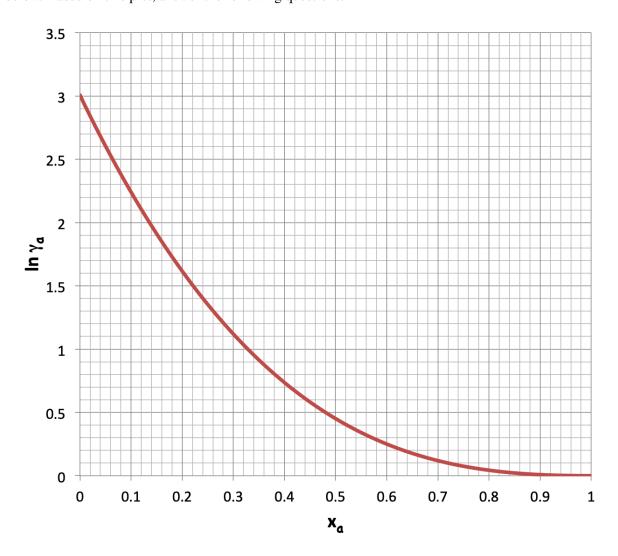
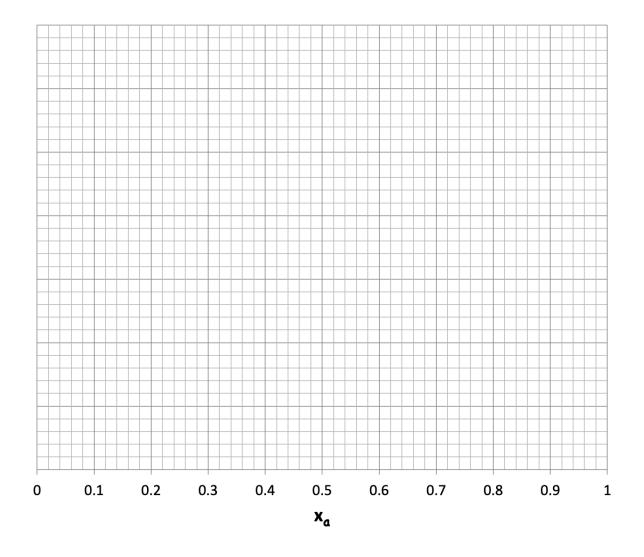
Consider a binary mixture of species a and species b at 300 K and 1 bar. The vapor pressure of pure a at 300 K is 80 kPa. A plot of the activity coefficient of species a vs. mole fraction of species a is shown below. Based on this plot, answer the following questions.



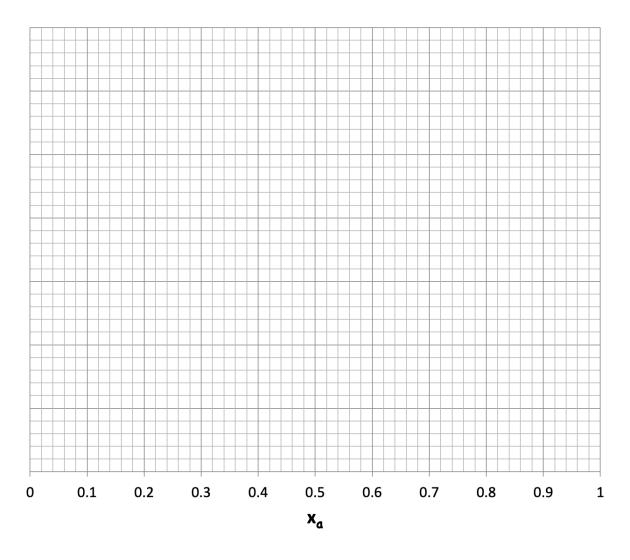
- 1. Specify the reference state for species a Lewis-Randall or Henry's. Calculate f_a , \mathcal{H}_a .
- 2. Calculate (a) the activity coefficient based on Lewis-Randall state in the infinite dilution, γ_a^{∞} and (b) the activity coefficient based on Henry's law for pure a, $\gamma_a^{\text{Henry's, pure } a}$
- 3. Add a plot $\ln \gamma_b$ vs x_a to the given graph. Fill the following table before plotting

x_a	0	0.02	0.04	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9	1.0
$\ln \gamma_a$													
$\ln \gamma_b$													

