



# Modeling Kinetics of Lead in the Human Body using First Order ODEs

**Iddrisu Hafsatu  
Ishmael Abdul-Hafiz  
Akintola Nurudeen  
Owusu Berchie Jacob**

*Supervisor*  
**Dr. E. Osei Frimpong**

May 20, 2014

# Outline of Presentation

- 1 Introduction
- 2 The Model
- 3 Analysis and Results
- 4 Conclusion and Recommendation
- 5 References

# Outline of Presentation

- 1 Introduction
- 2 The Model
- 3 Analysis and Results
- 4 Conclusion and Recommendation
- 5 References

# Background of study

## Lead

- Lead is a naturally occurring element found in the earth crust, its soft, blue-grey metallic element. It is very soft, highly malleable, ductile, and a relatively poor conductor of electricity.

## Exposure of lead

Today almost everyone is exposed to environmental lead. Lead is taking into the body through

- Inhalation
- Ingestion
- Dermal

## Problem Statement

Lead is a cumulative toxicant that affects multiple body systems and harmful to humans especially children and hence the need to study the behaviour and the concentration at the steady state to help reduce the lead content in human body through medical delivery.

## Objectives

- To determine the concentration of lead in the three compartments in the steady state
- To perform the sensitivity analysis on the model parameters

# Outline of Presentation

- 1 Introduction
- 2 The Model**
- 3 Analysis and Results
- 4 Conclusion and Recommendation
- 5 References

# Model Assumptions

- The rate at which lead transfers from the lungs to the blood stream is proportional to the rate at which lead enters the lungs.
- The rate at which lead transfers from the digestive tract to the blood stream is proportional to the rate at which lead enters the digestive tract.
- The rate ( $F_{ij}$ ) at which lead transfers from compartment  $i$  to compartment  $j$  is proportional to the amount of lead in the compartment  $i$ .
- Finally all lead transfer is measured in micrograms per day

# Model Development

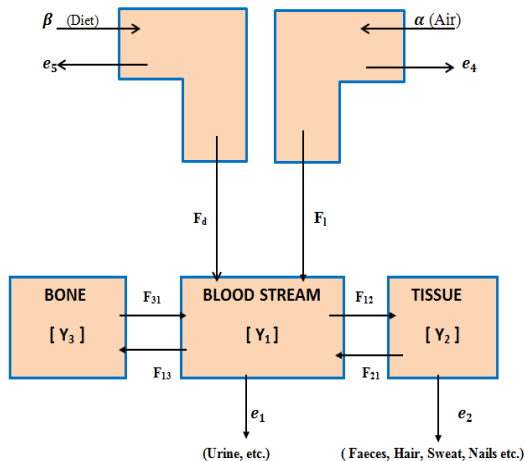


Figure 1: schematic of Lead Transfer



## System of ODEs

$$\begin{aligned}Y_1'(t) &= -A_{11}Y_1 + A_{12}Y_2 + A_{13}Y_3 + N \\Y_2'(t) &= A_{21}Y_1 - A_{22}Y_2 \\Y_3'(t) &= A_{31}Y_1 - A_{33}Y_3\end{aligned}\tag{1}$$

$$Y'(t) = AY + F\tag{2}$$

## Solution using Variation of Parameters

$$Y = Y_c + Y_p = Y_0 e^{At} + e^{At} \int_0^t e^{-At}(t)F(t)dt\tag{3}$$

$$Y = (Y_0 + A^{-1}F)e^{At} - A^{-1}F\tag{4}$$

# Outline of Presentation

- 1 Introduction
- 2 The Model
- 3 Analysis and Results
- 4 Conclusion and Recommendation
- 5 References

# Analysis and Results

## The system

$$Y_1'(t) = \frac{-13}{360} Y_1 + \frac{272}{21875} Y_2 + \frac{7}{200000} Y_3 + \frac{6162}{125}$$

$$Y_2'(t) = \frac{1}{90} Y_1 - \frac{1}{35} Y_2$$

$$Y_3'(t) = \frac{7}{1800} Y_1 - \frac{7}{200000} Y_3$$

## Long term Solution

By Gershgorian circle theorem, the eigenvalues of  $A$  are negative or have zero real parts.

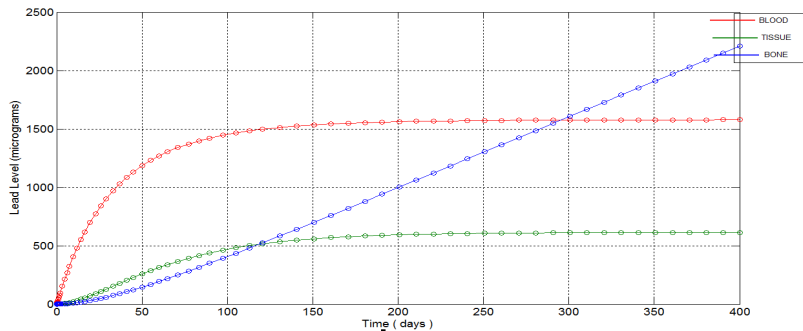
Hence the system is stable in the long run.

