

Q No 1

⇒ Newton's law of heating and cooling:-

①

$$\frac{dT}{T_0 - T_s} = -a dt$$

$$\Rightarrow \int \frac{dT}{T_0 - T_s} = -\int a dt$$

$$\Rightarrow T_0 - T_s = e^{-at+C}$$

$$\Rightarrow T_0 - T_s = Ce^{-at}$$

$$\Rightarrow T_0 = Ce^{-at} + T_s$$

$$\Rightarrow T(t) = Ce^{-at} + T_s$$

$$t=0 \quad T_s = 70^\circ\text{C} \quad T_0 = 90^\circ\text{C}$$

$$T(0) = Ce^{-a(0)} + 70$$

$$T(0) = C + 70$$

$$90 = C + 70$$

$$C = 20$$

$$\text{At } t = 1$$

$$T(1) = 80$$

$$T(1) = 20e^{-a(1)} + 70$$

$$T(1) = 20e^{-a} + 70$$

$$80 = 20e^{-a} + 70$$

$$80 - 70 = 20e^{-a}$$

$$e^{-a} = \frac{10}{20}$$

$$a = 0.69$$

⑥

temperature at 100°C

$$T(t) = 100^\circ\text{C}$$

$$100 = 20e^{0.69t}$$

$$\frac{30}{20} = e^{0.69t}$$

$$\ln(e^{0.69t}) = \ln(1.5)$$

$$0.69t = 0.405$$

$$t = 0.58 \text{ min}$$

$$\frac{dT}{T - T_0} = -k dt$$

$$T = Ce^{-kt} + T_s$$

At

$$t = 0 \quad T = 28^\circ\text{C}$$

⑥

$$\text{At } t = 0.004$$

$$T = 30 e^{-0.66(0.004)} + 25$$

$$T = 3(0.99) + 25$$

$$T = 27.94$$

$$t = 0.008$$

$$T = 30 e^{-0.66(0.008)} + 25$$

$$T = 27.99$$

$$t = 0.012$$

$$T = 30 e^{-0.66(0.012)} + 25$$

$$T = 27.97$$

Q No 7

⑦

$$\frac{dN}{dt} = kN \ln(N/N_0)$$

$$\int \frac{dN}{N} \ln(N/N_0) = \int k dt$$

Let  $u = \ln(M/N)$ ,  $du = 1/M/N^2 - M/N^3 dN$

$$du = N/M \times M/N^3 dN = \int k dt$$

$$-\int 1/u du = \int k dt$$

$$\ln|u| = -kt + C$$

$$u = e^{-kt+C}$$

$$|u| = ce^{-kt}$$

$$|u| = ce^{-kt} \rightarrow A$$

Putting values of  $u$

$$\ln(M/N) = ce^{-kt}$$

As  $t=0$

$$\ln(M/N) = ce^{-k(0)}$$

$$\ln(M/N) = c$$

Putting  $c$  in equation 1

$$u = \ln(M/N) e^{-kt}$$

$$N = Me^{-\ln(M/N) e^{-kt}}$$

QNO 3

Given

Half life of Carbon-14 = 5730 year

SolutionAS Analytical solution of  $\frac{dQ}{dt} = -rQ$ 

$$Q = Q_0 e^{-rt} \rightarrow (A)$$

$$Q = \frac{Q_0}{2} \text{ at } 5730 \text{ years.}$$

$$\frac{Q_0}{2} = e^{-rt}$$

$$\frac{Q_0}{2} = e^{-(0.000298)(5730)}$$

$$Q_0 = 11.0248 \rightarrow (B)$$

Now

$$Q = Q_0 e^{-rt}$$

we will find  $Q$  when amount is 20%

$$Q = (11.0248) (e^{-0.000298 \times 2168})$$

$$Q = (11.0248) (e^{-21680.6})$$

$$Q = 64.18 \rightarrow (C)$$

The equation (A)

$$Q = Q_0 e^{-rt}$$

$$64.18 = 11.0248 e^{-0.000298t}$$

$$e^{-0.000298t} = 5.8203$$

$$\ln[e^{-0.000298t}] = \ln 5.8203$$

$$0.000298t = 1.761$$

$$t = 591.06 \text{ years}$$

$$\underline{Q_{No4}}$$