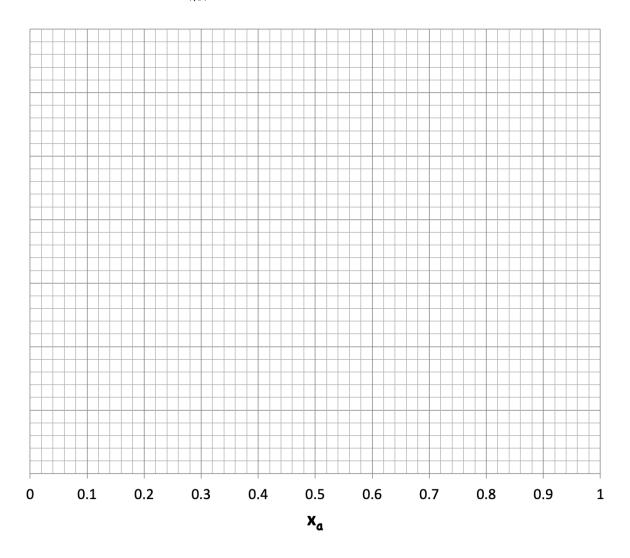
4. Plot the activities, a_a , a_b vs x_a



5. Plot $\ln \gamma_a^{\text{Henry's}}$, $\ln \gamma_b^{\text{Henry's}}$ vs x_a



6. Thermodynamic consistency test: Plot $\ln \frac{\gamma_a}{\gamma_b}$ vs x_a . Infer the value of $\int_0^1 \ln \left(\frac{\gamma_a}{\gamma_b}\right) dx_a$?



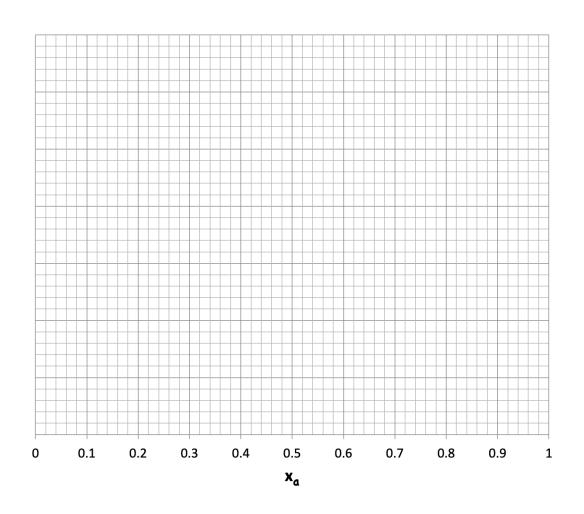
- 7. Consider a liquid mixture of 8 moles of a and 2 moles of b at 300 K and 1 bar. If a vapor mixture coexists (at equilibrium) with this liquid mixture what will be the composition of a in the vapor phase, y_a ?
- 8. Assume a two suffix Margules Calculate the two suffix Margules parameter *A*.

x_a	0	0.02	0.04	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9	1.0

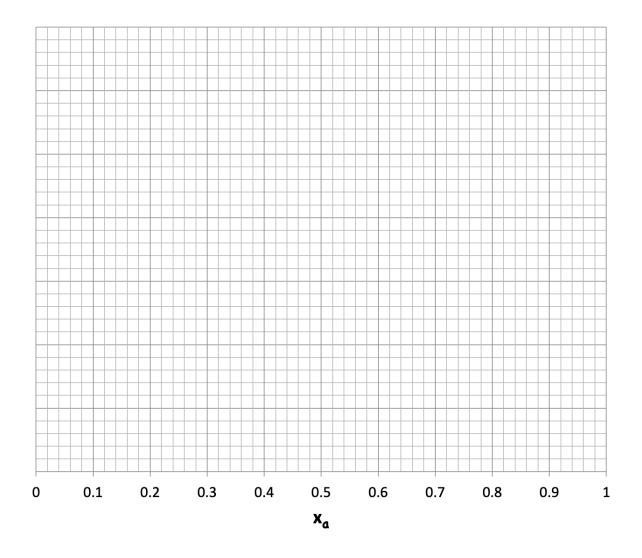
9. Assume a three suffix Margules model and calculate the model parameters A and B.

10. Assume $g_a = 50 J/mol$, $g_b = 100 J/mol$. Plot different contributions to the total Gibb's free energy. Assume two suffix Margules Equation for estimating excess Gibbs energy. (For clarity, plot the total in the next page).

x_a	0	0.02	0.04	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9	1.0
$x_a g_a + x_b g_b$													
Δg_{mix}^{ideal}													
g^E													
g^{total}													



11. Plot total Gibb's free energy as a function of x_a . Do you see two minima and a maximum? Draw a tangent line connecting the two minima. What does this tangent line tell you?



12. Assume that the two suffix Margues parameter that you have calculated is negative but of the same magnitude. Calculate the total Gibbs energy and plot it. What do you see now? Does phase separation happen?