$$Y = (Y_0 + A^{-1}F)e^{At} - A^{-1}F \quad (5)$$

$$Y = -A^{-1}F, t \longrightarrow \infty \quad (6)$$

$$Y = (1800\mu g, 700\mu g, 200000\mu g)$$

# Graphical analysis



**Figure 2:** After 400 days, blood and tissue levels go to equilibrium, but bone levels soar

# Sensitivity analysis

## Assumptions

In order to improve the subject condition, we applied the following Sensitivity analysis

- Non-Lead environment
- Small-Lead environment
- Medication
- Small-Lead environment plus Medication

# Sensitivity analysis (Small lead Environment)

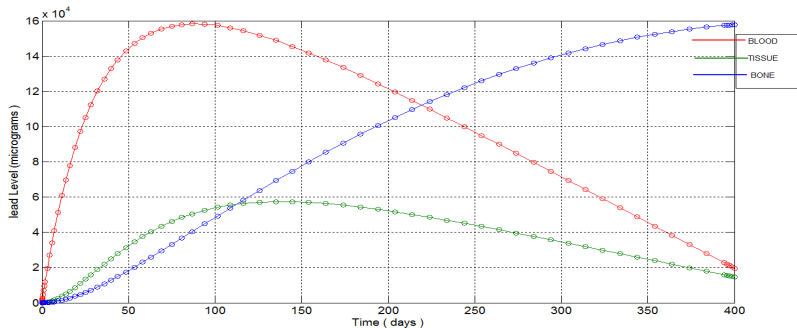


Figure 3: Lead level in the bone increases at a slower rate

# Sensitivity analysis (Medication)

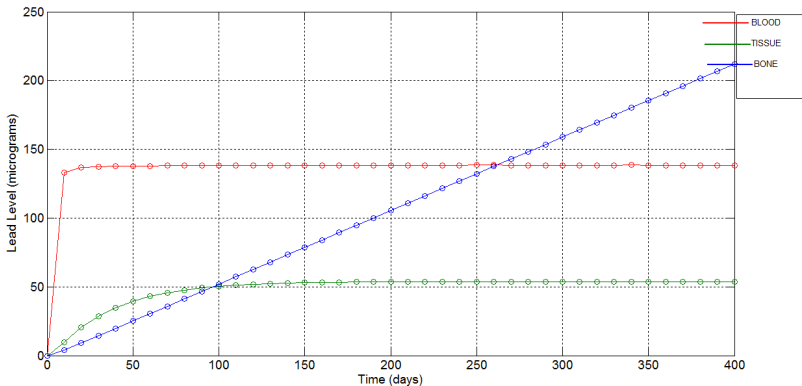


Figure 4: taking medication lowers the concentration in the blood and tissue but not in the bone.

# Small Lead environment plus Medication

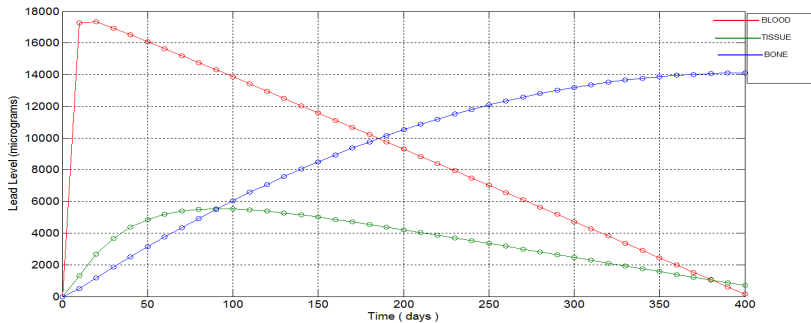


Figure 5: Reduced lead concentration even in the bone



# Outline of Presentation

- 1 Introduction
- 2 The Model
- 3 Analysis and Results
- 4 Conclusion and Recommendation**
- 5 References

# Conclusion

$Y(t) \longrightarrow -A^{-1}F$  as  $t \longrightarrow \infty$ .

Using matlab, the solution is given by  $Y = -\text{inv}(A) * F$

$Y = (1800\mu g, 700\mu g, 200000\mu g)$

Finally, we observed that moving the subject from the lead environment to a small lead environment with medication reduces the lead content in the system drastically. Therefore, from the graphical analysis, it is reasonable to apply those assumptions to real life situations

# Recommendation

- We recommend that further studies could be done to determine the kinetics of other metallic elements like Zinc in the human body.
- The data was obtained from the Journal of Mathematical biology. because of time and financial constraints, we therefore recommend that for any further studies, the data should be obtained from the research location.

# Outline of Presentation

- 1 Introduction
- 2 The Model
- 3 Analysis and Results
- 4 Conclusion and Recommendation
- 5 References

# References

- Agency for Toxic Substances and Disease Registry (ATSDR). Public Health Statement for Lead. Accessed at [www.atsdr.cdc.gov/toxprofiles/phs13.html](http://www.atsdr.cdc.gov/toxprofiles/phs13.html) on May 4, 2010.
- Agency for Toxic Substances and Disease Registry. 1999. Toxicological profile for lead. Atlanta: US Department of Health and Human Services, Public Health Service.
- Agency for Toxic Substances and Disease Registry. 2005. Toxicological profile for lead. Atlanta: US Department of Health and Human Services, Public Health Service.
- Alexander H, Checkoway H, van Netten C, et al. 1996. Semen quality of men employed at a lead smelter. *Occup Environ Med* 53:411-416

END OF PRESENTATION

THANK YOU