= 146454 \text{ J} = 146.45 \text{ kJ}
 \end{aligned}$$

$$\therefore W_{\text{isothermal}} < W_{\text{polytropic}} < W_{\text{isentropic}}$$

OCT 17

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8 Given Data  $m = 1 \text{ kg}$   $P_1 = 100 \text{ kPa}$ 

$$P_2 = 400 \text{ kPa} \quad T_1 = T_2 \quad T = \text{Constant}$$

$$T_1 = 27^\circ\text{C} \quad N = 100 \text{ rpm}$$

$$R = 0.287 \text{ kJ/kg}\cdot\text{K}$$

$$W = m R T_1 \ln \frac{P_2}{P_1}$$

$$= 1 \times 0.287 \times (27 + 273) \ln \left( \frac{400}{100} \right)$$

$$= 119.36 \text{ kJ}$$

Power required  $P = \frac{W \times N}{60}$

$$= \frac{119.36 \times 100}{60}$$

$$= \underline{\underline{198.93 \text{ kW}}}$$

OCT 17

9

Given Data

$$A = 10 \text{ m}^2$$

$$T_1 = 96^\circ\text{C}$$

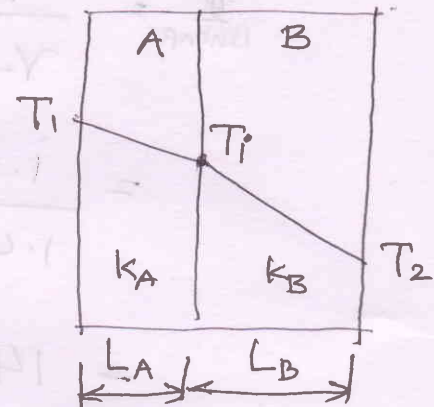
$$T_2 = 8^\circ\text{C}$$

$$k_A = 134 \text{ W/mK}$$

$$k_B = 60 \text{ W/mK}$$

$$L_A = 36 \text{ mm} = 0.036 \text{ m}$$

$$L_B = 42 \text{ mm} = 0.042 \text{ m}$$



Rate of heat transfer

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$$\Phi = \frac{T_1 - T_2}{\frac{L_A}{k_A \cdot A} + \frac{L_B}{k_B \cdot A}} = \frac{96 - 8}{\frac{0.036}{134 \times 10} + \frac{0.042}{60 \times 10}}$$

$$= 908475 \text{ W} = \underline{\underline{908.48 \text{ kW}}}$$

$$= 908.48 \text{ kJ/s}$$

To get interface temperature

$$\Phi = k_A \cdot A \frac{(T_1 - T_i)}{L_A}$$

$$908475 = 134 \times 10 \times \frac{(96 - T_i)}{0.036}$$

$$T_i = \underline{\underline{71.59^\circ \text{C}}}$$

APR 17

10 Given Data

$$T_1 = 24^\circ \text{C} \quad T_2 = 7^\circ \text{C}$$

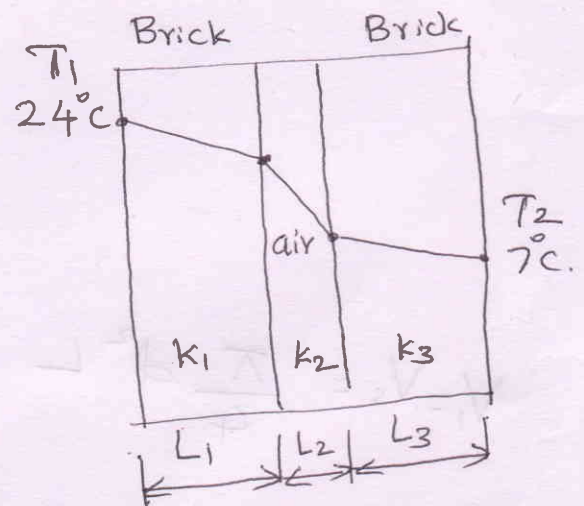
$$k_1 = 0.69 \text{ W/mK}$$

$$k_2 = 0.0605 \text{ W/mK}$$

$$k_3 = 1.038 \text{ W/mK}$$

$$A = 6.15 \text{ m} \times 5.5 \text{ m}$$

$$L_1 = L_3 = 0.15 \text{ m}, \quad L_2 = 0.04 \text{ m}$$





$$\dot{Q} = \frac{T_1 - T_2}{\frac{L_1}{k_1 A} + \frac{L_2}{k_2 A} + \frac{L_3}{k_3 A}} = \frac{T_1 - T_2}{\frac{1}{A} \left[ \frac{L_1}{k_1} + \frac{L_2}{k_2} + \frac{L_3}{k_3} \right]}$$

$$W = \frac{24 - 7}{\frac{1}{6.15 \times 5.5} \left[ \frac{0.15}{0.69} + \frac{0.04}{0.0605} + \frac{0.15}{1.038} \right]}$$

$$= 562.07 \text{ J/s.}$$

$$= 562.07 \times \frac{3600}{1000} \text{ kJ/hr}$$

$$= \underline{\underline{2023.45 \text{ kJ/h}}}$$

APR 17

11 Given Data  $d = 200 \text{ mm} = 0.2 \text{ m}$

$$L = 300 \text{ mm} = 0.3 \text{ m}$$

$$N = 150 \text{ rpm}$$

$$P_1 = 1 \text{ bar} \quad T_1 = 15^\circ \text{C}$$

$$P_2 = 10 \text{ bar}$$

$$n = 1.2$$

$$V_1 = V_2 = \frac{\pi}{4} d^2 L = \frac{\pi}{4} \times 0.2^2 \times 0.3 = \underline{\underline{9.425 \times 10^{-3} \text{ m}^3}}$$

$$W = \frac{n}{n-1} \times P_1 V_1 \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right]$$

$$W = \frac{1.2}{1.2-1} \times 100 \times 9.425 \times 10^{-3} \left[ \left( \frac{10}{1} \right)^{\frac{1.2-1}{1.2}} - 1 \right] \quad //$$

$$= 2.645 \text{ ~~kJ~~ kJ/cycle}$$

Power required  $P = \frac{W \times N}{60}$

$$= 2.645 \times \frac{150}{60}$$

$$= \underline{\underline{6.61 \text{ kW}}}$$